

Managing Aging Processes In Storage (MAPS) Report

Draft Report for Comment

AVAILABILITY OF REFERENCE MATERIALS IN NRC PUBLICATIONS

NRC Reference Material

As of November 1999, you may electronically access NUREG-series publications and other NRC records at the NRC's Public Electronic Reading Room at <http://www.nrc.gov/reading-rm.html>. Publicly released records include, to name a few, NUREG-series publications; *Federal Register* notices; applicant, licensee, and vendor documents and correspondence; NRC correspondence and internal memoranda; bulletins and information notices; inspection and investigative reports; licensee event reports; and Commission papers and their attachments.

NRC publications in the NUREG series, NRC regulations, and Title 10, "Energy," in the *Code of Federal Regulations* may also be purchased from one of these two sources.

1. The Superintendent of Documents

U.S. Government Publishing Office
Mail Stop SSOP
Washington, DC 20402-0001
Internet: <http://bookstore.gpo.gov>
Telephone: 1-866-512-1800
Fax: (202) 512-2104

2. The National Technical Information Service

5301 Shawnee Road
Alexandria, VA 22161-0002
<http://www.ntis.gov>
1-800-553-6847 or, locally, (703) 605-6000

A single copy of each NRC draft report for comment is available free, to the extent of supply, upon written request as follows:

U.S. Nuclear Regulatory Commission

Office of Administration
Publications Branch
Washington, DC 20555-0001
E-mail: distribution_resource@nrc.gov
Facsimile: (301) 415-2289

Some publications in the NUREG series that are posted at the NRC's Web site address <http://www.nrc.gov/reading-rm/doc-collections/nuregs> are updated periodically and may differ from the last printed version. Although references to material found on a Web site bear the date the material was accessed, the material available on the date cited may subsequently be removed from the site.

Non-NRC Reference Material

Documents available from public and special technical libraries include all open literature items, such as books, journal articles, transactions, *Federal Register* notices, Federal and State legislation, and congressional reports. Such documents as theses, dissertations, foreign reports and translations, and non-NRC conference proceedings may be purchased from their sponsoring organization.

Copies of industry codes and standards used in a substantive manner in the NRC regulatory process are maintained at—

The NRC Technical Library

Two White Flint North
11545 Rockville Pike
Rockville, MD 20852-2738

These standards are available in the library for reference use by the public. Codes and standards are usually copyrighted and may be purchased from the originating organization or, if they are American National Standards, from—

American National Standards Institute

11 West 42nd Street
New York, NY 10036-8002
<http://www.ansi.org>
(212) 642-4900

Legally binding regulatory requirements are stated only in laws; NRC regulations; licenses, including technical specifications; or orders, not in NUREG-series publications. The views expressed in contractor-prepared publications in this series are not necessarily those of the NRC.

The NUREG series comprises (1) technical and administrative reports and books prepared by the staff (NUREG-XXXX) or agency contractors (NUREG/CR-XXXX), (2) proceedings of conferences (NUREG/CP-XXXX), (3) reports resulting from international agreements (NUREG/IA-XXXX), (4) brochures (NUREG/BR-XXXX), and (5) compilations of legal decisions and orders of the Commission and Atomic and Safety Licensing Boards and of Directors' decisions under Section 2.206 of NRC's regulations (NUREG-0750).

DISCLAIMER: This report was prepared as an account of work sponsored by an agency of the U.S. Government. Neither the U.S. Government nor any agency thereof, nor any employee, makes any warranty, expressed or implied, or assumes any legal liability or responsibility for any third party's use, or the results of such use, of any information, apparatus, product, or process disclosed in this publication, or represents that its use by such third party would not infringe privately owned rights.

Managing Aging Processes In Storage (MAPS) Report

Draft Report for Comment

Manuscript Completed: October 2017
Date Published: October 2017

COMMENTS ON DRAFT REPORT

Any interested party may submit comments on this report for consideration by the staff of the U.S. Nuclear Regulatory Commission (NRC). Comments may be accompanied by additional relevant information or supporting data. Please specify the report number **NUREG-2214** in your comments, and send them by the end of the comment period specified in the Federal Register notice announcing the availability of this report. **Addresses:** You may submit comments by any one of the following methods. Please include Docket ID **NRC-2016-0238** in the subject line of your comments. Comments submitted in writing or in electronic form will be posted on the NRC Web site and on the Federal rulemaking Web site <http://www.regulations.gov>.

Federal Rulemaking Web site: Go to <http://www.regulations.gov> and search for documents filed under Docket ID **NRC-2016-0238**. Address questions about NRC dockets to Carol Gallagher at 301-415-3463 or by e-mail at Carol.Gallagher@nrc.gov.

Mail comments to: Cindy Bladey, Chief, Rules, Announcements, and Directives Branch, Division of Administrative Services, Office of Administration, Mail Stop: TWFN-8-D-36M, U.S. Nuclear Regulatory Commission, Washington, DC 20555 0001.

For any questions about the material in this report, please contact: John Wise, Senior Materials Engineer, 301-415-8085 or by e-mail at John.Wise@nrc.gov.

Please be aware that any comments that you submit to the NRC will be considered a public record and entered into the Agencywide Documents Access and Management System. Do not provide information you would not want to be publicly available.

ABSTRACT

This Managing Aging Processes in Storage (MAPS) Report provides guidance for the U.S. Nuclear Regulatory Commission (NRC) technical reviewer. It establishes a technical basis for the safety review of renewal applications for specific licenses of independent spent fuel storage installations and Certificates of Compliance for dry storage systems, as codified in Title 10 of the *Code of Federal Regulations* Part 72, "Licensing Requirements for the Independent Storage of Spent Nuclear Fuel, High-Level Radioactive Waste, and Reactor-Related Greater Than Class C Waste."

The MAPS Report evaluates known aging degradation mechanisms to determine if they could affect the ability of dry storage system components to fulfill their safety functions in the 20- to 60-year period of extended operation. The guidance also provides examples of aging management programs that are considered generically acceptable to address the credible aging mechanisms to ensure that the design bases of dry storage systems will be maintained. An applicant for a renewed license or Certificate of Compliance may reference the information in the MAPS Report to support its aging management review and proposed aging management programs.

Paperwork Reduction Act

This NUREG provides guidance for implementing the mandatory information collections in 10 CFR Part 72 that are subject to the Paperwork Reduction Act of 1995 (44 U.S.C. 3501 et. seq.). These information collections were approved by the Office of Management and Budget (OMB) under control number 3150-0132. Send comments regarding this information collection to the Information Services Branch, U.S. Nuclear Regulatory Commission, Washington, DC 20555-0001, or by e-mail to Infocollects.Resource@nrc.gov, and to the Desk Officer, Office of Information and Regulatory Affairs, NEOB-10202, (3150 -0132) Office of Management and Budget, Washington, DC 20503.

Public Protection Notification

The NRC may not conduct or sponsor, and a person is not required to respond to, a collection of information unless the document requesting or requiring the collection displays a currently valid OMB control number.

CONTENTS

1			
2			
3	ABSTRACT		iii
4	LIST OF FIGURES		ix
5	LIST OF TABLES		xi
6	ABBREVIATIONS AND ACRONYMS		xiii
7	1 INTRODUCTION		1-1
8	1.1 Purpose and use of the MAPS Report		1-1
9	1.2 Scope of report		1-2
10	1.3 Acknowledgments.....		1-3
11	1.4 References		1-4
12	2 DEFINITIONS		2-1
13	2.1 Materials		2-1
14	2.2 Environments.....		2-3
15	2.3 Aging mechanisms.....		2-4
16	2.4 Aging effects		2-9
17	3 EVALUATION OF AGING MECHANISMS		3-1
18	3.1 Introduction.....		3-1
19	3.2 Casks and internals		3-7
20	3.2.1 Steel (carbon, low-alloy, high-strength low-alloy).....		3-7
21	3.2.1.1 General corrosion		3-8
22	3.2.1.2 Pitting and crevice corrosion.....		3-10
23	3.2.1.3 Galvanic corrosion.....		3-11
24	3.2.1.4 Microbiologically influenced corrosion.....		3-11
25	3.2.1.5 Stress corrosion cracking		3-13
26	3.2.1.6 Creep		3-13
27	3.2.1.7 Fatigue		3-14
28	3.2.1.8 Thermal aging		3-15
29	3.2.1.9 Radiation embrittlement.....		3-16
30	3.2.1.10 Stress relaxation.....		3-17
31	3.2.1.11 Wear.....		3-17
32	3.2.2 Stainless steel		3-17
33	3.2.2.1 General corrosion		3-18
34	3.2.2.2 Pitting and crevice corrosion.....		3-18
35	3.2.2.3 Galvanic corrosion.....		3-19
36	3.2.2.4 Microbiologically influenced corrosion.....		3-19
37	3.2.2.5 Stress corrosion cracking		3-20
38	3.2.2.6 Creep		3-22
39	3.2.2.7 Fatigue		3-23
40	3.2.2.8 Thermal aging		3-23
41	3.2.2.9 Radiation embrittlement.....		3-25
42	3.2.2.10 Stress relaxation.....		3-26
43	3.2.2.11 Wear.....		3-26

1		3.2.3	Aluminum alloys	3-26
2		3.2.3.1	General corrosion	3-26
3		3.2.3.2	Pitting and crevice corrosion.....	3-27
4		3.2.3.3	Galvanic corrosion.....	3-28
5		3.2.3.4	Microbiologically influenced corrosion.....	3-29
6		3.2.3.5	Creep	3-29
7		3.2.3.6	Fatigue	3-31
8		3.2.3.7	Thermal aging	3-31
9		3.2.3.8	Radiation embrittlement.....	3-32
10		3.2.4	Nickel alloys	3-33
11		3.2.4.1	General corrosion	3-33
12		3.2.4.2	Pitting and crevice corrosion.....	3-33
13		3.2.4.3	Microbiologically influenced corrosion.....	3-34
14		3.2.4.4	Stress corrosion cracking	3-34
15		3.2.4.5	Fatigue	3-34
16		3.2.4.6	Radiation embrittlement.....	3-35
17		3.2.4.7	Stress relaxation.....	3-35
18		3.2.4.8	Wear.....	3-35
19		3.2.5	Copper alloys	3-36
20		3.2.5.1	General corrosion	3-36
21		3.2.5.2	Pitting and crevice corrosion.....	3-36
22		3.2.5.3	Microbiologically influenced corrosion.....	3-37
23		3.2.5.4	Radiation embrittlement.....	3-37
24		3.2.6	Lead	3-37
25		3.2.7	Depleted uranium	3-37
26		3.2.8	Coatings.....	3-38
27		3.2.9	References.....	3-38
28	3.3		Neutron shielding materials.....	3-49
29		3.3.1	Neutron-shielding materials.....	3-49
30		3.3.1.1	Boron depletion (borated materials).....	3-49
31		3.3.1.2	Thermal aging	3-50
32		3.3.1.3	Radiation embrittlement.....	3-50
33		3.3.2	References.....	3-51
34	3.4		Neutron poison materials	3-53
35		3.4.1	Borated stainless steel	3-53
36		3.4.1.1	Boron depletion	3-54
37		3.4.1.2	Creep	3-54
38		3.4.1.3	Thermal aging	3-55
39		3.4.1.4	Radiation embrittlement.....	3-55
40		3.4.2	Borated aluminum alloys and aluminum-based composites	3-55
41		3.4.2.1	General corrosion.....	3-56
42		3.4.2.2	Galvanic corrosion.....	3-56
43		3.4.2.3	Wet corrosion and blistering	3-56
44		3.4.2.4	Boron depletion	3-57
45		3.4.2.5	Creep	3-57
46		3.4.2.6	Thermal aging	3-58
47		3.4.2.7	Radiation embrittlement.....	3-58
48		3.4.3	References.....	3-59
49	3.5		Concrete overpacks, support pads, and ceramic fiber insulation.....	3-61
50		3.5.1	Concrete.....	3-62
51		3.5.1.1	Freeze and thaw.....	3-62

1		3.5.1.2 Creep	3-63
2		3.5.1.3 Reaction with aggregates	3-63
3		3.5.1.4 Differential settlement.....	3-64
4		3.5.1.5 Aggressive chemical attack	3-65
5		3.5.1.6 Corrosion of reinforcing steel	3-67
6		3.5.1.7 Shrinkage	3-68
7		3.5.1.8 Leaching of calcium hydroxide.....	3-69
8		3.5.1.9 Radiation damage	3-70
9		3.5.1.10 Fatigue	3-70
10		3.5.1.11 Dehydration at high temperature	3-71
11		3.5.1.12 Microbiological degradation	3-72
12		3.5.1.13 Delayed ettringite formation	3-73
13		3.5.1.14 Salt scaling	3-74
14	3.5.2	Ceramic fiber insulation	3-74
15		3.5.2.1 Radiation damage	3-75
16		3.5.2.2 Moisture absorption	3-76
17	3.5.3	References.....	3-76
18	3.6	Spent fuel assemblies.....	3-85
19	3.6.1	Cladding materials.....	3-86
20		3.6.1.1 Hydride reorientation (high burnup fuel).....	3-86
21		3.6.1.2 Delayed hydride cracking (high burnup fuel).....	3-89
22		3.6.1.3 Thermal creep (high burnup fuel).....	3-91
23		3.6.1.4 Low-temperature creep (high burnup fuel).....	3-93
24		3.6.1.5 Mechanical overload (high burnup fuel)	3-94
25		3.6.1.6 Oxidation	3-96
26		3.6.1.7 Pitting corrosion.....	3-96
27		3.6.1.8 Galvanic corrosion.....	3-97
28		3.6.1.9 Stress corrosion cracking	3-98
29		3.6.1.10 Radiation embrittlement.....	3-99
30		3.6.1.11 Fatigue	3-99
31	3.6.2	Assembly hardware materials.....	3-101
32		3.6.2.1 Creep	3-101
33		3.6.2.2 Hydriding	3-102
34		3.6.2.3 General corrosion.....	3-102
35		3.6.2.4 Stress corrosion cracking	3-103
36		3.6.2.5 Radiation embrittlement.....	3-104
37		3.6.2.6 Fatigue	3-104
38	3.6.3	References.....	3-104
39	4	ANALYSIS OF DRY STORAGE SYSTEMS AND SPENT FUEL ASSEMBLIES.....	4-1
40	4.1	Introduction	4-1
41	4.2	NUHOMS® systems: standardized and standardized advanced.....	4-3
42		4.2.1 System description	4-3
43		4.2.2 Dry shielded canister	4-3
44		4.2.3 Horizontal storage module.....	4-6
45		4.2.4 Transfer cask	4-8
46	4.3	HI-STORM 100 and HI-STAR 100 systems	4-91
47		4.3.1 System description	4-91
48		4.3.2 Multipurpose canister	4-91
49		4.3.3 HI-STORM concrete overpack.....	4-94
50		4.3.4 HI-STAR metal overpack.....	4-96

1		4.3.5 Transfer cask	4-99
2	4.4	TN-32 and TN-68 systems	4-151
3		4.4.1 System description	4-151
4		4.4.2 Bolted metal cask	4-151
5	4.5	NAC International Systems	4-169
6		4.5.1 System description	4-169
7		4.5.2 NAC-UMS	4-169
8		4.5.3 NAC-MPC	4-170
9		4.5.4 MAGNASTOR	4-171
10	4.6	FuelSolutions™ storage system	4-259
11		4.6.1 System description	4-259
12		4.6.2 W21 and W74 canisters	4-259
13		4.6.3 W150 storage cask	4-260
14		4.6.4 W100 transfer cask	4-262
15	4.7	Concrete pad	4-291
16	4.8	Spent fuel assemblies	4-297
17		4.8.1 Spent fuel assembly description	4-297
18		4.8.2 Fuel cladding and assembly hardware	4-297
19	4.9	References	4-303
20	5	TIME-LIMITED AGING ANALYSES	5-1
21		5.1 Introduction	5-1
22		5.2 Review	5-1
23		5.3 References	5-2
24	6	EXAMPLE AGING MANAGEMENT PROGRAMS	6-1
25		6.1 Introduction	6-1
26		6.2 Alternative approaches	6-1
27		6.3 Learning aging management	6-2
28		6.4 References	6-3
29		6.5 Localized corrosion and stress corrosion cracking of welded stainless	
30		steel dry storage canisters	6-5
31		6.6 Reinforced concrete structures	6-17
32		6.7 External surfaces monitoring of metallic components	6-29
33		6.8 Ventilation systems	6-35
34		6.9 Bolted cask seal leakage monitoring	6-43
35		6.10 Transfer casks	6-53
36		6.11 High-burnup fuel monitoring and assessment	6-59

LIST OF FIGURES

2	Figure 1-1	Use of the MAPS Report in the renewal process	1-3
3	Figure 4-1	NUHOMS dry storage system	4-4
4	Figure 4-2	NUHOMS-24PT2 DSC assembly–spacer disk design.....	4-4
5	Figure 4-3	NUHOMS-32PT DSC assembly–tube or plate design.....	4-5
6	Figure 4-4	Pressure and confinement boundaries for NUHOMS-32PT DSC	4-5
7	Figure 4-5	Air flow diagram for a typical standardized HSM design.....	4-7
8	Figure 4-6	Advanced NUHOMS horizontal storage module	4-7
9	Figure 4-7	Side elevation and end view of the DSC support structure.....	4-8
10	Figure 4-8	OS197L transfer cask	4-9
11	Figure 4-9	HI-STORM 100 and HI-STAR 100 systems	4-92
12	Figure 4-10	Cross section elevation view of MPC	4-92
13	Figure 4-11	Cross sectional views of different MPC designs.....	4-93
14	Figure 4-12	MPC confinement boundary.....	4-95
15	Figure 4-13	Cross sectional views of the HI-STORM 100 and 100S overpacks	
16		with an MPC inserted.....	4-95
17	Figure 4-14	HI-STAR 100 overpack elevation view.....	4-98
18	Figure 4-15	HI-STAR 100 overpack cross sectional view.....	4-98
19	Figure 4-16	Cross sectional views of the HI-TRAC 125 transfer cask with pool lid	
20		and transfer lid	4-100
21	Figure 4-17	Components of the TN-32 storage cask.....	4-152
22	Figure 4-18	TN-68 cask confinement boundary components	4-152
23	Figure 4-19	TN-32 cask seal pressure-monitoring system	4-153
24	Figure 4-20	Radial cross section of TN-32 cask showing basket, basket rails, and	
25		gamma and neutron shields	4-154
26	Figure 4-21	TN-32 cask shielding configuration	4-155
27	Figure 4-22	NAC-UMS	4-172
28	Figure 4-23	NAC-UMS transportable storage canister for PWR fuel	4-173
29	Figure 4-24	NAC-UMS VCC and transfer cask arrangement	4-173
30	Figure 4-25	NAC MAGNASTOR TSC and concrete cask.....	4-174
31	Figure 4-26	NAC MAGNASTOR concrete cask	4-174
32	Figure 4-27	Typical FuelSolutions™ W21 and W74 canisters	4-260
33	Figure 4-28	FuelSolutions™ W150 storage cask.....	4-261
34	Figure 4-29	FuelSolutions™ W100 transfer cask	4-263
35	Figure 4-30	Typical pressurized-water reactor fuel assembly.....	4-298
36	Figure 4-31	Boiling-water reactor GE14 fuel assembly	4-299
37			

LIST OF TABLES

2	Table 2-1	Use of terms for materials	2-1
3	Table 2-2	Use of terms for environments	2-3
4	Table 2-3	Use of terms for aging mechanisms	2-4
5	Table 2-4	Use of terms for aging effects.....	2-9
6	Table 3-1	Environment abbreviations	3-1
7	Table 3-2	Casks and internals aging mechanism evaluations	3-2
8	Table 3-3	Neutron shielding materials aging mechanism evaluations.....	3-4
9	Table 3-4	Neutron poison materials aging mechanism evaluations	3-4
10	Table 3-5	Concrete overpacks, support pads, and ceramic fiber insulation aging	
11		mechanism evaluations.....	3-5
12	Table 3-6	Spent fuel assembly aging mechanism evaluations	3-6
13	Table 4-1	Evaluated storage system designs	4-1
14	Table 4-2	Standardized NUHOMS dry shielded canister	4-10
15	Table 4-3	Standardized Advanced NUHOMS dry shielded canister	4-22
16	Table 4-4	Standardized NUHOMS horizontal storage module.....	4-33
17	Table 4-5	Standardized Advanced NUHOMS horizontal storage module	4-52
18	Table 4-6	NUHOMS transfer cask.....	4-69
19	Table 4-7	HI-STORM / HI-STAR multipurpose canister.....	4-101
20	Table 4-8	HI-STORM 100 overpack	4-110
21	Table 4-9	HI-STAR 100 overpack	4-129
22	Table 4-10	HI-TRAC transfer cask	4-139
23	Table 4-11	TN bolted metal casks.....	4-156
24	Table 4-12	NAC-UMS transportable storage canister.....	4-175
25	Table 4-13	NAC-UMS vertical concrete cask	4-184
26	Table 4-14	NAC-UMS transfer cask	4-194
27	Table 4-15	NAC-MPC transportable storage canister.....	4-200
28	Table 4-16	NAC-MPC vertical concrete cask	4-213
29	Table 4-17	NAC-MPC transfer cask	4-223
30	Table 4-18	MAGNASTOR transportable storage canister	4-228
31	Table 4-19	MAGNASTOR concrete cask	4-241
32	Table 4-20	MAGNASTOR transfer cask.....	4-250
33	Table 4-21	FuelSolutions™ canister	4-264
34	Table 4-22	FuelSolutions™ storage cask	4-274
35	Table 4-23	FuelSolutions™ transfer cask	4-282
36	Table 4-24	Concrete pad.....	4-292
37	Table 4-25	Spent fuel assemblies	4-300
38	Table 5-1	Examples of fatigue analyses contained within storage system design bases	5-2
39	Table 6-1	Example aging management programs.....	6-1
40	Table 6-2	Example aging management program for Localized Corrosion And Stress	
41		Corrosion Cracking Of Welded Stainless Steel Dry Storage Canisters	6-6
42	Table 6-3	Example aging management program for Reinforced Concrete Structures	6-18
43	Table 6-4	Example aging management program for External Surfaces Monitoring	
44		Of Metallic Components.....	6-30
45	Table 6-5	Example aging management program for Ventilation Systems.....	6-36
46	Table 6-6	Example aging management program for Bolted Cask Seal Leakage	
47		Monitoring	6-44
48	Table 6-7	Example aging management program for Transfer Casks.....	6-54
49	Table 6-8	Example aging management program for High-Burnup Fuel Monitoring	
50		and Assessment.....	6-60

ABBREVIATIONS AND ACRONYMS

ACI	American Concrete Institute
ADAMS	Agencywide Documents Access and Management System
AISC	American Institute of Steel Construction
AMP	aging management program
AMR	aging management review
ANSI	American National Standards Institute
ASCE	American Society of Civil Engineers
ASME	American Society of Mechanical Engineers
ASR	alkali-silica reaction
B&PV	boiler and pressure vessel
BWR	boiling-water reactor
CAP	corrective action program
CFR	<i>Code of Federal Regulations</i>
CoC	Certificate of Compliance
CISCC	chloride-induced stress corrosion cracking
CWSR	cold worked stress relieved
DBTT	ductile-to-brittle transition temperature
DEF	delayed ettringite formation
DHC	delayed hydride cracking
DOE	U.S. Department of Energy
DSC	dry shielded canister
DSS	dry storage system
EPRI	Electric Power Research Institute
FSAR	final safety analysis report
HBU	high burnup
HDRP	HBU Dry Storage Cask Research and Development Project
HSM	horizontal storage module
IFBA	integral fuel burnable absorber
IN	Information Notice
ISFSI	independent spent fuel storage installation
ISG	Interim Staff Guidance
MAPS	Managing Aging Processes in Storage
MIC	microbiologically influenced corrosion
MPC	multipurpose canister
NDE	nondestructive examination
NRC	U.S. Nuclear Regulatory Commission
PCMI	pellet-to-cladding mechanical interaction
PWR	pressurized-water reactor
QA	quality assurance

RIA	reactivity-initiated accident
RXA	recrystallized annealed
SCC	stress corrosion cracking
SNF	spent nuclear fuel
SSC	structure, system, and component
TC	transfer cask
TLAA	time-limited aging analysis
TMI	Three Mile Island
TN	Transnuclear Inc.
TS	technical specification(s)
TSC	transportable storage canister
VCC	ventilated concrete cask
VVM	vertical ventilated module

Units of Measure

atm	atmosphere (pressure)
C	Celsius
dpa	displacements per atom (radiation damage)
F	Fahrenheit
g	gram
gal	gallon
GWd/MTU	gigawatt-days per metric ton of uranium
in	inch
K	Kelvin
kGy	kilogray (absorbed radiation dose)
ksi	1,000 pounds per square inch
L	liter
mg	milligram, 0.001 grams
MPa	megapascal, 1×10^6 pascals (stress)
MeV	megaelectron-volt, 1×10^6 electron-volts (energy)
mil	one-thousandth of an inch, 0.001 inch
mpy	mils per year
mm	millimeter, 0.001 meter
n	neutrons
oz	ounce
ppm	parts per million
psi	Pounds per square inch
rad	(unit of absorbed radiation dose)
sec	second
μm	micrometer, 1×10^{-6} meter
yr	year

1 INTRODUCTION

1.1 Purpose and use of the MAPS Report

The U.S. Nuclear Regulatory Commission (NRC) licenses the storage of spent nuclear fuel (SNF) in dry storage systems (DSSs) under the regulations of Title 10 of the *Code of Federal Regulations* (10 CFR) Part 72, “Licensing Requirements for the Independent Storage of Spent Nuclear Fuel, High-Level Radioactive Waste, and Reactor-Related Greater Than Class C Waste.” To date, licenses for specific independent spent fuel storage installations (ISFSIs) or Certificates of Compliance (CoCs) for DSSs have been issued for initial terms of 20 years, although regulations currently allow an initial 40-year storage period. Licenses and CoCs can be renewed for additional terms not to exceed 40 years. In accordance with 10 CFR 72.42, “Duration of License; Renewal,” and 10 CFR 72.240, “Conditions for Spent Fuel Storage Cask Renewal,” renewal applications must include:

- i. time-limited aging analyses (TLAAs) that demonstrate that structures, systems, and components (SSCs) important to safety will continue to perform their intended function for the requested period of extended operation
- ii. aging management programs (AMPs) for management of issues associated with aging that could adversely affect SSCs important to safety

NUREG–1927, Revision 1, “Standard Review Plan for Renewal of Specific Licenses and Certificates of Compliance for Dry Storage of Spent Nuclear Fuel,” provides guidance for the staff’s review of TLAAs and AMPs (NRC, 2016).

This Managing Aging Processes in Storage (MAPS) Report is a technical basis document that provides additional guidance to NRC staff to improve the effectiveness and efficiency of the renewal process for the dry storage of SNF. The MAPS Report provides a generic evaluation of the aging mechanisms that have the potential to challenge the ability of DSS SSCs to fulfill their important-to-safety functions. The MAPS Report also describes acceptable generic AMPs that an applicant may use to maintain the approved design basis of its storage system during the period of extended operation (from 20 to 60 years of storage¹). An applicant for a renewed license or CoC may reference the information in the MAPS Report to support its design-specific aging management review (AMR) and proposed AMPs.

The content of the report is as follows.

- Chapter 1 briefly describes how the MAPS Report is to be used by the NRC staff.
- Chapter 2 defines the terms that are used throughout this report, including descriptions of materials, environments, aging mechanisms, and aging effects (the manifestations of aging mechanisms by degraded conditions or performance).
- Chapter 3 evaluates the aging mechanisms that may challenge the ability of SSCs to fulfill their important-to-safety function(s). Those mechanisms that are shown to have the potential to adversely affect an important-to-safety function in the 60-year timeframe

¹Because the NRC has granted, to date, initial storage licenses and CoCs for 20 years only, the MAPS Report considers the effects of aging for 40 years beyond the initial 20-year term (or 60 years total).

1 are identified as “credible.” This chapter provides the technical bases for the aging
2 management recommendations that appear in the AMR tables and AMPs in Chapters 4
3 and 5, respectively.

4 • Chapter 4 describes selected DSS designs and provides AMR tables for those designs.
5 The AMR tables identify the aging mechanisms and effects that could challenge the
6 capability of each SSC to fulfill its important-to-safety function(s) in the 20- to 60-year
7 period of extended operation. For those credible aging effects, the AMR tables
8 recommend aging management approaches (i.e., AMPs, TLAAAs, or other analyses).

9 • Chapter 5 provides guidance for identifying and evaluating time-limited aging analyses

10 • Chapter 6 contains example AMPs that an applicant may use to address the credible
11 aging effects identified in the AMR tables.

12 Figure 1-1 provides a flowchart that shows how the guidance in the MAPS Report supports the
13 renewal process.

14 The MAPS Report increases the efficiency of the licensing process by reducing redundant
15 reviews of the same topic. If an applicant credits the information in the MAPS Report in the
16 renewal application, the staff should ensure that the applicant demonstrates that the design
17 features, environmental conditions, and operating experience for the subject ISFSI or DSS are
18 bounded by those evaluated in the MAPS Report. Otherwise, the staff should ensure that the
19 applicant revises its AMR and AMPs, as appropriate, to address the design or operating
20 parameters applicable to its facility or storage system.

21 The MAPS Report contains one acceptable method to identify and manage credible aging
22 mechanisms and effects for specific-license and CoC renewals. An applicant may propose
23 alternatives for staff review. As such, the staff should not use the MAPS Report as a
24 requirement. Nevertheless, its use should facilitate both the preparation of a specific license or
25 CoC renewal application by an applicant and a timely, consistent review by the NRC staff.

26 Finally, the MAPS Report does not address the scoping of SSCs for specific-license or CoC
27 renewal; this is addressed in Chapter 2 of NUREG–1927, Revision 1. Although the MAPS
28 Report generically addresses SSCs for several storage system designs, scoping is design and
29 license specific. The inclusion of a certain SSC in the MAPS Report does not necessarily imply
30 that the particular SSC is within the scope of renewal for all ISFSIs or DSSs. Conversely, the
31 omission of a certain SSC in the MAPS Report does not imply that the particular SSC is not
32 within the scope of renewal for any ISFSI or DSS.

33 **1.2 Scope of report**

34 The MAPS Report addresses the aging mechanisms and effects associated with the following
35 DSS designs: Standardized and Advanced NUHOMS, HI-STORM 100, HI-STAR 100, TN-32
36 and -68, the NAC UMS, MPC, and MAGNASTOR systems, and the FuelSolutions storage
37 system. The selection of these systems addresses near-term renewal applications and a
38 variety of storage system designs. Although this report was written to specifically address those
39 designs, the staff may consider the general applicability of this guidance to other designs.

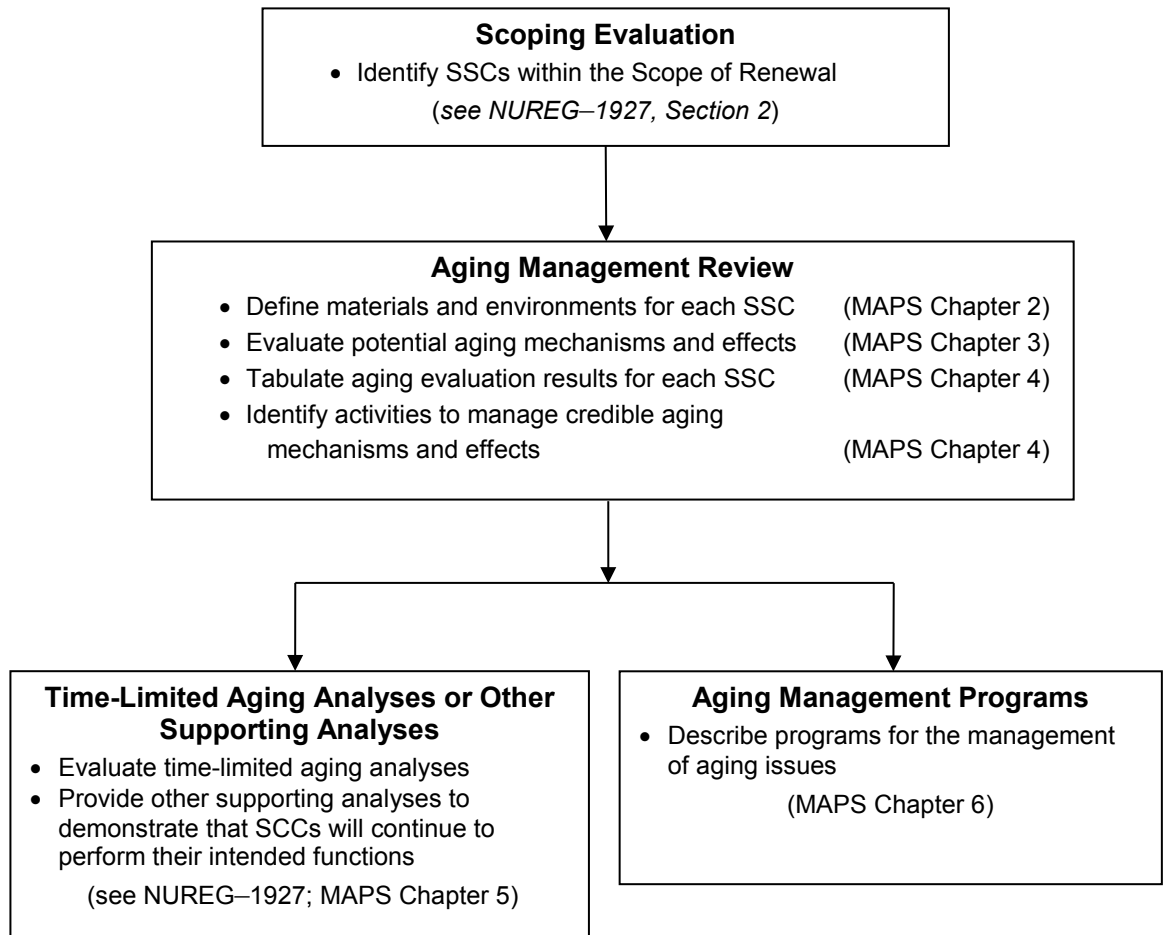


Figure 1-1 Use of the MAPS Report in the renewal process

1 **1.3 Acknowledgments**

2 The NRC would like to acknowledge the contributions of the staff at the Center for Nuclear
 3 Waste Regulatory Analyses at the Southwest Research Institute® for its role in developing the
 4 technical bases for the aging evaluations in this report. This includes the evaluations of the
 5 aging mechanisms in Chapter 3 and the associated AMR tables in Chapter 4. The staff at the
 6 CNWRA also assisted in the development of the introductory material and combining all
 7 portions of this document into a single, cohesive report.

8 The NRC also would like to acknowledge the contributions of Argonne National Laboratory in
 9 support of the U.S. Department of Energy (DOE) Used Fuel Disposition Campaign. Portions of
 10 the storage system descriptions in Chapter 4 of the MAPS Report were taken from the
 11 information contained in Chapter V of the Argonne/DOE report, “Managing Aging Effects on Dry
 12 Cask Storage Systems for Extended Long-Term Storage and Transportation of Used Fuel”
 13 (Chopra et al., 2014).

14

1 **1.4 References**

2 Chopra, O., D. Diercks, R. Fabian, Z. Han, and Y. Liu. "Managing Aging Effects on Dry Cask
3 Storage Systems for Extended Long-Term Storage and Transportation of Used Fuel."
4 FCRD-UFD-2014-000476. ANL-13/15, Rev. 2. Washington, DC.: U.S. Department of
5 Energy. 2014.

6 NRC. NUREG-1927, "Standard Review Plan for Renewal of Specific Licenses and Certificates
7 of Compliance for Dry Storage of Spent Nuclear Fuel." Revision 1. Washington, DC.:
8 U.S. Nuclear Regulatory Commission. Agencywide Documents Access and Management
9 System Accession No. ML16179A148. 2016.

1

2 DEFINITIONS

2 This chapter defines the usage of terms in the technical basis discussions in Chapter 3, the
3 aging management review (AMR) tables in Chapter 4, and the aging management programs in
4 Chapter 5. Selected definitions and usage are provided for the materials of construction,
5 service environments, aging mechanisms, and aging effects (the manifestations of aging
6 mechanisms by degraded conditions or performance).

7 **2.1 Materials**

8 Table 2-1 describes many of the terms used to describe the materials of construction for the dry
9 storage systems (DSSs).

Table 2-1 Use of terms for materials	
Term	Usage in This Document
Aluminum	Includes commercially pure aluminum 1100 and precipitation-hardened alloys 6061 and 6063.
BISCO NS-3	A castable cementitious material for neutron and gamma shielding applications that may be blended with boron fillers to enhance neutron attenuation. It is fully encased in a metal, such as aluminum or steel.
Boral [®]	A laminate composite that is used as a neutron poison material. It consists of a core of aluminum and boron-carbide powder sandwiched between sheets of aluminum. The boron-carbide content in the core ranges from 35 to 65 weight percent.
Boralyn [®] , Metamic [™]	Two variations of boron-carbide aluminum metal-matrix composite for neutron poison applications, one with billets produced by vacuum hot pressing (Boralyn [®]) and the second produced by cold isostatic pressing followed by vacuum sintering (Metamic [™]).
Borated aluminum	An aluminum alloy typically containing up to 4.5 weight percent boron. It is used as a neutron poison material. The boron is incorporated in the aluminum matrix as discrete particles of AlB ₂ or TiB ₂ (for alloys also containing titanium). Aluminum alloys 1100, 6063, and 6351 have been used as base materials for boron additions.
Borated polymers	Borated polymers include borated polyester resin and polypropylene for neutron shielding applications. Borated polyester resin is an unsaturated polyester crosslinked with styrene and typically contains about 50 weight percent mineral and fiberglass reinforcement.
Borated stainless steel	An austenitic chromium-nickel steel with boron additions up to 2.5 weight percent. It is used as a neutron poison material. The boron in the form of borides is dispersed in the Type 304 stainless steel matrix as an intermetallic phase.
Concrete	A mixture of hydraulic cement, aggregates, and water, with or without admixtures, fibers, or other cementitious materials.

Table 2-1 Use of terms for materials	
Term	Usage in This Document
Copper alloys	Copper alloys used in DSSs include bronzes (copper alloyed with tin) and brasses (copper alloyed with zinc).
Holtite-A™	A Holtec neutron shielding material consisting of epoxy polymer, B ₄ C added as a finely divided powder, and aluminum hydroxide. It is fully encased in a metal enclosure.
Nickel alloys	Nickel alloys include Inconel 718 and X750. Inconel is a family of austenitic nickel-chromium-based superalloys. Both Inconel 718 and X750 are precipitation-hardening alloys.
Stainless steel	Stainless steel includes Types 304, 316, XM-19, SA193-Gr. B8, SA351-Gr. CF3, and Nitronic 60 austenitic stainless steels and Type 630 precipitation-hardening martensitic stainless steel. Type 630 stainless steel is commonly referred to as 17-4PH and contains 15–17.5 percent chromium, 3–5 percent copper, and 3–5 percent nickel (in weight percent). Chrome-plated stainless steel is also included in the category of stainless steel.
Steel	Various carbon steels, alloy steels, and high-strength, low-alloy steels. Examples of steel designations included in this category are ASTM A36, ASTM A320-Gr. L43, ASTM F436, SA36, SA193-Gr. B7, SA203-Gr. D/E, SA266-Cl. 2, SA320-Gr. L43, SA350-Gr. LF2/LF3, SA414, SA508-Cl. 1A/3A, SA516-Gr. 70, SA533-Gr. B, SA537-Cl. 2, SA540-Gr. B23/24, SA620, and SA696-Gr. B. Galvanized steel, aluminum-coated steel, and electroless nickel-plated steel are also included in the category of steel.
Zirconium-based alloys	The materials of construction of fuel cladding and fuel assembly hardware. Various zirconium-based materials have been used in commercial reactor applications because of their low neutron cross section and excellent corrosion resistance to a variety of environmental conditions. The cladding types Zircaloy-2, Zircaloy-4, ZIRLO™, and M5® are included in this category. Zirconium-based cladding in high burnup (HBU) spent nuclear fuel refers to assembly-average burnups exceeding 45 GWd/MTU.

1 **2.2 Environments**

2 Table 2-2 defines many of the environments to which DSS SSCs are exposed.

Table 2-2 Use of terms for environments	
Term	Usage in This Document
Air–outdoor (OD)	Direct exposure to weather, including precipitation and wind; possibly salt laden. The indoor air environment to which transfer cask components are typically exposed is conservatively evaluated as outdoor air.
Demineralized water (DW)	Water that has been treated to remove dissolved minerals. Demineralized water is used as the liquid neutron shield in transfer casks.
Embedded in: Concrete (E-C) Metal (E-M) Neutron shielding (E-NS)	When one or more surfaces of a component are in contact with another component or material. This may prevent ingress of water and contaminants to the embedded surface, depending on the permeability of the embedding environment.
Fully encased or lined (FE)	The environment of some concrete structures that are fully enclosed inside another component or fully lined by another material (e.g., steel), which prevents ingress of water and contaminants. Also, ceramic fiber insulation is fully encased in foil-facing or jacketing.
Helium (HE)	The helium fill gas inside a canister or cask and trace quantities of other gases, such as nitrogen, oxygen, argon, and fission product gases. This environment applies to fuel, cladding, and other internal components inside a cask.
Groundwater/soil (GW)	Groundwater is subsurface water found in wells, tunnels, or drainage galleries, or water that flows naturally to the earth’s surface via seeps or springs. Soil is a mixture of organic and inorganic materials produced by the weathering of rock and clay minerals or the decomposition of vegetation. Voids containing air and moisture can occupy 30 to 60 percent of the soil volume. Below-grade concrete structures are assumed to be partially exposed to a groundwater or soil environment.
Sheltered (SH)	The environment outside a sealed canister but within the confined internal space of a shielding structure (e.g., overpack or horizontal storage module). The sheltered environment is open to outdoor air, but it is shielded from direct exposure to precipitation. This environment may contain moisture, salts, and other contaminants from the outdoor air.

3

1 **2.3 Aging mechanisms**

2 Table 2-3 defines the aging mechanisms that are evaluated in this report.

Table 2-3 Use of terms for aging mechanisms	
Term	Usage in This Document
Aggressive chemical attack	The degradation of concrete by strong acids. Chlorides and sulfates of potassium, sodium, and magnesium may attack concrete, depending on their concentrations in the soil/groundwater that comes into contact with the concrete. The minimum thresholds causing concrete degradation are 500 ppm chloride and 1,500 ppm sulfate.
Boron depletion	The degradation of the neutron-absorbing capacity of neutron poison and shielding materials when they are exposed to neutron fluence.
Corrosion	The electrochemical reaction of a metal or metal alloy in an environment that results in oxidation or wastage of the material.
Creep	Creep, for a metallic material, refers to a time-dependent continuous deformation process under constant stress. It is a thermally activated process and is generally a concern at temperatures greater than 40 percent of the material's absolute melting temperature. However, low-temperature creep is an athermal process that is considered as a potential degradation mechanism for some alloys, including zirconium-based alloys. In concrete, creep is related to the loss of absorbed water from the hydrated cement paste. It is a function of the modulus of elasticity of the aggregate.
Crevice corrosion	Localized corrosion in joints, connections, and other small, close-fitting regions that develop local aggressive environments.
Dehydration at high temperatures	Dehydration reactions of the hydrated cement paste in concrete when exposed to temperatures greater than 65 degrees C [149 degrees F]. Dehydration can degrade concrete strength and increase susceptibility to cracking. The degree of concrete degradation depends on several factors, including concrete mixing, aggregate type, curing, loading condition, moisture retention and content, and exposure time.
Delayed ettringite formation	During concrete curing, the naturally occurring ettringite (a calcium aluminum sulfate mineral) converts to monosulfoaluminate if curing temperatures are greater than about 70 degrees C [158 degrees F]. After concrete hardens, ettringite will reform if the temperature decreases below about 70 degrees C [158 degrees F], resulting in concrete cracking and spalling. The conditions necessary for the occurrence of delayed ettringite formation are excessive temperatures during concrete casting, the presence of internal sulfates, and a moist environment.

Table 2-3 Use of terms for aging mechanisms

Term	Usage in This Document
Delayed hydride cracking	The propagation of a crack in zirconium-based cladding materials as a result of diffusion of hydrogen to a crack tip and the embrittlement of the near-tip region due to hydride precipitation. The operability of the delayed-hydride-cracking mechanism in fuel cladding depends on the stress imposed on the cladding.
Erosion	Soil erosion, or removal, is primarily caused by rainfall and surface runoff, floods, or wind. Soil erosion can affect the stability of concrete structures, resulting in scouring that is a localized loss of soil, often around a foundation element. Factors that affect the erosion rates include soil structure and composition, climate, topography, and vegetation cover.
Fatigue	Also termed “cyclic loading” or “thermal/mechanical fatigue.” Fatigue is a phenomenon leading to fracture under repeated or fluctuating stresses having a maximum value less than the tensile strength of the material. Fatigue fractures are progressive and grow under the action of the fluctuating stress. Fatigue due to cyclic thermal loads is defined as the structural degradation that can occur from repeated stress/strain cycles caused by fluctuating loads and temperatures. After repeated cyclic loading of sufficient magnitude, microstructural damage may accumulate, leading to macroscopic crack initiation at the most vulnerable regions. Subsequent mechanical or thermal cyclic loading may lead to growth of the initiated crack.
Freeze-thaw	Repeated freezing and thawing of water can cause degradation of concrete, characterized by scaling, cracking, and spalling. The cause is water freezing within the pores of the concrete, creating hydraulic pressure.
Galvanic corrosion	Accelerated corrosion of a metal when in electrical contact with a more noble metal or nonmetallic conductor in a corrosive electrolyte.
General corrosion	Uniform loss of material due to corrosion, proceeding at approximately the same rate over a metal surface.
Hydride reorientation and hydride-induced embrittlement	The precipitation of radial hydrides results in embrittlement of zirconium-based cladding materials under pinch-load stresses at low-to-moderate temperatures. Reorientation of hydrides from the circumferential-axial to radial-axial direction is caused by heating and cooling of the cladding under sufficient cladding hoop tensile stresses.

Table 2-3 Use of terms for aging mechanisms

Term	Usage in This Document
Leaching of calcium hydroxide	The dissolution of calcium-containing concrete components (e.g., calcium hydroxide) when water passes through either cracks, inadequately prepared construction joints, or areas not sufficiently consolidated during placing. Once the calcium hydroxide has been leached away, other cementitious constituents become vulnerable to chemical decomposition, finally leaving only the silica and alumina gels behind and lowering the strength of the concrete. The water's aggressiveness in the leaching of calcium hydroxide depends on its salt content, pH, and temperature. This leaching action is effective only if the water flows through the concrete.
Mechanical overload	The overload of fuel cladding due to fuel pellet swelling. Fuel pellet swelling is the result of decay gas production in the pellet. Pellet swelling can increase stresses on the cladding.
Microbiological degradation	Biodegradation attack of concrete by organisms growing on its surfaces under favorable environmental conditions (e.g., moisture, near neutral pH, presence of nutrients), causing an increase in concrete porosity and permeability and the loss of material by spalling or scaling.
Microbiologically influenced corrosion	Any of the various forms of corrosion influenced by the activity of such microorganisms as bacteria, fungi, and algae, and/or the products of their metabolism. For example, anaerobic bacteria can establish an electrochemical galvanic reaction or disrupt a passive protective film; acid-producing bacteria can produce corrosive metabolites.
Oxidation	A corrosion reaction. In this report, oxidation also is a defined aging mechanism describing the reaction of zirconium alloy fuel rod cladding with water to form zirconium oxide..
Pitting corrosion	A localized form of corrosion that is confined to a point or small area of a metal surface. It takes the form of cavities called pits.
Radiation damage and radiation embrittlement	The loss of ductility, fracture toughness, and resistance to cracking of metals that may occur under exposure to neutron radiation. In concrete, radiation exposure can cause dissociation of water into hydrogen and oxygen, leading to decreased compressive and tensile strengths. The extent of radiation damage to concrete depends on the neutron and gamma fluence.
Reaction with aggregates	The presence of reactive alkalis in concrete can lead to subsequent reactions with aggregates that may lead to cracking, a loss of material, or an increase in porosity and permeability. These alkalis are introduced mainly by cement but also may come from admixtures, salt contamination, seawater penetration, or solutions of deicing salts. These reactions include alkali-silica reactions, cement-aggregate reactions, and aggregate-carbonate reactions.

Table 2-3 Use of terms for aging mechanisms

Term	Usage in This Document
Salt scaling	Salt scaling damage manifests as flaking of material from the concrete surface. Salt scaling takes place when concrete is exposed to freezing temperatures, moisture, and dissolved salts (e.g., deicing salts). This degradation mode affects mainly horizontal concrete surfaces where water ponding can be expected.
Settlement	Settlement of a concrete structure may occur due to changes in the site conditions (e.g., water table). The amount of settlement depends on the foundation material. In soil, loss of form due to settlement can occur during the first several years of placement. Factors that control soil settlement include the type of soil particles and particle packing, the amount of water used during the compaction process, and the height of soil fill.
Shrinkage	Shrinkage of concrete can result from cement hydration and loss of moisture during drying. Cracking and shortening of concrete due to shrinkage can occur early after concrete placement.
Stress corrosion cracking (SCC)	The cracking of a metal produced by the combined action of corrosion and a tensile stress (applied or residual). SCC is highly chemical specific in that certain alloys are likely to undergo SCC only when exposed to a small number of chemical environments.
Stress relaxation	A loss of preload in a heavily loaded bolt. Over time, the clamping force provided by a bolt may decrease due to atomic movement within the stressed bolt material (analogous to the metallic creep mechanism at elevated temperatures).
Thermal aging	Also termed “thermal aging embrittlement” or “thermal embrittlement.” Many materials are intentionally thermally aged during their manufacture to achieve desired mechanical properties. Continued exposure to elevated temperatures during operation can, in some cases, result in undesirable properties. For example, at operating temperatures of 300 to 400 degrees C [572 to 752 degrees F], austenitic stainless steel welds that contain ferrite exhibit a spinodal decomposition of the ferrite phase into ferrite-rich and chromium-rich phases. This may give rise to embrittlement (reduction in fracture toughness), depending on the amount, morphology, and distribution of the ferrite phase and the composition of the stainless steel.
Wear	The removal of surface material due to relative motion between two surfaces or under the influence of hard, abrasive particles. Wear occurs in parts that experience intermittent relative motion or frequent manipulation.

Table 2-3 Use of terms for aging mechanisms

Term	Usage in This Document
Wet corrosion and blistering	A degradation mechanism for neutron poison plates with open porosity as a result of water entering pores in the material during loading, leading to internal corrosion. Blisters occur from trapped hydrogen produced from corrosion reactions. Wet corrosion and blistering can cause dimensional changes affecting criticality considerations due to moderator displacement and may also hinder the retrieval of fuel assemblies.

1

1 **2.4 Aging effects**

- 2 An aging effect is the manifestation of an aging mechanism, as evidenced by a degraded
 3 condition or performance. Table 2-4 defines the aging effects described in this report.

Table 2-4 Use of terms for aging effects	
Term	Usage in This Document
Changes in dimension	A change in the size of a component resulting from creep of aluminum and zirconium-based alloys. Changes in dimension also can be caused by wet corrosion and blistering of Boral® neutron poison materials.
Cracking	Crack initiation and growth in metallic components as a result of SCC, fatigue, and delayed hydride cracking. Cracking in concrete is a complete or incomplete separation of concrete into two or more parts produced by breaking or fracturing.
Increase in porosity and permeability	An increase in the percentage of the volume of voids in a concrete material or an increase in the susceptibility of concrete to permit liquids or gasses to pass through.
Loss of bond	A loss of the interacting force that prevents slip of the reinforcing steel bars relative to the surrounding concrete in a reinforced concrete member.
Loss of criticality control	A diminishment of the capability of neutron poison materials to maintain the subcriticality of spent nuclear fuel.
Loss of form	A change in the shape or position of soil resulting from settlement due to poor soil consolidation. In addition, soil tends to absorb moisture with time and thus promotes loss of form.
Loss of fracture toughness and loss of ductility	A decrease in the ability of a material to resist fracture. This phenomenon results from thermal aging embrittlement, radiation embrittlement, or hydrogen embrittlement.
Loss of material	The destructive removal of material due to general corrosion, pitting corrosion, crevice corrosion, galvanic corrosion, microbiologically influenced corrosion, or aggressive chemical attack. In concrete structures, loss of material can result from local flaking, spalling, or peeling away of the near-surface portion of hardened concrete.
Loss of preload	A reduction in the clamping force in a mechanically loaded joint.
Loss of shielding	A diminishment of the capability of a material to shield radiation.
Loss of strength	A decrease in the ability of a material to support a mechanical load. In metals, loss of strength may be due to thermal aging or annealing. In concrete structures, loss of strength can also be caused by the leaching of calcium hydroxide or reaction with aggregates.
None	A term used in the AMR tables for certain material and environment combinations that may not be subject to credible aging mechanisms; thus, there are no relevant aging effects that require management.

Table 2-4 Use of terms for aging effects	
Term	Usage in This Document
Precursor to SCC	A material condition that initiates SCC. Both pitting and crevice corrosion are known to be precursors to SCC and, as such, can lead to cracking of stainless steel canisters.
Reduction of concrete pH (reducing corrosion resistance of steel embedments)	A decrease in the alkalinity of concrete. If the pH of concrete in which steel is embedded is reduced below 11.5 by intrusion of aggressive ions (e.g., chlorides > 500 ppm) in the presence of oxygen, embedded steel may corrode. A reduction in pH can be caused by carbonation.

1

3 EVALUATION OF AGING MECHANISMS

3.1 Introduction

This chapter evaluates known aging degradation mechanisms to determine which of those could adversely affect an important-to-safety function in the 20- to 60-year period of extended operation. These evaluations provide the technical bases for the recommendations in the aging management review (AMR) tables and aging management programs (AMPs) in Chapters 4 and 6, respectively. This chapter is first divided into major component areas (e.g., casks and internals, concrete overpacks), which in turn are subdivided into discussions of the aging mechanisms for each of the materials of construction (e.g., steel, aluminum).

Each evaluation in this chapter concludes with a determination of whether the aging mechanism is considered “credible” in the period of extended operation. A credible aging mechanism is one that could affect an important-to-safety function if the mechanism were not addressed by an aging management activity. The AMR tables in Chapter 4 recommend an AMP, time-limited aging analysis (TLAA), or other analysis to address the effects of aging.

Table 3-2 through Table 3-6 summarize the conclusions in this chapter. For each material, the tables show in which environments the aging mechanisms were determined to be credible and noncredible. Not all combinations of materials, environments, and aging mechanisms were evaluated in each major component area. This occurs because some material-environment combinations do not exist in every major component area or, in some instances, aging mechanisms were not considered to be reasonably plausible, and thus an evaluation was not performed. The reviewer should note that these conclusions are based only on a review of the specific storage system designs described in Section 1.2 and Chapter 4, and thus the reviewer should consider the credibility of aging mechanisms for other systems on a case-by-case basis.

Table 3-1 provides the environment abbreviations used in the summary tables.

Table 3-1 Environment abbreviations	
Outdoor air	OD
Demineralized water	DW
Embedded in concrete	E-C
Embedded in metal	E-M
Embedded in neutron shielding	E-NS
Fully encased or lined	FE
Helium	HE
Groundwater/soil	GW
Sheltered	SH

Table 3-2 Casks and internals aging mechanism evaluations

Material	Aging Mechanism	Credible Environments	Noncredible Environments	Section
Steel	General corrosion	OD, SH, DW, GW, E-C	E-M, E-NS, HE	3.2.1.1
	Pitting and crevice corrosion	OD, SH, DW, GW, E-C	E-M, E-NS, HE	3.2.1.2
	Galvanic corrosion*	OD, SH		3.2.1.3
	Microbiologically influenced corrosion (MIC)	GW, E-C	OD, SH, DW, E-M, E-NS, HE	3.2.1.4
	Stress corrosion cracking (SCC)		OD, SH	3.2.1.5
	Creep		OD, SH, DW, GW, E-M, E-NS, HE	3.2.1.6
	Fatigue	Evaluate design code TLAA, if applicable		3.2.1.7
	Thermal aging		OD, SH, DW, GW, E-M, E-NS, HE	3.2.1.8
	Radiation embrittlement		OD, SH, DW, GW, E-M, E-NS, HE	3.2.1.9
	Stress relaxation	SH	OD	3.2.1.10
	Wear	OD		3.2.1.11
Stainless Steel	General corrosion		OD, SH, DW, E-M, E-NS, HE	3.2.2.1
	Pitting and crevice corrosion [†]	OD, SH	DW, E-M, E-NS, HE	3.2.2.2
	Galvanic corrosion*	OD, SH		3.2.2.3
	MIC		OD, SH, DW, E-M, E-NS, HE	3.2.2.4
	SCC [‡]	OD, SH	DW, E-M, E-NS, HE	3.2.2.5
	Creep		OD, SH, DW, E-M, E-NS, HE	3.2.2.6
	Fatigue	Evaluate design code TLAA, if applicable		3.2.2.7
	Thermal aging	HE [§]	OD, SH, DW, E-M, E-NS	3.2.2.8
	Radiation embrittlement		OD, SH, DW, E-M, E-NS, HE	3.2.2.9
	Stress relaxation		OD, SH	3.2.2.10
	Wear	OD		3.2.2.11
*where dissimilar material galvanic couples exist				
†as a precursor to SCC				
‡SCC is credible at welds and other regions where sufficient stress exists; transfer cask components exposed to indoor/outdoor air are not considered to be susceptible to SCC because their surfaces are periodically rinsed with demineralized water.				
§thermal aging is credible only for precipitation-hardened martensitic stainless steels				

Table 3-2 Casks and internals aging mechanism evaluations (continued)				
Material	Aging Mechanism	Credible Environments	Noncredible Environments	Section
Aluminum Alloys	General corrosion*	SH	E-M, E-NS, HE	3.2.3.1
	Pitting and crevice corrosion	SH	E-M, E-NS, HE	3.2.3.2
	Galvanic corrosion†	SH	HE	3.2.3.3
	MIC		SH, E-M, E-NS, HE	3.2.3.4
	Creep	analyses required‡		3.2.3.5
	Fatigue	Evaluate design code TLAA, if applicable		3.2.3.6
	Thermal aging	analyses required‡		3.2.3.7
	Radiation embrittlement		SH, E-M, E-NS, HE	3.2.3.8
Nickel Alloys	General corrosion		OD	3.2.4.1
	Pitting and crevice corrosion		OD	3.2.4.2
	MIC		OD	3.2.4.3
	SCC		OD	3.2.4.4
	Fatigue	Evaluate design code TLAA, if applicable		3.2.4.5
	Radiation embrittlement		OD	3.2.4.6
	Stress relaxation		OD	3.2.4.7
	Wear	OD		3.2.4.8
Copper Alloys	General corrosion	OD		3.2.5.1
	Pitting and crevice corrosion		OD	3.2.5.2
	MIC		OD	3.2.5.3
	Radiation embrittlement		OD	3.2.5.4
Lead	All		E-M	3.2.6
Depleted Uranium	All		E-M	3.2.7
Coatings	Radiation embrittlement	analyses required		3.2.8
*general corrosion is not considered to be credible for anodized aluminum				
†where dissimilar metal couples exist				
‡creep and thermal aging are relevant only for load-bearing components.				

1

2

Table 3-3 Neutron shielding materials aging mechanism evaluations				
Material	Aging Mechanism	Credible Environments	Noncredible Environments	Section
Neutron Shielding	Boron depletion	analyses required		3.3.1.1
	Thermal aging	FE*		3.3.1.2
	Radiation embrittlement	FE*		3.3.1.3
*thermal aging and radiation embrittlement are credible only for polymer-based neutron-shielding materials.				

1
2

Table 3-4 Neutron poison materials aging mechanism evaluations				
Material	Aging Mechanism	Credible Environments	Noncredible Environments	Section
Borated Stainless Steels	General corrosion		HE	3.4.1
	Galvanic corrosion		HE	3.4.1
	Wet corrosion and blistering		HE	3.4.1
	Boron depletion		HE*	3.4.1.1
	Creep		HE	3.4.1.2
	Thermal aging		HE	3.4.1.3
	Radiation embrittlement		HE	3.4.1.4
Borated Aluminum and Aluminum-based Composites	General corrosion		HE	3.4.2.1
	Galvanic corrosion		HE	3.4.2.2
	Wet corrosion and blistering		HE	3.4.2.3
	Boron depletion		HE*	3.4.2.4
	Creep		HE [†]	3.4.2.5
	Thermal aging		HE [†]	3.4.2.6
	Radiation embrittlement		HE	3.4.2.7
*when a boron depletion analysis is included in the design basis, applicants must provide a TLAA to demonstrate that depletion will not challenge noncriticality in the period of extended operation [†] creep and thermal aging are relevant only for load-bearing aluminum components.				

3
4

Table 3-5 Concrete overpacks, support pads, and ceramic fiber insulation aging mechanism evaluations

Material	Aging Mechanism	Credible Environments	Noncredible Environments	Section
Concrete	Freeze and thaw	OD, GW (above freeze line)	SH, FE, GW (below freeze line)	3.5.1.1
	Creep		all	3.5.1.2
	Reaction with aggregates	all*		3.5.1.3
	Differential settlement	OD, SH, GW		3.5.1.4
	Aggressive chemical attack	OD, GW	SH, FE	3.5.1.5
	Corrosion of reinforcing steel	OD, GW	SH, FE	3.5.1.6
	Shrinkage		OD, SH, GW, FE	3.5.1.7
	Leaching of calcium hydroxide	OD, SH, GW	FE	3.5.1.8
	Radiation damage		OD, SH, GW, FE	3.5.1.9
	Fatigue		OD, SH, GW, FE	3.5.1.10
	Dehydration at high temperature		OD, SH, GW, FE	3.5.1.11
	Microbiological degradation	GW	OD, SH, FE	3.5.1.12
	Delayed ettringite formation		OD, SH, GW, FE	3.5.1.13
	Salt scaling	OD, GW (above freeze line)	SH, FE, GW (below freeze line)	3.5.1.14
Ceramic Fiber Insulation	Radiation damage	analysis required		3.5.2.1
	Moisture absorption		3.5.2.2	3.5.2.2
*where moisture is available				

1

Table 3-6 Spent fuel assembly aging mechanism evaluations

Material	Aging Mechanism	Credible Environments	Noncredible Environments	Section
Cladding Materials (Zirconium-based Alloys)	Hydride reorientation [†]	HE*		3.6.1.1
	Delayed hydride cracking [†]		HE	3.6.1.2
	Thermal creep [†]	HE		3.6.1.3
	Low-temperature creep [†]		HE	3.6.1.4
	Mechanical overload [†]		HE	3.6.1.5
	Oxidation		HE	3.6.1.6
	Pitting corrosion		HE	3.6.1.7
	Galvanic corrosion		HE	3.6.1.8
	SCC		HE	3.6.1.9
	Radiation embrittlement		HE	3.6.1.10
	Fatigue		HE	3.6.1.11
Assembly Hardware Materials (Zirconium-based, Inconel, and Stainless Steel Alloys)	Creep		HE	3.6.2.1
	Hydriding		HE	3.6.2.2
	General corrosion		HE	3.6.2.3
	SCC		HE	3.6.2.4
	Radiation embrittlement		HE	3.6.2.5
	Fatigue		HE	3.6.2.6
<p>*Although hydride reorientation and hydride-induced embrittlement of high-burnup cladding is credible, these mechanisms are only expected to potentially compromise intended functions under pinch-type loads. Such loads are not expected to be present during storage.</p> <p>[†]applicable to high-burnup fuel</p>				

1 **3.2 Casks and internals**

2 “Casks and internals” include various metallic subcomponents of the storage casks or canisters,
3 the fuel baskets and other internal subcomponents (other than spent fuel assemblies), the
4 storage modules or overpacks, and the transfer casks. These subcomponents are exposed to
5 several environments within and outside the dry storage systems (DSSs), such as sheltered
6 environments, indoor air, outdoor air, demineralized water, groundwater or soil, helium, and
7 embedded environments. The spent nuclear fuel (SNF) also exposes subcomponents to
8 elevated temperatures and radiation, with heat exposure and dose depending on the
9 subcomponent location and the SNF characteristics (e.g., burnup and age of spent fuel). The
10 materials of construction for these subcomponents include steel, stainless steel, aluminum
11 alloys, nickel alloys, copper alloys, and lead.

12 A set of known aging mechanisms for metallic cask and internal subcomponents was
13 established by first broadly identifying all potential mechanisms through a review of gap
14 assessments for DSSs, technical literature, and operating experience from nuclear and
15 nonnuclear applications (NRC, 2014, 2010a; Chopra et al., 2014; Hanson et al., 2012;
16 Sindelar et al., 2011; NWTRB, 2010). The known environmental, thermal, mechanical, and
17 irradiation-induced aging mechanisms are as follows:

- 18 • general corrosion
- 19 • pitting and crevice corrosion
- 20 • galvanic corrosion
- 21 • MIC
- 22 • SCC (including hydrogen embrittlement)
- 23 • creep
- 24 • fatigue
- 25 • thermal aging
- 26 • radiation embrittlement
- 27 • stress relaxation
- 28 • wear

29 Not all of these mechanisms are considered to be credible for each structure, system, and
30 component (SSC). For example, temperatures are not considered sufficiently high to cause
31 creep of steel and stainless steel subcomponents. Also, general corrosion is not considered to
32 be a credible aging mechanism for subcomponents fabricated from stainless steels, because
33 these materials exhibit passive behavior and negligible general corrosion rates. Detailed
34 discussions regarding potential aging mechanisms for each material and the technical bases for
35 those requiring aging management follow.

36 **3.2.1 Steel (carbon, low-alloy, high-strength low-alloy)**

37 In DSSs, steel subcomponents are commonly used and are exposed to sheltered environments,
38 outdoor air, helium, demineralized water, and groundwater or soil, and also may be embedded
39 in concrete or neutron-shielding materials. The exterior surfaces of some steel subcomponents
40 are coated with epoxy or inorganic zinc to mitigate corrosion; however, these coatings can
41 degrade, resulting in exposure of steel to the atmosphere. Steels used to construct transfer
42 casks are predominately exposed to an indoor environment, except for short periods of outdoor
43 exposure during transfer operations. For such air-indoor/outdoor environment exposure, aging

1 effects from aqueous corrosion processes are expected to be bounded by the outdoor
2 environment. As such, the indoor air environment is not discussed separately.

3 3.2.1.1 *General corrosion*

4 General corrosion, also known as uniform corrosion, proceeds at approximately the same rate
5 over a metal surface (Phull, 2003b). Freely exposed steel surfaces in contact with moist air or
6 water are subject to general corrosion. The corrosion rate depends on solution composition,
7 pH, and temperature. The iron Pourbaix diagram shows that iron undergoes active corrosion
8 forming Fe^{2+} or Fe^{3+} ions at pH values lower than 8.5 to 9 (Kodama, 2005). At higher values of
9 pH, iron can be passive, leading to a very low corrosion rate.

10 Steel subcomponents exposed to outdoor and sheltered environments

11 If steel is placed in a completely dry atmosphere, oxide film growth is so small that the corrosion
12 rate is virtually negligible. However, in outdoor conditions, rain, fog, snow, and dew
13 condensation can generate moisture layers on the steel surface that cause general corrosion.
14 Atmospheric corrosion rates can vary from 0 to 0.2 millimeters/year (mm/yr) [0 to 7.9 mils/yr]
15 depending on relative humidity, temperature, and levels of chloride and pollutants in the
16 atmosphere (NACE, 2002). Rates can be more significant in industrial and marine
17 environments (McCuen and Albrecht, 1994).

18 In a sheltered environment, deliquescence of airborne salts below the dew point also could
19 generate an aqueous electrolyte initiating general corrosion. These salts may be chloride rich
20 and originate from marine environments, deicing salts, and condensed water from cooling
21 towers, as well as a range of other nonchloride-rich species originating from industrial,
22 agricultural, and commercial activities. Studies have shown that $MgCl_2$, a component of sea salt
23 with a low deliquescence relative humidity, would deliquesce below 52 degrees C
24 [126 degrees F] under realistic absolute humidities in nature (He et al., 2014). The heat
25 generated by the radioactive decay of spent fuel decreases over time. Time-temperature
26 profiles calculated for the stainless steel canister shell suggest that, while initial temperatures
27 are high, the threshold temperature for deliquescence of some salts on the external surface of
28 the shell could be reached during the 60-year timeframe (EPRI, 2006; Meyer et al., 2013).

29 Because steel subcomponents exposed to sheltered environments are usually located farther
30 away from the fuel compared to the stainless steel canister shell, they are expected to reach
31 these threshold temperatures for deliquescence at an earlier time. As such, the potential for
32 general corrosion of steel subcomponents exposed to a sheltered environment is present.

33 Because aqueous electrolytes initiating general corrosion of steels exposed to outdoor and
34 sheltered environments are potentially present, and corrosion rates may be sufficient to affect
35 component intended functions, general corrosion is considered to be credible, and therefore,
36 aging management is required during the 60-year timeframe.

37 Steel subcomponents exposed to demineralized water

38 Demineralized water is used in the steel water jacket of some transfer casks for radiation
39 shielding. In some cases, 25-percent ethylene glycol is added to the water to decrease the
40 freezing point, and this is expected to decrease the corrosivity of water (van Bodegom et al.,
41 1987). The iron Pourbaix diagram shows that iron undergoes active corrosion at neutral pH, as
42 long as water is present (Kodama, 2005). The corrosion rate for iron is approximately

1 0.1 mm/yr [3.9 mils/yr] in stagnant fresh water at atmospheric temperatures (Kodama, 2005). In
2 60 years of continuous exposure in such water, the material thinning is expected to be
3 approximately 6 mm [0.2 in]. This is a conservative estimate of the corrosion of steel water
4 jackets, as the jackets are not necessarily filled when the transfer cask is not in use. However,
5 general corrosion of steels exposed to demineralized water is nonetheless considered to be
6 credible, and therefore, aging management is required during the 60-year timeframe.

7 Steel subcomponents exposed to groundwater or soil

8 The corrosion rate of steel in groundwater or soil depends on many factors, such as the oxygen
9 level; resistivity; pH, buffer capacity; redox potential; and the presence of chlorides, sulfides,
10 neutral salts, and sulfates. Soils may be acidic, neutral, or alkaline, with pH values typically
11 ranging from 4.5–8.5 (Kodama, 2005), which is in the range of active corrosion discussed
12 previously. Corrosion rate data for iron artifacts buried in soil show that most corrosion rates
13 are 0.1 to 10 micrometers (μm)/yr [0.004 to 0.4 mils/yr], despite the variety of artifacts in terms
14 of origin and environmental conditions (David et al, 2002). In 60 years of continuous soil
15 exposure, the material thinning is expected to be approximately 0.006 to 3.6 mm [0.2 to
16 142 mils]. As such, general corrosion of steels exposed to groundwater or soil is considered to
17 be credible, and therefore, aging management is required during the 60-year timeframe.

18 Steel subcomponents exposed to an embedded (concrete) environment

19 In overpacks, some steel subcomponents are embedded in concrete. The concrete is in contact
20 with air or soil. When the concrete is intact, the alkaline concrete solution passivates the steel.
21 As concrete degrades with time, embedded steel can be exposed to water containing dissolved
22 carbonates and chlorides, and general corrosion can be significant, as discussed previously. As
23 such, general corrosion of steels exposed to an embedded (concrete) environment is
24 considered to be credible, and therefore, aging management is required during the 60-year
25 timeframe.

26 Steel subcomponents exposed to an embedded (neutron-shielding) environment

27 In DSSs, some polymer-based, neutron-shielding materials are poured into a steel structure,
28 leaving one side of the steel embedded. The neutron-shielding materials include Holtite™ and
29 BISCO NS-3. Because the embedded side of the steel has limited exposure to water and
30 oxygen, general corrosion is not considered to be credible, and therefore, aging management is
31 not required during the 60-year timeframe.

32 Steel subcomponents exposed to helium

33 As mentioned previously, the iron Pourbaix diagram shows that iron undergoes active corrosion
34 at neutral pH as long as water is present (Kodama, 2005). However, there is very little residual
35 water in internal environments following drying and refilling with inert gas, and thus the corrosion
36 reaction with steel will be limited. Jung et al. (2013) show that the relative humidity inside the
37 system after drying is no more than 5 percent at the beginning of storage and is less than
38 0.5 percent in 60 years. Furthermore, some steel subcomponents are coated by aluminum or
39 electroless nickel, which are more corrosion resistant than steel. As such, general corrosion of
40 steel exposed to helium is not considered to be credible, and therefore, aging management is
41 not required during the 60-year timeframe.

1 3.2.1.2 *Pitting and crevice corrosion*

2 Pitting corrosion is a localized form of corrosion that is confined to a point or small area of a
3 metal surface (Frankel, 2003). It takes the form of cavities called pits. Crevice corrosion is
4 another localized form of corrosion that occurs in a wetted environment when a crevice exists
5 (Kelly, 2003). It occurs more frequently in connections, lap joints, splice plates, bolt threads,
6 under bolt heads, or at points of contact between metals and nonmetals. Crevice corrosion is
7 associated with stagnant or low-flow solutions. As discussed previously, the common form of
8 corrosion for steel is general corrosion. However, steel is also known to be susceptible to pitting
9 and crevice corrosion in an oxidizing and alkaline environment, especially in the presence of
10 chlorides. The exterior surfaces of some subcomponents are coated with epoxy or inorganic
11 zinc to mitigate corrosion (e.g., the outer shell of the bolted cask system). Depending on the
12 quality and chemical composition of the coating, water and corrosive agents can permeate
13 coating defects, initiating pitting. After initiation of a coating defect, the coating could function as
14 a crevice former and initiate crevice corrosion.

15 *Steel subcomponents exposed to outdoor and sheltered environments, demineralized water,*
16 *groundwater or soil, and embedded (concrete) environments*

17 As discussed in Section 3.2.1.1, the potential to form aqueous electrolytes on surfaces exposed
18 to outdoor and sheltered environments is present, either via direct exposure to precipitation or
19 through deliquescence of deposited salts. These electrolytes, demineralized water, and
20 groundwater or soil could be conducive to pitting and crevice corrosion of steel. For steel
21 embedded in concrete, as concrete degrades with time, steel can be exposed to water
22 containing dissolved carbonates and chlorides, which could be conducive to pitting and crevice
23 corrosion as well.

24 Localized corrosion of steels is attributed to the presence of macro-galvanic cells, where local
25 differences in electrochemical potential are created by conditions such as chemical composition
26 differences within the steel matrix, discontinuous surface films (e.g., mill scale), and differences
27 in oxygen supply (Revie, 2000).

28 Because steel subcomponents exposed to outdoor and sheltered environments are likely to
29 come into contact with aqueous electrolytes, and the localized corrosion in these environments
30 is possible, loss of material due to pitting and crevice corrosion is considered to be credible.
31 Therefore, aging management is required during the 60-year timeframe.

32 *Steel subcomponents exposed to embedded (neutron-shielding materials) environments*

33 Because of the limited water and oxygen in embedded environments, pitting and crevice
34 corrosion are not considered to be credible, and therefore, aging management is not required
35 during the 60-year timeframe.

36 *Steel subcomponents exposed to helium*

37 Inside DSSs, there is very little residual water following drying, and thus the corrosion reaction
38 with steel will be limited. Jung et al. (2013) show that the relative humidity inside the system is
39 no more than 5 percent at the beginning of storage and is less than 0.5 percent in 60 years.
40 Furthermore, some steel subcomponents are coated by aluminum or electroless nickel, which
41 are more corrosion resistant than steel. As such, localized corrosion of steel exposed to helium

1 is considered to be insignificant, and therefore, aging management is not required during the
2 60-year timeframe, regardless of the coating.

3 3.2.1.3 *Galvanic corrosion*

4 Galvanic corrosion occurs when two dissimilar metals or conductive materials are in physical
5 contact in the presence of a conducting solution (Baboian, 2003; Hack, 1993). Under these
6 conditions, an electrolytic cell is formed, transmitting an electrical current between an anode
7 (i.e., less noble material) and a cathode (i.e., more noble material). Oxidation occurs at the
8 anode, and reduction occurs at the cathode. The relative nobility of different materials has been
9 most commonly constructed from measurements in seawater (Baboian, 2003). With certain
10 exceptions, it is broadly applicable to other natural waters and in uncontaminated atmospheres.
11 It is used here to infer the relative nobility of the canister materials during extended storage
12 (e.g., steel is less noble than stainless steel, graphite, nickel, and brass). The extent of galvanic
13 corrosion depends on potential differences between the two metals, surface area ratio of the
14 anode and cathode, environment, reaction kinetics, corrosion products, and other factors
15 (Baboian, 2003). In DSSs, galvanic coupling exists between steel and other more noble
16 materials such as stainless steel, graphite, nickel, and brass. These galvanic couples can be
17 exposed to sheltered and outdoor air environments.

18 Steel subcomponents exposed to outdoor and sheltered environments

19 Aqueous electrolytes for subcomponents exposed to outdoor and sheltered environments are
20 present during the 60-year timeframe. Because these electrolytes could initiate steel corrosion,
21 and corrosion of steel is expected to be enhanced under galvanic coupling, loss of material due
22 to galvanic corrosion of steel is considered to be credible in dissimilar metal couples, and
23 therefore, aging management is required during the 60-year timeframe.

24 3.2.1.4 *Microbiologically influenced corrosion*

25 MIC is corrosion caused or promoted by the metabolic activity of microorganisms
26 (Dexter, 2003). Active microbial metabolism requires water in the form of water vapor,
27 condensation, or deliquescence, and available nutrients to support microbial activity (Horn and
28 Meike, 1995). Biofilms can form even under radiation environments (Bruhn et al., 2009).
29 Bacteria resistant to radiation include *Micrococcus radiodurans*, which can tolerate 10 kilograys
30 (kGy) [10^6 rads] of irradiation. MIC is limited where relative humidity is below 90 percent and
31 negligible for relative humidity below 60 percent (King, 2009). MIC has been found to be
32 operable within a temperature range of -5 degrees C to 110 degrees C [23 to 230 degrees F].

33 Several types of microbes can exist within a biofilm. For instance, sulfate-reducing bacteria are
34 of primary concern in wet, cool, and anoxic environments (Little and Wagner, 1996). Another
35 type of microbe is the acid-producing bacteria, which can promote depassivation of oxide films
36 on metals. Other types of bacteria are created by ammonia production, metal deposition, and
37 hydrogen production (Walch and Mitchell, 1983; Little and Wagner, 1996). Although most of the
38 evidence of MIC for metallic components is from conditions under which the metal surface is
39 kept continuously wet, microorganisms can live in many environments, such as water, soil, and
40 air, where aerobic bacteria (e.g., iron-manganese oxidizing bacteria, sulfur/sulfide oxidizing
41 bacteria, methane producers, organic acid-producing bacteria), fungi, and algae can develop.
42 This is borne out by research studies on MIC in soils (Jack et al., 1996) and in tropical
43 environments (Caprio et al., 1995).

1 Steel subcomponents exposed to groundwater/soil and embedded (concrete) environments

2 For soils, MIC rates for steel and iron have been correlated with the pH, oxidation reduction
3 potential, resistivity, and water content of the soil, as well as with the type of soil. Moist, aerobic
4 soils, where oxygen can readily reach exposed steel, show MIC rates typically in the range of
5 0.04 to 0.2 mm/yr [2 to 8 mils/yr] (Jack et al., 1996). Anaerobic soil environments show
6 intermediate MIC rates of steel on the order of 0.002 to 0.01 mm/yr [0.08 to 0.3 mils/yr]. Typical
7 MIC rates of metal loss for unprotected line pipe steel in a sulfate-reducing bacteria/FeS
8 environment are 0.2 mm/yr [8 mils/yr] for general corrosion and 0.7 mm/yr [28 mils/yr] for pitting
9 corrosion. When steel is embedded in concrete, it can be exposed to groundwater or soil, as
10 concrete degrades with time, which could be conducive to MIC as well. As such, MIC of steel in
11 soil and concrete environments is considered to be credible, and therefore, aging management
12 is required during the 60-year timeframe.

13 Steel subcomponents exposed to sheltered and outdoor environments

14 As discussed in Section 3.2.1.1, the potential to form aqueous electrolytes for subcomponents
15 exposed to outdoor and sheltered environments is present, either from direct exposure to
16 precipitation or by deliquescence of deposited salts. These electrolytes have the potential to
17 support microbial activity.

18 A limited number of research studies have shown that MIC may occur on steel surfaces
19 exposed to tropical and polluted atmospheric conditions (Caprio et al., 1995; Parra et al., 1996;
20 Maruthamuthu et al., 2008). However, there is no operating experience of MIC degradation of
21 steel engineering components that are exposed to environments similar to those of dry cask
22 storage systems, where continuous exposure to a relative humidity above 90 percent is not
23 expected. The operating experience of MIC for metallic components is largely from instances in
24 which the metal surface was kept continuously wet. Because there is no applicable operating
25 experience of MIC damage of steel under relevant atmospheric conditions, MIC is not
26 considered to be credible, and therefore, aging management is not required during the 60-year
27 timeframe.

28 Steel subcomponents exposed to demineralized water

29 The transfer cask water jackets are filled with demineralized water and drained during each
30 loading campaign. If any bacteria are introduced during these operations, the concentration is
31 expected to be insignificant. Microbial metabolism and growth depends upon adequate supplies
32 of essential macro and micro nutrients. Critical nutrients, such as carbon, nitrogen, and
33 phosphorous, must be present in appropriate concentrations (Dragun, 1988). It is expected that
34 the concentrations of these species in demineralized water are well below the critical values. As
35 such, MIC of steel in this environment is considered to be insignificant, and therefore, aging
36 management is not required during the 60-year timeframe.

37 Steel subcomponents exposed to helium and embedded (neutron shielding) environments

38 Because of the limited amount of water and nutrients in the helium environments within casks
39 and canisters, and the limited water in embedded environments, MIC of steel is not credible for
40 the 60-year timeframe, and therefore, aging management is not required.

1 3.2.1.5 *Stress corrosion cracking*

2 SCC is the cracking of a metal produced by the combined action of corrosion and tensile stress
3 (applied or residual) (Jones, 1992). SCC is highly chemical specific in that certain alloys are
4 likely to undergo SCC only when exposed to a small number of chemical environments. SCC is
5 the result of a combination of three factors: (1) a susceptible material, (2) exposure to a
6 corrosive environment, and (3) tensile stresses. High-strength steels with yield strengths
7 greater than or equal to 150,000 pounds per square inch [150 ksi] have been found to be
8 susceptible to SCC under exposure to aqueous electrolytes, particularly when containing H₂S)
9 (Jones, 2003; McMahon, 2001; EPRI, 2007).

10 *Steel subcomponents exposed to sheltered and outdoor environments*

11 In DSSs, some steels with moderately high strength are used as bolting material, such as the lid
12 bolts for the direct-load bolted cask systems. These steel subcomponents are exposed to
13 sheltered and outdoor environments, and thus an aqueous electrolyte necessary to support
14 SCC could be present.

15 SCC also requires the presence of a sufficient tensile stress. Calculations using the approach
16 proposed by Baggerly (1999) show that the stress threshold to initiate SCC of steel bolts is
17 usually larger than 70 percent of the bolting material's minimum yield strength, while the Electric
18 Power Research Institute (EPRI, 2007) states that stresses near the yield strength are required
19 to initiate SCC. The high-strength steel bolting in DSSs is expected to be loaded to stresses
20 much lower than these SCC thresholds. For example, under normal conditions, the stress
21 experienced by the lid bolts of bolted cask systems is primarily from the bolt preload applied to
22 seat, or engage, the lid gaskets, and these preloads are well below the bolting material's yield
23 strength. Also, in the Standardized NUHOMS system, the high-strength structural bolts in the
24 horizontal storage module (HSM) are installed "snug tight" and are not loaded close to critical
25 stresses.

26 Because of the low applied stresses, SCC of steel bolts exposed to sheltered and outdoor
27 environments is not considered to be credible, and therefore, aging management is not required
28 during the 60-year timeframe.

29 3.2.1.6 *Creep*

30 Creep is the time-dependent inelastic deformation that takes place at an elevated temperature
31 and a constant stress (Gibeling, 2000). Because the deformation processes that produce creep
32 are thermally activated, the rate of this time-dependent deformation is a strong function of the
33 temperature. The creep rate also depends on the applied stress but does not generally vary
34 with the environment. As a general rule of thumb, at temperatures below 0.4T_m, where T_m is the
35 melting point of the metal in Kelvin (K), thermal activation is insufficient to produce significant
36 creep (Cadek, 1988). With a melting point of 1,789 K (1,516 degrees C [2,760 degrees F]),
37 temperatures of at least 716 K (443 degrees C [829 degrees F]) are required to initiate creep in
38 steels. However, the 0.4T_m rule of thumb underestimates the minimum creep temperature for
39 steels, as temperatures above 500 degrees C [932 degrees F] have been found to be required
40 for creep in steels (Samuels, 1988).

1 Steel subcomponents exposed to helium

2 The highest temperatures within the DSSs are at locations close to the fuel rods. The maximum
3 expected temperature of fuel cladding has been estimated to be 400 degrees C [752 degrees F]
4 at the beginning of storage (Jung et. al., 2013). This cladding temperature is expected to
5 decrease to around 266 degrees C [510 degrees F] after 20 years and to approximately
6 127 degrees C [261 degrees F] after 60 years. These estimates depend on many factors, such
7 as the initial heat load of the SNF. Because the fuel rods are the only heat source within the
8 system, these temperatures provide upper temperature limits for all subcomponents. It is
9 apparent from these temperatures that internal subcomponents will not approach the minimum
10 500 degrees C [932 degrees F] temperature that has been found to be required for significant
11 creep to occur in steels. Hence, creep of steel internals exposed to helium is not expected to be
12 credible, and therefore, aging management is not required during the 60-year timeframe.

13 Steel subcomponents exposed to sheltered, outdoor air, demineralized water, groundwater or
14 soil, and embedded (all) environments

15 Because steel subcomponents exposed to sheltered, outdoor air, demineralized water,
16 groundwater or soil, and embedded environments experience significantly lower temperatures
17 than those experienced by the internal subcomponents, creep of these steel subcomponents is
18 not considered to be credible, and therefore, aging management is not required during the
19 60-year timeframe.

20 3.2.1.7 *Fatigue*

21 Fatigue is the progressive structural damage that occurs when a metal is subjected to cyclic
22 loading (Hoepfner, 1996). Because spent fuel storage is a static application, cyclic loading by a
23 purely mechanical means is largely limited to transfer cask lifting trunnions, which are loaded
24 each time a canister is moved from the spent fuel pool to the dry storage pad. Other
25 subcomponents, however, could experience cyclic loads due to thermal effects.

26 The reviewer should review all fatigue analyses contained in the applicant's design basis
27 documents to determine whether the renewal application adequately addresses the implications
28 of extending the operating period to 60 years. This re-examination of the original fatigue
29 analyses are defined as TLAAs.

30 As described in greater detail in Chapter 5 of this report, the reviewer should review the design
31 codes and standards to identify any required fatigue analyses and ensure that the applicant
32 addresses those analyses with a TLAA. For example, components that were designed in
33 accordance with the American Society of Mechanical Engineers Boiler and Pressure Vessel
34 Code (ASME Code) Section III, Division 1, Subsections NB or NC (ASME, 2007a) were
35 evaluated for the effects of cyclic loading per subparagraphs NB-3222.4 and NC-3219.2,
36 respectively. Also, the designs of some steel support structures may be performed in
37 accordance with the American Institute of Steel Construction (AISC) Standard 360,
38 "Specifications for Structural Steel Buildings" (AISC, 2010). Appendix 3 of AISC 360, "Design
39 for Fatigue," provides criteria for the evaluation of cyclic loading.

40 The staff's guidance for the review of TLAAs is provided in NUREG-1927, Revision 1, and
41 Chapter 5 of this report. In its evaluation of a TLAA, an applicant may conclude that an analysis
42 can no longer support a determination that aging will not adversely affect an important-to-safety

1 function in the 60-year timeframe of the period of extended operation. In that case, the
2 applicant may manage the aging of the associated SCC with an AMP.

3 The AMR tables in Chapter 4 recommend that applicants address any applicable TLAAAs
4 associated with components with a structural function. If no fatigue analysis was performed in
5 support of the component design, no action is required of the applicant.

6 3.2.1.8 Thermal aging

7 The microstructures of most steels will change, given sufficient time at temperature, and this
8 can affect mechanical properties. This process is commonly called thermal aging. The effect of
9 thermal aging will depend on the time at temperature and the microstructure and carbon content
10 of the steel subcomponents.

11 Steel subcomponents exposed to helium

12 The maximum expected temperature of fuel cladding has been estimated to be 400 degrees C
13 [752 degrees F] at the beginning of storage (Jung et. al., 2013). This upper-bound cladding
14 temperature is expected to decrease to around 266 degrees C [510 degrees F] after 20 years
15 and to approximately 127 degrees C [261 degrees F] after 60 years. Although the temperature
16 of steel components within the cask internal environment will be lower than that of the fuel
17 cladding, consideration of the cladding temperatures provides a conservative estimate of the
18 effects of thermal aging.

19 Carbon steels in the normalized condition (ferrite/pearlite microstructures) are commonly used
20 in the petroleum and chemical industry with exposure temperatures similar to those in DSS
21 internal environments, approximately 400 degrees C [752 degrees F] and lower
22 (ASM International, 1998). ASME Code Section II, Part D, provides allowable operating
23 stresses for carbon steels at these temperatures (ASME, 2007b).

24 The ASME Code also provides for the use of hardened (quenched and tempered) alloy steels at
25 temperatures typically expected within storage systems during the 20- to 60-year period of
26 extended operation. For example, ASME type SA-537 Grade 2 alloy steel receives a tempering
27 heat treatment of at least 595 degrees C [1,100 degrees F] following quenching, and the
28 ASME Code provides allowable operating stresses up to 371 degrees C [700 degrees F]. This
29 compares to the estimated upper-bound 266 degrees C [510 degrees F] temperature during the
30 period of extended operation. Some hardened alloy steels can experience reductions in
31 fracture toughness when tempered at temperatures greater than 200 degrees C
32 [392 degrees F]. The degree of the reduction in toughness depends on the carbon content and
33 the tempering conditions that were employed during processing (Krauss, 2005).

34 The effects of elevated storage temperatures on material properties are evaluated during the
35 initial license application (typically first 20 years of storage). Although the temperatures inside
36 the canister after 20 years may still have the capacity to alter mechanical properties, it is likely
37 that the steel tempering that occurs during manufacture and the higher temperatures present
38 during the initial storage period would dominate any effects of tempering at the lower
39 temperatures during the period of extended operation.

40 It can thus be concluded that thermal aging generally is not expected to produce degradation of
41 the mechanical properties of steels in the period of extended operation, and therefore, aging

1 management is not required during the 60-year timeframe. Nevertheless, the reviewer should
2 verify this conclusion on a case-by-case basis.

3 Steel subcomponents exposed to sheltered, outdoor air, demineralized water, groundwater or
4 soil, and embedded (all) environments

5 As stated above, undesired material property changes due to tempering of hardened steels
6 could occur at temperatures greater than 200 degrees C [392 degrees F]. The temperatures of
7 steel subcomponents exposed to sheltered, outdoor air, demineralized water, groundwater or
8 soil, and embedded environments are bounded by the stainless steel canister shell temperature,
9 because these subcomponents are located farther away from the fuel. Time-temperature
10 profiles calculated for the stainless steel canister shell estimate that the peak temperature is
11 below 200 degrees C [392 degrees F] (EPRI, 2006; Meyer et al., 2013). Because the peak
12 temperatures for steel subcomponents exposed to sheltered, outdoor air, demineralized water,
13 and embedded environments are below the temperature required to cause reductions in
14 toughness, thermal aging is not considered to be credible for these subcomponents, and
15 therefore, aging management is not required during the 60-year timeframe.

16 3.2.1.9 *Radiation embrittlement*

17 Embrittlement of metals may occur under exposure to neutron radiation. Depending on the
18 neutron fluence, radiation can cause changes in mechanical properties, such as loss of ductility,
19 reduced fracture toughness, and decreased resistance to cracking.

20 Neutron irradiation has the potential to increase the tensile and yield strength and decrease the
21 toughness of carbon and alloy steels (Nikolaev et al., 2002). Neutron fluence levels greater
22 than 10^{19} neutrons/square centimeter (n/cm^2) [6.5×10^{19} n/in^2] are required to produce a
23 measureable degradation of the mechanical properties (Nikolaev et al., 2002; Odette and
24 Lucas, 2001).

25 For dry cask storage, a neutron flux of 10^4 – 10^6 n/cm^2 -s [6.5×10^4 – 6.5×10^6 n/in^2 -s] is typical
26 (Sindelar et al., 2011). At these flux levels, the accumulated neutron fluence after 60 years is
27 about 10^{13} – 10^{15} n/cm^2 [6.5×10^{13} – 6.5×10^{15} n/in^2]. To verify the conservatism of this estimate,
28 the NRC staff performed an independent calculation of the maximum potential accumulated
29 neutron fluence on DSS components. The staff considered components most directly exposed
30 to the radiation source (middle of the fuel basket) and assumed fuel is loaded immediately after
31 it is removed from the reactor vessel and stored for 100 years. To further provide a bounding
32 estimate, the staff assumed a cask design that uses 40 Westinghouse 17×17 PWR fuel
33 assemblies with an average burnup of 70 GWd/MTU and 4.0 fuel enrichment. The staff
34 calculated the neutron source term for neutrons with energy at or greater than 1 MeV using the
35 Origen/Arp computer code of the SCALE 6.1 computer code system. At this location, the total
36 accumulated neutron fluence after 100 years of storage was calculated to be 2.63×10^{16} n/cm^2
37 [1.70×10^{17} n/in^2]. This worst-case estimate is greater than that calculated using the flux levels
38 reported in Sindelar, however, the NRC estimated fluence level is still three orders of magnitude
39 below the levels reported to degrade the fracture resistance of carbon and alloy steels.

40 Thus, radiation embrittlement of steel exposed to any environment is not a credible aging
41 mechanism, and therefore, aging management is not required during the 60 year timeframe.

1 3.2.1.10 *Stress relaxation*

2 Stress relaxation of bolting or other tightening subcomponents is the steady loss of elastic
3 stress in a loaded part due to atomic movement at elevated temperature (Earthman, 2000). It
4 results in a loss of clamping forces or preload in a heavily loaded joint. In the stress relaxation
5 process, the total strain is constant and the stress reduction at constant temperature occurs as
6 an elastic strain is converted to an inelastic strain. Stress relaxation is a strong function of
7 temperature and bolt material. It also depends on geometry of the bolt and thread quality
8 (Sachs and Evans, 1973). It decreases with time, as the tensile stress in the bolt decreases
9 (Kulak et al., 2001). Steel bolting is used in several DSS applications in sheltered and outdoor
10 environments, such as in the NUHOMS canister support structure and the HI-STORM overpack
11 lid.

12 *Steel subcomponents exposed to sheltered environments*

13 Bickford (2008) demonstrated that the residual stress of carbon steel bolts due to relaxation is
14 about 85 percent of the initial applied stress at temperatures greater than about 100 degrees C
15 [212 degrees F]. Meyer et al. (2013) show that the external surface temperature of storage
16 canisters can be greater than 200 degrees C [392 degrees F] at the beginning of the storage
17 period. Thus, stress relaxation of steel bolting exposed to sheltered environments adjacent to
18 the canister is considered to be credible, and therefore, aging management is required during
19 the 60-year timeframe.

20 *Steel subcomponents exposed to outdoor environments*

21 Bolting in outdoor environments is not considered to be exposed to sufficiently high
22 temperatures to cause stress relaxation. Similarly, transfer cask bolting in indoor/outdoor
23 environments is not considered to be exposed to high temperatures for a sufficient amount of
24 time to cause stress relaxation. Thus, for steel bolting exposed to outdoor environments, aging
25 management is not required during the 60-year timeframe.

26 3.2.1.11 *Wear*

27 Rolling contact wear results from the repeated mechanical stressing of the surface of a body
28 rolling on another body (Blau, 1992). For the HI-TRAC transfer cask exposed to indoor and
29 outdoor air, ASME SA36 steel is used to construct the transfer lid wheel track, which could
30 experience rolling contact during SNF loading and unloading operations. Thus, wear of these
31 steel subcomponents is considered to be credible, and therefore, aging management is required
32 during the 60-year timeframe.

33 **3.2.2 Stainless steel**

34 Austenitic, ferritic, martensitic, duplex, and precipitation-hardened stainless steels are used in
35 constructing DSS subcomponents. They are exposed to outdoor, sheltered, embedded, helium,
36 and demineralized water environments. Some stainless steels are used to construct the
37 transfer cask, which is predominately exposed to an indoor environment or otherwise encased
38 without direct air ingress, except for short periods of air exposure during transfer operations.
39 For such air-indoor/outdoor environments, the aging mechanisms from aqueous corrosion
40 processes are expected to be bound by the outdoor environment, because it is more corrosive.
41 As such, the indoor air environment is only discussed separately for the evaluation of SCC,

1 where periodic rinsing of the transfer cask external surfaces is expected to minimize halide
2 deposition.

3 3.2.2.1 *General corrosion*

4 Stainless steels exhibit passive behavior in all DSS environments, resulting in negligible general
5 corrosion rates (Grubb, 2005). As such, general corrosion of stainless steel exposed to all
6 environments is not considered to be credible, and therefore, aging management is not required
7 during the 60-year timeframe.

8 3.2.2.2 *Pitting and crevice corrosion*

9 As discussed in Section 3.2.1.2, pitting corrosion is a localized form of corrosion that is confined
10 to a point or small area of a metal surface (Frankel, 2003), and crevice corrosion occurs in a
11 wetted environment when a crevice exists that allows a corrosive environment to develop in a
12 component (Kelly, 2003). In DSSs, crevice corrosion may occur (i) where the canister contacts
13 the support rails for horizontal canister designs and (ii) between canister and guide rails or the
14 support pedestal in some vertical designs. Stainless steels are susceptible to pitting and
15 crevice corrosion, with chloride being the most common agent for initiation (Grubb et al., 2005).
16 Other halides, notably bromides, and hypochlorites are also initiation agents (EPRI, 2007).

17 *Stainless steel subcomponents exposed to outdoor and sheltered environments*

18 As discussed in Section 3.2.1.1, the potential to form aqueous electrolytes for subcomponents
19 exposed to outdoor and sheltered environments is present, either via direct exposure to
20 precipitation or by deliquescence of deposited salts. These electrolytes could be conducive to
21 pitting and crevice corrosion of stainless steel. Atmospheric corrosion of stainless steels
22 typically proceeds in the form of localized corrosion (Cook et al., 2010; Shirai et al., 2011;
23 Tani et al., 2009). However, experimentally measured penetration rates for pitting and crevice
24 corrosion are quite low. Stainless steel exposed to a saturated NaCl steam mist at
25 60 degrees C [140 degrees F] and 95 percent relative humidity (NWTRB, 2010) yielded
26 maximum penetration rates of 0.02 mm/yr [8 mils/yr] for pitting and 0.03 mm/yr [11 mils/yr] for
27 crevice corrosion. These maximum rates suggest that penetration of a 15-mm [0.59-in]-thick
28 canister wall by pitting or crevice corrosion would require 750 years and 495 years, respectively.
29 Davison et al. (1987) reported pitting penetration of 0.028 mm [1.1 mils] after 15 years, which
30 yields a penetration rate of 0.0019 mm/yr [0.075 mils/yr]. Using the penetration depth versus
31 time equation in Eq. (3.2-1) from NRC (2014):

$$d = At^n \text{ and } n = 0.33 \text{ to } 0.5, \tag{3.2-1}$$

32 the penetration rate in Davison et al. (1987), and $n = 0.5$ yields a penetration time for a 15-mm
33 [0.59-in]-thick canister wall of 19,000 years. Based on these penetration rates, the canister wall
34 would not be penetrated in the 60-year timeframe. The rate of pit propagation can be much
35 higher in aggressive environments. Morrison (1972) reported pit penetrations exceeding
36 0.5 mm [20 mils] in 304 and 316 stainless steels after a 28-month exposure at the Kennedy
37 Space Center, Florida. However, the pitting rates measured under aggressive marine
38 environments would require more than 250 years to penetrate 12.7-mm [0.5-in]-thick stainless
39 steel. Hence, neither pitting nor crevice corrosion itself is expected to produce damage to the
40 stainless steel subcomponents in the 60-year timeframe.

1 However, both pitting and crevice corrosion are known to be precursors to SCC. He et al.
2 (2014) observed that all the SCC cracks started at the bottom of the pits. Therefore, pitting and
3 crevice corrosion are also considered to be credible during the 60-year timeframe, due to their
4 role as precursors to atmospheric SCC, and aging management is required accordingly.

5 Stainless steel subcomponents exposed to helium, demineralized water, and embedded (all)
6 environments

7 Stainless steel exposed to helium and demineralized water is not susceptible to pitting and
8 crevice corrosion due to the lack of halides. Because of limited water and oxygen, stainless
9 steel is also not susceptible to pitting and crevice corrosion in embedded environments. As
10 such, pitting and crevice corrosion of stainless steel exposed to helium, demineralized water,
11 and embedded environments are not considered to be credible, and therefore, aging
12 management is not required during the 60-year timeframe.

13 3.2.2.3 *Galvanic corrosion*

14 As discussed in Section 3.2.1.3, galvanic corrosion occurs when two dissimilar metals or
15 conductive materials are in physical contact in the presence of a conducting solution
16 (Baboian, 2003; Hack, 1993). In DSSs, graphite is used to lubricate stainless steel
17 subcomponents such as the stainless steel upper trunnion for the TN-68 bolted cask and the
18 interface between the NUHOMS canister shell and support structure, resulting in galvanic
19 contact between stainless steel and graphite. Because graphite is strongly cathodic and the
20 contact is close, the galvanic coupling effect is expected to be strong. These galvanic couples
21 are exposed to sheltered and outdoor environments.

22 Because these electrolytes conducive to galvanic corrosion exist in both sheltered and outdoor
23 environments, galvanic corrosion of stainless steel in contact with graphite lubricants is
24 considered to be credible, and therefore, aging management is required during the 60-year
25 timeframe.

26 3.2.2.4 *Microbiologically influenced corrosion*

27 As discussed in Section 3.2.1.4, MIC is caused or promoted by the metabolic activity of
28 microorganisms (Dexter, 2003). Microorganisms can live in many environments, such as water,
29 soil, and air, where aerobic bacteria (e.g., iron-manganese oxidizing bacteria, sulfur/sulfide
30 oxidizing bacteria, methane producers, organic acid-producing bacteria), fungi, and algae
31 can develop.

32 Stainless steel subcomponents exposed to sheltered and outdoor environments

33 As discussed in Section 3.2.1.1, the potential to form aqueous electrolytes for subcomponents
34 exposed to outdoor and sheltered environments is present during the 60-year timeframe, either
35 from direct exposure to precipitation or by deliquescence of deposited salts. These electrolytes
36 could support microbial activity; however, there has not yet been any operating experience of
37 MIC in atmospheric environments where stainless steel surfaces are only intermittently wetted.
38 Due to the absence of any operating experience of MIC damage of stainless steel under
39 atmospheric conditions, MIC is not considered to be credible, and therefore, aging management
40 is not required during the 60-year timeframe.

41

1 Stainless steel subcomponents exposed to demineralized water

2 The transfer cask water jackets are filled with demineralized water and drained during each
3 loading campaign. If any bacteria are introduced during these operations, the concentration is
4 expected to be insignificant. Microbial metabolism and growth depends upon adequate supplies
5 of essential macro and micro nutrients. Critical nutrients such as carbon, nitrogen, and
6 phosphorous must be present in appropriate concentrations (Dragun, 1988). It is expected that
7 the concentrations of these species in demineralized water are well below the critical values. As
8 such, MIC of stainless steel exposed to demineralized water is not considered to be credible,
9 and therefore, aging management is not required during the 60-year timeframe.

10 Stainless steel subcomponents exposed to helium and embedded (all) environments

11 Because of the limited amount of water and nutrients in the helium environments within casks
12 and canisters, and the limited water in embedded environments, MIC of stainless steel is not
13 credible for the 60-year timeframe, and therefore, aging management is not required.

14 3.2.2.5 *Stress corrosion cracking*

15 SCC is the cracking of a metal produced by the combined action of corrosion and tensile stress
16 and is highly chemical specific (Jones, 1992, 2003). Most ferritic and duplex stainless steels are
17 either immune or highly resistant to SCC; however, all austenitic grades, especially Types 304,
18 304L, 304LN, 316, 316L, and 316LN, have long been reported in the literature to be susceptible
19 to chloride-induced SCC in the normal wrought condition (Grubb et al., 2005; Morgan, 1980;
20 Kain, 1990). This susceptibility increases when the material is sensitized (He et al., 2014). In
21 the welded condition, the heat-affected zone, which is a thin band located adjacent to the weld,
22 can be sensitized by the precipitation of carbides that extract chromium out of the metal matrix.

23 The Electric Power Research Institute (EPRI, 2005, 2006) and the Nuclear Decommissioning
24 Authority in the United Kingdom (Nuclear Decommissioning Authority, 2007) published review
25 reports on SCC of stainless steel. More recently, the NRC released Information Notice
26 (IN) 2012-20, "Potential for Chloride-Induced Stress Corrosion Cracking of Austenitic Stainless
27 Steel and Maintenance of Dry Cask Storage Systems" (NRC, 2012). The IN describes several
28 incidents in commercial nuclear power plants where SCC of austenitic stainless steel
29 components was attributed to atmospheric chloride exposure (NRC, 1999, 2010c; FPL, 2005;
30 Alexander et al., 2010). These events involved components such as emergency core cooling
31 system piping, SNF pool cooling lines, and outdoor tanks. The IN notes that chlorides may be
32 present in the atmosphere, not only in marine environments but also near cooling towers, salted
33 roads, or other locations. The susceptibility of austenitic stainless steels to SCC tends to
34 increase as the chloride concentration in the solution increases, but the level of chlorides
35 required to produce SCC is very low and is dependent on the type of chloride salts present.
36 The material is more resistant to SCC in NaCl solutions but cracks readily in MgCl₂ solutions
37 (Grubb et al., 2005). Increased temperature and the presence of oxygen tend to aggravate
38 chloride-induced SCC.

39 Stainless steel subcomponents exposed to outdoor and sheltered environments

40 As discussed in Section 3.2.1.1, the potential to form electrolytes for subcomponents exposed
41 to outdoor and sheltered environments is present, either via direct exposure to precipitation or
42 by deliquescence of deposited salts. These electrolytes could be conducive to SCC of stainless
43 steel. SCC also requires the presence of a tensile stress, which commonly exists at welds

1 originating from fabrication processes, contacts between components, and bolted structures.
2 Fuhr et al. (2013) stated that stresses well below yield can cause SCC and the required stress
3 for SCC initiation decreases as chloride concentration and temperature increase. SCC tests
4 were performed with Type 304L C-ring specimens strained to 0.4 or 1.5 percent (He et al.,
5 2014). At the strain of 0.4 percent, the stress on the C-ring specimen was approximately equal
6 to the material yield stress. SCC initiation was observed on specimens deposited with 1 or
7 10 grams/square meter (g/m^2) [0.003 or 0.03 ounces/square foot (oz/ft^2)] of simulated sea salt at
8 both strain levels. Constant load tensile tests were performed on Type 304 between 0.5 and
9 1.75 times the material yield stress (Mayuzumi et al., 2008). Surface chloride concentration was
10 estimated to exceed 10 g/m^2 [0.03 oz/ft^2], while test conditions were 80 degrees C
11 [176 degrees F] at 35 percent relative humidity. Specimens failed at the stress level of
12 0.5 times the yield stress.

13 For DSS subcomponents, the stainless steel canister shell is welded. Welds also exist in other
14 subcomponents, such as the cover plates for the vent and drain ports, grapple ring and grapple
15 support, and the Nitronic 60 support rail plate of the NUHOMS system used to support the
16 canister. Fuhr et al. (2013) concluded that the driving stress for SCC of the welded canister is
17 expected to be weld residual stress, considering that the applied stresses are low and residual
18 compressive stresses are believed to be present on the shell outer diameter due to rolling.
19 Their calculations indicate that residual stresses parallel to the weld are tensile through-wall and
20 significantly above the original yield strength of the base metal, while those transverse to the
21 weld are either compressive along the outer canister surface or slightly tensile on the outer
22 diameter but compressive along the midwall. Based on these calculated residual weld stresses,
23 it was concluded that through-wall SCC is most likely to occur transverse to the weld direction.
24 Weld residual stress modeling conducted by the NRC (2013) also indicates that through-wall
25 tensile stresses of sufficient magnitude to support SCC are likely to exist in the weld
26 heat-affected zone.

27 Because sufficient weld residual stresses and more susceptible material conditions are present
28 near the welds, and aqueous electrolytes conducive to SCC are present in sheltered and
29 outdoor environments, the potential for SCC of the welds in the canister shell and other
30 stainless steel subcomponents is present in the 60-year timeframe. Additionally, the SCC
31 initiation times are relatively short (NWTRB, 2010) with reported crack growth rates of austenitic
32 stainless steels at the weld heat-affected zones ranging from 0.1 mm/yr [3.9 mils/yr]
33 (Hosler, 2010) to 0.67 mm/yr [26.1 mils/yr] (Basson and Wicker, 2002). As a result,
34 through-wall penetration could occur during the 60-year timeframe. This is consistent with the
35 observation of outer-diameter-initiated through-wall SCC in stainless steel piping after 20 to
36 30 years of exposure in marine environments (Fuhr et al., 2013). As such, atmospheric SCC of
37 stainless steel subcomponents with welds exposed to sheltered and outdoor air is considered to
38 be credible, and therefore, aging management is required during the 60-year timeframe.

39 For weld-free austenitic stainless steel subcomponents or regions away from welds, such as the
40 canister body, atmospheric SCC is a likely aging mechanism if sufficient stress exists. Its
41 significance and corresponding aging management requirement will need to be assessed case
42 by case, based on applied and residual stresses, operating temperatures, and the presence of
43 chlorides in the environment.

44

45

1 Stainless steel subcomponents exposed to indoor/outdoor environments and
2 demineralized water

3 Stainless steel transfer casks are exposed to indoor environments during their storage between
4 cask loading campaigns, and thus an aqueous electrolyte is not likely to be present on the
5 transfer cask external surfaces for extended periods. Also, the transfer cask external surfaces
6 are periodically rinsed with demineralized water as they are removed from the spent fuel pool,
7 which would be expected to remove any halides present. As a result, SCC is not considered to
8 be a credible degradation mechanism. In the demineralized water environments of transfer
9 cask neutron shields, SCC is also not considered to be a credible degradation mechanism
10 because of the lack of halides. Therefore, aging management of stainless steel subcomponents
11 exposed to an indoor environment and demineralized water is not required during the 60-year
12 timeframe.

13 Stainless steel subcomponents exposed to helium and embedded (all) environments

14 Because of the lack of halides and the small amount of water in helium and embedded
15 environments, SCC of stainless steel is not considered to be credible. Therefore, aging
16 management of stainless steel subcomponents exposed to helium and embedded environments
17 is not required during the 60-year timeframe.

18 3.2.2.6 Creep

19 As discussed in Section 3.2.1.6, as a general rule of thumb, thermal activation is insufficient to
20 produce significant creep at temperatures below $0.4T_m$, where T_m is the melting point of the
21 metal in Kelvin (Cadek, 1988). The term “stainless steel” covers a wide range of compositions
22 and microstructures, including austenitic, ferritic, martensitic, duplex, and precipitation
23 hardening stainless steels. This discussion will focus on the austenitic or 300 series stainless
24 steels, because they are most commonly used in DSSs and have the lowest melting point and
25 minimum creep temperature. With a melting point of 1,698 K [1,425 degrees C;
26 2,597 degrees F], temperatures of at least 679 K [406 degrees C; 763 degrees F] are required
27 to initiate creep in the austenitic stainless steels.

28 Stainless steel subcomponents Exposed to helium

29 The highest temperatures within the DSSs are at locations close to the fuel rods where the
30 environment is helium. The maximum expected temperature of fuel cladding has been
31 estimated to be 400 degrees C [752 degrees F] at the beginning of storage (Jung et. al., 2013).
32 This cladding temperature is expected to decrease to around 266 degrees C [510 degrees F]
33 after 20 years and to approximately 127 degrees C [261 degrees F] after 60 years. These
34 estimates depend on many factors, such as the initial heat load of the SNF. Because the fuel
35 rods are the only heat source within the canister, these temperatures provide upper temperature
36 limits for all subcomponents within the canister. It is apparent from these temperatures that
37 subcomponents within the canister will not reach the 406 degrees C [763 degrees F] minimum
38 temperature that is required for significant creep to occur in austenitic stainless steels.
39 Similarly, significant creep would also not be expected to occur in the other classes of stainless
40 steel, which all have higher minimum creep temperatures. Hence, creep of stainless steel
41 internals exposed to helium is not credible, and therefore, aging management is not required
42 during the 60-year timeframe.

1 Stainless steel subcomponents exposed to sheltered, outdoor air, demineralized water, and
2 embedded (all) environments

3 Because stainless steel subcomponents exposed to sheltered, outdoor air, demineralized water,
4 and embedded environments experience significantly lower temperatures than those
5 experienced by the internal subcomponents, creep of these stainless steel subcomponents is
6 not considered to be credible, and therefore, aging management is not required during the
7 60-year timeframe.

8 3.2.2.7 *Fatigue*

9 As discussed previously in Section 3.2.1.7, because spent fuel storage is a static application,
10 cyclic loading by a purely mechanical means is largely limited to transfer cask lifting trunnions,
11 which are loaded each time a canister is moved from the spent fuel pool to the dry storage pad.
12 Other subcomponents, however, could experience cyclic loads due to thermal effects, such as
13 those caused by daily and seasonal fluctuations in the temperature of the external environment.

14 The NRC reviewer should review the fatigue analyses contained in the applicant's original
15 design-basis documents to determine whether the renewal application adequately addresses
16 the implications of extending the operating period to 60 years. This reexamination of the
17 original fatigue analyses would typically be defined as TLAA's.

18 As described in greater detail in Chapter 5 of this report, the reviewer should review the design
19 codes and standards to identify any required fatigue analyses and ensure that the applicant
20 addresses those analyses with a TLAA. For example, components that were designed in
21 accordance with the American Society of Mechanical Engineers Boiler and Pressure Vessel
22 Code (ASME Code) Section III, Division 1, Subsections NB or NC (ASME, 2007a) were
23 evaluated for the effects of cyclic loading per subparagraphs NB 3222.4 and NC 3219.2,
24 respectively.

25 The staff's guidance for the review of TLAA's is provided in NUREG-1927, Revision 1, and
26 Chapter 5 of this report. In its evaluation of a TLAA, an applicant may conclude that an analysis
27 can no longer support a determination that aging will not adversely affect an important-to-safety
28 function in the 60-year timeframe of the period of extended operation. In that case, the
29 applicant may manage the aging of the associated SCC with an AMP.

30 The AMR tables in Chapter 4 recommend that applicants address any applicable TLAA's
31 associated with components with a structural function. If no fatigue analysis was performed in
32 support of the component design, no action is required of the applicant.

33 3.2.2.8 *Thermal aging*

34 The microstructures of most stainless steels will change, given sufficient time at temperature,
35 and these changes may alter the material's strength and fracture toughness. This process is
36 commonly called thermal aging. For stainless steel subcomponents, the thermal aging process
37 differs for welded and nonwelded subcomponents.

38 Welded austenitic stainless steel subcomponents exposed to helium

39 The ferrite present in austenitic stainless steel welds can transform by spinodal decomposition
40 to form Fe-rich alpha and Cr-rich alpha prime phases, and further aging can produce an

1 intermetallic G-phase. The spinodal decomposition and the formation of the intermetallic
2 G-phase takes place during extended exposure to temperatures between 300 and
3 400 degrees C [572 and 752 degrees F] (Alexander and Nanstad, 1995; Chandra et al., 2012).
4 The maximum expected temperature of fuel cladding has been estimated to be 400 degrees C
5 [752 degrees F] at the beginning of storage (Jung et. al., 2013). This cladding temperature is
6 expected to decrease to around 266 degrees C [510 degrees F] after 20 years and to
7 approximately 127 degrees C [261 degrees F] after 60 years. Based on these temperature
8 estimates, subcomponents located inside the canister and near the fuel could be above the
9 300 degrees C [572 degrees F] minimum temperature required for these phase changes.
10 Because the phase transformations take place only within the ferrite phase, they increase the
11 hardness and reduce the toughness of the ferrite phase but do not alter the mechanical
12 properties of the austenite phase. Hence, the degree of embrittlement of a weld will depend on
13 a number of factors, including the amount and distribution of ferrite present in the weld and the
14 time spent within the 300 to 400 degrees C [572 and 752 degrees F] temperature range.

15 Based on Charpy impact toughness testing of cast duplex stainless steels, Kim and Kim (1998)
16 concluded that ferrite levels above 15 percent are required for significant embrittlement,
17 because ferrite resides in discrete islands below this level and does not provide a continuous
18 low-toughness fracture path. Because most welds contain around 4 to 15 percent ferrite
19 (Gavendra et al., 1996), substantial embrittlement of austenitic stainless steel welds is not
20 expected. Gavendra et al. (1996) in NUREG/CR-6428, "Effects of Thermal Aging on Fracture
21 Toughness and Charpy-Impact Strength of Stainless Steel Pipe Welds," concluded that thermal
22 aging produced moderate decreases (no more than 25 percent) in the upper shelf Charpy
23 impact energy and relatively small decreases in the fracture toughness of a wide range of
24 austenitic welds. Although the phase changes associated with thermal embrittlement of
25 austenitic stainless steel welds could take place in subcomponents near the fuel within the
26 60-year timeframe, the minor reductions in fracture toughness that would be produced in the
27 weld indicate that this is not a credible aging mechanism for subcomponents in proximity to the
28 fuel rods, and therefore, aging management is not required.

29 Subcomponents near the internal wall of a canister or cask would experience temperatures
30 lower than those close to the fuel rods. Time-temperature profiles calculated for a canister
31 surface (EPRI, 2006; Meyer et al., 2013) suggest that maximum canister temperatures would be
32 well below the 300 degrees C [572 degrees F] minimum temperature required for the embrittling
33 phase changes. Hence, thermal aging would not produce any degradation in these
34 subcomponents constructed from austenitic stainless steel, and therefore, aging management is
35 not required during the 60-year timeframe.

36 Nonwelded austenitic stainless steel subcomponents exposed to helium

37 Because the phase changes described previously occur only within the ferrite-containing,
38 heat-affected zone of a weld, embrittlement will not occur in austenitic stainless steel
39 subcomponents that do not contain a weld. The only significant thermal aging possible in
40 nonwelded austenitic stainless steels would be a decrease in strength due to a decrease in
41 dislocation density, recrystallization, and an increase in grain size. These processes occur
42 during annealing at temperatures above 1,000 degrees C [1,832 degrees F]. The temperatures
43 of less than 400 degrees C [752 degrees F] that will be experienced by cask internal
44 subcomponents will not degrade nonwelded austenitic stainless steels. Thus, thermal aging of
45 nonwelded austenitic stainless steel is not credible, and therefore, aging management is not
46 required during the 60-year timeframe.

1 Precipitation-hardened martensitic stainless steel subcomponents exposed to helium

2 Type 17-4 precipitation-hardened (PH) martensitic stainless steel with Cu and Nb additions is
3 used to construct some fuel basket subcomponents. Operating experience has shown that this
4 material is susceptible to thermal embrittlement, in both welded and nonwelded conditions, at
5 temperatures above 243 degrees C [470 degrees F] (Andresen et al., 2007; Olender et al.,
6 2015; NRC, 2007). The embrittlement mechanism arises from an intra-granular decomposition
7 of the martensitic matrix into two phases, α and α' , which are rich in iron and chromium,
8 respectively, and formation of copper rich ϵ -phase upon further aging. This process leads to an
9 increase in hardness, but decrease in fracture toughness. Olender et al. (2015) reviewed
10 reactor operating experience with 17-4 PH stainless steels. Susceptibility to thermal
11 embrittlement is dependent on several factors including the alloy composition within the
12 allowable specifications, the initial heat treatment and the operating temperature. For operating
13 temperatures between 243 and 316 degrees C [470 to 600 degrees F] Olender et al (2015)
14 recommends an evaluation of conditions on a per-component basis considering operating
15 temperature, exposure time, operating environment, stress levels, and material composition.
16 Above 316 degrees C [600 degrees F] the use of 17-4 PH stainless steel in any condition is not
17 recommended. Subcomponents located inside the canister and near the fuel could be above
18 the temperature threshold for thermal aging. As such, thermal aging of Type 17-4 PH stainless
19 steel is considered to be credible.

20 Although the above generic evaluation identifies thermal aging of Type 17-4 PH stainless steel
21 as a credible aging mechanism, the degree of embrittlement of a specific SSC will depend on
22 the service temperature and time duration, as well as the initial heat treatment condition of the
23 SSC. As such, a review of the thermal aging effects should be performed on a case-by-case
24 basis for all subcomponents constructed from Type 17-4 PH stainless steel. The reviewer
25 should ensure that the application provides a bounding analysis to show that reduction in
26 mechanical properties due to thermal aging is not expected to compromise the SSC's intended
27 function during the period of extended operation.

28 Stainless steel subcomponents exposed to sheltered, outdoor, demineralized water, and
29 embedded (all) environments

30 Because the peak temperatures for stainless steel subcomponents exposed to sheltered,
31 outdoor air, demineralized water, and embedded environments are below the temperature
32 required for the phase changes associated with thermal embrittlement of stainless steels,
33 thermal aging is not considered to be credible for these subcomponents, and therefore, aging
34 management is not required during the 60-year timeframe.

35 3.2.2.9 *Radiation embrittlement*

36 Embrittlement of metals may occur under exposure to neutron radiation. Depending on the
37 neutron fluence, radiation can cause changes in stainless steel mechanical properties, such as
38 loss of ductility, fracture toughness, and resistance to cracking (Was et al., 2006).

39 Cracking has been observed in boiling-water reactor oxygenated water at fluences above
40 2 to 5×10^{20} n/cm² [1.3 to 3.2×10^{21} n/in²] (Was et al., 2006). Gamble (2006) found that neutron
41 fluence levels greater than 1×10^{20} n/cm² [6.5×10^{20} n/in²] are required to produce
42 measureable degradation of the mechanical properties. Caskey et al. (1990) also indicates that
43 neutron fluence levels of up to 2×10^{21} n/cm² [1.3×10^{22} n/in²] were not found to enhance
44 SCC susceptibility.

1 As discussed in Section 3.2.1.9 of this report, the maximum potential accumulated neutron
2 fluence on DSS components after 100 years was calculated to be 2.63×10^{16} n/cm²
3 [1.70×10^{17} n/in²]. This fluence level is four orders of magnitude below the level that would
4 degrade the mechanical properties of stainless steels. As such, radiation embrittlement of
5 stainless steel exposed to any environment is not credible.

6 3.2.2.10 *Stress relaxation*

7 In DSSs, some stainless steel bolts or screws are used in applications exposed to sheltered and
8 outdoor environments. Section 3.2.1.10 explained that stress relaxation of bolting is the steady
9 loss of stress due to atomic movement at elevated temperature in a loaded part with dimensions
10 that are fixed (Earthman, 2000). The loss of initial applied stress in austenitic stainless steel
11 bolting due to stress relaxation is negligible at temperatures below 300 degrees C
12 [572 degrees F] (Bickford, 2008). This temperature is significantly below those expected in
13 sheltered and outdoor environments. Thus, stress relaxation of stainless steel subcomponents
14 exposed to sheltered and outdoor environments is not considered to be credible, and therefore,
15 aging management is not required during the 60-year timeframe.

16 3.2.2.11 *Wear*

17 Adhesive wear occurs when two metallic components slide against each other under an applied
18 load where no abrasives are present (Magee, 1992). For the NUHOMS transfer cask exposed
19 to indoor and outdoor air, Nitronic® 60 stainless steel (UNS S21800) is used to construct the
20 rails in the cask cavity. The additions of silicon and manganese make this alloy best known for
21 its wear and galling resistance, even in the annealed condition (Magee, 1992). The rails could
22 experience repeated sliding contact over multiple canister transfer operations. Thus, wear of
23 these stainless steel rails is considered to be credible, and therefore, aging management is
24 required during the 60-year timeframe.

25 3.2.3 **Aluminum alloys**

26 In DSSs, aluminum and its 6000 series alloys are commonly used in canister internals to
27 transfer heat because of aluminum's good thermal conductivity. For example, in the NUHOMS
28 HSM, anodized Al 1100 is used to construct part of the heat shield assemblies, which are
29 exposed to a sheltered environment. In the TN-32 and 68 systems, the lid seal is a double
30 metallic O-ring exposed to a sheltered environment, where the outer jacket of the O-ring is
31 aluminum. Also, Al 6063-T5 is used in the TN systems to hold the radial neutron shield
32 material, in which one side of the aluminum is embedded in borated polyester resin and the
33 other side is in contact with steel.

34 3.2.3.1 *General corrosion*

35 General corrosion, also known as uniform corrosion, proceeds at approximately the same rate
36 over a metal surface (Phull, 2003b). Freely exposed aluminum surfaces in contact with moist
37 air or water are subject to general corrosion. The corrosion rate depends on solution
38 composition, pH, and temperature. The corrosion rate of aluminum is normally controlled by the
39 formation of a passive film of Al₂O₃ at the metal and water interface. The Pourbaix diagram for
40 aluminum shows that aluminum is passive in the pH range of approximately 4 to 8.5 at
41 25 degrees C [77 degrees F] (Kaufman, 1999). However, the aluminum passive film is reported
42 to be more porous than the chromium oxide film that passivates stainless steel materials
43 (Bass, 1956).

1 Aluminum subcomponents exposed to helium

2 Above a temperature of about 230 degrees C [446 degrees F], an aluminum protective film no
3 longer develops in the presence of water or steam (Ghali 2010; 2011). As such, general
4 corrosion of aluminum is possible if exposed to moisture, because initial temperatures near the
5 spent fuel are above 200 degrees C [392 degrees F]. However, there is very little residual water
6 in the cask internal environment following drying. Assuming a residual water content of 1 liter
7 (L) [0.26 gallon (gal)], Jung et al. (2013) calculated that oxidation of all aluminum in the basket
8 assembly is limited to just 0.54 g [0.019 oz], which is equivalent to a 20- or 2- μm
9 [0.79- or 0.079-mils]-thick layer of aluminum over a surface area of 100 or 1,000 cm^2
10 [15.5 or 155 in^2]. This suggests that material thinning from oxidation is a very small fraction of
11 the millimeter-thick [tens of mils-thick] aluminum materials used inside the system. As a result,
12 sufficient general corrosion to challenge SSC functions is not credible, and therefore, aging
13 management is not required during the 60-year timeframe in helium environments.

14 Aluminum subcomponents exposed to sheltered and embedded (all) environments

15 Section 3.2.1.1 discussed how an aqueous electrolyte can be developed under a sheltered
16 environment through deliquescence of deposited salts. The deliquescent brine can be
17 concentrated and acidic, initiating general corrosion. Therefore, general corrosion of aluminum
18 lid seals exposed to a sheltered environment is considered to be credible, and aging
19 management is required during the 60-year timeframe.

20 Anodized aluminum, in which a surface oxide film is deliberately formed in an electrochemical
21 process, can increase the resistance to corrosion (Vargel, 2004). The successful formation of a
22 protective oxide during manufacture depends on the anodizing solution, applied voltages, and
23 sealing operations. Because of its anodized film and the relatively low temperatures present,
24 general corrosion of the NUHOMS aluminum heat shield is not considered to be credible.
25 However, if defects develop in the anodized film, deep pitting in the underlying metal could
26 occur, and this is discussed below in Section 3.2.3.2. In the embedded environment, because it
27 is moisture free, general corrosion is also not considered to be credible. Therefore, aging
28 management is not required during the 60-year timeframe for anodized aluminum exposed to a
29 sheltered environment and standard aluminum exposed to embedded environments.

30 3.2.3.2 Pitting and crevice corrosion

31 As discussed in Section 3.2.1.2, pitting corrosion is a localized form of corrosion that is confined
32 to a point or small area of a metal surface (Frankel, 2003), and crevice corrosion occurs in a
33 wetted environment when a crevice exists that allows a corrosive environment to develop in a
34 component (Kelly, 2003). Aluminum and its alloys form a passive film on the surface. Localized
35 corrosion in the form of pitting or crevice corrosion could occur for these passive aluminum
36 materials, especially in the presence of halides.

37 Aluminum subcomponents exposed to sheltered environments

38 Section 3.2.1.1 discussed how an aqueous electrolyte can be developed on a stainless steel
39 canister surface in a sheltered environment through deliquescence of deposited salts. The
40 aluminum heat shield would be expected to be cooler than the canister surface, because it is
41 farther away from the fuel, and thus the time to reach the critical temperatures for the
42 development of an aqueous electrolyte in sheltered environments is much lower.

1 The protection of aluminum against corrosion, especially the anodized material, depends on the
2 stability of the passivating oxide films. In chloride-rich environments, the passive layer breaks
3 down and pitting corrosion becomes the predominant corrosion mode (Foley, 1986; Nguyen and
4 Foley, 1979). Analyses of surface deposits demonstrate that aluminum exposed to sheltered
5 environments accumulates adherent particles containing large concentrations of chloride and
6 sulfate ions (Munier, 1982). Pitting corrosion rates on the order of 25 $\mu\text{m}/\text{yr}$ [0.98 mils/yr] have
7 been reported in seawater (Summerson et al., 1957). In 1 molar NaCl solution, crevice
8 corrosion rates of aluminum can be as large as 1.3 mm/yr [51 mils/yr] (Baumgattner and
9 Kaesche, 1988).

10 Because temperatures of aluminum heat-shield surfaces are expected to drop below the
11 deliquescence threshold for airborne salts during the 60-year timeframe, and the corrosion rate
12 is not negligible, pitting and crevice corrosion of aluminum in sheltered environments is
13 considered to be credible, and therefore, aging management is required.

14 Aluminum subcomponents exposed to helium and embedded environments

15 Pitting and crevice corrosion of aluminum is not considered to be credible in helium and
16 embedded environments because of (i) the lack of moisture and halides in helium environments
17 within the cask or canister and (ii) low moisture and oxygen in the embedded environment.
18 Therefore, aging management of pitting and crevice corrosion is not required for aluminum
19 exposed to helium and embedded environments during the 60-year timeframe.

20 3.2.3.3 Galvanic corrosion

21 As discussed in Section 3.2.1.3, galvanic corrosion occurs when two dissimilar metals or
22 conductive materials are in physical contact in the presence of a conducting solution
23 (Baboian, 2003; Hack, 1993). In DSSs, galvanic coupling exists between aluminum and steel,
24 stainless steel, and nickel (where aluminum is less noble in each case). For example, the
25 aluminum lid seal is in contact with stainless steel in the TN-32 and TN-68 systems and an
26 aluminum plate is in contact with the stainless steel fuel compartment within the TN-32
27 bolted cask.

28 Aluminum subcomponents exposed to sheltered environments

29 Section 3.2.1.1 discussed how an aqueous electrolyte conducive to corrosion can be developed
30 in sheltered environments through deliquescence of deposited salts. Caseres (2007) reported
31 corrosion rates of aluminum coupled to carbon steel of about 0.2 mm/yr [8 mils/yr] in solutions
32 containing chloride ions. The galvanic corrosion rate of aluminum coupling to stainless steel is
33 expected to be larger, because the corrosion potential difference between stainless steel and
34 aluminum is larger than carbon steel and aluminum. Because an aqueous electrolyte conducive
35 to corrosion may be present and corrosion of aluminum is expected to be enhanced under
36 galvanic coupling, loss of material due to galvanic corrosion of aluminum is considered to be
37 credible, and therefore, aging management is required during the 60-year timeframe.

38 Aluminum subcomponents exposed to helium

39 There is very little residual water within a cask or canister following drying. Assuming a residual
40 water content of 1 L [0.26 gal], Jung et al. (2013) calculated that oxidation of all aluminum in
41 the basket assembly is limited to 0.54 g [0.019 oz], which is equivalent to a 20 or 2- μm
42 [0.79- or 0.079-mils]-thick layer of aluminum over a surface area of 100 or 1,000 cm^2

1 [15.5 or 155 in²]. This suggests that material thinning from oxidation is a very small fraction of
2 the aluminum materials used inside the system. In conclusion, loss of material due to galvanic
3 corrosion in helium environments is not considered to be credible, and therefore, aging
4 management is not required during the 60-year timeframe.

5 **3.2.3.4 *Microbiologically influenced corrosion***

6 As discussed in Section 3.2.1.4, MIC is corrosion caused or promoted by the metabolic activity
7 of microorganisms (Dexter, 2003). Microorganisms can live in many environments, such as
8 water, soil, and air, where aerobic bacteria (e.g., iron-manganese oxidizing bacteria,
9 sulfur/sulfide oxidizing bacteria, methane producers, organic acid-producing bacteria), fungi,
10 and algae can develop.

11 **Aluminum subcomponents exposed to sheltered environments**

12 Section 3.2.1.1 discussed how an aqueous electrolyte conducive to corrosion can be developed
13 in sheltered environments through deliquescence of deposits. This electrolyte also has the
14 potential to support microbial activity.

15 A single research study found MIC on an aluminum compact disc exposed to tropical
16 atmospheres (Garcia-Guinea et al., 2001). However, there is no operating experience of MIC
17 degradation of aluminum engineering components that operate in environments similar to those
18 of dry cask storage systems. All of the operating experience of MIC for metallic components is
19 from conditions in which the metal surface is kept continuously wet. Due to the absence of any
20 applicable experience of MIC damage of aluminum components under atmospheric conditions,
21 MIC is not considered to be significant in sheltered environments, and therefore, aging
22 management is not required during the 60-year timeframe.

23 **Aluminum subcomponents exposed to helium and embedded (all) environments**

24 Because of the limited amount of water and nutrients in the helium environments within casks
25 and canisters, and because of the limited water in embedded environments, MIC of aluminum is
26 not credible for the 60-year timeframe, and therefore, aging management is not required.

27 **3.2.3.5 *Creep***

28 Section 3.2.1.6 explained that, as a general rule of thumb, thermal activation is insufficient to
29 produce significant creep at temperatures below $0.4T_m$, where T_m is the melting point of the
30 metal in Kelvin (Cadek, 1988). With melting points of 911 to 930 K [638 to 657 degrees C ;
31 1,180 to 1,215 degrees F], temperatures of at least 364 to 372 K [91 to 99 degrees C
32 ; 196 to 210 degrees F] are required to initiate significant creep in aluminum. These
33 temperatures are consistent with Sindelar et al. (2011), which indicates that creep in aluminum
34 is possible at temperatures greater than 100 degrees C [212 degrees F]. Microstructure also
35 plays a significant role in a metal's resistance to creep. Hence, while this 100 degrees C
36 [212 degrees F] minimum temperature for creep is representative for pure aluminum, creep in
37 precipitation hardened aluminum alloys does not become significant until about 200 degrees C
38 [392 degrees F] (Samuels, 1988). Additionally, at temperatures near these threshold values,
39 high stresses are required to produce creep.

1 Aluminum subcomponents exposed to helium

2 The highest temperatures within the DSSs are at locations close to the fuel rods, where the
3 environment is helium. The maximum expected temperature of fuel cladding has been
4 estimated to be 400 degrees C [752 degrees F] at the beginning of storage (Jung et. al., 2013).
5 This cladding temperature is expected to decrease to around 266 degrees C [510 degrees F]
6 after 20 years and to approximately 127 degrees C [261 degrees F] after 60 years. These
7 estimates depend on many factors, such as the initial heat load of the SNF. Because the fuel
8 rods are the only heat source within the cask or canister, these temperatures provide upper
9 temperature limits for all subcomponents. It is apparent from these temperatures that
10 subcomponents within the cask or canister could be exposed to temperatures above the
11 minimum creep temperatures for aluminum during at least the first 40 years.

12 Because the minimum creep temperature will be exceeded during a portion of the 60-year
13 period, it is necessary to consider the load applied to the subcomponent to determine whether
14 creep deformation will occur and whether the creep affects safety. Subcomponents that do not
15 serve a structural function are not expected to be under loads other than their own weight, and
16 in many instances, their weight is also supported by adjacent structures. Due to the minimal
17 applied loads, creep of nonstructural subcomponents will not produce significant damage during
18 the 60-year timeframe. Conversely, aluminum subcomponents that serve a structural function
19 may experience loads that are high enough to produce sufficient creep deformation to affect the
20 subcomponents' safety functions.

21 Aluminum subcomponents exposed to sheltered and embedded (all) environments

22 Aluminum subcomponents exposed to sheltered and embedded environments experience lower
23 temperatures than those experienced by the internal subcomponents. Time-temperature
24 profiles calculated for the canister surface (EPRI, 2006; Meyer et al., 2013) suggest that
25 temperatures in excess of 200 degrees C [392 degrees F] could initially be present on portions
26 of the canister surface and temperatures above 100 degrees C [212 degrees F] could persist for
27 30 years. Based on these temperatures, creep is credible during the 60-year timeframe but only
28 on aluminum subcomponents that are attached directly to the canister shell or cask wall and
29 have a structural function.

30 The NRC reviewer should review the creep analyses for aluminum structural components that
31 are exposed to the elevated temperatures discussed above, as contained in the applicant's
32 original design-bases documents, to determine whether the renewal application adequately
33 addresses the implications of extending the operating period to 60 years. This reexamination of
34 the original analyses would typically be defined as TLAAs in the renewal application. The staff's
35 guidance for the review of TLAAs is provided in NUREG-1927, Revision 1. If the original design
36 basis does not include the pertinent analyses, the reviewer nevertheless should ensure that the
37 application addresses this potential aging mechanism.

38 If the TLAA or other supplemental analyses demonstrate that creep does not have the potential
39 to challenge an important-to-safety function, aging management is not required during the
40 60-year timeframe.

41 Conversely, an applicant may conclude that an analysis cannot support a determination that
42 creep damage will not challenge an important-to-safety function in the 60-year timeframe of the
43 period of extended operation. In that case, the applicant may manage the aging of the
44 associated SSC with an AMP.

1 3.2.3.6 *Fatigue*

2 As discussed previously in Section 3.2.1.7, because spent fuel storage is a static application,
3 cyclic loading by a purely mechanical means is largely limited to transfer cask lifting trunnions.
4 Some aluminum subcomponents, however, could experience cyclic loads due to thermal
5 effects, such as those caused by daily and seasonal fluctuations in the temperature of the
6 external environment.

7 The NRC reviewer should review the fatigue analyses contained in the applicant's original
8 design-basis documents to determine whether the renewal application adequately addresses
9 the implications of extending the operating period to 60 years. This reexamination of the
10 original fatigue analyses would typically be defined as TLAA's.

11 As described in greater detail in Chapter 5 of this report, the reviewer should review the design
12 codes and standards to identify any required fatigue analyses and ensure that the applicant
13 addresses those analyses with a TLAA. For example, components that were designed in
14 accordance with the American Society of Mechanical Engineers Boiler and Pressure
15 Vessel Code (ASME Code) Section III, Division 1, Subsections NB or NC (ASME, 2007a)
16 were evaluated for the effects of cyclic loading per subparagraphs NB 3222.4 and
17 NC 3219.2, respectively.

18 The staff's guidance for the review of TLAA's is provided in NUREG-1927, Revision 1, and
19 Chapter 5 of this report. In its evaluation of a TLAA, an applicant may conclude that an analysis
20 can no longer support a determination that aging will not adversely affect an important-to-safety
21 function in the 60-year timeframe of the period of extended operation. In that case, the
22 applicant may manage the aging of the associated SCC with an AMP.

23 The AMR tables in Chapter 4 recommend that applicants address any applicable TLAA's
24 associated with components with a structural function. If no fatigue analysis was performed in
25 support of the component design, no action is required of the applicant.

26 3.2.3.7 *Thermal aging*

27 The microstructures of many aluminum alloys will change, given sufficient time at temperature.
28 This process is commonly called thermal aging. The effect of the thermal aging on mechanical
29 properties will depend on the time at temperature and the microstructure and chemical
30 composition of the aluminum components. In some DSSs, Al 1100 and its 6000 series alloys
31 are used inside and outside the system to transfer heat because of their good thermal
32 conductivity.

33 *Aluminum subcomponents exposed to helium, sheltered, and embedded (all) environments*

34 The 6000 series aluminum alloys, such as 6061 and 6063 used in the system internals, are
35 precipitation-hardened alloys. The precipitation treatment is performed between 163 and
36 204 degrees C [325 and 399 degrees F] (ASM International, 1991). Prolonged elevated
37 temperature exposure is known to significantly reduce the strength of these alloys due to
38 microstructural changes. For example, Farrell (1995) shows that, when alloy 6061-T6 is held at
39 200 degrees C (392 degrees F), its yield strength drops from approximately 18 ksi at 10,000
40 hours (1.14 years) to approximately 11.5 ksi at 100,000 hours (11.4 years). Because of this
41 sensitivity to exposure time, ASME B&PV Code Section II requires that time-dependent
42 properties be used for exposures above 177 degrees C (350 degrees F) for this alloy.

1 The maximum expected temperature of fuel cladding has been estimated to be 400 degrees C
2 [752 degrees F] at the beginning of storage (Jung et. al., 2013). This cladding temperature is
3 expected to decrease to around 266 degrees C [510 degrees F] after 20 years and to
4 approximately 127 degrees C [261 degrees F] after 60 years. It is apparent from these
5 temperatures that the 6061 and 6063 aluminum alloys may experience significant overaging at
6 a higher temperature than that for precipitation treatment, leading to loss of strength. This loss
7 of strength could be an issue for any subcomponents that perform a structural function.
8 Because Al 1100 aluminum is not a precipitation-hardened alloy, it will not experience any
9 overaging. However, if it is used in the cold worked state, it will anneal at temperatures above
10 300 degrees C [572 degrees F] (ASM International, 1991). This annealing will reduce strength,
11 which could be significant for subcomponents that serve a structural function.

12 Aluminum subcomponents exposed to sheltered and embedded environments experience lower
13 temperatures than the internal subcomponents. Time-temperature profiles calculated for the
14 canister surface (EPRI, 2006; Meyer et al., 2013) suggest that temperatures in excess of
15 200 degrees C [392 degrees F] could initially be present on portions of the canister surface and
16 temperatures above 100 degrees C [212 degrees F] could persist for 30 years. Based on these
17 temperatures, thermal aging could occur on aluminum subcomponents that have a structural
18 function and are attached directly to the canister shell or cask wall.

19 Because thermal aging of aluminum is a possible aging mechanism, the NRC reviewer should
20 review any aging analyses for aluminum structural components that are exposed to the elevated
21 temperatures discussed above, as contained in the applicant's original design-bases
22 documents, to determine whether the renewal application adequately addresses the
23 implications of extending the operating period to 60 years. This reexamination of the original
24 analyses would typically be defined as TLAAs in the renewal application. The staff's guidance
25 for the review of TLAAs is provided in NUREG-1927, Revision 1. If the original design basis
26 does not include the pertinent analyses, the reviewer nevertheless should ensure that the
27 application addresses the potential for thermal aging to adversely affect the structural function of
28 aluminum components.

29 3.2.3.8 *Radiation embrittlement*

30 Embrittlement of metals may occur under exposure to neutron radiation. Depending on the
31 neutron fluence, radiation can cause changes in mechanical properties, such as loss of ductility,
32 fracture toughness, and resistance to cracking.

33 Farrell and King (1973) showed that pure aluminum had increased strength but decreased
34 ductility after being irradiated to fast fluences in the range of 1 to 3×10^{22} n/cm²
35 [6.5 to 19.4×10^{22} n/in²] from a research reactor for 8 years. Alexander (1999) showed that
36 irradiation at 10^{22} n/cm² [6.5×10^{22} n/in²] simulating reactor conditions affected the mechanical
37 properties of aluminum alloy 6061-T651.

38 Some results from radiation testing of aluminum-based neutron poisons are reported in the
39 literature (EPRI, 2009a). Gamma, thermal neutron, and fast neutron radiation testing of an
40 aluminum-based laminate composite in water for 9 years and exposed to up to 7×10^{11} rad
41 gamma, 3.6×10^{18} n/cm² [2.2×10^{19} n/in²] fast neutron fluence, and 2.7×10^{19} n/cm²
42 [1.7×10^{20} n/in²] thermal neutron fluence showed no change in ultimate strength and no other
43 signs of physical deterioration except for severe oxidation because of the presence of water.
44 Also, radiation testing of an aluminum-based, sintered composite subjected to up to
45 1.5×10^{20} n/cm² [9.7×10^{20} n/in²] fast neutron fluence and a maximum of 3.8×10^{11} rad gamma

1 exposure showed little change in the yield strength and ultimate strength (EPRI, 2009a).
2 Finally, neutron radiation of borated aluminum to fluences of 10^{17} n/cm² [6.5×10^{17} n/in²]
3 showed no dimensional change or radiation damage (EPRI, 2009a). These test conditions are
4 expected to be more severe than those experienced by aluminum alloys in the extended
5 storage application (EPRI, 2009a).

6 As discussed in Section 3.2.1.9 of this report, the maximum potential accumulated neutron
7 fluence on DSS components after 100 years was calculated to be 2.63×10^{16} n/cm²
8 [1.70×10^{17} n/in²]. This fluence is well below the levels that have been found degrade the
9 mechanical properties of aluminum alloys. Thus, radiation embrittlement of aluminum
10 subcomponents exposed to any environment is expected to be insignificant, and therefore,
11 aging management is not required during the 60-year timeframe.

12 **3.2.4 Nickel alloys**

13 Nickel alloys are used in only a few DSS applications. In the HI-STAR overpack, nickel
14 alloy 718 (ASME SB637) is used to construct closure plate bolts and trunnion bolts, and nickel
15 alloy X750 is used to construct seals. These components are exposed to an outdoor
16 environment. Nickel alloy 718 (ASME SB637) is also used to construct the trunnion for the
17 HI-TRAC transfer cask, which is predominantly exposed to an indoor environment or otherwise
18 encased without direct air ingress except for short periods of air exposure during transfer
19 operations. For such air-indoor/outdoor environments, the aging effects from aqueous corrosion
20 processes are expected to be bounded by those from the outdoor environment. Both nickel
21 alloys 718 and X750 are precipitation-hardened alloys that contain chromium to form a passive
22 oxide film on the surface (Crook, 2005).

23 *3.2.4.1 General corrosion*

24 The high chromium contents of alloys 718 and X750 (greater than 17 and 14 weight percent,
25 respectively), make these alloys very resistant to general corrosion, even in such reducing acids
26 as hydrochloric acid (Crook, 2005). Because of its passive behavior and high corrosion
27 resistance, general corrosion of nickel alloys exposed to outdoor environments is not
28 considered to be credible, and therefore, aging management is not required during the 60-year
29 timeframe.

30 *3.2.4.2 Pitting and crevice corrosion*

31 As discussed in Section 3.2.1.2, pitting corrosion is a localized form of corrosion that is confined
32 to a point or small area of a metal surface (Frankel, 2003) and crevice corrosion occurs in a
33 wetted environment when a crevice exists that allows a corrosive environment to develop in a
34 component (Kelly, 2003).

35 Section 3.2.1.1 discussed how an aqueous electrolyte can be developed in outdoor air. This
36 electrolyte could contain chemical species such as halides and sulfides. Localized corrosion in
37 the form of pitting and/or crevice corrosion may occur for some passive nickel alloys, but overall,
38 nickel alloys are more resistant to localized corrosion than stainless steels (Crook, 2005).
39 Nickel alloy 718 is used in sea water applications, where the chloride concentration is much
40 higher than that from outdoor air. Furthermore, for many nickel alloys in different environmental
41 systems, localized corrosion growth is often observed to slow down or stop, which is referred to
42 as the stifling and arrest phenomena (He and Dunn, 2007). Because of the high corrosion
43 resistance, pitting or crevice corrosion of nickel subcomponents exposed to outdoor air is not

1 considered to be credible, and therefore, aging management is not required during the 60-year
2 timeframe.

3 3.2.4.3 *Microbiologically influenced corrosion*

4 As discussed in Section 3.2.1.4, MIC is corrosion caused or promoted by the metabolic activity
5 of microorganisms (Dexter, 2003). Microorganisms can live in many environments, such as
6 water, soil, and air, where aerobic bacteria (e.g., iron-manganese oxidizing bacteria,
7 sulfur/sulfide oxidizing bacteria, methane producers, organic acid-producing bacteria), fungi,
8 and algae can develop.

9 Although the moisture necessary to support microbial activity may be present on surfaces
10 exposed to outdoor environments, all of the operating experience of MIC of metallic components
11 is from conditions where the surface is continuously wet. Furthermore, there is no operational
12 or experimental evidence of MIC degradation of nickel-chromium alloys similar to 718 and X750
13 (Little and Lee, 2009). Due to the absence of any operating experience of MIC damage to
14 nickel alloys under atmospheric conditions, MIC of nickel subcomponents exposed to outdoor
15 air is not considered to be credible, and therefore, aging management is not required during the
16 60-year timeframe.

17 3.2.4.4 *Stress corrosion cracking*

18 As discussed in Section 3.2.1.5, SCC is the cracking of a metal produced by the combined
19 action of corrosion and tensile stress (applied or residual) (Jones, 1992, 2003). SCC of nickel
20 alloys has been experienced in high-temperature water and hot caustic solutions (Phull, 2003).
21 These conditions do not exist in the outdoor air environment of DSSs. Although
22 chloride-containing electrolytes could develop in outdoor air, as discussed in Section 3.2.1.1,
23 nickel-based alloys are known to be highly resistant to the chloride-induced SCC that affects
24 stainless steels. In indoor air, the probability of developing a corrosive aqueous electrolyte is
25 negligible. Because alloys 718 and X750 are not susceptible to the dry storage outdoor air
26 environments, SCC is not expected to be credible. Therefore, aging management is not
27 required during the
28 60-year timeframe.

29 3.2.4.5 *Fatigue*

30 As discussed previously in Section 3.2.1.7, because spent fuel storage is a static application,
31 cyclic loading by a purely mechanical means is largely limited to transfer cask lifting trunnions,
32 which are loaded each time a canister is moved from the spent fuel pool to the dry storage pad.
33 Other subcomponents, however, could experience cyclic loads due to thermal effects, such as
34 those caused by daily and seasonal fluctuations in the temperature of the external
35 environment. The NRC reviewer should review the fatigue analyses contained in the applicant's
36 original design-basis documents to determine whether the renewal application adequately
37 addresses the implications of extending the operating period to 60 years. This reexamination of
38 the original fatigue analyses would typically be defined as TLAA's.

39 In some cases, fatigue analyses may have been performed to support the original design but
40 are not explicitly discussed in the design basis documentation. As a result, the reviewer should
41 review the design codes and standards to identify any required fatigue analyses and ensure that
42 the applicant addresses those analyses with a TLAA. For example, components that were
43 designed in accordance with the American Society of Mechanical Engineers Boiler and

1 Pressure Vessel Code (ASME Code) Section III, Division 1, Subsections NB or NC
2 (ASME, 2007a) were evaluated for the effects of cyclic loading per subparagraphs NB 3222.4
3 and NC 3219.2, respectively.

4 The staff's guidance for the review of TLAA's is provided in NUREG-1927, Revision 1, and
5 summarized in Chapter 5 of this report. In its evaluation of a TLAA, an applicant may conclude
6 that an analysis can no longer support a determination that aging will not adversely affect an
7 important-to-safety function in the 60-year timeframe of the period of extended operation. In
8 that case, the applicant may manage the aging of the associated SCC with an AMP.

9 The AMR tables in Chapter 4 recommend that applicants address any applicable TLAA's
10 associated with components with a structural function. If no fatigue analysis was performed in
11 support of the component design, no action is required of the applicant.

12 3.2.4.6 *Radiation embrittlement*

13 Depending on the neutron fluence, radiation can cause changes in mechanical properties such
14 as loss of ductility, fracture toughness, and resistance to cracking. Nickel-based alloys
15 experienced significant reductions in tensile ductility during neutron irradiation at elevated
16 temperatures of 400–600 degrees C [752–1,112 degrees F] for neutron doses approaching
17 10–15 displacements per atom (dpa), which corresponds to a neutron fluence of about
18 10^{21} – 10^{22} n/cm² [6.5×10^{21} – 6.5×10^{22} n/in²] (Was et al., 2006; Rowcliffe, 2009). Nickel
19 alloy X-750 cracking has been observed extensively in nuclear power plant applications after
20 attaining an end-of-life fluence of 1 to 10×10^{21} n/cm² [6.5 to 65×10^{21} n/in²] (Was et al., 2006).

21 As discussed in Section 3.2.1.9 of this report, the maximum potential accumulated neutron
22 fluence on DSS basket components after 100 years was calculated to be 2.63×10^{16} n/cm²
23 [1.70×10^{17} n/in²]. This fluence is five to six orders of magnitude below the level at which the
24 mechanical properties of nickel have been observed to be degraded. In addition, for the nickel
25 overpack and transfer cask subcomponents, the neutron exposure is significantly lower than the
26 calculated exposure for the basket components in Section 3.2.1.9. Thus, radiation
27 embrittlement of nickel alloys is expected to be insignificant, and therefore, aging management
28 is not required during the 60-year timeframe.

29 3.2.4.7 *Stress relaxation*

30 Section 3.2.1.10 explained that stress relaxation of bolting is the steady loss of stress due to
31 atomic movement at elevated temperature in a loaded part where dimensions are fixed
32 (Earthman, 2000). The service temperature limit for nickel alloy 718 is 649 degrees C
33 [1,200 degrees F] (Bickford, 2008), which is much higher than the external temperature of the
34 HI-STAR overpack in which nickel bolts are used. Below the service temperature limit, the bolts
35 are expected to maintain their original clamping force. Thus, stress relaxation of nickel alloy
36 subcomponents exposed to the outdoor environment is not considered to be credible, and
37 therefore, aging management is not required during the 60-year timeframe.

38 3.2.4.8 *Wear*

39 Fretting wear is the repeated cyclical rubbing between two surfaces. For the HI-TRAC transfer
40 cask exposed to air-indoor/outdoor environments, the nickel alloy used to construct the lifting
41 trunnion may experience cyclic rubbing during loading and unloading. Thus, wear of the nickel

1 alloy is considered to be credible, and therefore, aging management is required during the
2 60-year timeframe.

3 **3.2.5 Copper alloys**

4 Copper alloys are used in only a few DSS applications. In the HI-STAR overpack, brass, which
5 is a copper-zinc alloy containing more than 50 percent copper, is used as the rupture disk
6 material. In the NUHOMS HSM, copper is used to construct the lightning protection system.
7 Both subcomponents are exposed to outdoor air.

8 *3.2.5.1 General corrosion*

9 General corrosion, also known as uniform corrosion, proceeds at approximately the same rate
10 over a metal surface (Phull, 2003b). Freely exposed copper surfaces in contact with moist air or
11 water are subject to general corrosion. The corrosion rate depends on solution composition,
12 pH, and temperature. The copper Pourbaix diagram (Pourbaix, 1974) indicates that copper and
13 copper alloys are reactive with water in the presence of oxygen, but the low corrosion rate has
14 allowed their wide use in industrial, marine, and rural atmospheres (Cohen, 2005). General
15 corrosion of copper and its alloys is the predominant corrosion mode, because they do not form
16 a truly passive oxide film on the surface.

17 Atmospheric corrosion of copper has been observed and studied extensively (Leidheiser, 1974;
18 Rozenfeld, 1972). The corrosion rate of copper is strongly dependent on relative humidity and
19 the concentration of pollutants in the air (e.g., chlorides, sulfur dioxide, hydrogen sulfide). The
20 presence of NaCl in a marine environment has a strong corrosive effect toward copper under
21 thin electrolyte layers and in alternating wet and dry cyclic conditions. Copper corrosion rates
22 usually decrease with time, following an exponential decay law (Feliu et al., 1993). Typical
23 corrosion rates of copper exposed to marine and industrial environments are 0.6–2.5 $\mu\text{m}/\text{yr}$
24 [0.024–0.098 mils/yr] and 1.3 $\mu\text{m}/\text{yr}$ [0.051 mils/yr], respectively (Tracy, 1955; Herman and
25 Castillo, 1974). Fonseca et al. (2004) recorded copper corrosion in marine environments as
26 high as 7.8 $\mu\text{m}/\text{yr}$ [0.31 mils/yr]. In atmospheric marine environments, copper corrosion is on
27 the order of 16 $\mu\text{m}/\text{yr}$ [0.62 mils/yr] (Farro et al., 2009). Assuming a corrosion rate of 10 $\mu\text{m}/\text{yr}$
28 [0.39 mils/yr], the metal loss could be 0.6 mm [23.6 mils] over 60 years. As such, general
29 corrosion of copper alloys exposed to an outdoor air environment is considered to be credible,
30 and therefore, aging management is required during the 60-year timeframe.

31 *3.2.5.2 Pitting and crevice corrosion*

32 As discussed in Section 3.2.1.2, pitting corrosion is a localized form of corrosion that is confined
33 to a point or small area of a metal surface (Frankel, 2003), and crevice corrosion occurs in a
34 wetted environment when a crevice exists that allows a corrosive environment to develop in a
35 component (Kelly, 2003).

36 The common form of atmospheric corrosion for copper exposed to outdoor air is general
37 corrosion, because copper alloys do not have a true protective film (Cohen, 2005). In an
38 oxidizing environment, copper could experience surface roughening, initially appearing like
39 localized corrosion; however, localized corrosion tends to converge with general corrosion
40 (i.e., the penetration front of localized corrosion merges with that of general corrosion).
41 Long-term tests of copper alloys show that the average pit depth does not continually increase
42 with extended times of exposure (Cohen, 2005). Copper has been commonly used for
43 architectural components exposed to outdoor air for many years, such as when used for roofing,

1 building fronts, and statues, where localized corrosion is not shown to be evident. Because
2 localized corrosion is not a primary corrosion mechanism for copper alloys exposed to outdoor
3 air, and it tends to converge with general corrosion, it is not considered to be credible, and
4 therefore, aging management is not required during the 60-year timeframe.

5 3.2.5.3 *Microbiologically influenced corrosion*

6 As discussed in Section 3.2.1.4, MIC is corrosion caused or promoted by the metabolic activity
7 of microorganisms (Dexter, 2003). Although the moisture necessary to support microbial
8 activity may be present on surfaces exposed to the outdoor environment, all of the operating
9 experience of MIC of metallic materials is from conditions under which the surface is
10 continuously wet, and it is unclear whether these rates could be sustained if the conditions to
11 support MIC are only present on an intermittent basis. Furthermore, there is no experimental
12 evidence of MIC degradation of copper alloys under atmospheric conditions. Due to the
13 absence of any operating experience of MIC damage of copper alloys under atmospheric
14 conditions, MIC is not considered to be significant, and therefore, aging management is not
15 required during the 60-year timeframe.

16 3.2.5.4 *Radiation embrittlement*

17 Depending on the neutron fluence, radiation can cause changes in mechanical properties, such
18 as loss of ductility, fracture toughness, and resistance to cracking. Radiation hardening and
19 embrittlement of pure copper and copper-based alloys have been observed at temperatures in
20 the range of 60–90 degrees C [140–194 degrees F] in the dose range of 10^{-3} – 10^{-1} dpa
21 (Fabritsiev et al., 2004). Blewitt et al. (1957) observed yield drop on stress–strain curves,
22 hardening, and a decrease in uniform and total elongation upon irradiation of pure copper at
23 60 degrees C [140 degrees F] to doses of 10^{19} n/cm² [6.5×10^{19} n/in²].

24 As discussed in Section 3.2.1.9 of this report, the maximum potential accumulated neutron
25 fluence on DSS basket components after 100 years was calculated to be 2.63×10^{16} n/cm²
26 [1.70×10^{17} n/in²]. This fluence is at least three orders of magnitude below the level at which
27 the mechanical properties of copper alloys have been reported to be degraded. In addition, for
28 locations outside of the overpack where copper alloys are used, the accumulated dose is much
29 lower than the level calculated in Section 3.2.1.9. Thus, radiation embrittlement of copper alloys
30 exposed to outdoor air is expected to be insignificant, and therefore, aging management is not
31 required during the 60-year timeframe.

32 3.2.6 **Lead**

33 Lead is used as gamma radiation shielding in the NUHOMS and Holtec transfer casks, as well
34 as some NUHOMS dry shielded canister designs. In each case, the lead is encased in steel or
35 stainless steel and thus is not exposed to water or atmospheric contaminants. Lead is well
36 known to be very resistant to corrosion in a variety of environments (Alhasan, 2005). Because
37 there are no credible aging mechanisms that could challenge the ability of lead to perform its
38 shielding (and, in some cases, heat transfer) functions, aging management of this material is not
39 required during the 60-year timeframe.

40 3.2.7 **Depleted uranium**

41 Depleted uranium is used as a shield plug in the FuelSolutions canister. The material is
42 encased in steel or stainless steel and thus is not exposed to water or atmospheric

1 contaminants. Uranium is known to be resistant to corrosion in a variety of environments
2 (Lillard and Hanrahan, 2005). Because there are no credible aging mechanisms that could
3 challenge the ability of depleted uranium to perform its shielding functions, aging management
4 of this material is not required during the 60-year timeframe.

5 **3.2.8 Coatings**

6 Coatings in DSSs are used primarily for corrosion mitigation, to facilitate decontamination, and
7 to improve heat-rejection capability by increasing the emissivity of cask internal components. A
8 wide array of coating materials is used to fulfill these functions, such as organic epoxy, inorganic
9 zinc-rich coatings, galvanized zinc, aluminum, nickel, and cadmium. However, coatings are
10 often present for operational purposes and may not be credited as supporting an important-to-
11 safety function. Thus, the reviewer should examine the DSS design-basis documentation to
12 verify that the renewal applicant appropriately identified the coatings that meet the renewal
13 scoping criteria in NUREG-1927, Revision 1.

14 Coatings are exposed to outdoor air, indoor/outdoor air (transfer cask), and sheltered
15 environments, which are characterized by elevated temperature and radiation exposure. As
16 discussed in greater detail for neutron shielding materials in Section 3.3.1, polymeric materials
17 may be susceptible to heat- and radiation-induced molecular scission (breaking) and cross-
18 linking that can cause embrittlement and cracking.

19 The variety of coatings and the proprietary nature of many coating systems make a generic
20 evaluation of specific degradation mechanisms impractical. Nevertheless, the NRC recognizes
21 that coatings may degrade, either through aging or inappropriate application methods, and
22 recommends in-service condition assessments of coatings to ensure that they continue to
23 support their important-to-safety functions. NRC Regulatory Guide 1.54, Revision 2, "Service
24 Level I, II, and III Protective Coatings Applied to Nuclear Power Plants," references American
25 Society for Testing and Materials (ASTM) standards that are considered appropriate guidance
26 for coating maintenance in nuclear power plants (NRC, 2010d). The ASTM standards typically
27 recommend periodic visual inspections for blistering, cracking, flaking/peeling, and rusting,
28 which may be followed by physical tests when degradation is identified. Regulatory Guide 1.54
29 also notes that the Electric Power Research Institute (EPRI) Report 1019157, "Guideline on
30 Nuclear Safety-Related Coatings," Revision 2, provides additional information on the
31 maintenance of coatings (EPRI, 2009b).

32 The reviewer should verify that the renewal applicant has an existing coating maintenance
33 program or proposes a new aging management program for coatings that are credited with
34 performing an important-to-safety function or protecting an important-to-safety component. The
35 AMR tables identify thermal and radiation effects as most likely to degrade coatings, and a
36 site-specific AMP consistent with ASTM guidelines is recommended to manage aging.

37 **3.2.9 References**

38 AISC. ANSI/AISC 360-10, "Specification for Structural Steel Buildings." Chicago, Illinois:
39 American Institute of Steel Construction. 2010.

40 Alexander, D.J. "Effects of Irradiation on the Mechanical Properties of 6061-T651 Aluminum
41 Base Metal and Weldments." ASTM Special Technical Publication. Vol. 1325.
42 pp. 1,027-1,044. 1999.

- 1 Alexander, D.J. and R.K. Nanstand. "The Effects of Aging for 50,000 Hours at 343°C on the
2 Mechanical Properties of Type 308 Stainless Steel Weldments." Proceedings of the Seventh
3 International Symposium on Environmental Degradation of Materials in Nuclear Power
4 Systems—Water Reactors. Breckenridge, Colorado. NACE. Houston, Texas. pp. 747–758.
5 1995.
- 6 Alexander, D., P. Doubell, and C. Wicker. "Degradation of Safety Injection Systems and
7 Containment Spray Piping and Tank—Fracture Toughness Analysis." Presentation at
8 Fontevraud 7, *Contribution of Materials Investigations to Improve the Safety and Performance of*
9 *LWRs*, September 26–30, 2010. Avignon, France. 2010.
- 10 Alhasan, S.J. "Corrosion of Lead and Lead Alloys." In ASM Handbook, Vol. 13B, *Corrosion:*
11 *Materials*. Materials Park, Ohio: ASM International. pp. 195–204. 2005.
- 12 Andresen, P.L., F.P. Ford, K. Gott, R.L. Jones, P.M. Scott, T. Shoji, Staehle, and R.L. Tapping.
13 "Expert Panel Report on Proactive Materials Degradation Assessment." NUREG/CR-6923.
14 Washington, DC: U.S. Nuclear Regulatory Commission. 2007.
- 15 ASM International. *Metals Handbook, Desk Edition, Second Edition*. Materials Park, Ohio:
16 ASM International. pp. 280–285. 1998.
- 17 _____. "Heat Treating of Aluminum Alloys." In ASM Handbook, Vol. 4, *Heat Treating*.
18 Materials Park, Ohio: ASM International. pp. 841–879. 1991.
- 19 ASME. Boiler and Pressure Vessel (B&PV) Code, Section III, "Rules for Construction of
20 Nuclear Facility Components," Division 1, Subsection NB, "Class 1 Components," and
21 Subsection NC, "Class 2 Components"; American Society of Mechanical Engineers. 2007a.
- 22 _____. Boiler and Pressure Vessel (B&PV) Code, Section II, "Materials," Part D, "Properties,"
23 American Society of Mechanical Engineers. 2007b.
- 24 Baboian, R. "Galvanic Corrosion." In ASM Handbook, Vol. 13A, *Corrosion: Fundamentals,*
25 *Testing, and Protection*. Materials Park, Ohio: ASM International. pp. 210–213. 2003.
- 26 Baggerly, R. "Environmental Failures of High Strength Bolts, in Case Histories on Integrity and
27 Failures in Industry." V., ed. *Proceedings of an International Symposium on Case Histories on*
28 *Integrity and Failures in Industry*, September 28–October 2, 1999. Milan, Italy. 1999.
- 29 Bass, H.K. "The Corrosion of Aluminum in Boric Acid Solutions." Master's thesis.
30 Agricultural and Mechanical College of Texas. College Station, Texas. 1956.
- 31 Basson, J.P. and C. Wicker. "Environmentally Induced Transgranular Stress Corrosion
32 Cracking of 304L Stainless Steel Components at Koeberg." Fontevraud 5 International
33 Symposium, *Contributions of Materials Investigations to Resolution of Problems Encountered in*
34 *Pressurized Water Reactors*. Société Française d'Énergie Nucléaire–SFEN. Paris, France.
35 Vol. 1–2. 1,175p. September 2002.
- 36 Baumgattner, M. and H. Kaesche. "The Nature of Crevice Corrosion of Aluminum in Chloride
37 Solutions." *Werkstoffe und Korrosion*. Vol. 39. pp. 129–135. 1988.

- 1 Bickford, J.H. *Introduction to the Design and Behavior of Bolted Joints*. 4th Edition.
2 Boca Raton, Florida: CRC Press. 2008.
- 3 Blau, P.J. "Rolling Contact Wear." In ASM Handbook Vol. 18, *Friction, Lubrication, and Wear*
4 *Technology*. Materials Park, Ohio: ASM International. pp. 257–262. 1992.
- 5 Blewitt, T.H., R.R. Coltman, C.E. Klabunde, and T.S. Noggle. "Low-Temperature Reactor
6 Irradiation Effects in Metals." *Journal of Applied Physics*. Vol. 28. pp. 639–644. 1957.
- 7 Bruhn, D.F., S.M. Frank, F.F. Roberto, P.J. Pinhero, and S.G. Johnson. "Microbial Biofilm
8 Growth on Irradiated, Spent Nuclear Fuel Cladding." *Journal of Nuclear Materials*. Vol. 384,
9 No. 2. pp. 140–145. 2009.
- 10 Cadek, J. *Creep of Metallic Materials*. Elsevier Science Publishing Company, Inc. 1988.
- 11 Caprio, J.J., A. Parra, and L. Martinez. "Scanning Electron Microscopy and Infrared
12 Spectroscopic Studies of Marine Atmospheric Corrosion Products of Steel." Paper No. 242.
13 Houston, Texas: NACE International. 1995.
- 14 Caseres, L. "Electrochemical behavior of aluminized steel type 2 in scale-forming waters."
15 Ph.D. dissertation. Tampa, Florida: University of South Florida. 2007.
- 16 Caskey, G.R., R.S. Ondrejcin, P. Aldred, R.B. Davis, and S.A. Wilson. "Effects of Irradiation on
17 Intergranular Stress Corrosion Cracking of Type 304 Stainless Steel." *Proceedings of 45th*
18 *NACE Annual Conference*, April 23–27, 1990, Las Vegas, Nevada. 1990.
- 19 Chandra, K., K. Vivekanand, V.S. Raja, R. Tewari, and G.K. Dey. "Low Temperature Thermal
20 Ageing Embrittlement of Austenitic Stainless Steel Welds and its Electrochemical Assessment."
21 *Corrosion Science*. Vol. 54. pp. 278–290. 2012.
- 22 Chopra, O., D. Diercks, R. Fabian, Z. Han, and Y. Liu. "Managing Aging Effects on Dry Cask
23 Storage Systems for Extended Long-Term Storage and Transportation of Used Fuel."
24 FCRD–UFD–2014–000476. ANL–13/15, Rev. 2. Washington, DC: U.S. Department of
25 Energy. 2014.
- 26 Code of Federal Regulations. Title 10, Energy," Part 50, "Domestic Licensing of Production and
27 Utilization Facilities," Appendix H, "Reactor Vessel Material Surveillance Program
28 Requirements." Washington, DC: U.S. Government Printing Office. 2015.
- 29 Cohen, A. "Corrosion of Copper and Copper Alloys." In ASM Handbook, Vol. 13B, *Corrosion:*
30 *Materials*. Materials Park, Ohio: ASM International. pp. 125–163. 2005.
- 31 Cook, A., J. Duff, N. Stevens, S. Lyon, A. Sherry, and T.J. Marrow. "Preliminary Evaluation of
32 Digital Image Correlation for *In-Situ* Observation of Low Temperature Atmospheric-Induced
33 Chloride Stress Corrosion Cracking in Austenitic Stainless Steels." *ECS Transactions*. Vol. 25,
34 No. 37. pp. 119–132. 2010.
- 35 Crook, P. "Corrosion of Nickel and Nickel-Base Alloys." In ASM Handbook, Vol. 13B,
36 *Corrosion: Materials*. Materials Park, Ohio: ASM International. pp. 228–251. 2005.

- 1 David, D., C. Lemaitre, and C. Crusset. "Archaeological Analogue Studies for the Prediction of
2 Long-Term Corrosion on Buried Metals." D. Feron and D. D. Macdonald, eds. EFC Series
3 Vol. 36, *Prediction of Long-Term Corrosion Behavior in Nuclear Waste Systems*. 242p.
4 Maney, London, United Kingdom. European Federation of Corrosion Publications. 2002.
- 5 Davison, R.M., T. DeBold, and M.J. Johnson. "Corrosion of Stainless Steels." In
6 ASM Handbook Vol. 13, *Corrosion*. Materials Park, Ohio: ASM International. pp. 547–565.
7 1987.
- 8 Dexter, S.C. "Microbiologically Influenced Corrosion." In ASM Handbook, Vol. 13A, *Corrosion:
9 Fundamentals, Testing, and Protection*. Materials Park, Ohio: ASM International.
10 pp. 398–416. 2003.
- 11 Dragun, J. "The Soil Chemistry of Hazardous Materials." Silver Spring, Maryland:
12 Hazardous Materials Control Research Institute. pp. 325–445. 1988.
- 13 Earthman, J.C. "Introduction to Creep and Stress-Relaxation Testing." In ASM Handbook.
14 Vol. 8, *Mechanical Testing and Evaluation*. Materials Park, Ohio: ASM International.
15 pp. 361–362. 2000.
- 16 EPRI. "Handbook of Neutron Absorber Materials for Spent Nuclear Fuel Transportation and
17 Storage Applications," Report 1019110. Palo Alto, California: Electric Power Research
18 Institute. 2009a.
- 19 _____. "Guideline on Nuclear Safety-Related Coatings," Revision 2, Report 1019157.
20 Palo Alto, California: Electric Power Research Institute. 2009b.
- 21 _____. "Aging Effects for Structures and Structural Components (Structural Tools)."
22 Report 1015078. Palo Alto, California: Electric Power Research Institute. 2007.
- 23 _____. "Climatic Corrosion Considerations for Independent Spent Fuel Storage Installations in
24 Marine Environments." Report 1013524. Palo Alto, California: Electric Power Research
25 Institute. 2006.
- 26 _____. "Effects of Marine Environments on Stress Corrosion Cracking of Austenitic Stainless
27 Steels." Report 1011820. Palo Alto, California: Electric Power Research Institute. 2005.
- 28 Fabritsiev, S.A., A.S. Pokrovsky, and S.E. Ostrovsky. "Effect of the Irradiation–Annealing–
29 Irradiation Cycle on the Mechanical Properties of Pure Copper and Copper Alloy." *Journal of
30 Nuclear Materials*. Vol. 324. pp. 23–32. 2004.
- 31 Ferrell, K., "Assessment of Aluminum Structural Materials for Service Within the ANS Reflector
32 Vessel," ORNL/TM-13049, Oak Ridge National Laboratory, August, 1995.
- 33 Farrell, K. and R.T. King. "Radiation-Induced Strengthening and Embrittlement in Aluminum."
34 *Metallurgical Transactions A. Physical Metallurgy and Materials Science*. Vol. 4, Issue 5.
35 pp. 1,223–1,231. 1973.
- 36 Farro, N.W., L. Veleva, and P. Aguilar. "Copper Marine Corrosion: I. Corrosion Rates in
37 Atmospheric and Seawater Environments of Peruvian Port." *The Open Corrosion Journal*.
38 Vol. 2. pp. 130–138. 2009.

- 1 Feliu, S., M. Morcillo, and S. Feliu, Jr. "The Prediction of Atmospheric Corrosion from
2 Meteorological and Pollution Parameters-II, Long-Term Forecasts." *Corrosion Science*.
3 Vol. 34, No. 3. pp. 415–422. 1993.
- 4 Foct, F. and J.-M. Gras. "Semi-Empirical Model for Carbon Steel Corrosion in Long Term
5 Geological Nuclear Waste Disposal." D. Feron and D.D. Macdonald, eds. EFC Series. Vol. 36.
6 *Prediction of Long-Term Corrosion Behavior in Nuclear Waste Systems*. Maney, London,
7 United Kingdom. 91p. 2002.
- 8 Foley, R.T. "Localized Corrosion of Aluminum Alloys—A Review." *Corrosion*. Vol. 42.
9 pp. 277–288. 1986.
- 10 Fonseca, I.T.E., R. Picciochi, M.H. Mendonca, and A.C. Ramos. "The Atmospheric Corrosion of
11 Copper at Two Sites in Portugal: A Comparative Study." *Corrosion Science*. Vol. 46.
12 pp. 547–561. 2004.
- 13 FPL. "Turkey Point Nuclear Plant Unit 3, Docket No. 50-250, 10 CFR 50.55a, Request for
14 Temporary Non-Code Repair, Spent Fuel Pool Cooling Line." Florida Power and Light.
15 ADAMS Accession No ML052780060. 2005.
- 16 Frankel, G.S. "Pitting Corrosion." In ASM Handbook, Vol. 13A, *Corrosion: Fundamentals,*
17 *Testing, and Protection*. Materials Park, Ohio: ASM International. pp. 236–241. 2003.
- 18 Fuhr, K., J. Gorman, J. Broussard, and G. White. "Failure Modes and Effects Analysis (FMEA)
19 of Welded Stainless Steel Canisters for Dry Cask Storage Systems." Palo Alto, California:
20 Electric Power Research Institute. 2013.
- 21 Gamble, R. "BWRVIP-100-A: BWR Vessel and Internal Project, Updated Assessment of the
22 Fracture Toughness of Irradiated Stainless Steel for BWR Core Shrouds." EPRI-1013396.
23 Palo Alto, California: Electric Power Research Institute. 2006.
- 24 Garcia-Guinea, J., V. Cardenes, A.T. Martinez, and M.J. Martinez. "Fungal Bioturbation Paths
25 in a Compact Disk." *Naturwissenschaften (The Science of Nature)*. Vol. 88. pp. 351–354.
26 2001.
- 27 Gavendra, D.J., W.F. Michaud, T.M. Galvin, W.F. Burke, and O.K. Chopra. NUREG/CR–6428,
28 "Effects of Thermal Aging on Fracture Toughness and Charpy-Impact Strength of Stainless
29 Steel Pipe Welds." Washington, DC: U.S. Nuclear Regulatory Commission. May 1996.
- 30 Ghali, E. "Aluminum and Aluminum Alloys." In *Uhlig's Corrosion Handbook*. 3rd Edition.
31 R.W. Revie, eds. John Wiley & Sons, Inc. pp. 715–745. 2011.
- 32 _____. *Corrosion Resistance of Aluminum and Magnesium Alloys Understanding,*
33 *Performance, and Testing*. Hoboken, New Jersey: John Wiley & Sons, Inc. 2010.
- 34 Gibeling, J.C. "Creep Deformation of Metals, Polymers, Ceramics, and Composites." In
35 ASM Handbook, Vol. 8, *Mechanical Testing and Evaluation*. Materials Park, Ohio:
36 ASM International. pp. 363–368. 2000.

- 1 Grubb, J.F., T. DeBold, and J.D. Fritz. "Corrosion of Wrought Stainless Steels." In
2 ASM Handbook. Vol. 13B. *Corrosion: Materials*. Materials Park, Ohio: ASM International.
3 pp. 54–77. 2005.
- 4 Hack, H.P. *Galvanic Corrosion Test Methods*. Houston, Texas: NACE International. 1993.
- 5 Hanson, B., H. Alsaed, C. Stockman, D. Enos, R. Meyer, and K. Sorenson. "Used Fuel
6 Disposition Campaign: Gap Analysis to Support Extended Storage of Used Nuclear Fuel,
7 Rev. 0." Richland, Washington: Pacific Northwest National Laboratory. 2012.
- 8 He, X., T. Mintz, R. Pabalan, L. Miller, and G. Oberson. NUREG/CR–7170, "Assessment of
9 Stress Corrosion Cracking Susceptibility for Austenitic Stainless Steels Exposed to Chloride and
10 Non-chloride Atmospheric Salts." Washington, DC: U.S. Nuclear Regulatory Commission.
11 2014.
- 12 He, X. D. Dunn. "Crevice Corrosion Penetration Rates of Alloy 22 in Chloride-Containing
13 Waters." *Corrosion*. Vol. 63. pp. 145–158. 2007.
- 14 Herman, R.S. and A.P. Castillo. ASTM-STP 558, "Short-Term Atmospheric Corrosion of
15 Various Copper-Base Alloys—Two- and Four-Year Results." West Conshohocken,
16 Pennsylvania: ASTM International. pp. 82–96. 1974.
- 17 Hoepfner, D.W. "Industrial Significance of Fatigue Problems." In ASM Handbook, Vol. 19.
18 *Fatigue and Fracture*." Materials Park, Ohio: ASM International. pp. 3–4. 1996.
- 19 Horn, J.M. and A. Meike. "Microbial Activity at Yucca Mountain." UCRL-ID-122256.
20 Livermore, California: Lawrence Livermore National Laboratory. 1995.
- 21 Hosler, R. "Screening Criteria for ID and OD-Initiated SCC of Pressure Boundary Stainless
22 Steel Components (Phase 1 of I&E Guideline Development)." AREVA document
23 51-9142337-000. October 18, 2010.
- 24 Jack, T.R., M.J. Wilmott, R.L. Sutherby, and R.G. Worthingham. "External Corrosion of Line
25 Pipe—A Summary of Research Activities." *Materials Performance*. Vol. 35. pp. 18–24. 1996.
- 26 Jones, R.H. "Stress corrosion Cracking." In ASM Handbook, Vol. 13A, *Corrosion:
27 Fundamentals, Testing, and Protection*. Materials Park, Ohio: ASM International.
28 pp. 346–366. 2003.
- 29 Jones, R.H. *Stress corrosion Cracking*. Materials Park, Ohio: ASM International: 1992.
- 30 Jung, H., P. Shukla, T. Ahn, L. Tipton, K. Das, X. He, and D. Basu. "Extended Storage and
31 Transportation: Evaluation of Drying Adequacy." San Antonio, Texas: Center for Nuclear
32 Waste Regulatory Analyses. 2013.
- 33 Kain, R. "Marine Atmospheric Stress Corrosion Cracking of Austenitic Stainless Steel."
34 *Materials Performance*. Vol. 29, No. 12. pp. 60–62. 1990.
- 35 Kaufman, J.G. *Properties of Aluminum Alloys: Tensile, Creep, and Fatigue Data at High and
36 Low Temperatures*. Materials Park, Ohio. ASM International. 1999.

- 1 Kelly, R.G. "Crevice Corrosion." In ASM Handbook, Vol. 13A, *Corrosion: Fundamentals,*
2 *Testing, and Protection.* Materials Park, Ohio: ASM International. pp. 242–247. 2003.
- 3 Kim, S. and Y. Kim. "Estimation of Thermal Aging Embrittlement of LWR Primary Pressure
4 Boundary Components." *Journal of the Korean Nuclear Society.* Vol. 30, No. 6. pp. 609–616.
5 1998.
- 6 King F. "Microbiologically Influenced Corrosion of Nuclear Waste Containers." *Corrosion.*
7 Vol. 65. pp. 233–251. 2009.
- 8 Kodama, T. "Corrosion of Wrought Carbon Steels." In ASM Handbook, Vol. 13B, *Corrosion:*
9 *Materials.* Materials Park, Ohio: ASM International. pp. 5–10. 2005.
- 10 Krauss, G., *Steels: Processing, Structure, and Performance.* Materials Park, Ohio.
11 ASM International. pp. 396–402. 2005.
- 12 Kulak, G.L, J.W. Fisher, and J.H.A. Struik. *Guide to Design Criteria for Bolted and Riveted*
13 *Joints.* 2nd ed. Chicago, Illinois: AISC Inc. 2001.
- 14 Leidheiser, H. *The Corrosion of Copper, Tin, and Their Alloys.* New York, New York:
15 John Wiley & Sons, Inc. 1974.
- 16 Lillard, J.A. and R.J. Hanrahan, Jr. "Corrosion of Uranium and Uranium Alloys, Corrosion:
17 Materials." Vol 13B, ASM Handbook. ASM International. pp. 370–384. 2005.
- 18 Little, B.J. and P.A. Wagner. "An Overview of Microbiologically Influenced Corrosion of Metals
19 and Alloys Used in the Storage of Nuclear Wastes." *Canadian Journal of Microbiology.* Vol. 42.
20 pp. 367–374. 1996.
- 21 Little, B.J. and J.S. Lee. "Microbiologically Influenced Corrosion." U.S. Naval Research
22 Laboratory Report NRL/BC/7303-08-8209. 2009.
- 23 Magee, J.H. "Wear of Stainless Steels." In ASM Handbook, Vol. 18, *Friction, Lubrication, and*
24 *Wear Technology.* Materials Park, Ohio: ASM International. pp. 710–724. 1992.
- 25 Manaktala, H.K. "Degradation Modes in Candidate Copper-Based Materials for High-Level
26 Radwaste Canisters." *Corrosion/90.* Paper No. 512. Las Vegas, Nevada: NACE. 1990.
- 27 Maruthamuthu, S., N. Muthukumar, M. Natesan, and N. Palaniswamy. "Role of Air Microbes on
28 Atmospheric Corrosion." *Current Science.* Vol. 94. pp. 359–363. 2008.
- 29 Mayuzumi, M., J. Tani, and T. Arai. "Chloride Induced Stress Corrosion Cracking of Candidate
30 Canister Materials for Dry Storage of Spent Fuel." *Nuclear Engineering and Design.* Vol. 238,
31 No. 5. pp. 1,227–1,232. 2008.
- 32 McCuen, R.H. and P. Albrecht. "Composite Modeling of Atmospheric Corrosion Penetration
33 Data." STP 1194, *Application of Accelerated Corrosion Testing to Service Life Prediction of*
34 *Materials.* ASTM International. West Conshohocken, Pennsylvania. 1994.
- 35 McMahon, C.J. "Hydrogen-Induced Intergranular Fracture of Steels." *Engineering Fracture*
36 *Mechanics.* Vol. 68. pp. 773–788. 2001.

- 1 Meyer, R.M., A.F. Pardini, J.M. Cuta, H.E. Adkins, A.M. Casella, A. Qiao, A.A. Diaz, and
2 S.R. Doctor. "NDE to Manage Atmospheric SCC in Canisters for Dry Storage of Spent Fuel:
3 An Assessment." PNNL-22495. Richland, Washington: Pacific Northwest National Laboratory.
4 2013.
- 5 Morgan, J.D. "Report on Relative Corrosivity of Atmospheres at Various Distances From the
6 Seacoast." NASA Report MTB 099-74. National Aeronautics and Space Administration.
7 Cape Canaveral, Florida: Kennedy Space Center. 1980.
- 8 Morrison, J.D. "Corrosion Study of Bare and Coated Stainless Steel." NASA TN D-6519.
9 Washington, DC: National Aeronautics and Space Administration. 1972.
- 10 Munier, G.B., L.A. Psota-Kelty, and J.D. Sinclair. *Atmospheric Corrosion*. W.H. Ailor, ed.
11 Wiley-Interscience. New York, New York. 275p. 1982.
- 12 NACE. *Corrosion Engineer's Reference Book*. Third Edition. Edited by R. Baboian.
13 Houston, Texas: NACE International. 2002.
- 14 Nguyen, T.H. and R.T. Foley. "On the Mechanism of Pitting of Aluminum." *Journal of*
15 *Electrochemical Society*. Vol. 126. pp. 1,855-1,860. 1979.
- 16 Nikolaev, Yu., A.V. Nikolaeva, and Ya.I. Shtrombakh. "Radiation Embrittlement of Low-Alloy
17 Steels." *International Journal of Pressure Vessels and Piping*. Vol. 79. pp. 619-636. 2002.
- 18 NRC. NUREG-1927, "Standard Review Plan for Renewal of Specific Licenses and Certificates
19 of Compliance for Dry Storage of Spent Nuclear Fuel." Revision 1. Washington, DC:
20 U.S. Nuclear Regulatory Commission. ADAMS Accession No. ML16179A148. 2016.
- 21 _____. "Identification and Prioritization of the Technical Information Needs Affecting Potential
22 Regulation of Extended Storage and Transportation of Spent Nuclear Fuel." Washington, DC:
23 U.S. Nuclear Regulatory Commission. May 2014.
- 24 _____. "Finite Element Analysis of Weld Residual Stresses in Austenitic Stainless Steel Dry
25 Cask Storage System Canisters." NRC Technical Letter Report. Washington, DC:
26 U.S. Nuclear Regulatory Commission. ADAMS Accession No. ML13330A512. 2013.
- 27 _____. "Potential Chloride-Induced Stress Corrosion Cracking of Austenitic Stainless Steel and
28 Maintenance of Dry Cask Storage System Canisters." NRC Information Notice 2012-20.
29 Washington, DC: U.S. Nuclear Regulatory Commission. ADAMS Accession
30 No. ML12319A440. 2012.
- 31 _____. NUREG-1801, "Generic Aging Lessons Learned (GALL) Report." Rev. 2. Washington,
32 DC: U.S. Nuclear Regulatory Commission. ADAMS Accession No. ML103490041. 2010a.
- 33 _____. NUREG-1536, "Standard Review Plan for Spent Fuel Dry Storage Systems at a
34 General License Facility." Rev. 1. Washington, DC: U.S. Nuclear Regulatory Commission.
35 2010b.
- 36 _____. "Outside Diameter Initiated Stress Corrosion Cracking Revised Final White Paper."
37 PA-MS-0474." Letter (October 14) to NRC From M.L. Arey, Jr. (PWROG Owners Group).

1 Washington, DC: U.S. Nuclear Regulatory Commission. ADAMS Accession
2 No. ML110400241. 2010c.

3 _____. Regulatory Guide 1.54, "Service Level I, II, and III Protective Coatings Applied to
4 Nuclear Power Plants," Rev. 2. Washington, DC: U.S. Nuclear Regulatory Commission.
5 2010d.

6 _____. "Failure of Control Rod Drive Mechanism Lead Screw Male Coupling at a Babcock and
7 Wilcox-designed Facility." NRC Information Notice 2007-02. Washington, DC: U.S. Nuclear
8 Regulatory Commission. ADAMS Accession No. ML070100459. 2007.

9 _____. NUREG-1567, "Standard Review Plan for Spent Fuel Dry Storage Facilities," Rev. 0.
10 Washington, DC: U.S. Nuclear Regulatory Commission. ADAMS Accession No.
11 ML003686776. 2000.

12 _____. "ECCS Suction Header Leaks Result in Both ECCS Trains Inoperable and TS 3.0.3
13 Entry." Licensee Event Report 1999-003-00. ADAMS Legacy Library Accession
14 No. 9905130085. Washington, DC: U.S. Nuclear Regulatory Commission. April 1999.

15 Nuclear Decommissioning Authority. "Literature Review of Atmospheric Stress Corrosion
16 Cracking of Stainless Steels Report to Nirex." Report No. NR3090/043.
17 Cumbria, United Kingdom: Nuclear Decommissioning Authority. 2007.

18 NWTRB. "Evaluation of the Technical Basis for Extended Dry Storage and Transportation of
19 Used Nuclear Fuel." Washington, DC: Nuclear Waste Technical Review Board. 2010.

20 Odette, G.R. and G.E. Lucas. "Embrittlement of Nuclear Reactor Pressure Vessels." *Journal of*
21 *Metals*. Vol. 53, Issue 7. pp.18-22. 2001.

22 Olender, A., J. Gorman, C. Marks, and G. Ilevbare. "Recent Operating Experience Issues with
23 17-4 PH in LWRs." Fontevraud 8: Conference on Contribution of Materials Investigations and
24 Operating Experience to LWRs' Safety, Performance and Reliability. France. 2015.

25 Parra, A., J. Carpio, and L. Martinez. "Microbial Corrosion of Metals Exposed to Air in Tropical
26 Marine Environments." *Materials Performance*. Vol. 35. pp. 44-50. 1996.

27 Phull, B. "Evaluating Stress corrosion Cracking." In ASM Handbook, Vol. 13A, *Corrosion:*
28 *Fundamentals, Testing, and Protection*. Materials Park, Ohio: ASM International.
29 pp. 575-616. 2003a.

30 _____. "Evaluating Uniform Corrosion." In ASM Handbook, Vol. 13A, *Corrosion:*
31 *Fundamentals, Testing, and Protection*. Materials Park, Ohio: ASM International. pp. 542-544.
32 2003b.

33 Pourbaix, M. *Atlas of Electrochemical Equilibria in Aqueous Solutions*. 2nd ed.
34 Houston, Texas: NACE. 1974.

35 Revie, R.W. *Uhlig's Corrosion Handbook*. Second Edition. Hoboken, New Jersey:
36 John Wiley and Sons. 2000.

- 1 Rowcliffe, A.F., L.K. Mansur, D.T. Hoelzer, and R.K. Nanstad. "Perspectives on Radiation
2 Effects in Nickel-Base Alloys for Applications in Advanced Reactors." *Journal of Nuclear*
3 *Materials*. Vol. 392. pp. 341–352. 2009.
- 4 Rozenfeld, I.L. "Atmospheric Corrosion of Metals." Houston, Texas: NACE. 1972.
- 5 Sachs, K. and D.G. Evans. "The Relaxation of Bolts at High Temperatures." Report C364/73.
6 Wolverhampton, United Kingdom: GKN Group Technological Center. 1973.
- 7 Samuels, I.E. *Metals Engineering: A Technical Guide*. Metals Park, Ohio: ASM International.
8 1988.
- 9 Shirai, K., J. Tani, T. Arai, M. Wataru, H. Takeda, and T. Saegusa. "SCC Evaluation Test of a
10 Multi-Purpose Canister." Presentation at the *13th International High-Level Radioactive Waste*
11 *Management Conference*, Albuquerque, New Mexico, April 10–14, 2011. LaGrange Park,
12 Illinois: American Nuclear Society. 2011.
- 13 Sindelar, R.L., A.J. Duncan, M.E. Dupont, P.-S. Lam, M.R. Louthan, Jr., and T.E. Skidmore.
14 NUREG/CR-7116, "Materials Aging Issues and Aging Management for Extended Storage and
15 Transportation of Spent Nuclear Fuel." Washington, DC: U.S. Nuclear Regulatory Commission.
16 2011.
- 17 Summerson, T.J., M.J. Pryor, D.S. Keir, and R.J. Hogan. "Pit Depth Measurements as a Means
18 of Evaluating the Corrosion Resistance of Aluminum in Seawater." ASTM STP 196.
19 pp. 157–175. West Conshohocken, Pennsylvania: ASTM International. 1957.
- 20 Tani, J.I., M. Mayuzurmi, and N. Hara. "Initiation and Propagation of Stress Corrosion Cracking
21 of Stainless Steel Canister for Concrete Cask Storage of Spent Nuclear Fuel." *Corrosion*.
22 Vol. 65, No. 3. pp. 187–194. 2009.
- 23 Tator, K.B. "Degradation of Protective Coatings." *Corrosion: Materials*. Vol 13B.
24 ASM Handbook. ASM International. pp. 589–599. 2005.
- 25 Tracy, A.W. "Effect of Natural Atmospheres on Copper Alloys: 20 Year Test." *Atmospheric*
26 *Corrosion of Nonferrous Metals*. ASTM-STP 175. 67p. West Conshohocken, Pennsylvania:
27 ASTM International. 1955.
- 28 Vargel, C. *Corrosion of Aluminum*. San Diego, California: Elsevier, Inc. 2004.
- 29 van Bodegom, L., K. van Gelder, M.K.F. Paksa, and L. van Raam. "Effect of Glycol and
30 Methanol on CO₂ Corrosion of Carbon Steel." *Proceeding of CORROSION Conference*.
31 Paper No. 55. Houston, Texas: NACE International. 1987.
- 32 Walch, M. and R. Mitchell. "The Role of Microorganisms in Hydrogen Embrittlement of Metals."
33 *Proceeding of CORROSION Conference*. Paper No. 249. Houston, Texas:
34 NACE International. 1983.
- 35 Was, G.S., J. Busby, and P.L. Andresen. "Effect of Irradiation on Stress corrosion Cracking and
36 Corrosion in Light Water Reactors." In ASM Handbook, Vol. 13C, *Corrosion: Environments and*
37 *Industries*. Materials Park, Ohio: ASM International. pp. 386–414. 2006.

- 1 West, G.A. and C.D. Watson. "Gamma Radiation Damage and Decontamination Evaluation of
- 2 Protective Coatings and Other Materials for Hot Laboratory and Fuel Processing Facilities."
- 3 ORNL-3589. Oak Ridge, Tennessee: Oak Ridge National Laboratory. 1965.

1 **3.3 Neutron shielding materials**

2 Neutron shielding typically is provided by either borated or nonborated polymeric or
3 cementitious materials. Hydrogen and oxygen reduce the energy of the neutrons such that the
4 neutrons are more effectively absorbed by the boron. The degradation and possible relocation
5 of shielding materials may be mitigated by encasing or reinforcing materials. For example,
6 shielding is often cast within a metal liner, which prevents ingress of water and contaminants.
7 Also, some shielding materials include reinforcements (e.g., fiberglass) for stability.

8 A set of known aging mechanisms with the potential to affect the performance of shielding
9 materials was identified from reviews of a range of information; sources of the information
10 include gap assessments for DSSs, relevant technical literature, and operating experience from
11 nuclear applications (NRC, 2014a, 2010; Chopra et al., 2014; Hanson et al., 2012;
12 Sindelar et al., 2011; NWTRB, 2010; EPRI, 2011). These mechanisms, which are induced by
13 thermal and irradiation conditions, include boron depletion, thermal aging, and radiation
14 embrittlement. Detailed discussions regarding each of these aging mechanisms follow.

15 **3.3.1 Neutron-shielding materials**

16 Polymer based

17 The TN-32 and TN-68 systems use both a borated polyester resin and polypropylene for
18 shielding, while Holtec's HI-STAR overpack and HI-TRAC transfer cask use Holtite-A.TM
19 Holtite-ATM is a composite material consisting of an epoxy polymer, boron carbide powder, and
20 aluminum hydroxide.

21 Cement based

22 The cementitious BISCO NS-3 material is used in one of the NUHOMS transfer cask designs for
23 neutron shielding. The structural concrete used to construct overpacks also serves as neutron
24 and gamma shielding; the degradation of such concrete is discussed separately in Section 3.5.

25 *3.3.1.1 Boron depletion (borated materials)*

26 The boron concentration in the neutron shields decreases as boron atoms in the borated
27 materials absorb neutrons. Boron-10 nuclei capture neutrons, yielding excited Boron-11 nuclei,
28 which in turn decay into high-energy alpha particles and Lithium-7 nuclei. The neutron shielding
29 material will lose one boron-10 atom per such a reaction. Significant depletion of boron-10
30 atoms may occur over time, if the shielding material is exposed to sufficient neutron fluence.

31 The NRC reviewer should ensure that the applicant provides a bounding analysis to show that
32 boron-10 depletion is not a credible aging mechanism for its specific DSS design. The reviewer
33 should review any boron depletion analyses contained in the applicant's original design-bases
34 documents, if present, to determine whether the design-basis analysis or license renewal
35 application adequately addresses the implications of extending the operating period to 60 years.
36 This reexamination of the original analyses would typically be defined as TLAAs in the renewal
37 application. The staff's guidance for the review of TLAAs is provided in NUREG-1927,
38 Revision 1 (NRC, 2016). If the original design basis does not include an analysis for loss of
39 boron-10, the reviewer nevertheless should ensure that the renewal application adequately
40 addresses this aging mechanism.

1 Rather than demonstrating performance through an analysis, an applicant may choose to
2 manage loss of neutron shielding, such as through radiation monitoring, to confirm the
3 shielding's continued effectiveness. In that case, the reviewer should refer to NRC guidance on
4 the review of AMPs in NUREG-1927, Revision 1.

5 *3.3.1.2 Thermal aging*

6 Polymers may be susceptible to heat-induced changes to material properties and configuration
7 due to a number of mechanisms. At elevated temperatures, the long chain backbone of a
8 polymer can undergo molecular scission (breaking) and cross linking. Also, gaseous products
9 may be formed, including H₂, CH₄, and CO₂. These reactions may cause embrittlement,
10 shrinkage, decomposition, and changes in physical configuration (e.g., loss of hydrogen or
11 water) (EPRI, 2002; McManus and Chamis, 1996). Shrinkage and embrittlement can locally
12 displace shielding material and potentially diminish shielding effectiveness, although this may be
13 mitigated in part by reinforcement materials within the polymer matrix and the support provided
14 by the encasing metal. Because many polymers are known to degrade at elevated
15 temperatures, thermal aging for polymer-based neutron-shielding materials is a credible aging
16 mechanism. Therefore, either a supporting analysis for the material's continued use or an AMP
17 is required during the 60-year timeframe.

18 The cementitious BISCO NS-3 shielding material used in one of the NUHOMS transfer cask
19 designs may experience some loss of hydrogen (neutron moderator) when exposed to elevated
20 temperatures. However, the material is subjected to elevated temperatures only during
21 relatively brief periods when the storage canister is being transported from the spent fuel pool to
22 the storage pad. Thus, the time of thermal exposure in the transfer cask is minimal compared to
23 the continuous thermal exposure NS-3 experiences in other NRC-approved applications
24 (e.g., the MC-10 metal storage cask) (NRC, 2005). As a result, thermal aging of the NS-3
25 shielding material is not considered to be a credible aging mechanism in the transfer cask, and
26 therefore, aging management is not required during the 60-year timeframe.

27 *3.3.1.3 Radiation embrittlement*

28 Similar to the thermal aging mechanism discussed above, radiation can alter polymer structures
29 by molecular scission and cross linking to reduce ductility, fracture toughness, and resistance to
30 cracking (Fu, et al., 1988; Cota, et al., 2007). For example, the threshold for radiation
31 embrittlement has been found to be about 10⁶ rad for polyethylene and significantly lower for
32 other polymers, such as polytetrafluoroethylene (EPRI, 1998). Depending on the DSS design
33 and the specific SNF, this dose can be reached in 10–100 years (EPRI, 1998). Embrittlement
34 can locally displace shielding material and potentially reduce shielding effectiveness, although
35 this may be mitigated, in part, by reinforcement materials within the polymer matrix and the
36 support provided by the encasing metal. As a result, radiation embrittlement of polymer-based
37 neutron-shielding materials is a credible aging mechanism, and therefore, either a supporting
38 analysis for the material's continued use or an AMP is required during the 60-year timeframe.
39 An acceptable AMP may include monitoring and trending of radiation dose to confirm the
40 absence of an decreasing trend in shielding effectiveness.

41 An analysis of the effects of radiation on the shielding properties of BISCO NS-3 has shown that
42 both the gamma and neutron radiation dose the shielding material receives over 60 years in the
43 NUHOMS transfer cask are several orders of magnitude below the material's exposure limit
44 (BISCO, 1986; NRC, 2014b). As a result, radiation embrittlement of the NS-3 shielding material

1 is not considered to be a credible aging mechanism, and therefore, aging management is not
2 required during the 60-year timeframe.

3 3.3.2 References

4 BISCO Products, Inc. "NS-3 Specification Sheet." (ADAMS ML110730731), June 23, 1986.

5 Chopra, O., D. Diercks, R. Fabian, Z. Han, and Y. Liu. "Managing Aging Effects on Dry Cask
6 Storage Systems for Extended Long-Term Storage and Transportation of Used Fuel."
7 FCRD-UFD-2014-000476. ANL-13/15, Rev. 2. Washington, DC: U.S. Department of
8 Energy. 2014.

9 Cota, S.S., V. Vasconcelos, M. Senne, Jr., L.O.L. Carvalho, D.B. Rezende, and R.F. Cõrrea.
10 "Changes in Mechanical Properties Due to Gamma Irradiation of High-Density
11 Polyethylene." *Brazilian Journal of Chemical Engineering*. Volume 24, No. 02. pp. 259-265.
12 2007.

13 EPRI. "Extended Storage Collaboration Program (ESCP) Progress Report and Review of Gap
14 Analyses." Report 1022914. Palo Alto, California: Electric Power Research Institute. 2011.

15 _____. "Technical Bases for Extended Dry Storage of Spent Nuclear Fuel." Report 1003416.
16 Palo Alto, California: Electric Power Research Institute. 2002.

17 _____. "Data Needs for Long-Term Dry Storage of LWR Fuel." Report TR-108757.
18 Palo Alto, California: Electric Power Research Institute. 1998.

19 Fu, L., R.A. Fouracre, and H.M. Banford. "An Investigation of Radiation Damage in Cured
20 Epoxy Resin System Using Regression Experiment Design, Electrical Insulation and Dielectric
21 Phenomena." *1988 Annual Report, Conference on Electrical Insulation and Dielectric
22 Phenomena*. IEEE Dielectrics and Electrical Insulation Society. 1988.

23 Hanson, B., H. Alsaed, C. Stockman, D. Enos, R. Meyer, and K. Sorenson. "Used Fuel
24 Disposition Campaign: Gap Analysis to Support Extended Storage of Used Nuclear Fuel,
25 Rev. 0." Richland, Washington: Pacific Northwest National Laboratory. 2012.

26 McManus, H.L. and C.C. Chamis. "Stress and Damage in Polymer Matrix Composite Materials
27 Due to Material Degradation at High Temperatures." NASA Technical Memorandum 4682.
28 Cambridge, Massachusetts: Massachusetts Institute of Technology. 1996.

29 NRC. NUREG-1927, "Standard Review Plan for Renewal of Specific Licenses and Certificates
30 of Compliance for Dry Storage of Spent Nuclear Fuel." Revision 1. ADAMS Accession
31 No. ML16179A148. Washington, DC: U.S. Nuclear Regulatory Commission. 2016.

32 _____. "Identification and Prioritization of the Technical Information Needs Affecting Potential
33 Regulation of Extended Storage and Transportation of Spent Nuclear Fuel." Washington, DC:
34 U.S. Nuclear Regulatory Commission. 2014a.

35 _____. "Safety Evaluation Report for License Renewal: Calvert Cliffs Nuclear Power Plant
36 Independent Spent Fuel Storage Installation." Washington, DC: U.S. Nuclear Regulatory
37 Commission. 2014b.

- 1 _____ . NUREG-1801, "Generic Aging Lessons Learned (GALL) Report." Rev. 2.
2 Washington, DC: U.S. Nuclear Regulatory Commission. 2010.
- 3 _____ . "Safety Evaluation Report for License Renewal: Surry Independent Spent Fuel Storage
4 Installation." Washington, DC: U.S. Nuclear Regulatory Commission. ADAMS Accession No.
5 ML050590266. 2005.
- 6 NWTRB. "Evaluation of the Technical Basis for Extended Dry Storage and Transportation of
7 Used Nuclear Fuel." Washington, DC: Nuclear Waste Technical Review Board. 2010.
- 8 Sindelar, R.L., A.J. Duncan, M.E. Dupont, P.-S. Lam, M.R. Louthan, Jr., and T.E. Skidmore.
9 NUREG/CR-7116, "Materials Aging Issues and Aging Management for Extended Storage and
10 Transportation of Spent Nuclear Fuel." Washington, DC: U.S. Nuclear Regulatory Commission.
11 2011.

1 **3.4 Neutron poison materials**

2 Subcriticality of the SNF in DSSs may be maintained, in part, by the placement of neutron
3 absorbers, or poison plates, around the fuel assemblies. Commonly used neutron poisons
4 include borated stainless steel, borated aluminum alloys, aluminum metal-matrix composites
5 such as Metamic™ and Boralyn®, and aluminum-boron carbide laminate composites, commonly
6 referred to as cermet, such as Boral®. These materials are exposed to helium environments,
7 where temperature and radiation levels are expected to be high because of their proximity to the
8 fuel assemblies. This environment also could include small amounts of water left after the
9 drying operations.

10 A list of known aging mechanisms that have the potential to affect the performance of neutron
11 poison plates was identified from reviews of a range of information sources, including gap
12 assessments for DSSs, relevant technical literature, and operating experience from nuclear and
13 nonnuclear applications (NRC, 2014, 2010; Chopra et al., 2014; Hanson et al., 2012;
14 Sindelar et al., 2011; NWTRB, 2010). These mechanisms, which are induced by various
15 physicochemical, thermal-mechanical, and irradiation conditions, include general corrosion,
16 galvanic corrosion, wet corrosion and blistering, creep, thermal aging, radiation embrittlement,
17 and boron depletion.

18 **3.4.1 Borated stainless steel**

19 The Type 304 borated stainless steels used as neutron poison plates are similar in composition
20 to standard Type 304 stainless steels used in other engineering applications, except that the
21 borated steels contain boron, which has a much higher thermal neutron absorption cross
22 section. ASTM A887–89 defines eight types of borated stainless steels (304B and 304B1–
23 304B7) with natural boron concentrations (including both B-10 and B-11 isotopes) ranging from
24 0.2 to 2.25 weight percent (ASTM International, 2009). Boron is essentially insoluble in
25 stainless steel, and thus it is present as iron and chromium borides (Fe₂B, Cr₂B) in a steel
26 matrix.

27 Of the identified aging mechanisms for neutron poison plates discussed in Section 3.4 above,
28 the following were removed from consideration for aging effects of borated stainless steels,
29 because they were determined not to be reasonably credible: (i) general corrosion, (ii) galvanic
30 corrosion and (iii) wet corrosion and blistering. The technical justifications for the decisions to
31 eliminate these aging mechanisms follow.

- 32 • General corrosion: Similar to other austenitic stainless steel alloys, borated stainless
33 steel exhibits passive behavior in helium environments, and thus, general corrosion
34 rates are expected to be negligible.
- 35 • Galvanic corrosion: Borated stainless steel could be coupled to steel, aluminum, or
36 other stainless steel alloys. The galvanic corrosion behavior of stainless steel is
37 complicated by the fact that its relative nobility with respect to other materials may
38 depend on whether a passivating oxide film is present. Nevertheless, both passivated
39 and nonpassivated stainless steels are generally more noble than steel and aluminum
40 (Baboian, 2003). In addition, there is no aqueous electrolyte inside the cask or canister
41 to support galvanic corrosion in the helium environment.

42

1 • Wet corrosion and blistering: Because borated stainless steel is solid without porosity,
2 no water can be trapped inside the material. Thus, wet corrosion and blistering are not
3 considered to be credible.

4 More detailed discussions regarding the other aforementioned potential aging mechanisms for
5 borated stainless steel are provided below.

6 3.4.1.1 *Boron depletion*

7 Boron depletion in boron-based neutron poison plates refers to the loss of boron and hence the
8 loss of the neutron-absorbing capacity of a material when it is exposed to neutron fluence. For
9 example, under a neutron fluence, boron-10 nuclei capture neutrons, yielding excited Boron-11
10 nuclei, which in turn decay into alpha particles and Lithium-7 nuclei. In this nuclear reaction,
11 one neutron absorption reaction results in the loss of one boron-10 atom. Significant depletion
12 of boron-10 atoms may occur if the poison material is exposed to sufficient neutron fluence.

13 Borated stainless steel typically has an areal density of 10^{19} to 10^{21} boron-10 atoms/cm²
14 [6.5×10^{19} to 10^{21} boron-10 atoms/in²] (EPRI, 2009). The boron areal density can reach this
15 level by adjusting the thickness of the poison plate, by adjusting the weight fraction of added
16 boron, and through the use of enriched boron (i.e., boron-10) (EPRI, 2009). A neutron flux of
17 10^4 – 10^6 n/cm²-s [6.5×10^4 – 6.5×10^6 n/in²-s] is typical for dry cask storage (Sindelar et al.,
18 2011). At a typical neutron flux and boron-10 concentration, the neutron poison plates would
19 deplete at most 0.0002 percent of the available boron-10 atoms after 60 years of storage.
20 Using the highest expected neutron flux and the lowest boron-10 concentration as a most
21 conservative scenario, only 0.02 percent of the available boron-10 atoms would be depleted
22 after 60 years, an amount too small to decrease the criticality control function of the neutron-
23 absorbing materials. As such, boron depletion is not considered to be credible, and therefore,
24 aging management is not required during the 60-year timeframe.

25 Although boron depletion in borated stainless steel is not generally considered to be a credible
26 aging mechanism, the reviewer nevertheless should ensure that the renewal application
27 addresses any depletion analyses that exist in the original design basis to consider the
28 implication of extending the operating period to 60 years. Staff guidance for the review of such
29 TLAs is provided in NUREG–1927.

30 3.4.1.2 *Creep*

31 As discussed in Section 3.2.1.6, as a general rule of thumb, significant creep can occur at
32 temperatures above $0.4T_m$, where T_m is the melting point of the metal in Kelvin (Cadek, 1988).
33 At these temperatures, plastic deformation or distortion can occur over long times, even under
34 stresses that normally would not be considered sufficient to cause yielding of the material.
35 Robino and Cieslak (1997) show that borated stainless steel has a melting range of
36 1,250–1,340 degrees C [2,282–2,444 degrees F], corresponding to the melting of borides and
37 the austenitic structure, which is slightly lower than standard nonborated stainless steel.
38 Applying the $0.4T_m$ rule, a temperature range of 336–372 degrees C [637–702 degrees F] is
39 required to initiate significant creep in borated stainless steels, which is below the estimated
40 peak fuel cladding temperature of 400 degrees C [752 degrees F] at the beginning of the
41 storage period (Jung et al, 2013). The maximum cladding temperature is estimated to drop
42 below the creep range (336 degrees C [637 degrees F]) in fewer than 9 years, well before the
43 period of extended operation. Also, the borated stainless steel poison plates, which are used in
44 the verticle DSSs, are not expected to be under loads other than their own weight, and in many

1 instances, their weight is also supported by adjacent structures. As such, creep of borated
2 stainless steel is not considered to be credible, and therefore, aging management is not
3 required during the 60-year timeframe.

4 3.4.1.3 Thermal aging

5 As previously discussed in Section 3.2.2.8, the microstructures of most stainless steels will
6 change, given sufficient time at elevated temperatures, and this can affect its mechanical
7 properties. The thermal aging resistance highly depends on material chemical composition and
8 microstructure. Borated stainless steel alloys consist of $(\text{Fe,Cr})_2\text{B}$ precipitates dispersed in an
9 austenite stainless steel matrix. Robino and Cieslak (1997) showed that the estimated peak fuel
10 cladding temperature of 400 degrees C [752 degrees F] in storage (Jung et al, 2013) is well
11 below the temperatures that are needed to cause a change in the boride precipitates. Also, as
12 discussed in Section 3.2.2.8, the austenite matrix is not expected to be susceptible to
13 microstructure changes until temperatures exceed 1,000 degrees C [1,832 degrees F]. As
14 such, thermal aging of borated stainless steel is not considered to be credible, and therefore,
15 aging management is not required during the 60-year timeframe.

16 3.4.1.4 Radiation embrittlement

17 Embrittlement of metals occurs when radiation displaces atoms in metal crystal structures,
18 creating defects. Neutron radiation (rather than gamma radiation) has the greatest potential to
19 cause this phenomenon. Depending on the neutron fluence, radiation can cause changes in
20 mechanical properties such as loss of ductility, fracture toughness, and resistance to cracking.

21 Neutron embrittlement effects on the mechanical properties and the microstructures of borated
22 stainless steel were studied by irradiating borated stainless steel to different radiation levels,
23 from 10^{13} to 10^{17} n/cm² [6.5×10^{13} to 10^{17} n/in²] (Soliman et al., 1991). Tests included samples
24 manufactured by both powder metallurgical and conventional wrought processes. The energy
25 of the neutron source was such that approximately 20 percent of the neutron flux had an energy
26 above 0.1 megaelectron-volt (MeV), meaning that a significant portion of the flux contained the
27 most damaging intermediate or fast neutrons. The investigators reported that there was almost
28 no change in mechanical properties with the fluence level up to 10^{17} n/cm² [6.5×10^{17} n/in²]. As
29 discussed in Section 3.2.1.9, for dry cask storage, the maximum potential accumulated neutron
30 fluence on DSS basket components after 100 years was calculated to be 2.63×10^{16} n/cm²
31 [1.70×10^{17} n/in²], which is about one order of magnitude below the level of that used in the
32 tests by Soliman et al. (1991). In addition, neutron flux decreases with time during storage,
33 which will limit the radiation effects. As such, radiation embrittlement of borated stainless steel
34 is not considered to be credible, and therefore, aging management is not required during the
35 60-year timeframe.

36 3.4.2 Borated aluminum alloys and aluminum-based composites

37 As in stainless steels, boron is essentially insoluble in aluminum. In borated aluminum, boron is
38 present in the form of aluminum or titanium boride precipitates (AlB_2 , TiB_2) that reside in an
39 aluminum matrix. In aluminum metal-matrix composites, boron is in the form of boron carbides
40 (B_4C) in an aluminum matrix. The laminate composites (e.g., Boral[®]) consist of (i) a core of
41 uniformly distributed boron carbide and aluminum alloy particles and (ii) a surface cladding of
42 aluminum alloy on both sides of the core.

1 Of the identified potential aging mechanisms for neutron poison plates listed in Section 3.4
2 above, wet corrosion and blistering are considered to be credible only for Boral[®], because only
3 this material has porosity that can trap water and initiate this mechanism. Detailed discussions
4 of all aging mechanisms for aluminum-based poison materials are provided below.

5 *3.4.2.1 General corrosion*

6 Because aluminum is present as a continuous matrix (borated aluminum and aluminum
7 metal-matrix composites) or used as an outer cladding (Boral[®]), the degree of general corrosion
8 of each of the neutron poison plate materials is considered to be largely governed by the
9 corrosion of aluminum. As discussed in Section 3.2.3.1 for other aluminum components,
10 aluminum forms a protective oxide film at temperatures below approximately 230 degrees C
11 [446 degrees F]. Above this temperature, the protective film no longer forms if water or steam is
12 present. As such, general corrosion of aluminum is possible if aluminum were exposed to
13 moisture in the internal helium environment. However, there is very little residual water in the
14 cask or canister internal environment following drying. Assuming a residual water content of 1 L
15 [0.26 gal], Jung et al. (2013) calculated that oxidation of all aluminum in the basket assembly is
16 limited to 0.54 g [0.019 oz], which is equivalent to a 2- μ m [0.079-mils]-thick layer of aluminum
17 over a surface area of 1,000 cm² [155 in²]. Thus, the potential for material thinning from
18 oxidation is a very small fraction of the aluminum poison materials used inside the system. As a
19 result, general corrosion is not considered to be credible, and therefore, aging management is
20 not required during the 60-year timeframe.

21 *3.4.2.2 Galvanic corrosion*

22 Galvanic corrosion occurs when two dissimilar metals or conductive materials are in physical
23 contact in the presence of a conducting solution (Baboian, 2003; Hack, 1993). The
24 aluminum-based neutron poison materials used inside DSSs can be in galvanic contact with
25 stainless steel, where aluminum is less noble.

26 As discussed above in the evaluation of general corrosion, there is very little residual water
27 within a cask or canister following drying. Thus, there is a limited potential for the presence of a
28 conducting solution that can support galvanic corrosion. As a result, loss of material due to
29 galvanic corrosion is not considered to be credible, and therefore, aging management is not
30 required during the 60-year timeframe.

31 *3.4.2.3 Wet corrosion and blistering*

32 The core of aluminum-boron carbide laminate composites is not fully sintered and, as a result,
33 can have a porosity of 1 to 8 percent with varying degrees of interconnectivity among pores.
34 This may allow water ingress into the core, where the water can react with the aluminum to form
35 aluminum oxide and hydrogen gas (EPRI, 2009; 2012). Blistering has been observed in the
36 Boral[®] cladding in wet and dry storage applications. Tests simulating the wetting and vacuum
37 drying cycles during canister closure operations show that Boral[®] can form blisters in the
38 aluminum cladding because of water ingress through its exposed edges (EPRI, 2004). The
39 blisters are characterized by a local area where the aluminum cladding separates from the
40 underlying boron carbide-aluminum core, and the cladding is physically deformed outward.

41 Although wet corrosion and blistering may occur, this aging mechanism has not been observed
42 to reduce the neutron absorbing capability of Boral[®] in spent fuel pool surveillance coupons
43 (EPRI, 2009). It is important to note that, because only a trace amount of water will be left in a

1 dry storage cask after dehydration and helium backfill, the occurrence of wet corrosion and
2 blistering will be minimal in a dry cask environment during the period of extended operation.
3 Therefore, wet corrosion and blistering are not considered to be an aging mechanism requiring
4 aging management and aging management is not required for Boral® in the DSSs with respect
5 to criticality safety during the 60-year timeframe.

6 3.4.2.4 *Boron depletion*

7 Boron depletion refers to the loss of the capability of a material to absorb neutrons when the
8 neutron fluence significantly consumes boron-10 atoms. Neutron poison plates typically contain
9 10^{19} to 10^{21} boron-10 atoms/cm² [6.5×10^{19} to 10^{21} boron-10 atoms/in²] (EPRI, 2009). A neutron
10 flux of 10^4 – 10^6 n/cm²-s [6.5×10^4 – 6.5×10^6 n/in²-s] is typical for dry cask storage (Sindelar et
11 al., 2011). Under a neutron flux, boron-10 nuclei capture neutrons, yielding excited boron-11
12 nuclei, which, in turn, decay into high-energy alpha particles and lithium-7 nuclei. In this nuclear
13 reaction, one neutron would deplete one boron-10 atom. At typical levels of neutron flux and
14 boron-10 concentration, the neutron dose after 60 years would deplete at most 0.0002 percent
15 of the available boron-10 atoms. Using the highest expected neutron flux and the lowest boron-
16 10 concentration as a worst case scenario, only 0.02 percent of the available boron-10 atoms
17 would be depleted after 60 years, which is too small to challenge the criticality control function of
18 the neutron poisons. As such, boron depletion for borated aluminum alloys, aluminum metal
19 matrix composites, and Boral® is not expected to result in significant changes in the criticality
20 control function. As such, boron depletion is not considered to be credible, and therefore, aging
21 management is not required during the 60-year timeframe.

22 Although the above generic evaluation does not identify boron depletion as a significant aging
23 mechanism, the reviewer nevertheless should ensure that the renewal application addresses
24 any depletion analyses that exist in the original design basis to consider the implication of
25 extending the operating period to 60 years. Staff guidance for the review of such TLAAAs is
26 provided in NUREG–1927.

27 3.4.2.5 *Creep*

28 As discussed in Section 3.2.1.6, as a general rule of thumb, significant creep occurs at
29 temperatures above $0.4T_m$, where T_m is the melting point of the metal in Kelvin (Cadek, 1988).
30 At these temperatures, plastic deformation or distortion can occur over long times, even under
31 stresses that normally would not be considered sufficient to cause yielding of the material.
32 Because aluminum is present as a continuous matrix and as an external cladding in the neutron
33 poison plates, and aluminum has a lower melting point than the other portions of the material
34 microstructures (e.g., aluminum and titanium borides, boron carbides), the creep behavior of
35 poison materials is considered to be governed by the behavior of aluminum. Applying the $0.4T_m$
36 rule, the critical creep temperature for aluminum is 100 degrees C [212 degrees F].

37 The highest temperatures within DSSs are at locations close to the fuel rods. For example, the
38 maximum expected temperature of the cladding on the fuel rods has been estimated to be
39 400 degrees C [752 degrees F] at the beginning of the storage period, and the cladding
40 temperatures are expected to decrease to approximately 266 degrees C [510 degrees F] after
41 20 years and 127 degrees C [261 degrees F] after 60 years (Jung et al., 2013). These
42 estimates depend on many factors, such as the initial heat load of the SNF. It is apparent from
43 these temperatures that subcomponents within the cask or canister could be exposed to
44 temperatures above the minimum creep temperatures for aluminum during at least the first
45 40 years.

1 Because temperatures within DSSs have the potential to exceed the minimum creep
2 temperature of aluminum, it is necessary to consider the load applied to the subcomponent to
3 determine whether significant creep deformation will occur, as well as the specific application to
4 determine whether the creep affects safety. Typically, neutron poison plates do not serve a
5 structural function and are thus not expected to be under loads other than their own weight.
6 Also, in many instances, their weight is also supported by adjacent structures. For example, the
7 neutron poison plates in the Holtec HI-STORM 100 system are completely enclosed in stainless
8 steel sheathing (Holtec International, 2014). Due to the minimal applied loads and presence of
9 adjacent supporting structures, the impact of creep on the criticality control function of the
10 neutron poisons is not considered to be credible, and therefore, aging management is not
11 required during the 60-year timeframe.

12 3.4.2.6 *Thermal aging*

13 Prolonged exposure to elevated temperatures can lead to a loss of fracture toughness and
14 ductility in some materials as a result of changes to their microstructure. Testing of
15 aluminum-based neutron poison plates, however, has shown that these materials typically
16 increase in ductility when they are aged at high temperatures. For example, a series of
17 elevated temperature tensile tests on an aluminum metal-matrix composite (METAMIC™) found
18 an increase in elongation to break (a measure of ductility) when the material was aged at
19 399 degrees C [750 degrees F] for 8,523 hours (EPRI, 2009). These and other material
20 qualification tests performed on neutron poisons demonstrate that microstructural changes
21 induced by aging typically make the aluminum softer and more ductile as it is annealed, while
22 the boride and carbide particulates are thermally stable at cask internal temperatures.

23 Also, as discussed above for the creep mechanism, decreases in strength due to thermal aging
24 are not expected to affect the criticality control function of the poison plates, because they
25 typically do not serve a structural function and may be supported by adjacent structures.
26 Consequently, thermal aging of neutron poison materials is not considered to be credible, and
27 therefore, aging management is not required over the 60-year timeframe.

28 3.4.2.7 *Radiation embrittlement*

29 As discussed in Section 3.4.1.4 above, embrittlement of metals may occur under exposure to
30 radiation. Neutron radiation (rather than gamma radiation) has the greatest potential to cause
31 this phenomenon.

32 Depending on the neutron fluence, radiation can cause changes in mechanical properties such
33 as loss of ductility, fracture toughness, and resistance to cracking. Farrell and King (1973)
34 showed that pure aluminum had increased strength but decreased ductility after being irradiated
35 to fast neutron fluences (energy greater than 0.1 MeV) in the range of 1 to 3×10^{22} n/cm²
36 [6.5 to 19.4×10^{22} n/in²] from a research reactor for 8 years. However, these radiation levels
37 are six orders of magnitude higher than the estimated fluence after dry storage for 100 years as
38 discussed in Section 3.2.1.9.

39 Some results from radiation testing of aluminum-based neutron poisons are reported in the
40 literature (EPRI, 2009). Gamma, thermal neutron, and fast neutron radiation testing of Boral® in
41 water was performed for 9 years. With exposures of to up to 7×10^{11} rad of gamma,
42 3.6×10^{18} n/cm² [2.3×10^{19} n/in²] fast neutron fluence, and 2.7×10^{19} n/cm² [1.7×10^{20} n/in²]
43 thermal neutron fluence, the specimen showed no change in ultimate strength and no other
44 signs of physical deterioration, except for severe oxidation because of the presence of water.

1 Also, radiation testing of a sintered composite subjected to up to 1.5×10^{20} n/cm²
2 [9.7×10^{20} n/in²] fast neutron fluence and a maximum of 3.8×10^{11} rad gamma exposure
3 showed little change in the yield strength and ultimate strength (EPRI, 2009). Finally, neutron
4 radiation of borated aluminum to fluences of 10^{17} n/cm² [6.5×10^{17} n/in²] showed no dimensional
5 change or radiation damage (EPRI, 2009). These test conditions are expected to be more
6 severe than those experienced by the aluminum-based neutron poison materials in the
7 extended storage application (EPRI, 2009). Therefore, radiation embrittlement of borated
8 aluminum alloys, aluminum metal-matrix composites, and Boral[®] is not expected to be credible.
9 Consequently, aging management is not required during the 60-year timeframe.

10 **3.4.3 References**

- 11 Alexander, D.J. "Effects of Irradiation on the Mechanical Properties of 6061-T651
12 Aluminum Base Metal and Weldments." ASTM Special Technical Publication. Vol. 1325.
13 pp. 1,027–1,044. 1999.
- 14 ASTM International. "Standard Specification for Borated Stainless Steel Plate,
15 Sheet, and Strip for Nuclear Application." ASTM A887–89. West Conshohocken,
16 Pennsylvania: ASTM International. 2009.
- 17 Baboian, R. "Galvanic Corrosion." In ASM Handbook, Vol. 13A, *Corrosion: Fundamentals,*
18 *Testing, and Protection*. Materials Park, Ohio: ASM International. pp. 210–213. 2003.
- 19 Cadek, J. *Creep in Metallic Materials*. Elsevier Science Publishing Company, Inc. 1988.
- 20 Chopra, O., D. Diercks, R. Fabian, Z. Han, and Y. Liu. "Managing Aging Effects on Dry Cask
21 Storage Systems for Extended Long-Term Storage and Transportation of Used Fuel."
22 FCRD–UFD–2014–000476. ANL–13/15, Rev. 2. Washington, DC: U.S. Department of
23 Energy. 2014.
- 24 EPRI. "Strategy for Managing the Long-Term Use of BORAL[®] in Spent Fuel Storage Pools."
25 Report 1025204. Palo Alto, California: Electric Power Research Institute. 2012.
- 26 _____. "Handbook of Neutron Absorber Materials for Spent Nuclear Fuel Transportation and
27 Storage Applications." Report 1019110. Palo Alto, California: Electric Power Research
28 Institute. 2009.
- 29 _____. "BORAL[®] Behavior Under Simulated Cask Vacuum Drying. Part 2 Test Results."
30 Report 1009696. Palo Alto, California: Electric Power Research Institute. 2004.
- 31 Farrell, K. and R.T. King. "Radiation-Induced Strengthening and Embrittlement in Aluminum."
32 *Metallurgical Transactions A. Physical Metallurgy and Materials Science*. Vol. 4.
33 pp. 1,223–1,231. 1973.
- 34 Gibeling, J.C. "Creep Deformation of Metals, Polymers, Ceramics, and Composites." In
35 ASM Handbook. Vol. 8. *Mechanical Testing and Evaluation*. Materials Park, Ohio:
36 ASM International. pp. 363–368. 2000.
- 37 Hack, H.P. *Galvanic Corrosion Test Methods*. Houston, Texas: NACE International. 1993.

- 1 Hanson, B., H. Alsaed, C. Stockman, D. Enos, R. Meyer, and K. Sorenson. "Gap Analysis to
2 Support Extended Storage of Used Nuclear Fuel." Rev. 0. FCRD–USED–2011–000136.
3 PNNL–20509. Washington, DC: U.S. Department of Energy. 2012.
- 4 Holtec International. "Final Safety Analysis Report for the HI-STORM 100 Cask System,
5 Revision 12." Holtec Report No. HI-2002444. USNRC Docket No. 72-1014. pp. 1.2-18.
6 ADAMS Accession No. ML14086A410. 2014
- 7 Jung, H., P. Shukla, T. Ahn, L. Tipton, K. Das, X. He, and D. Basu. "Extended Storage and
8 Transportation: Evaluation of Drying Adequacy." ADAMS Accession No. ML13169A039.
9 San Antonio, Texas: Center for Nuclear Waste Regulatory Analyses. 2013.
- 10 Kaufman, J.G. *Properties of Aluminum Alloys—Tensile, Creep, and Fatigue Data at High and*
11 *Low Temperatures*. Materials Park, Ohio: ASM International. 1999.
- 12 NRC. "Identification and Prioritization of the Technical Information Needs Affecting Potential
13 Regulation of Extended Storage and Transportation of Spent Nuclear Fuel." Washington, DC:
14 U.S. Nuclear Regulatory Commission. May 2014.
- 15 _____. NUREG–1801, "Generic Aging Lessons Learned (GALL) Report." Rev. 2.
16 Washington, DC: U.S. Nuclear Regulatory Commission. 2010.
- 17 NWTRB. "Evaluation of the Technical Basis for Extended Dry Storage and Transportation of
18 Used Nuclear Fuel." Washington, DC: Nuclear Waste Technical Review Board. 2010.
- 19 Robino, C.V. and M.J. Cieslak. "Fusion Welding of a Modern Borated Stainless Steel." *Welding*
20 *Journal*. Vol. 76, No. 1. pp. 11-s – 23-s. 1997.
- 21 Sindelar, R.L., A.J. Duncan, M.E. Dupont, P.-S. Lam, M.R. Louthan, Jr., and T.E. Skidmore.
22 NUREG/CR–7116, "Materials Aging Issues and Aging Management for Extended Storage and
23 Transportation of Spent Nuclear Fuel." Washington, DC: U.S. Nuclear Regulatory Commission.
24 2011.
- 25 Soliman, S.E., D.L. Youchison, A.J. Baratta, and T.A. Ballreht. "Neutron Effects on Borated
26 Stainless Steel." *Nuclear Technology*. Vol. 96. pp. 346–352. 1991.

27

1 **3.5 Concrete overpacks, support pads, and ceramic fiber insulation**

2 Concrete overpacks and support pads include various structural subcomponents constructed of
3 concrete and reinforcing steel, as well as pad-supporting materials constructed of engineered
4 fill, natural soil, or treated soil. These subcomponents may be exposed to several
5 environments, such as outdoor air, groundwater or soil, and flowing water, or they may be
6 sheltered or embedded in concrete or steel. The environment may also include elevated
7 temperatures due to heat released by the SNF and radiation, with dose rates depending on the
8 SNF characteristics (e.g., burnup and age of fuel), exposure time, and location of the
9 subcomponent.

10 Potential aging mechanisms for the concrete overpack and pad subcomponents were identified
11 from reviews of gap assessments of DSSs, relevant technical literature, American Concrete
12 Institute (ACI) guides and reports, and operating experience from nuclear and nonnuclear
13 applications (NRC, 2014, 2011a, 2010a; Chopra et al., 2014; Hanson et al., 2011;
14 NWTRB, 2010). Additional mechanisms were identified during a recent NRC concrete expert
15 panel workshop (NRC, 2015). Thermal, mechanical, chemical, and irradiation-induced
16 degradation mechanisms were identified as follows:

- 17 • freeze and thaw
- 18 • creep
- 19 • reaction with aggregates
- 20 • aggressive chemical attack
- 21 • corrosion of reinforcing steel
- 22 • differential settlement
- 23 • shrinkage
- 24 • leaching of calcium hydroxide
- 25 • radiation damage
- 26 • fatigue
- 27 • dehydration at high temperature
- 28 • microbiological degradation
- 29 • delayed ettringite formation
- 30 • salt scaling

31 In addition, a review of known degradation modes for ceramic fiber insulation was performed,
32 which resulted in consideration of the following:

- 33 • radiation damage
- 34 • moisture absorption

35 Potential mechanisms were refined by considering the thermal, mechanical, chemical, and
36 irradiation conditions specific to each subcomponent. This process eliminated several
37 mechanisms from consideration for some subcomponents in the AMR tables in Chapter 4.
38 Detailed discussions regarding potential aging mechanisms for each material and the technical
39 bases for those requiring aging management are included in the following sections.

40 These discussions do not consider potential synergistic effects, if any, due to coupled
41 degradation mechanisms. Coupled degradation mechanisms in concrete refer to degradation
42 modes that can interact, affecting their relative times for initiation and progression
43 (e.g., freeze-thaw cracking that leads to water ingress and subsequent leaching of calcium

1 hydroxide). Few in-depth studies have been published on the effects of concrete damage
2 caused by these potential coupled degradation mechanisms. However, the staff expects that an
3 AMP is an adequate approach for addressing potential synergistic effects due to coupled
4 degradation mechanisms. The example of an AMP for concrete structures in Chapter 6 relies
5 on the licensee's corrective action program to ensure that conditions that may lead to a loss of
6 intended function will be reviewed and dispositioned by trained personnel. If a particular aging
7 effect is detected, part of the licensee's corrective action may include a root-cause evaluation to
8 determine the cause of the aging effect. If the root-cause evaluation determines that the rate of
9 degradation is being accelerated by the effects of coupled degradation modes, followup
10 corrective actions may include a review of the inspection or monitoring procedures to ensure
11 that aging management activities remain adequate for the remaining period of extended
12 operation.

13 **3.5.1 Concrete**

14 *3.5.1.1 Freeze and thaw*

15 Concretes exposed to outdoor and groundwater/soil (below-grade) environments above the 16 freeze line

17 Concretes that are nearly or fully saturated with water can be damaged by repeated freezing
18 and thawing cycles in environments with weathering indexes (i.e., the product of the average
19 annual number of freezing cycle days and the average annual winter rainfall in inches) on the
20 order of 100 day-in/yr or greater (NRC, 2010a). For environments with weathering indexes less
21 than 100 day-in/yr, freeze and thaw degradation is not considered to be significant. The
22 weathering index for the continental United States and adequate data sourcing for determining
23 the weathering for any locality can be found in ASTM C216 (ASTM, 2016). For below-grade
24 concrete structures above the freeze line, water that resides in soil can also be subject to
25 freezing conditions, potentially promoting freeze and thaw damage.

26 Freeze and thaw damage has been observed in outdoor concrete structures in nuclear power
27 plants (NRC, 1995, 2012). Because water expands when freezing, fully or mostly saturated
28 concrete will experience internal stresses from the expanding ice, which can cause concrete
29 cracking or scaling when pressures exceed the concrete tensile strength (ACI, 2008c; Pigeon,
30 1994; Marchand et al., 1994; Sawan, 1987; Fagerlund, 1977).

31 The degradation mode would initiate at the outer concrete surface of the DSS exposed to
32 outdoor environments, primarily at horizontal surfaces where water ponding can occur.
33 Operating experience has identified freeze and thaw damage in the roofs of the concrete
34 storage modules at the Three Mile Island Unit 2 (TMI-2) and the Millstone independent spent
35 fuel storage installation (ISFSI) (NRC, 2012).

36 Therefore, freeze and thaw damage is considered credible in concrete exposed to outdoor and
37 groundwater or soil (below-grade) environments above the freeze line, and aging management
38 is required during the 60-year timeframe.

39

40

1 Concretes exposed to sheltered environments, fully encased (lined) in steel, and exposed to
2 groundwater/soil (below-grade) environments under the freeze line

3 Freeze and thaw degradation of concrete exposed to sheltered environments with low water
4 availability is not considered credible; the heat load from the fuel in the DSS is expected to aid
5 in drying the interior concrete surfaces of the overpacks, preventing freeze and thaw damage.

6 Freeze and thaw degradation of concrete exposed to groundwater or soil (below-grade)
7 environments at temperatures above freezing is not considered credible.

8 Freeze and thaw damage also is not considered credible for concrete fully encased in metallic
9 liners (not in direct contact with outdoor environments or proven to be protected from water
10 ingress); the lack of water transfer from the outside environment into the concrete prevents the
11 degradation mechanism.

12 Therefore, aging management of concrete for freeze and thaw degradation in these
13 environments is not required.

14 3.5.1.2 *Creep*

15 Creep in concrete is the time-dependent deformation resulting from sustained loads (Wang and
16 Salmon, 1998). Cement paste in concrete exhibits creep due to its porous structure and a large
17 internal surface area that is sensitive to water movements. Creep manifests as cracking on the
18 concrete outer surfaces and causes redistributions of internal forces. Factors affecting creep
19 are concrete constituents (composition and fineness of the cement; admixtures; and size,
20 grading, and mineral content of aggregates), water content and water-cement ratio, curing
21 temperature, relative humidity, concrete age at loading, duration and magnitude of loading,
22 surface-volume ratio, and slump (Wang and Salmon, 1998; Neville and Dilger, 1970). However,
23 the most important parameter controlling creep is concrete sustained loading. Creep increases
24 with increasing load and temperature (McDonald, 1972). However, the creep rate decreases
25 exponentially with time (Branson, 1977; NRC, 2014; Wang and Salmon, 1998). In summary, in
26 the case of a given concrete mix design, concrete creep is generally understood to be a
27 phenomenon that would affect concrete structures early in the service life under sustained
28 loading. Thus, the age of concrete and the magnitude and duration of sustained loading are the
29 primary factors that determine the magnitude of the creep of concrete (Neville and Dilger, 1970).
30 For example, if a sustained load is applied on 2-year-old and 40-year-old concrete, the
31 2-year-old concrete will have significantly more creep. Also, the creep in concrete could largely
32 be mitigated by proper design practices, in accordance with ACI 318-05 (ACI, 2005) or
33 ACI 349-06 (ACI, 2007). Furthermore, creep-induced concrete cracks are not generally large
34 enough to reduce the compressive strength of concrete, cause deterioration of concrete, or
35 cause exposure of reinforcing steel to the environment. In a DSS, the initial sustained load is
36 normally low, and no significant change of load is expected during the 40-year timeframe
37 beyond initial licensing. Thus, creep is not considered credible for any environment, and aging
38 management is not required during the 60-year timeframe.

39 3.5.1.3 *Reaction with aggregates*

40 The two most common alkali-aggregate reactions are alkali-silica reaction (ASR) and
41 alkali-carbonate reaction, with ASR being the most common and damaging. ASR is a chemical
42 reaction between hydroxyl ions (present in the alkaline cement pore solution) and reactive forms
43 of silica present in some aggregates (e.g., opal, chert, chalcedony, tridymite, cristabolite,

1 strained quartz). An aggregate that presents a large surface area for reaction (i.e., amorphous,
2 glassy) is susceptible to ASR (Poole, 1992). The resulting chemical reaction produces an alkali-
3 silica gel that swells with the absorption of moisture, exerting expansive pressures within the
4 concrete (Figg, 1987). ASR damage in the concrete manifests as a characteristic map cracking
5 on the concrete surface (ACI, 2008a). The internal damage results in the degradation of
6 concrete mechanical properties, and in severe cases, the expansion can result in undesirable
7 dimensional changes and popouts. In reinforced concrete, cracks tend to align parallel to the
8 direction of maximum restraint and rarely progress below the level of the reinforcement. In
9 general, ASR is a slow degradation mechanism that can cause serviceability issues and may
10 exacerbate other deterioration mechanisms.

11 The requisite conditions for initiation and propagation of ASR include (i) a sufficiently high alkali
12 content of the cement (or alkali from other sources, such as deicing salts, seawater, and
13 groundwater), (ii) a reactive aggregate, and (iii) available moisture, generally accepted to be
14 relative humidity greater than 80 percent (Pedneault, 1996; Stark, 1991). A study by the
15 California Department of Transportation (Glauz et al., 1996) revealed that ASR increases
16 proportionally to the cement content, alkali content greater than 0.6 percent can accelerate
17 ASR, high calcium oxide content can promote ASR, and the use of various types of admixtures
18 in certain doses can mitigate ASR (ACI, 2008a; ASTM, 1998). At higher concentrations of alkali
19 hydroxides, even the more stable forms of silica are susceptible to ASR attack (Xu, 1987).
20 Repeated cycles of wetting and drying can accelerate ASR (ACI, 1998). As a result, it is
21 desirable to minimize both available moisture and wet-dry cycles by providing good drainage.
22 Moreover, concretes exposed to warm environments are more susceptible to ASR than those
23 exposed to colder environments (Perenchio et al., 1991).

24 As mentioned earlier, ASR is generally a slow degradation mechanism. ASR may take from
25 3 to more than 25 years to develop in concrete structures, depending on the nature
26 (reactivity level) of the aggregates, the moisture and temperature conditions to which the
27 structures are exposed, and the concrete alkali content (Thomas et al., 2013). The delay in
28 exhibiting deterioration indicates that there may be less reactive forms of silica that can
29 eventually cause deterioration (Mindess and Young, 1981). Recent operating experience has
30 revealed degradation of the concrete in the Seabrook reactor containment as a result of ASR
31 (NRC, 2011b). The concrete used at the Seabrook plant passed all industry standard ASR
32 screening tests (ASTM, 2007, 2012) at the time of construction. However, ASR-induced
33 degradation was identified in August 2010. In addition, ASR screening tests are not conducted
34 on each aggregate source but rather in select batches, which increases the risk for use of
35 aggregates of different reactivities when procured from different sources. Due to the
36 uncertainties in screening tests that can effectively be used to eliminate the potential for ASR
37 and previous ASR operating experience at a nuclear facility, the aging mechanism is considered
38 credible in concrete exposed to any environment with available moisture, and therefore, aging
39 management is required during the 60-year timeframe.

40 3.5.1.4 *Differential settlement*

41 Differential settlement is a result of the uneven deformation of the supporting foundation soil
42 (Das, 1999; NAVFAC, 1986). The factors affecting structural settlement include the type of
43 foundation soil (e.g., clayey soil, sandy soil) and its physical properties, thickness of soil layers,
44 water-table level, depth of foundation mat below the ground surface, liquefaction during seismic
45 events, and load. Differential settlement, which causes distortion (loss of form) and damage
46 (cracking) to concrete structures, is a function of the uniformity of the soil, stiffness of the

1 structure, stiffness of the soil, and distribution of loads within the structure (U.S. Department of
2 the Army, 1990; NAVFAC, 1996).

3 The settlement of saturated cohesive soil consists of three components: (1) immediate
4 settlement occurring due to the applied load, (2) consolidation settlement occurring gradually
5 due to dissipation of the excess pore pressures generated by the applied load, and
6 (3) secondary compression that depends on the composition and structure of the soil skeleton
7 (NAVFAC, 1986). The settlement of course-grained granular soils subject to applied load
8 occurs immediately, primarily from the compression of the soil skeleton due to rearrangement
9 of particles. However, most settlement issues involving a combination of immediate
10 settlement and progressing long-term settlement are typically discovered in less than 1 year
11 of construction.

12 Differential settlement is addressed during the design-basis calculations. The analyses
13 generally include calculations to predict differential settlement based on the sequential DSS
14 placement; the analyses are used to determine an optimum DSS placement sequence to limit
15 differential settlement of the ISFSI support pad. However, operating experience has shown that
16 it can occur; periodic walkdowns ensure these limited occurrences are evaluated on a
17 case-by-case basis. NUREG-1522, "Assessment of In-service Conditions of Safety-Related
18 Nuclear Plant Structures" (NRC, 1995), stated that foundation settlement of concrete structures
19 at Point Beach and Beaver Valley, inspected during walkdowns, experienced appreciable
20 differential settlement. In addition, the loads on the concrete pad are expected to increase over
21 time as more loaded DSSs are placed on the pad. Therefore, differential settlement of
22 concretes exposed to sheltered, outdoor, and groundwater or soil (below-grade) environments
23 is considered credible, and aging management is required during the 60-year timeframe.

24 3.5.1.5 *Aggressive chemical attack*

25 The intrusion of aggressive ions or acids into the pore network of the concrete can cause
26 various degradation phenomena. The aggressive chemical attack typically originates from an
27 external source of sulfate or magnesium ions as well as acidic environmental conditions.
28 Depending on the type of aggressive chemical, the degradation of concrete can manifest in the
29 form of cracking, loss of strength, concrete spalling and scaling, and reduction in concrete pH.

30 Concretes exposed to outdoor and groundwater/soil (below-grade) environments

31 *External sulfate attack*

32 External sulfate attack is a process whereby ions in species such as K_2SO_4 , Na_2SO_4 , $CaSO_4$,
33 and $MgSO_4$, which are present in groundwater, seawater, and rainwater, penetrate the concrete
34 and chemically react with alkali and calcium ions to form a precipitate of calcium sulfate in
35 addition to other forms of calcium and sulfate-based compounds (e.g., ettringite). The
36 manifestation of sulfate attack is cracking, increase in concrete porosity and permeability, loss
37 of strength, and surface scaling generated by the expansion associated with the formation of
38 ettringite within the concrete and the pressure generated by the precipitated calcium and
39 sulfate-base compounds inside the concrete pore network (Poe, 1998; NWTRB, 2010). Unlike
40 the alkali sulfates, no decalcification of the calcium silicate hydrate phase occurs in the $CaSO_4$
41 attack. On the other hand, the $MgSO_4$ attack is significantly faster and more thorough than the
42 attack by the other sulfate compounds because of the limited solubility of magnesium hydroxide
43 ($Mg(OH)_2$) in the high pH of concrete (Drimalas et al., 2010). In addition, magnesium ions

1 present in deicing salts can react with calcium silicate hydrate, gradually converting it to
2 magnesium silicate hydrate, which is not cementitious in nature.

3 A service life model for sulfate attack in concrete was developed by Atkinson and Hearne
4 (1990). Cases of sulfate attack in the field are fairly uncommon, mainly because most
5 transportation regulatory agencies have adopted specifications aimed at preventing this damage
6 mode
7 (Weiss et al., 2009; Van Dam and Peshkin, 2009). In particular, degradation due to external
8 sulfate attack has not been reported in nuclear applications. Atkinson and Hearne (1990)
9 developed a concrete service life model to assess degradation due to sulfate attack. Using
10 aggressive soil and groundwater conditions (sulfate concentration of 1,500 ppm as specified in
11 ASME Code Section XI, Subsection IWL (ASME, 1995)) and typical concrete properties
12 (i.e., elastic modulus, roughness factor, Poisson's ratio, and concrete porosity), the model
13 predicts that sulfate damage can occur within 60 years of exposure (Berntz et al., 2001).

14 *Magnesium attack*

15 Magnesium ions can rapidly replace calcium ions in the silica hydrate compounds. In
16 groundwater, magnesium ions are commonly found in the form of $MgSO_4$. The magnesium ion
17 attack is more commonly observed in arid western U.S. areas and in below-grade structures. At
18 present, there is no stipulation on the threshold concentration of magnesium ions needed to
19 promote damage to concrete structures for nuclear and nonnuclear applications. Because
20 magnesium attack could be part of the sulfate attack, the timeframe implications and exposure
21 conditions are expected to be comparable to those of sulfate attack.

22 *Acid attack*

23 Acids with a pH less than 3 can dissolve both hydrated and unhydrated cement compounds
24 (e.g., calcium hydroxide, calcium silicate hydrates, and calcium aluminate hydrates) as well as
25 calcareous aggregate in concrete without any significant expansion reaction (Gutt and Harrison,
26 1997; Mehta, 1986). In most cases, the chemical reaction forms water-soluble calcium
27 compounds, which are then leached away by aqueous solutions. The dissolution of concrete
28 commences at the surface and propagates inward as the concrete degrades. The signs of
29 acidic attack are loss of alkalinity (also disturbing of electrochemical passive conditions for the
30 embedded steel reinforcement), loss of material (i.e., concrete cover), and loss of strength.

31 The extent and rate of concrete degradation depends on the type, concentration and pH of the
32 acidic solution, concrete permeability, calcium content in the cement, the water-to-cement ratio,
33 and the type of cement and mineral admixtures (Pavlik and Uncik, 1997). Sulfuric acid is
34 particularly aggressive to concrete, because the calcium sulfate formed from the acid reaction
35 will also deteriorate concrete via sulfate attack (Pavlik, 1994). Even slightly acidic solutions that
36 are lime deficient can attack concrete by dissolving calcium from the paste, leaving behind a
37 deteriorated paste consisting primarily of silica gel.

38 Acids can come from groundwater as well as from acid rain containing SO_2 , NO_x , and HCl from
39 polluted regions, which can compromise the durability of concrete (Webster and Kukacka,
40 2009). Ueda et al. (2001) proposed a model for acid rain deterioration, which is dependent on
41 the amount of acid absorption into the concrete, type of acid, mix proportion, and contact time or
42 interval of rainfalls. The model can predict the depth of concrete damage as a function of
43 environmental pH. A study by Manjeeth and Rama (2015) found that the compressive strength
44 and mass loss of concrete samples decreased after 28 days of exposure to sulfuric acid

1 solutions with pH ranging from 1 to 7. As such, this degradation mode is expected to affect the
2 concrete shortly after the concrete surface is in contact with the acid solution.

3 In summary, aggressive chemical attack of concretes exposed to outdoor and groundwater or
4 soil (below-grade) environments is considered to be credible, and therefore, aging management
5 is required during the 60-year timeframe.

6 Concretes exposed to sheltered and fully encased (lined) in steel environments

7 With regard to concrete in sheltered environments and fully encased (lined) in steel, external
8 sources of sulfate, magnesium, and acid entering concrete are considered to be insignificant. In
9 addition, the heat load from the fuel in the DSS is expected to aid in drying the interior concrete
10 surfaces, thus decreasing water availability at the concrete surface, which is necessary to
11 promote this degradation mode. Thus, aggressive chemical attack of sheltered and fully
12 encased (lined) concrete is not considered credible, and therefore, aging management is not
13 required during the 60-year timeframe.

14 3.5.1.6 *Corrosion of reinforcing steel*

15 Concretes exposed to outdoor and groundwater/soil (below-grade) environments

16 Corrosion of the reinforcing steel embedded in the concrete is mainly caused by the presence of
17 chloride ions in the concrete pore solution and carbonation of the concrete. Chloride attack of
18 concrete structures is well established in the literature (Cheung et al., 2009). The highly alkaline
19 environment provided by the concrete (normally with pore water pH>13.0) results in the
20 formation of a metal-adherent oxide film on the reinforcement steel bar surface, which
21 passivates the steel (Page, 1982). However, chloride ions may penetrate the concrete matrix
22 and break down the steel passive layer, once the chloride concentration at the reinforcing steel
23 surface exceeds a threshold value, triggering corrosion of the reinforcing steel and shortening
24 the service life of a concrete structure. For instance, chlorides may already exist at low levels
25 within the base mix constituents. In most practical situations, chloride ions penetrate from the
26 outside environment, such as when using deicing salts, from groundwater, and in marine
27 environments (Tang and Sandberg, 1996). The presence of corrosion products at the steel
28 surface can generate internal stresses within the concrete matrix, causing cracks and spalling of
29 the concrete cover with consequent structural damage.

30 The threshold chloride concentration in concrete required to promote corrosion of the reinforcing
31 steel depends on the pH of the concrete pore solution. The onset of corrosion can be enhanced
32 when acid attack or concrete carbonation¹ reduces the concrete pH at the steel surface. Thus,
33 the chloride-to-hydroxide ratio is an important parameter in evaluating the steel corrosion. The
34 present literature does not provide a clear agreement on the value of the critical chloride ion
35 concentration required for corrosion initiation. Glass and Buenfeld (1997) have reviewed the
36 chloride threshold values reported for steel embedded in concrete structures. From this
37 investigation, it was concluded that a universal, well-defined chloride threshold value does not
38 exist. The lowest limit of chloride threshold value in concrete ranged from 0.2 to 2.5 percent
39 (by weight of cement). Factors such as the chemical composition of the rebar, as well as its

¹Carbonation results from the chemical reaction between the hydrated cement material and atmospheric carbon dioxide, which lowers the pH of the concrete and reduces the passivation effect of calcium hydroxide in preventing the corrosion of reinforcing steel. The carbonation rate depends on the external CO₂ concentration, concrete type, temperature, time of wetness of the concrete surface, and degree of moisture (Bertolini et al., 2004).

1 surface roughness, can influence the chloride threshold (Szkłarska-Smiałowska, 1986).
2 Groundwater aggressiveness is defined based on the chloride threshold concentration of
3 500 ppm [milligram (mg)/kilogram (kg)] with a pH less than 5.5 (ASME, 1995, Section XI,
4 Subsection IWL). This value is consistent with those recommended in ACI 201.2R-08
5 (ACI, 2008c).

6 Concrete durability is directly related to the quality of the concrete, the external concentration of
7 chlorides on the concrete surface, and the reinforcement material. The service life of concretes
8 exposed to chloride attack depends on the concrete cover, the surface chloride concentration,
9 the chloride diffusion coefficient, the type of cementitious material, and the reinforcing steel
10 material. Several service life models have been proposed to determine the durability of
11 concrete subject to chloride-induced corrosion (Schiessl et al., 2006; DuraCrete, 2000;
12 Berntz et al., 2001). For example, for a constant surface chloride concentration of 0.05 percent
13 by weight of concrete (i.e., the maximum chloride concentration in soil and groundwater per
14 ASME Code Section XI, Subsection IWL ASME, (ASME, 1995)), a 2.54-cm [1-in] concrete
15 cover, and a chloride threshold of 0.03 percent by weight of concrete, the onset of
16 chloride-induced corrosion in concrete occurs in about 6, 20, and 120 years for constant
17 chloride diffusion coefficients of 6.45×10^{-7} cm²/second (sec) [10^{-7} in²/sec] (poor concrete
18 quality), 6.45×10^{-8} cm²/sec [10^{-8} in²/sec] (moderate concrete quality), and 6.45×10^{-9} cm²/sec
19 [10^{-9} in²/sec] (good concrete quality), respectively (Berntz et al., 2001).

20 Although no cases of corrosion-induced damage have been reported, the results of the
21 durability model presented by Berntz et al. (2001) show that corrosion of the reinforcing steel in
22 concrete can potentially initiate and propagate within the 60-year timeframe for concretes of
23 moderate to low quality. Thus, corrosion of reinforcing steel in concrete exposed to outdoor and
24 groundwater or soil (below-grade) environments is considered to be credible, and therefore,
25 aging management is required during the 60-year timeframe.

26 Concretes exposed to sheltered environments and fully encased (lined) in steel

27 Chloride ingress is expected to be insignificant for steel reinforcement embedded in concrete in
28 sheltered environments with limited exposure to water. In addition, the heat load from the fuel in
29 the DSS is expected to aid in drying the interior concrete surfaces, thus decreasing water
30 availability at the concrete surface, which is necessary to promote this degradation mode.
31 Chloride ingress will also be impeded in concrete fully encased (lined) in steel. Thus, corrosion
32 of reinforcing steel is not considered credible for concrete in these environments, and therefore,
33 aging management is not required during the 60-year timeframe.

34 3.5.1.7 Shrinkage

35 Shrinkage occurs when hardened concrete dries from a saturated condition to a state of
36 equilibrium in about 50 percent relative humidity (NRC, 2012). As excess concrete water
37 evaporates, tensile stresses are induced in the concrete due to internal pressure from the
38 capillary action of water movement, which results in cracking. The factors affecting shrinkage
39 are cement content, water-to-cement ratio, degree of hydration, elastic modulus of aggregates,
40 amount and characteristics of concrete admixtures, temperature and humidity during curing, and
41 size and shape of concrete (NRC, 2014; Branson, 1977; Mindess and Young, 1981).

42 The maximum shrinkage is in the range of 400×10^{-6} to 780×10^{-6} cm/cm [400×10^{-6} to
43 780×10^{-6} in/in] (NRC, 2014; Branson, 1977) and decreases exponentially with time
44 (Branson, 1977). Shrinkage of concrete occurs initially during curing, which can be controlled

1 through concrete formulation and the density and distribution of internal reinforcement
2 (ACI, 2005, 2007). According to ACI 209R-92 (ACI, 2008b), over 90 percent of the shrinkage
3 occurs during the first year, reaching 98 percent by the end of the first 5 years. Thus, shrinkage
4 is not expected to influence concrete performance after the initial storage or licensing period,
5 because most of the shrinkage will take place early on in the life of the concrete. As a result,
6 shrinkage of concretes exposed to sheltered, outdoor, groundwater or soil (below grade), and
7 fully encased environments is not considered to be credible, and therefore, aging management
8 is not required during the 60-year timeframe.

9 *3.5.1.8 Leaching of calcium hydroxide*

10 *Concretes exposed to outdoor, sheltered, and groundwater/soil (below-grade) environments*

11 A constant or intermittent flux of water through a concrete surface can result in the removal or
12 leaching of calcium hydroxide (Hanson et al., 2011). Calcium hydroxide leaching is observed in
13 the form of white leachate deposits (calcium carbonate) on the concrete surface. Calcium
14 hydroxide leaching causes loss of concrete strength, converting the cement into gels that have
15 no strength. Leaching also increases the concrete porosity and permeability, making it more
16 susceptible to other forms of aggressive attack. In addition, leaching of calcium hydroxide in
17 concrete lowers the concrete pH, affecting the integrity of the protective oxide film of the
18 reinforcing steel (EPRI, 2007).

19 The extent of the leaching depends on the environmental salt content and temperature
20 (NRC, 2011a), and it can take place above and below ground. However, the leaching rate is
21 generally slow and controlled by diffusion (Berner, 1992). For example, interior inspections
22 conducted at the Calvert Cliffs ISFSI revealed the presence of white-colored stalactite debris in
23 the gap between the heat shield and the concrete ceiling of two sheltered DSS concrete
24 structures after 15–20 years in service. Stalactites are formed when water leaches calcium
25 hydroxide out of the concrete, which precipitates as calcium carbonate on contact with carbon
26 dioxide in the air. The licensee concluded that water entering the outlet vent stack promoted
27 calcium hydroxide leaching (Gellrich, 2012). Other exterior inspections conducted at the TMI-2
28 ISFSI revealed efflorescence growth on multiple DSS concrete structures exposed to an
29 outdoor environment. The licensee concluded that the efflorescence deposits were formed by
30 water entering freeze and thaw cracks in the anchor blockout holes on the roof of the HSMs.
31 The licensee conducted core sample testing to verify concrete compressive strength.
32 Therefore, operating experience indicates that leaching of calcium hydroxide is a mechanism
33 that can be exacerbated by other degradation mechanisms or designs that do not adequately
34 prevent ingress of precipitation into the sheltered structure. As such, leaching of calcium
35 hydroxide in concrete exposed to outdoor, sheltered, and groundwater or soil (below-grade)
36 environments is considered to be credible, and therefore, aging management is required during
37 the 60-year timeframe.

38 *Concretes fully encased (lined) in steel*

39 Leaching of calcium hydroxide is not considered a credible mechanism for concrete fully
40 encased (lined) in steel because of the lack of water ingress, and therefore, aging management
41 is not required during the 60-year timeframe.

1 3.5.1.9 *Radiation damage*

2 Radiation effects on concrete properties will depend on the gamma and neutron radiation
3 doses, temperature, and exposure period. Gamma radiation can decompose and evaporate
4 water in concrete (Bouniol and Aspart, 1998). Because most of the water is contained in the
5 cement paste, the effect of gamma radiation on cement paste is more significant than on the
6 aggregates. Gamma radiation can also decompose the SiO bond within calcium silicate hydrate
7 (Kontani et al., 2010). Neutron radiation deteriorates concrete by reducing stiffness, forming
8 cracks by swelling, and changing the microstructure of the aggregates. This consequently
9 reduces concrete strength (Kontani et al., 2010). The changes in aggregate microstructure also
10 can lead to higher reactivity of aggregates to certain aggressive chemicals.

11 NUREG/CR-7171, “A Review of the Effects of Radiation on Microstructure and Properties of
12 Concretes Used in Nuclear Power Plants,” provides a comprehensive review of the effects of
13 gamma and neutron radiation on the microstructure and properties of concrete used in nuclear
14 power plants (NRC, 2013). Concrete structures have been regarded as being sound as long as
15 the cumulative radiation does not exceed critical levels over the life of the structure. In general,
16 the critical radiation levels to reduce concrete strength and elastic modulus are considered to be
17 approximately 1×10^{19} n/cm² [6.5×10^{19} n/in²] for fast neutrons (neutron energy >1 MeV) and
18 $1\text{-}2 \times 10^{10}$ rad [$1\text{-}2 \times 10^8$ grays] for gamma rays (Hilsdorf et al., 1978; EPRI, 2012; IAEA, 1998;
19 ASME, 2007).

20 As discussed in Section 3.2.1.9, the maximum potential accumulated neutron fluence on DSS
21 basket components after 100 years was calculated to be 2.63×10^{16} n/cm² [1.70×10^{17} n/in²],
22 which is three orders of magnitude below the level that would lead to a reduction of concrete
23 strength and elastic modulus. The gamma dose is also expected to be several orders of
24 magnitude less than the limits defined in the above references, per the specific DSS design
25 bases. Thus, radiation damage is not considered credible for concrete, and therefore, aging
26 management is not required during the 60-year timeframe.

27 3.5.1.10 *Fatigue*

28 Concrete fatigue strength is defined as the maximum stress that the concrete can sustain
29 without failure under a given number of stress cycles (NRC, 2014). Because dry storage is a
30 static application, mechanical cyclic loading is not expected. However, restraint of the concrete
31 from expanding and contracting as it is exposed to rapid changes in temperature will lead to
32 internal stresses in the structure. If the changes in temperature are severe and the resulting
33 strains are sufficient, local plastic deformation can occur. Repeated application of this thermal
34 loading can lead to crack initiation and propagation in low-cycle fatigue.

35 Concrete fatigue in the DSS reinforced concrete may be caused by diurnal and seasonal
36 temperature gradients through the wall of the DSS assembly. The inside surface of the
37 concrete wall is hotter than the outside surface of the concrete wall, which causes compressive
38 stresses in the DSS concrete near the inside of the concrete wall and tensile stresses in the
39 rebar near the outside of the concrete wall.

40 Extreme seasonal temperature variations are expected to be significantly higher than diurnal
41 variations; these would be capable of producing higher cyclic stress amplitudes. Assuming
42 ambient temperatures of -40 degrees C [-40 degrees F] (winter) and 52 degrees C
43 [125 degrees F] (summer), the maximum thermal gradient across the DSS concrete is expected

1 to be less than 16 degrees C [60 degrees F]. The number of extreme seasonal temperature
2 cycles, conservatively postulated to occur 10 times per year, is 600 over 60 years.

3 Diurnal temperature fluctuations in ambient air temperatures are assumed to occur once per
4 day. For conservatism, it is assumed that the diurnal temperature fluctuations are 25 degrees C
5 (the largest mean daily change of temperature in the United States). Therefore, the total
6 number of thermal cycles due to diurnal temperature variations in ambient temperatures over
7 60 years is 21,900 thermal cycles. Thus, the total number of thermal cycles due to seasonal
8 and daily variations over 60 years is 22,500 cycles. The thermally induced stress, σ , defined in
9 Section 3.2.1.7, can be used to determine the stress in the concrete during each
10 temperature cycle. Using a thermal expansion coefficient of 1.1×10^{-5} cm/cm/degrees C
11 [6.5×10^{-6} in/in/degrees F] and an elastic modulus of 2.764×10^4 megapascals (MPa)
12 [4.035×10^3 ksi], which are typical for concretes, the computed values of σ are 7.53 MPa
13 [1.09 ksi] and 9.99 MPa [1.45 ksi] for the diurnal and seasonal temperature
14 fluctuations, respectively.

15 The seasonal change in stress is assumed bounding for the cumulative number of cycles of
16 both diurnal and seasonal temperature fluctuations. Assuming that these cyclic stresses are the
17 only cyclic mechanical loading experienced by the DSS (an adequate assumption for a passive
18 system), the ratio of the concrete compressive stress to its design strength is less than 0.29
19 (i.e., 1.45 ksi/5 ksi). This calculated ratio at 22,500 cycles is lower than the lowest
20 stress/cycles-to-failure (S-N) curve for concrete reported in ACI 215R (ACI, 1997). Thus,
21 fatigue of concrete exposed to sheltered, outdoor, groundwater or soil (below-grade), and fully
22 encased environments is not considered to be credible, and therefore, aging management is not
23 required during the 60-year timeframe.

24 Notwithstanding the conclusion above, the NRC reviewer must review any fatigue analyses for
25 concrete structures contained in the applicant's original design-bases documents to determine
26 whether the renewal application adequately addresses the implications of extending the
27 operating period to 60 years. This reexamination of the original fatigue analyses should be
28 defined as TLAAs in the renewal application. The staff's guidance for the review of TLAAs is
29 provided in NUREG-1927, Revision 1, and is summarized in Chapter 5 of this report.

30 3.5.1.11 *Dehydration at high temperature*

31 Exposure of concrete to elevated temperatures can affect its mechanical and physical
32 properties (Phan and Carino, 2000). It is well known that concretes can degrade at high
33 temperatures due to dehydration of the hydrated cement paste, thermal incompatibility between
34 the cement and aggregates, and physicochemical deterioration of the aggregates (NRC, 2006).
35 As the temperature increases to about 105 degrees C [221 degrees F], all evaporable water is
36 removed from the concrete. At temperatures above 105 degrees C [221 degrees F], the
37 strongly absorbed and chemically combined water are gradually lost, with the dehydration
38 essentially complete at 850 degrees C [1,562 degrees F] (Harmathy, 1970). High-temperature
39 degradation in concrete manifests as a change in compressive strength and stiffness, as well as
40 an increase in concrete shrinkage and transient creep, resulting in the formation of cracks
41 (Naus, 1981, 1988; Schneider et al., 1981). The effect of the elevated temperature is
42 most significant on the concrete's modulus of elasticity, which can decrease up to 40 percent
43 (Freskakis, 1979). Concretes in the temperature range of 20 to 200 degrees C [68 to
44 392 degrees F] show small changes in compressive strength. Beyond 350 degrees C
45 [662 degrees F], concrete compressive strength decreases rapidly (NRC, 2006).

1 NUREG–1536, “Standard Review Plan for Spent Fuel Dry Storage Systems at a General
2 License Facility” (NRC, 2010b), provides staff guidance for acceptable temperature limits during
3 operation of DSS concrete structures. By design, general or local concrete temperatures should
4 be kept below 93 degrees C [200 degrees F] to avoid mechanical deterioration. For DSS
5 concrete designs that satisfy additional acceptance criteria, the maximum temperature during
6 operation can exceed 93 degrees C [200 degrees F] but should remain less than 149 degrees C
7 [300 degrees F]. Therefore, the effects of thermal dehydration are addressed during the initial
8 ISFSI licensing or DSS approval. Because the fuel temperature decreases over time, the
9 design temperature considerations in NUREG–1536 are expected to continue to be adequate.
10 Thus, dehydration of concrete at high temperature is not considered to be credible in sheltered,
11 outdoor, groundwater or soil (below-grade), and fully encased (lined) environments, and
12 therefore, aging management is not required during the 60-year timeframe.

13 3.5.1.12 *Microbiological degradation*

14 Concretes exposed to groundwater/soil (below-grade) environments

15 Biodeterioration is caused by colonization of microbes and microorganisms that grow on
16 concrete surfaces that offer favorable environmental conditions (e.g., available moisture, near
17 neutral pH, presence of nutrients). Conducive environments may have elevated relative
18 humidity (i.e., greater than about 60 percent), long cycles of humidification and drying, freezing
19 and thawing, high carbon dioxide concentrations, high concentrations of chloride ions or other
20 salts, or high concentrations of sulfates and small amounts of acids (Wei et al., 2013).
21 According to Sanchez-Silva and Rosowsky (2008), biodeterioration may lead to reduction of the
22 protective cover depth and increase both concrete porosity and the transport of aggressive
23 chemicals. In addition, this degradation mode can promote a reduction in concrete pH, loss of
24 concrete strength, and spalling/scaling.

25 Evidence shows that a wide variety of organisms can cause concrete deterioration in polluted
26 soils and groundwater. The biodeterioration of concrete typically is confined to the surface. The
27 rate of deterioration is slow, but the degradation mode has been observed within 40 years of
28 exposure (Hu et al., 2011). Recent observations in Texas, Alabama, Georgia, and Mississippi
29 have identified several sites where microorganisms have caused deterioration of the columns of
30 concrete bridges embedded in soil (Trejo et al., 2008). Giannantonio et al. (2009),
31 Magniont et al. (2011), Vollertsen et al. (2008), and Ghafoori and Mathis (1997) provide a list of
32 microorganisms that can promote degradation in concrete in soils and waters. According to
33 Bastidas-Arteaga et al. (2008), biodeterioration of concrete is mainly caused by bacteria, fungi,
34 algae and lichens, and mussels (Perez et al., 2003). Once the pH of the surface of the concrete
35 drops below 9 in the presence of sufficient nutrients, moisture, and oxygen, some species of
36 sulfur bacteria, such as *Thiobacillus sp.*, can attach to the concrete surface and reproduce
37 (Mori et al., 1992). As the pH continues to fall to moderate or weakly acidophilic conditions,
38 *T. novellus*, *T. neapolitanus*, and *T. intermedius* establish on the surface of concrete
39 (Milde et al., 1983). The type of bacteria is strongly dependent on the concrete pH and
40 environmental conditions (Okabe et al., 2007).

41 Although no cases of microbiological degradation of concrete have been reported in nuclear
42 applications, the degradation mode is considered credible, as below-grade environments may
43 be conducive to microbe and bacteria growth. Thus, microbiological degradation of concrete
44 structures exposed to groundwater or soil (below-grade) environments is considered credible,
45 and therefore, aging management is required during the 60-year timeframe.

1 Concretes exposed to outdoor, sheltered, and fully encased (lined) environments

2 The outdoor and sheltered environments may provide favorable conditions for microbiological
3 degradation mechanisms because of the potential presence of moisture. However, the
4 conditions may be intermittent, and there is no evidence that actual concrete subcomponents in
5 the DSS environment microbiologically degrade. In addition, fully encased concrete is
6 considered to be largely protected from moisture intrusion. Thus, microbiological degradation of
7 concretes exposed to outdoor, sheltered, and fully encased (lined) environments is not
8 considered credible, and therefore, aging management is not required during the 60-year
9 timeframe.

10 3.5.1.13 *Delayed ettringite formation*

11 At the initial stage of fresh concrete curing, ettringite,¹ commonly referred to as “naturally
12 occurring ettringite,” is formed by the reaction of tricalcium aluminate and gypsum in the
13 presence of water. The formation of naturally occurring ettringite in fresh concrete is not
14 detrimental to the overall concrete performance. At the still-early stage of concrete curing, the
15 naturally occurring ettringite may convert to monosulfoaluminate if curing temperatures are
16 greater than about 70 degrees C [158 degrees F] (Fu, 1996). After concrete hardens, if the
17 temperature decreases below this value, the monosulfoaluminate becomes unstable and, in the
18 presence of sulfates released by the C-S-H gel, ettringite will reform. This mechanism is called
19 “delayed ettringite formation” (DEF), which results in volume expansion and increased internal
20 pressures in the concrete (Fu, 1996). Because the concrete has hardened at this stage, the
21 volume expansion leads to cracking and spalling, with greatest severity commonly observed in
22 below-ground structures with elevated temperatures from curing and heat of hydration
23 (Shayan and Quick, 1992; Hobbs, 1999). DEF has been reported in precast concrete railroad
24 ties in Sweden (Sahu and Thaulow, 2004), cast-in-place concrete structures in the southern
25 United States after 10 years in service (Thomas et al., 2008), and mass concretes with high
26 cement contents in the United Kingdom (Hobbs, 1999; Johansen and Thaulow, 1999).
27 However, to date, no operating experiences exist of DEF degradation for concrete structures at
28 nuclear power plants.

29 The conditions necessary for the occurrence of DEF are excessive temperatures during
30 concrete placement and curing, the presence of internal sulfates, and a moist environment.
31 ACI 318-05 (ACI, 2005) indicates that inspection reports shall document concrete temperature
32 and protection during placement when the ambient temperature is above 35 degrees C
33 [95 degrees F]. Protection measures during concrete placement include lowering the
34 temperature of the batch water, cement, and aggregates as referenced in ACI 305R-10
35 (ACI, 2010). As such, following the ACI 318-05, ACI 305R-10, and ACI 308R-01 (ACI, 2008d)
36 guidelines during concrete placement and curing can effectively limit the concrete temperature
37 to below 70 degrees C [158 degrees F], therefore preventing the development of DEF.
38 NUREG-1536 (NRC, 2010b) cites ACI 349 (ACI, 2007) and ACI 318 as applicable codes for the
39 design and construction of concrete structures of the DSSs. In addition to the adequate
40 placement and curing standards, no occurrences of DEF-related degradation of concrete have
41 been reported in nuclear applications. Thus, DEF of concrete is not considered credible in

¹Ettringite ($3\text{CaO}\cdot\text{Al}_2\text{O}_3\cdot 3\text{CaSO}_4\cdot 32\text{H}_2\text{O}$) is the product of the reaction of gypsum and other sulfate compounds with calcium aluminate in the cement within the first few hours after mixing with water.

1 outdoor, sheltered, groundwater or soil (below-grade), and fully encased (lined) environments,
2 and therefore, aging management is not required during the 60-year timeframe.

3 3.5.1.14 Salt scaling

4 Concretes exposed to outdoor environments and groundwater/soil (below-grade) environments 5 above the freeze line

6 Salt scaling is defined as superficial damage caused by freezing a saline solution on the surface
7 of a concrete body. The damage is progressive and consists of the removal of small chips or
8 flakes of material. Similar to freeze and thaw damage, salt scaling takes place when concrete is
9 exposed to freezing temperatures, moisture, and dissolved salts. The degradation is maximized
10 at a moderate concentration of salt (e.g., from deicing salts), called the pessimum concentration
11 (Marchand et al., 1999). Verbeck and Klieger (1957) reported that the pessimum concentration
12 is independent of the types of salt species and is about 3 to 4 percent of the solute by weight.
13 The most common deicing salts are sodium chloride and calcium chloride. Other deicing
14 chemicals include magnesium chloride, urea, potassium chloride, ammonium sulfate, and
15 ammonium nitrate.

16 Salt scaling of concrete roadways, pavements, sidewalks, driveways, decks, and other slabs is
17 a common problem in locations exposed to cyclic freezing and thawing and deicing salts. For
18 vertical surfaces, this damage mechanism is not expected to be operative unless the DSS
19 concrete structure is surrounded by standing water containing salts. Therefore, this degradation
20 mode is only expected to initiate and manifest in horizontal structures exposed to outdoor
21 environments where water ponding can occur. Because salt scaling is closely related to freeze
22 and thaw damage, the timeframe associated with the initiation of salt scaling of concrete could
23 be relevant for both short- and long-term exposures. Thus, salt scaling damage is considered
24 credible within the 60-year timeframe for DSS concrete structures exposed to outdoor and
25 groundwater or soil (below-grade) environments above the freeze line, and therefore, aging
26 management is required during the 60-year timeframe.

27 Concretes exposed to sheltered environments, fully encased (lined) in steel, and exposed to 28 groundwater/soil (below-grade) environments under the freeze line

29 Concretes exposed to sheltered environments with low water availability or below-grade
30 concrete maintained above freezing temperatures are not susceptible to salt scaling
31 degradation. The heat load from the emplaced fuel in DSSs is expected to aid in drying the
32 internal concrete surface, preventing the development of salt scaling inside the DSSs' concrete
33 structure. Salt scaling damage is also expected to be insignificant for concretes fully encased
34 by liners (e.g., metallic compartments)—even under freezing conditions—due to the lack of
35 water and salt transfer between the concrete and the outside environment. Thus, interior DSS
36 concrete surfaces, below-grade concretes maintained under the freeze line, and fully encased
37 (lined) concrete not in direct contact with outdoor environments are not expected to undergo salt
38 scaling damage within the 60-year timeframe, and therefore, aging management is not required.

39 3.5.2 Ceramic fiber insulation

40 The HI-STORM 100U underground system uses a divider shell to separate the intake cooling air
41 from the heated air that streams up around the canister. This shell is insulated to minimize the
42 preheating of the intake cooling air, with Kaowool® ceramic fiber insulation being a preferred
43 insulation material in this DSS design.

1 3.5.2.1 *Radiation damage*

2 Neutron radiation has been shown to affect the strength and thermal diffusivity of ceramic fiber
3 insulation. The effects will generally depend on the radiation dose, moisture content,
4 temperature, and exposure period.

5 Snead et al. (1992) provide an example of the effects of neutron irradiation on ceramic-fiber
6 interfacial strength. Results comparing unirradiated and 1-dpa neutron-irradiated ceramic fiber
7 insulation samples (SiC/C/Nicalon) exhibited a marked decrease in both interfacial shear
8 strength and frictional resistance to sliding. The decrease in interfacial shear strength resulted
9 in the decrease of the ultimate strength of the ceramic fibers by about 25 percent. Similarly, the
10 decrease in frictional resistance resulted in increased fiber toughness. The changes in the
11 mechanical properties were attributed to the fiber shrinkage that causes a partial debonding of
12 the fiber and matrix interface.

13 Other research provides examples of the effects of neutron irradiation on the thermal diffusivity
14 of ceramic fiber insulation (Akiyoshi and Yano, 2008; Snead et al., 2000; Akiyoshi, 2009;
15 Akiyoshi et al., 2006; Yano et al., 2000; and Snead et al., 2005). For example, Akiyoshi and
16 Yano (2008) showed a degradation of thermal diffusivity in neutron-irradiated specimens by
17 studying the macroscopic property changes in as-irradiated and annealed specimens under
18 different temperatures from 373 to 766 degrees C [703 to 1,411 degrees F] and different
19 neutron doses from 0.4 to 8.0×10^{22} n/cm² [2.6 to 51.6×10^{22} n/in²]. The thermal diffusivity of
20 as-irradiated specimens showed dependence on the neutron-irradiation dose and the irradiation
21 temperature. Snead et al. (2000) have also demonstrated that the thermal conductivity of most
22 ceramic fiber insulation materials undergoes a rapid reduction with irradiation when subjected to
23 a fast-neutron fluence up to about 3.4×10^{21} n/cm² [2.2×10^{22} n/in²] and irradiation temperature
24 of about
25 200–700 degrees C [392–1,292 degrees F]. Gamma irradiation also results in a permanent
26 decrease in the volume and surface resistivity of ceramic fibers at gamma values of around
27 1×10^9 rads [1×10^7 grays] (Davies, 1966). In general, the reduction of thermal diffusivity of
28 ceramic fiber insulation should result in improved thermal insulation performance.

29 While the reduction of strength of ceramic fiber insulation due to radiation is not expected to
30 compromise the SSC's intended function, a review of the radiation effects should be performed
31 on a case-by-case basis.

32 The NRC reviewer should review the analyses contained in the applicant's original
33 design-bases documents to determine whether the renewal application adequately addresses
34 radiation damage of ceramic fiber insulation for an extended operating period of 60 years. This
35 reexamination of the original analyses would typically be defined as TLAAs in the renewal
36 application. The staff's guidance for the review of TLAAs is provided in NUREG–1927,
37 Revision 1. If the original design basis does not include an analysis for an SSC that could
38 reasonably be expected to be subject to radiation damage in the 60-year timeframe, the
39 reviewer nevertheless should ensure that the application addresses this potential aging effect.

40 An applicant may conclude that an analysis cannot support a determination that fatigue will not
41 challenge an important-to-safety function in the 60-year timeframe of the period of extended
42 operation. In that case, the applicant may manage the aging of the associated SSC with
43 an AMP.

1 3.5.2.2 *Moisture absorption*

2 Ceramic fiber insulation materials are generally porous (either open- or closed-pore network)
3 and filled with atmospheric air in the dry condition. In nonencased SSCs, moisture transport
4 through the insulation can be realized by diffusion and/or capillary suction. Vafai and Sarkar
5 (1986) first modeled the transient heat and moisture transfer with condensation. The effect of
6 condensates on the effective thermal conductivity and radiative heat transfer have also been
7 considered in a transient model in porous media (Fan et al., 2000). This model suggests that
8 the initial water content, service temperature, and insulation thickness are key factors
9 influencing the insulation performance. Other parameters, such as the water vapor resistance,
10 the thermal conductivity, and the insulation porosity were found to have smaller effects. The
11 presence of moisture can significantly increase the insulation thermal conductivity
12 (Cai et al., 2012).

13 The ceramic fiber insulation is foil faced or jacketed and therefore encased and protected from
14 moisture. The high zinc content in the coating of the adjacent divider shell in the
15 HI-STORM 100U system provides protection for the foil/jacket from galvanic corrosion. In
16 addition, SCC of the foil/jacket is not a credible aging mechanism due to low stresses derived
17 from the dead weight of the foil or jacket. Therefore, the integrity of the foil or jacket is not
18 expected to be compromised, which will prevent moisture entering the ceramic fiber insulation.
19 As such, moisture absorption of ceramic fiber insulation is not considered to be credible, and
20 therefore, aging management is not required during the 60-year timeframe.

21 **3.5.3 References**

22 ACI. ACI 305R-10, "Guide to Hot Weather Concreting." Farmington Hills, Michigan:
23 American Concrete Institute. 2010.

24 _____. ACI 221.1R-98, "State-of-the-Art Report on Alkali-Aggregate Reactivity."
25 Farmington Hills, Michigan: American Concrete Institute. 2008a.

26 _____. ACI 209R-92, "Prediction of Creep, Shrinkage, and Temperature Effects in Concrete
27 Structures (Reapproved 2008)." Farmington Hills, Michigan: American Concrete Institute.
28 2008b.

29 _____. ACI 201.2R-08, "Guide to Durable Concrete." Farmington Hills, Michigan:
30 American Concrete Institute. 2008c.

31 _____. ACI 308R-01, "Guide to Curing Concrete." Farmington Hills, Michigan:
32 American Concrete Institute. 2008d.

33 _____. ACI 349-06, "Evaluation of Existing Nuclear Safety-Related Concrete Structures."
34 Farmington Hills, Michigan: American Concrete Institute. 2007.

35 _____. ACI 318-05, "Building Code Requirements for Structural Concrete and Commentary."
36 Farmington Hills, Michigan: American Concrete Institute. 2005.

37 _____. ACI 221.1R-98, "State-of-the-Art Report on Alkali-Aggregate Reactivity."
38 Farmington Hills, Michigan: American Concrete Institute. 1998.

1 _____. ACI 215R-74, "Considerations for Design of Concrete Structures Subjected to Fatigue
2 Loading." Farmington Hills, Michigan: American Concrete Institute. 1997.

3 Akiyoshi, M. "Thermal Diffusivity of Ceramics at the Neutron Irradiation Temperature Estimated
4 from Post-Irradiation Measurements at 123–413 K." *Journal of Nuclear Materials*.
5 Vol. 386–388. pp. 303–306. 2009.

6 Akiyoshi, M. and T. Yano. "Neutron-Irradiation Effect in Ceramics Evaluated from Macroscopic
7 Property Changes in As-Irradiated and Annealed Specimens." *Progress in Nuclear Energy*.
8 Vol. 50. pp. 567–574. 2008.

9 Akiyoshi, M., I. Takagi, T. Yano, N. Akasaka, and Y. Tachi. "Thermal Conductivity of Ceramics
10 During Irradiation." *Fusion Engineering and Design*. Vol. 81. pp. 321–325. 2006.

11 ASME. "ASME Boiler and Pressure Vessel Code, Section III, Division 2." New York,
12 New York: American Society of Mechanical Engineers. 2007.

13 _____. "ASME Boiler and Pressure Vessel Code, Section XI, Subsection IWL." New York, New
14 York: American Society of Mechanical Engineers. 1995.

15 ASTM International. ASTM C33, "Standard Specification for Concrete Aggregates."
16 West Conshohocken, Pennsylvania: American Society for Testing and Materials. 2013.

17 _____. ASTM C295, "Standard Guide for Petrographic Examination of Aggregates for
18 Concrete." West Conshohocken, Pennsylvania: American Society for Testing and Materials.
19 2012.

20 _____. ASTM C216, "Standard Specification for Facing Brick (Solid Masonry Units made from
21 Clay or Shale)." West Conshohocken, Pennsylvania: American Society for Testing and
22 Materials. 2016.

23 _____. ASTM C289, "Standard Test Method for Potential Alkali-Silica Reactivity of Aggregates
24 (Chemical Method)." West Conshohocken, Pennsylvania: American Society for Testing and
25 Materials. 2007.

26 _____. ASTM C618, "Standard Specification for Coal Fly Ash and Raw or Calcined Natural
27 Pozzolan for Use as a Mineral Admixture in Concrete." West Conshohocken, Pennsylvania:
28 American Society for Testing and Materials. 1998.

29 Atkinson, A. and J.A. Hearne. "Mechanistic Model for the Durability of Concrete Barriers
30 Exposed to Sulphate-Bearing Groundwaters." *Proceedings of the Materials Research Society
31 Conference*. Symposium Proceedings. Pittsburgh, Pennsylvania: Materials Research Society.
32 Vol. 176. pp. 149–156. 1990.

33 Bastidas-Arteaga, E., M. Sanchez-Silva, A. Chateauneuf, and M. Ribas-Silva. "Coupled
34 Reliability Model of Biodeterioration." *Chloride Ingress and Cracking for Reinforced Concrete
35 Structures, Structural Safety*. Vol. 30. pp. 110–129. 2008.

36 Berner, U.R. "Evolution of Pore Water Chemistry During Degradation of Cement in a
37 Radioactive Waste Repository Environment." *Waste Management*. Vol. 12. pp. 201–219.
38 1992.

- 1 Berntz, D.P, M.A. Ehlen, C.F. Ferraris, and E.J. Garboczi. "Sorptivity-Based Service Life
2 Predictions for Concrete Pavements." *7th International Conference on Concrete Pavements,*
3 *Proceedings*, Vol. 1, Orlando, Florida, September 9–13, 2001. International Society for
4 Concrete Pavements. pp. 181–193. 2001.
- 5 Bertolini, L., B. Elsener, P. Pedferri, and R.P. Polder. *Corrosion of Steel in Concrete:*
6 *Prevention, Diagnosis, Repair*, 2nd Edition, Wiley-VCH. p. 409. 2004.
- 7 Bickford, J.H. *An Introduction to the Design and Behavior of Bolted Joints*. 3rd Edition.
8 New York, New York: Marcel Decker. 1995.
- 9 Bouniol, P. and A. Aspart. "Disappearance of Oxygen in Concrete Under Irradiation: The Role
10 of Peroxides in Radiolysis." *Cement and Concrete Research*. Vol. 28. pp. 1,669–1,681. 1998.
- 11 Branson, D.E. *Deformation of Concrete Structures*. New York, New York: McGraw-Hill
12 International Book Company. 1977.
- 13 Cai, S., L. Cremaschi, and A.J. Ghajar. "Moisture Accumulation and its Impact on the Thermal
14 Performance of Pipe Insulation for Chilled Water Pipes in High Performance Buildings."
15 *International Refrigeration and Air Conditioning Conference at Purdue*, Indiana.
16 July 16–19, 2012. 2012
- 17 Cheung, M.M.S., J. Zhao, and Y.B. Chan. "Service Life Prediction of RC Bridge Structures
18 Exposed to Chloride Environments." *Journal of Bridge Engineering*. Vol. 14. pp. 164–178.
19 2009.
- 20 Chopra, O., D. Diercks, R. Fabian, Z. Han, and Y. Liu. "Managing Aging Effects on Dry Cask
21 Storage Systems for Extended Long-Term Storage and Transportation of Used Fuel."
22 FCRD–UFD–2014–000476. ANL–13/15, Rev. 2. Washington, DC: U.S. Department of
23 Energy. 2014.
- 24 Das, B.J. *Principles of Foundation Engineering*. 4th Edition. Pacific Grove, California:
25 Brooks/Cole Publishing Company. 1999.
- 26 Davies, N.F. "Developmental Irradiation Test of SNAP 8 Electrical Components (HF-8),
27 North American Rockwell Corp." NAA-SR- 11924, AT(11- 1)-Gen-8. Canoga Park, California:
28 Atomics International. p. 25. 1966.
- 29 Drimalas T., J.C. Clement, K.J. Folliard, R. Dhole, and M.D.A. Thomas. "Laboratory and Field
30 Evaluations of External Sulfate Attack in Concrete." Austin, Texas: Center for Transportation
31 Research, The University of Texas at Austin. 2010.
- 32 DuraCrete R17. "Final Technical Report, Probabilistic Performance Based Durability Design of
33 Concrete Structures." BE95-1347/R17. CUR, Gouda, The Netherlands. The European
34 Union–Brite EuRam III. 2000.
- 35 EPRI. "Effect of Radiation on Concrete—A Literature Survey and Path Forward."
36 Report 1025584. Palo Alto, California: Electric Power Research Institute. 2012.
- 37 _____. "Aging Effects for Structures and Structural Components (Structural Tools)." Rev. 2,
38 Report 1015078, Palo Alto, California: Electric Power Research Institute. 2007.

- 1 Fagerlund, G. "The International Cooperative Test of the Critical Degree of Saturation Method
2 of Assessing the Freeze/Thaw Resistance of Concrete." *Materials and Structures*. Vol. 10.
3 pp. 231–253. 1977.
- 4 Fan, J., Z. Luo, and Y. Li. "Heat and Moisture Transfer with Sorption and Condensation in
5 Porous Clothing Assemblies and Numerical Simulation." *International Journal of Heat Mass*
6 *Transfer*. Vol. 43. pp. 2,989–3,000. 2000.
- 7 Figg, J. "ASR-Inside Phenomena and Outside Effects (Crack Origin and Pattern)."
8 Concrete Alkali-Aggregate Reactions. E. Patrick, eds. *Proceedings of the 7th International*
9 *Conference*. Grattan-Bellew and E. Patrick, eds. Park Ridge, New Jersey: 7th International
10 Conference Organizers. pp. 152–156. 1987.
- 11 Freskakis, G.N. "Strength Properties of Concrete at Elevated Temperature." Civil Engineering
12 Nuclear Power. Vol. 1. ASCE National Convention. Boston, Massachusetts:
13 American Society of Civil Engineers. 1979.
- 14 Fu, Y. "Delayed ettringite formation in portland cements products." Thesis (Ph.D.). Dept. Civil
15 Engineering. University of Ottawa. Ottawa, Ontario, Canada. 1996.
- 16 Gellrich, G. "Calvert Cliffs Nuclear Power Plant." Letter to U.S. Nuclear Regulatory
17 Commission, Response to Request for Supplemental Information. RE: Calvert Cliffs
18 Independent Spent Fuel Storage Installation License Renewal Application (TAC No. L24475).
19 ADAMS Accession No. ML12212A216. 2012.
- 20 Ghafoori, N. and R. Mathis. "Sulfate Resistance of Concrete Pavers." *Journal of Materials in*
21 *Civil Engineering*. Vol. 9. pp. 35–40. 1997.
- 22 Giannantonio, D.J., J.C. Kurth, K.E. Kurtis, and P.A. Sobecky. "Effects of Concrete Properties
23 and Nutrients on Fungal Colonization and Fouling." *International Biodeterioration and*
24 *Biodegradation*. Vol. 63. pp. 252–259. 2009.
- 25 Glass, G.K. and N.R. Buenfeld. "The Presentation of the Chloride Threshold Level for
26 Corrosion of Steel in Concrete." *Corrosion Science*. Vol. 39. pp. 1,001–1,013. 1997.
- 27 Glauz, D.L., D. Roberts, V. Jain, H. Moussavi, R. Llewellyn, and B. Lenz. "Evaluate the Use of
28 Mineral Admixtures in Concrete to Mitigate Alkali-Silica Reactivity." Report
- 29 FHWA/CA/OR 97-01. Sacramento, California: Office of Materials Engineering and Testing
30 Services. California Department of Transportation. 1996.
- 31 Gutt, W.H. and W.H. Harrison. "Chemical Resistance of Concrete." *Concrete*. Vol. 11.
32 pp. 35–37. 1997.
- 33 Hanson, B., H. Alsaed, C. Stockman, D. Enos, R. Meyer, and K. Sorenson. "Gap Analysis to
34 Support Extended Storage of Used Nuclear Fuel, Rev. 0." FCRD–USED–2011–000136.
35 PNNL–20509. 2011.
- 36 Harmathy, T.Z. "Thermal Properties of Concrete at Elevated Temperatures." *Journal of*
37 *Materials*. Vol. 5. pp. 47–74. 1970.

- 1 Hilsdorf, H.R., J. Kroop, and H.J. Koch. "The Effects of Nuclear Radiation on the Mechanical
2 Properties of Concrete." *Douglas McHenry International Symposium on Concrete and Concrete*
3 *Structures*. American Concrete Institute Publication SP-55. 1978.
- 4 Hobbs, D.W. "Expansion and Cracking in Concrete Associated with Delayed Ettringite
5 Formation." *Ettringite, the Sometimes Host of Destruction*. B. Erlin, ed. SP177
6 Farmington Hills, Michigan: American Concrete Institute International. pp. 159–181. 1999.
- 7 Hu, J., D. Hahn, W. Rudzinski, Z. Wang, and L. Estrada. "Evaluation, Presentation and Repair
8 of Microbial Acid-Produced Attack of Concrete." Report No. FHWA/TX-11/0-6137-1.
9 Texas Department of Transportation Research and Technology Implementation Office. 2011.
- 10 IAEA. "Assessment and Management of Ageing of Major Nuclear Power Plant Components
11 Important to Safety: Concrete Containment Buildings." IAEA-TECDOC-1025. Vienna, Austria.
12 1998.
- 13 Johansen, V. and N. Thaulow. "Heat Curing and Late Formation of Ettringite." ACI SP-177.
14 Bernard Erlin, ed. Farmington Hills, Michigan: American Concrete Institute. pp. 199–206.
15 1999.
- 16 Kontani, O., Y. Ichikawa, A. Ishizawa, M. Takizawa, and O. Sato. "Irradiation Effects on
17 Concrete Structures." *Proceedings of International Symposium on the Ageing Management &*
18 *Maintenance of Nuclear Power Plants*. pp. 173–182. 2010.
- 19 Magniont, C., M. Coutand, A. Bertron, X. Cameleyre, C. Lafforgue, S. Beaufort, and
20 G. Escadeillas. "A New Test Method to Assess the Bacterial Deterioration of Cementitious
21 Materials." *Cement Concrete Research*. Vol. 41. pp. 429–438. 2011.
- 22 Manjeeth K.V. and J.S.K. Rama. "An Experimental Investigation on the Behavior of Portland
23 Cement Concrete and Geopolymer Concrete in Acidic Environment." *SSRG International*
24 *Journal of Civil Engineering*. Vol. 2, Issue 5. 2015.
- 25 Marchand J., M. Pigeon, D. Bager, and C. Talbot. "Influence of Chloride Solution Concentration
26 of Salt Scaling Deterioration of Concrete." *ACI Materials Journal*. pp. 429–435. 1999.
- 27 Marchand, J., E.J. Sellevold, and M. Pigeon. "Deicer Salt Scaling Deterioration—An Overview."
28 SP-145. American Concrete Institute. pp. 1–46. 1994.
- 29 McDonald, J.E. "An Experimental Study of Multiaxial Creep in Concrete." American Concrete
30 Institute Special Publication No. 34. Detroit, Michigan: Concrete for Nuclear Reactors.
31 pp. 732–768. 1972.
- 32 Mehta, P.K. *Concrete, Structure, Properties and Materials*. Upper Saddle River, New Jersey:
33 Prentice-Hall, Inc.. 1986.
- 34 Milde, K., W. Sand, W. Wolff, and E. Bock. "Thiobacilli of the Corroded Concrete Walls of the
35 Hamburg Sewer System." *Journal of General Microbiology*. Vol. 129. pp. 1,327–1,333. 1983.
- 36 Mindess S. and J.F. Young. *Concrete*. Englewood Cliffs, New Jersey: Prentice-Hall, Inc.
37 1981.

- 1 Mori, T., T. Nonaka, K. Tazak, M. Koga, Y. Hikosaka, and S. Nota. "Interactions of Nutrients,
2 Moisture, and pH on Microbial Corrosion of Concrete Sewer Pipes. *Water Research*. Vol. 26.
3 pp. 29–37. 1992.
- 4 Naus, D.J. "A Review of the Effects of Elevated Temperature on Concrete Materials and
5 Components with Particular Reference to the Modular High-Temperature Gas-Cooled Reactor
6 (MHTGR)." ORNL/NRC/LTR-88/2, LTR Report CTP-88-01. Concrete Technology Program.
7 Oak Ridge, Tennessee: Oak Ridge National Laboratory. 1988.
- 8 _____. "Concrete Properties in Nuclear Environment—A Review of Concrete Material Systems
9 for Application to Pre-stressed Concrete Pressure Vessels." ORNL/TM-7632. Oak Ridge,
10 Tennessee: Oak Ridge National Laboratory. 1981.
- 11 NAVFAC. "Foundations and Earth Structures." Design Manual NAVFAC DM-7.02.
12 Alexandria, Virginia: U.S. Naval Facilities Engineering Command. 1996.
- 13 _____. "Soil Mechanics." Design Manual NAVFAC DM-7.01. Alexandria, Virginia: U.S. Naval
14 Facilities Engineering Command. 1986.
- 15 Neville, A.M. and W. Dilger. "Creep of Concrete: Plain, Reinforced and Prestressed."
16 Amsterdam, Holland: North-Holland Publishing Co. 1970.
- 17 NRC. "Expert Panel Workshop on Degradation of Concrete in Spent Nuclear Fuel Dry Cask
18 Storage Systems, Official Transcript of Proceedings." ADAMS Accession Nos. ML15093A003,
19 ML15093A004. Washington, DC: U.S. Nuclear Regulatory Commission. 2015.
- 20 _____. NUREG–1801, "Generic Aging Lessons Learned (GALL) Report." Rev. 2.
21 Washington, DC: U.S. Nuclear Regulatory Commission. 2010a.
- 22 _____. NUREG–1536, "Standard Review Plan for Spent Fuel Dry Storage Systems at a
23 General License Facility." Rev. 1. Washington, DC: U.S. Nuclear Regulatory Commission.
24 2010b.
- 25 _____. NUREG/CR-6900, "The Effect of Elevated Temperature on Concrete Materials and
26 Structures—A Literature Review." Washington, DC: U.S. Nuclear Regulatory Commission.
27 2006.
- 28 _____. NUREG–1557, "Summary of Technical Information and Agreements from Nuclear
29 Management and Resources Council Industry Reports Addressing License Renewal."
30 Washington, DC: U.S. Nuclear Regulatory Commission. 1996.
- 31 _____. NUREG–1522, "Assessment of In-service Conditions of Safety-Related Nuclear Plant
32 Structures." Washington, DC: U.S. Nuclear Regulatory Commission. 1995.
- 33 Nuclear Waste Technical Review Board. "Evaluation of the Technical Basis for Extended Dry
34 Storage and Transportation of Used Nuclear Fuel." Washington, DC: Nuclear Waste Technical
35 Review Board. 2010.
- 36 Okabe, S., O. Mitsunori, I. Tsukasa, and S. Hisashi. "Succession of Sulfur-Oxidizing Bacteria in
37 the Microbial Community on Corroding Concrete in Sewer Systems." *Applied Environmental
38 Microbiology*. Vol. 73. pp. 971–980. 2007.

- 1 Page, C.L. *Nature*. Vol. 297, No. 5,862. pp. 109–115. 1982.
- 2 Pavlik, V. “Corrosion of Hardened Cement Paste by Acetic and Nitric Acids: Part I. Calculation
3 of Corrosion Depth.” *Cement and Concrete Research*. Vol. 24. pp. 551–562. 1994.
- 4 Pavlik, V. and S. Uncik. “The Rate of Corrosion of Hardened Cement Pastes and Mortars with
5 Additive of Silica Fume in Acids.” *Cement and Concrete Research*. Vol. 27. pp. 1,731–1,745.
6 1997.
- 7 Pedneault, A. “Development of testing and analytical procedures for the evaluation of the
8 residual potential of reaction, expansion, and deterioration of concrete affected by ASR.”
9 M.Sc. Memoir. Laval University. Québec City, Canada. p. 133. 1996.
- 10 Perenchio, W.F., I. Kaufman, and R. J. Krause. “Concrete Repair in a Desert Environment.”
11 *Concrete International*. Vol. 13, No. 2. Farmington Hills, Michigan: American Concrete
12 Institute. pp. 23–25. 1991.
- 13 Perez, M., M. Garcia, L. Transversa and M. Stupak. “Concrete Deterioration by Golden
14 Mussels.” *Proceedings of International RILEM Conference on Microbial Impact on Building
15 Materials*. M. Ribas Silva ed. Lisbon, Portugal. pp. 39-47. 2003.
- 16 Phan, L.T. and N.J. Carino. “Fire Performance of High Strength Concrete: Research Needs.”
17 *Advanced Technology in Structural Engineering. ASCE/SEI Structures Congress 2000*.
18 Proceedings. Philadelphia, Pennsylvania. 2000.
- 19 Pigeon, M. “Frost Resistance, A Critical Look.” Concrete Technology, Past, Present, and
20 Future. *Proceedings of V. Mohan Malhotra Symposium*. American Concrete Institute.
21 SP-144. pp. 141–158. 1994.
- 22 Poe, W.L. “Final Long-Term Degradation of Concrete Facilities Presently Used for Storage
23 of Spent Nuclear Fuel and High-Level Waste.” Rev. 1. Tetra Tech NUS, Inc.
24 Aiken, South Carolina: Degradation Mechanisms for Concrete and Reinforcing Steel. 1998.
- 25 Poole, A.B. “Introduction to Alkali-Aggregate Reaction in Concrete.” R.N. Swamy and
26 R. Van Nostrand, eds. New York, New York: *The Alkali-Silica Reaction in Concrete*. 1992.
- 27 Sahu, S. and N. Thaulow “Delayed Ettringite Formation in Swedish Concrete Railroad Ties.”
28 *Cement and Concrete Research*. Vol. 34. pp. 1,675–1,681. 2004.
- 29 Sanchez-Silva, M. and D. Rosowsky. “Biodeterioration of Construction Materials: State of the
30 Art and Future Challenges.” *Journal of Materials in Civil Engineering*. Vol. 20. pp. 352–365.
31 2008.
- 32 Sawan, J. “Cracking Due to Frost Action in Portland Cement Concrete Pavements—A Literature
33 Survey, Concrete Durability.” *Proceedings of Katharine and Bryant Mather International
34 Conference*. American Concrete Institute. SP-100. pp. 781–802. 1987.
- 35 Schiessl, P., P. Bamforth, V. Baroghel-Bouny, G. Corley, M. Faber, J. Forbes, C. Gehlen,
36 P. Helene, S. Helland, T. Ishida, G. Markeset, L. Nilsson, S. Rostam, A.J.M. Siemes, and
37 J. Walraven. “Model Code for Service Life Design.” Lausanne, Switzerland:
38 Fib Bulletin No. 34. 2006

- 1 Schneider, U., U. Diederichs, and C. Ehm. "Effect of Temperature on Steel and Concrete for
2 PCRV's." *Nuclear Engineering and Design*. Vol. 67. pp. 245–258. 1981.
- 3 Shayan, A. and G.W. Quick. "Microscopic Features of Cracked and Uncracked Concrete
4 Railway Sleepers." *ACI Materials Journal*. Vol. 89. pp. 348–361. 1992.
- 5 Sindelar, R.L., A.J. Duncan, M.E. Dupont, P.S. Lam, M.R. Louthan, and T.E. Skidmore.
6 NUREG/CR-7116, "Materials Aging Issues and Aging Management for Extended Storage and
7 Transportation of Spent Nuclear Fuel—Draft Report." Washington, DC: U.S. Nuclear
8 Regulatory Commission. 2011.
- 9 Snead, L.L., S.J. Zinkle, and D.P. White. "Thermal Conductivity Degradation of Ceramic
10 Materials Due to Low Temperature, Low Dose Neutron Irradiation." *Journal of Nuclear
11 Materials*. Vol. 340. pp. 187–202. 2005.
- 12 Snead, L.L., D. Steiner, and S.J. Zinkle. "Measurement of the Effect of Radiation Damage to
13 Ceramic Composite Interfacial Strength." *Journal of Nuclear Materials*. Vol. 191–194.
14 pp. 566–570. 1992.
- 15 Snead, L.L., R. Yamada, K. Noda, Y. Katoh, S.J. Zinkle, W.S. Eatherly, and A.L. Qualls. "In Situ
16 Thermal Conductivity Measurement of Ceramics in a Fast Neutron Environment." *Journal of
17 Nuclear Materials*. Vol. 283–287. pp. 545–550. 2000.
- 18 Stark, D. "The Moisture Condition of Field Concrete Exhibiting Alkali-Silica Reactivity."
19 *CANMET/ACI Second International Conference on Durability of Concrete*. SP-126.
20 Farmington Hills, Michigan. American Concrete Institute. pp. 973–987. 1991.
- 21 Szklarska-Smialowska, Z. *Pitting Corrosion of Metals*. Houston, Texas: National Association
22 of Corrosion Engineers. 1986.
- 23 Tang, L. and P. Sandberg. "Chloride Penetration into Concrete Exposed Under Different
24 Conditions." *Durability of Building Materials and Components 7*. Vol. 1. C. Sjöström, eds.
25 Stockholm, Sweden. 1996.
- 26 Thomas, M.D.A., B. Fournier, and K.J. Folliard. *Alkali-Aggregate Reactivity (AAR) Facts Book*.
27 Austin, Texas: The Transtec Group, Inc. 2013.
- 28 Thomas, M., K. Folliard, T. Drimalas, and T. Ramlochan "Diagnosing Delayed Ettringite
29 Formation in Concrete Structures." *Cement and Concrete Research*. Vol. 38. pp. 841–847.
30 2008.
- 31 Trejo, D., P.D. Figueiredo, M. Sanchez, C. Gonzalez, S. Wei, and L. Li. "Analysis and
32 Assessment of Microbial Biofilm-Mediated Concrete Deterioration." Texas Transportation
33 System. Texas Transportation System. The Texas A&M University System. 2008.
- 34 Ueda, H., Y. Kimachi, S. Ushijima, and K. Shyuttoh. "Deterioration Model of Acid-Rain-Affected
35 Concrete and Test Results of Ordinary and Super Quality Concrete." *26th Conference on Our
36 World in Concrete & Structures*. Singapore. 2001.
- 37 U.S. Department of the Army. "Engineering and Design: Settlement Analysis."
38 EM 1110-1-1904. Washington, DC: U.S. Army Corps of Engineers. September 30, 1990.

- 1 Vafai, K. and S. Sarkar. "Condensation Effects in a Fibrous Insulation Slab. *Journal of Heat*
2 *Transfer*. Vol. 108, No. 8. pp. 667–675. 1986.
- 3 Van Dam, T. and D. Peshkin. "Concrete Aggregate Durability Study." Final Report 5756.
4 Urbana, Illinois: Applied Pavement Technology, Inc. 2009.
- 5 Verbeck, C.J. and P. Klieger. "Studies of Salt Scaling of Concrete." *Highway Research Bulletin*.
6 Vol. 150. pp. 1–17. 1957.
- 7 Vollertsen, J., A.H. Nielsen, H.S. Jensen, W.A. Tove, and H.J. Thorkild. "Corrosion of Concrete
8 Sewers—The Kinetics of Hydrogen Sulfide Oxidation." *Science of the Total Environment*.
9 Vol. 394. pp. 162–170. 2008.
- 10 Wang, C.K and C.G. Salmon. *Reinforced Concrete Design*. 6th Edition. New York, New York:
11 Addison-Wesley. 1998.
- 12 Webster, R.P. and L.E. Kukacka. "Effects of Acid Deposition on Portland Cement Concrete."
13 *Materials Degradation Caused by Acid Rain*. ACS Symposium Series. Vol. 318. pp. 239–249.
14 2009.
- 15 Wei, S., Z. Jiang, H. Liu., D. Zhou, and M. Sanchez-Silva. "Microbiologically Induced
16 Deterioration of Concrete—A Review." *Brazilian Journal of Microbiology*. Vol. 44.
17 pp. 1,001–1,007. 2013.
- 18 Weiss, C. A., Jr., M.C. Sykes, T.S. Poole, J.G. Tom, B.H. Green, B.D. Neeley, and P.G. Malone.
19 "Controlling Sulfate Attack in Mississippi Department of Transportation Structures."
20 Vicksburg, Mississippi: U.S. Army Engineer Research and Development Center. 2009.
- 21 Xu, H. "On the Alkali Content of Cement in AAR." *Concrete Alkali Aggregate Reactions*.
22 *Proceedings of the 7th International Conference*. Grattan-Bellew and E. Patrick, eds.
23 Park Ridge, New Jersey: 7th International Conference Organizers. pp. 451–455. 1987.
- 24 Yano, T., K. Ichikawa, M. Akiyoshi, and Y. Tachi. "Neutron Irradiation Damage in Aluminum
25 Oxide and Nitride Ceramics Up to a Fluence of 4.2×10^{26} n/m²" *Journal of Nuclear Materials*.
26 Vol. 283–287. pp. 947–951. 2000.

1 **3.6 Spent fuel assemblies**

2 The SNF assembly components evaluated in this section include the zirconium-based cladding
3 and fuel assembly hardware that provide structural support to ensure that the spent fuel is
4 maintained in a known geometric configuration. The safety analyses for the ISFSI or DSS rely
5 on the fuel assembly having a specific configuration (e.g., geometric form, a certain number of
6 fuel rods or solid replacement filler rods in the assembly lattice). Although the spent fuel
7 assembly is not an SSC of the ISFSI or DSS, depending on the particular design bases, the
8 spent fuel must remain in its analyzed configuration during the period of extended operation for
9 continuation of the approved design bases. Therefore, for these ISFSIs and DSSs, the
10 condition of the SNF assembly and cladding are within the scope of renewal and are reviewed
11 for aging mechanisms and effects that may lead to a change in the analyzed fuel configuration.

12 The experimental confirmatory basis that low-burnup fuel (≤ 45 gigawatt days per metric ton of
13 uranium (GWd/MTU)) will remain in its analyzed configuration during the period of extended
14 operation was provided in NUREG/CR-6745, "Dry Cask Storage Characterization Project—
15 Phase 1; CASTOR V/21 Cask Opening and Examination" (Bare and Torgerson, 2001), and
16 NUREG/CR-6831, "Examination of Spent PWR Fuel Rods after 15 Years in Dry Storage"
17 (Einziger et al., 2003). This research demonstrated that low-burnup fuel cladding and other
18 cask internals had no deleterious effects after 15 years of storage and confirmed the basis for
19 the guidance on creep deformation and radial hydride reorientation in Interim Staff Guidance
20 (ISG)-11, "Cladding Considerations for the Transportation and Storage of Spent Fuel,
21 Revision 3" (NRC, 2003). In ISG-11, Revision 3, the NRC staff indicated that the spent fuel
22 configuration is expected to be maintained as analyzed in the safety analyses for the ISFSI or
23 DSS, provided certain acceptance criteria (regarding maximum fuel clad temperature and
24 thermal cycling) are met and the fuel is stored in a dry inert atmosphere. The research results
25 in NUREG/CR-6745 and NUREG/CR-6831 support a determination that degradation of
26 low-burnup fuel cladding and assembly hardware should not result in changes to the approved
27 design bases during the first period of extended operation, provided that the cask/canister
28 internal environment is maintained. The U.S. Department of Energy (DOE) is expected to
29 gather similar experimental confirmatory data to support the technical basis for storage of
30 high-burnup (HBU) fuel during the first period of extended operation (EPRI, 2014).

31 The staff reviewed gap assessments for DSS, relevant technical literature, and operating
32 experience from nuclear applications (NRC, 2014a; Chopra et al., 2014; Hanson et al., 2012;
33 Sindelar et al., 2011; NWTRB, 2010) to identify potential degradation mechanisms in
34 consideration of the materials and condition of the SNF at loading and the environment in dry
35 storage. The SNF cladding materials are zirconium-based alloys. The primary components of
36 the fuel assembly hardware are spacer grids, end fittings, guide tubes (PWR only), and
37 assembly channels (BWR only). The materials of construction for these components include
38 zirconium-based alloys, nickel alloys, and stainless steel. The staff's assessment of the
39 condition of the SNF assembly at loading considered changes to the fuel pellets and the
40 zirconium-based cladding during reactor service, including hydrogen absorption by the cladding,
41 swelling of the fuel pellets, increased rod pressurization due to helium and fission gas release,
42 and pellet-cladding interactions. The storage environment is helium or an alternative cover gas
43 in high radiation and temperature. A minimal amount of water (about 0.43 gram mole) is also
44 considered to be retained inside the cask/canister (NRC, 2010). This moisture content is based
45 on a design-basis drying process that evacuates the cask/canister to less than or equal to 3 torr
46 [0.06 psi] and maintains a constant pressure for 30 minutes before closure.

1 The aging mechanisms considered for high burnup zirconium-based cladding (i.e., average
2 assembly burnups exceeding 45 GWd/MTU) include hydride reorientation, delayed hydride
3 cracking, thermal and athermal (low-temperature) creep and localized mechanical overload.
4 These mechanisms are primarily driven by cladding hoop stresses, which are lower in low
5 burnup fuel. The technical bases for these mechanisms (Sections 3.6.1.1–3.6.1.5) considered
6 cladding hoop stresses for high burnup fuel, therefore these discussions are considered
7 bounding to low burnup fuel. In addition, the demonstration program discussed in
8 NUREG/CR-6745 (Bare and Torgerson, 2001) and NUREG/CR-6831 (Einziger et al., 2003)
9 provided confirmation that hydride reorientation and creep will not compromise the configuration
10 of low burnup fuel during the renewal period. Other aging mechanisms considered for both low
11 and high burnup zirconium-based cladding include radiation embrittlement, fatigue, oxidation,
12 pitting corrosion, galvanic corrosion, and SCC and MIC. Of these potential mechanisms, MIC
13 was not considered to be applicable, as the aging mechanism is not expected to be operable
14 under the inert atmosphere of dry storage. Detailed discussions regarding each of the potential
15 aging mechanisms for zirconium-based cladding are provided in Section 3.6.1.

16 The degradation mechanisms considered for the assembly hardware include creep, fatigue,
17 hydriding, general corrosion, SCC, and radiation embrittlement. Detailed discussions regarding
18 each of these applicable aging mechanisms for assembly hardware are provided in
19 Section 3.6.2.

20 **3.6.1 Cladding materials**

21 *3.6.1.1 Hydride reorientation (high burnup fuel)*

22 In reactor service, the zirconium-based fuel cladding absorbs hydrogen, which leads to the
23 precipitation of hydride platelets as the dissolved hydrogen exceeds the solubility limit of the
24 cladding. The primary source of the hydrogen is water-side corrosion (oxidation) of the cladding
25 (Hanson et al., 2012; IAEA, 1993). The total concentration of hydrogen absorbed by the
26 cladding (i.e., dissolved in the zirconium matrix and in precipitated hydrides) increases with
27 burnup and varies axially across the fuel rods. For burnups above 45 GWd/MTU and up to
28 62 GWd/MTU (the current NRC licensing limit), the total hydrogen content for Zircaloy-2 is
29 expected to be in the range of 260–300 weight parts per million [wppm] (NRC, 2015a;
30 Geelhood and Luscher, 2014), 200–1,200 wppm for Zircaloy-4 (Mardon et al., 2010;
31 Thomazet et al., 2005; King et al., 2002; Bossis et al., 2007; Hanson, 2016), ≤ 100 wppm for
32 M5[®] (King et al., 2002; Bossis et al., 2007; Mardon et al., 2010; Thomazet et al., 2005,
33 Billone, 2013, Hanson, 2016), and up to 550± 300 wppm for ZIRLO[™] (Billone et al., 2013,
34 Billone et al., 2015). When discharged from the reactor and during wet storage, the hydride
35 platelets are mostly oriented in the circumferential-axial direction, with a smaller fraction
36 oriented in the radial-axial direction.

37 Once the SNF assemblies are removed from wet storage and loaded into a DSS, the
38 cask/canister cavity is vacuum dried and backfilled with an inert gas. During vacuum drying, the
39 temperature of the SNF assemblies and the temperature-dependent solubility limit of hydrogen
40 in the cladding will also increase. As a result, some of the hydrides present in the cladding will
41 redissolve as hydrogen. The amount of dissolved hydrogen will depend on the peak cladding
42 temperature during the vacuum drying operations, which, per ISG-11, Revision 3 (NRC, 2003),
43 is not to exceed 400 degrees C [752 degrees F] for HBU fuel. For example, the maximum
44 dissolved hydrogen at 400 degrees C [752 degrees F] is approximately 200 wppm based on
45 representative solubility correlations (Kammenzind et al., 1996; Kearns et al., 1967). Once the
46 loaded cask/canister is dried and backfilled, the cladding temperature will decrease over time,

1 and upon a sufficient temperature drop (~65 degrees C [117 degrees F]), some of the hydrogen
2 in solution will reprecipitate as new hydrides. During this process, the orientation of these
3 precipitated hydrides may change from the circumferential-axial to the radial-axial direction.
4 The degree of reorientation is primarily driven by the metallurgical microstructure of the cladding
5 alloy and the cladding hoop stresses during drying operations and subsequent cooling, which
6 are determined by the rod internal pressure at a given gas temperature.

7 Cladding with a high concentration of radial hydrides (determined by the DSS drying conditions)
8 has been shown to have reduced ductility under pinch-load stresses at sufficiently low
9 temperatures, thereby affecting the ability to retrieve the HBU fuel (Billone et al., 2013; Aomi et
10 al., 2008). The degradation of the mechanical properties at a particular temperature (described
11 as the “ductile-to-brittle transition temperature” or DBTT) depends on the interconnectivity and
12 number density of radial hydrides (as determined by their length, distribution, and orientation),
13 and the thickness of the outer-surface hydride rim. This phenomenon has led the staff to
14 express concern about potential cladding failures when subjected to pinch-load stresses higher
15 than the fuel’s mechanical limit, if the cladding temperature decreases below the corresponding
16 DBTT (NRC, 2015b). Therefore, as the cladding cools down during the 60-year timeframe, the
17 extent of radial hydride reorientation and the DBTT are important for evaluating the cladding
18 performance and ensuring that the HBU fuel remains in the analyzed configuration.

19 The primary driving force for radial hydride reorientation is the cladding hoop stresses, which
20 are determined by the peak cladding temperature during drying operations. A review indicates
21 that there is no consensus in the literature on minimum level or threshold hoop stresses needed
22 to reorient hydrides for a given cladding alloy and temperature, as discussed in the following
23 references:

- 24 • Zircaloy-4: Data from Chung (2004), Daum et al. (2006), and Chu et al. (2008) suggest
25 that the threshold hoop stress for hydride reorientation in Zircaloy-4 is about 90 MPa
26 [13 ksi] for peak temperatures at or near 400 degrees C [752 degrees F] for both
27 irradiated and unirradiated rods. Other data obtained from irradiated cladding (Einziger
28 and Kohli, 1984; Cappelaere, et al., 2001; and Goll, et al., 2001) suggest that hoop
29 stresses greater than 120 MPa [17 ksi] may be required. Most recently, Kim et al
30 (2015a) showed threshold stresses for hydride reorientation in unirradiated Zircaloy-4 of
31 60 ± 5 MPa
32 $[8.7 \pm 0.7$ ksi] at 400 degrees C [752 degrees F], 68 ± 5 MPa $[9.8 \pm 0.7$ ksi] at
33 335 degrees C [635 degrees F], 75 ± 6 MPa $[10.9 \pm 0.9$ ksi] at 300 degrees C
34 [572 degrees F], and 90 ± 6 MPa $[13.0 \pm 0.9$ ksi] at 235 degrees C [455 degrees F].
35 Kamimura (2010) also reported a threshold stress for Zircaloy-4 of about 100 MPa
36 [16 ksi] at 275 degrees C [527 degrees F] for a nominal burnup of 48 GWd/MTU.
- 37 • Zircaloy-2: Kamimura (2010) reported a threshold hoop stress of 70 MPa [10 ksi] for
38 Zircaloy-2 (no zirconium liner) of nominal burnup of 40 GWd/MTU at 200 degrees C
39 [392 degrees F], and 70 MPa [10 ksi] for Zircaloy-2 (with zirconium liner) of nominal
40 50 GWd/MTU and 55 GWd/MTU burnups at 300 degrees C [572 degrees F].
- 41 • Advanced alloys: Kamimura (2010) reported a threshold stress of 90 MPa [13 ksi] for
42 ZIRLO™ at 250 degrees C [482 degrees F] for a nominal burnup of 55 GWd/MTU.
43 Billone et al. (2013) reported reorientation of M5® cladding at their lowest studied hoop
44 stress of 90 MPa [16 ksi] for a peak cladding temperature of 400 degrees C
45 [752 degrees F] and nominal burnup of 68 GWd/MTU.

1 These threshold hoop stresses for hydride reorientation were compared to estimated hoop
2 stresses for representative BWR and PWR fuel assemblies. Raynaud and Einziger (2015)
3 estimated the hoop stresses for 10×10 BWR and 17×17 PWR fuel assemblies as a function
4 of decay gas release and fuel pellet swelling, which accounted for decay gas released to the
5 pellet-clad gap. The maximum calculated hoop stress during drying operations for the BWR
6 cladding was approximately 40 MPa [5.8 ksi] at a peak cladding temperature close to
7 400 degrees C [752 degrees F]. Similarly, the maximum calculated hoop stress during drying
8 operations for PWR cladding was approximately 100 MPa [14.5 ksi] at 400 degrees C
9 [752 degrees F], which rapidly decays and falls well below 50 MPa after a few decades in dry
10 storage. These calculations did not account for ZIRLO™-clad integral fuel burnable absorber
11 (IFBA) rods with hollow and solid blanket pellets; however, these rods are expected to
12 experience higher maximum hoop stresses (Bratton et. al, 2015). Since the calculated hoop
13 stresses exceed the experimental values in the literature for when radial hydride reorientation
14 was observed, the staff considers that the radial hydride precipitation is credible in both in BWR
15 and PWR fuel claddings in dry storage.

16 The cladding alloy and corresponding fabrication process are also important factors for defining
17 the extent of hydride reorientation. Two predominant cladding microstructures are produced
18 during fabrication: (1) recrystallized annealed (RXA) and (2) cold worked stress relieved
19 (CWSR) annealed. Zircaloy-4 (PWR) and ZIRLO™ (PWR) are generally CWSR, whereas
20 Zircaloy-2 and M5® are RXA. Because hydrides tend to precipitate in the grain boundaries,
21 RXA claddings are more susceptible to hydride reorientation, since these cladding types have a
22 larger fraction of grain boundaries in the radial direction (equiaxed grains) relative to CWSR
23 claddings (which have more elongated grains).

24 The staff also considered the effect of the cladding cooling rate on the degree of hydride
25 reorientation. The cooling rate post-drying and under dry storage is expected to be in the range
26 of 10^{-3} to 10^{-5} degrees C/hr [1.8×10^{-3} to 1.8×10^{-5} degrees F/hr]. Most of the experimental
27 studies reported in the literature have used cooling rates in the range of 0.6–30 degrees C/hr
28 [1.08–54 degrees F/hr] (Aomi et al., 2008). However, an analysis of ductility data collected at
29 different cooling rates in Aomi et al. does not show a clear trend. Chan (1996) also developed a
30 micromechanical model to determine the effect of slow cooling rates on hydride reorientation
31 and morphology, including volume fraction of both radial and circumferential hydrides and
32 continuity of the hydride network. Using experimental data to validate the model, Chan
33 concluded that the cooling rate exerts no direct influence on radial hydride precipitation; instead,
34 hydride orientation is dictated by the cladding stresses during hydride precipitation, regardless
35 of the cooling rate. Therefore, the staff concludes that the slow cooling rates experienced post-
36 drying and during dry storage are not expected to inhibit the precipitation of radial hydrides.

37 Available DBTT data on HBU fuel cladding samples with radial hydrides have been obtained
38 under conservative conditions and acceptance criteria (e.g. testing was performed on defueled
39 samples, which do not account for the composite pellet-clad mechanical behavior) (Fuketa et
40 al., 2003; Billone et al., 2013; Aomi et al., 2008). For example, Billone et al. showed that
41 Zircaloy-4, ZIRLO™, and M5® cladding samples subjected to a radial hydride reorientation
42 treatment exhibited lower ductility under pinch-load stresses at low relative temperatures
43 (less than 200 degrees C [392 degrees F]). The radial hydride treatment was designed to
44 simulate drying and storage conditions (i.e., peak cladding temperature of 400 degrees C
45 [752 degrees F] and peak hoop stresses of ~110 MPa [16.0 ksi] and ~140 MPa [20.3 ksi]).
46 General conclusions from Billone et al. were that: (1) the DBTT generally increases with
47 increasing hoop stresses (i.e., the degradation of cladding mechanical properties shifts to higher
48 cladding temperature), (2) both the susceptibility to radial hydride precipitation and degradation

1 of mechanical properties depend on cladding type and initial hydrogen content, and
2 (3) depending on the cladding and test conditions, the DBTT can occur at temperatures in the
3 range of approximately 20 degrees C to 185 degrees C [68 to 328 degrees F]. The results for
4 as-irradiated Zircaloy-4 are consistent with studies by Wisner and Adamson (1998) and Bai et
5 al. (1994).

6 Considering the hydrogen content, peak drying temperatures, and corresponding hoop stresses,
7 the staff concludes that hydride reorientation in zirconium-based HBU cladding is credible
8 during the 60-year timeframe. Further, depending on the specific fuel contents, it is possible for
9 some of the cladding to reach temperatures near or below the DBTTs reported in the literature.
10 Therefore, degradation of mechanical properties during pinch-type stresses due to hydride
11 reorientation is considered a credible aging mechanism for HBU fuel claddings.

12 The degradation of mechanical properties due to hydride reorientation is only expected to
13 potentially compromise the ability to maintain the analyzed fuel configuration during pinch-type
14 loads. These loads are only expected during fuel retrieval operations, if the design bases of the
15 DSS or ISFSI rely on retrievability of the HBU fuel on a single-assembly basis. Pinch-type loads
16 are not expected to be present during normal, off-normal, and accident conditions of storage.
17 More specifically, the tensile stress field associated with potential inertial rod bending during
18 storage is expected to be parallel to both radial and circumferential hydrides and not expected
19 to compromise the structural integrity of the cladding. The NRC is sponsoring confirmatory
20 research to this effect at Oak Ridge National Laboratory, and the results will be
21 publicly-available soon (see Wang and Wang (2015) for details on the experimental protocol).

22 Until the Oak Ridge data is publicly-available, the staff has proposed two alternatives for
23 demonstrating that the safety analyses pertaining to the analyzed spent fuel configuration will
24 not be compromised by the effects of hydride reorientation. The first approach relies on the
25 applicant performing a defense-in-depth analysis, assuming credible reconfiguration based on
26 1 percent fuel failure for normal conditions of storage, 10 percent failure for off-normal
27 conditions of storage, and 100 percent or other justifiable value for accident conditions. The
28 staff has issued a generic consequence analysis for both vertical and horizontal storage
29 configurations in NUREG/CR-7203 (Scaglione et al., 2015), which can be used by applicants in
30 the development of their defense-in-depth analysis. A second approach relies on the evaluation
31 of data from a demonstration (surrogate) program consistent with the guidance in Appendix D of
32 NUREG-1927, Revision 1 (NRC, 2016). For example, destructive examination from the
33 DOE/EPRI cask demonstration project (EPRI, 2014) may be used as confirmation that hydride
34 reorientation has not compromised the ability to retrieve the spent fuel on a single-assembly
35 basis. An example AMP consistent with the guidance in Appendix D of NUREG-1927,
36 Revision 1 is provided in Chapter 5.

37 3.6.1.2 *Delayed hydride cracking (high burnup fuel)*

38 Delayed hydride cracking (DHC) is a time-dependent mechanism traditionally thought to occur
39 by the diffusion of hydrogen to an incipient crack tip (notch, flaw) in the cladding, followed by
40 nucleation, growth, and subsequent fracture of the precipitated hydrides at the crack tip
41 (Hanson et al., 2012). Hydrogen dissolved in the cladding (see Section 3.6.1.1) can diffuse up a
42 stress gradient in the crystalline lattice, or into the stress field at the core of an edge dislocation
43 (Cox, 1997). The concentration gradient established by the stress gradient may lead to
44 hydrogen supersaturation (i.e., solubility limit being exceeded) leading to the precipitation of
45 hydrides at the crack tip. The precipitated hydride will continue to grow by the dissolution of
46 hydrides in the low-stress regions of the material and by the continued diffusion of hydrogen up

1 the stress gradient. Once the hydride reaches a critical size, it will crack and propagate to the
2 end of the hydride, where it will blunt. The cycle could then repeat, until the crack propagates
3 through the thickness of the material. DHC of spent fuel cladding has been studied under
4 thermal transients representative of reactor operation (Kubo, 2012; Kim, 2009b) and
5 representative of dry storage (Sasahara and Matsumura, 2008; EPRI, 2002).

6 Requisite conditions for DHC are the presence of: (i) hydrides, (ii) existing crack tips
7 (notch, flaws) that act as initiating sites, and (iii) sufficient cladding hoop stresses. Regarding
8 requisite hydrides, a threshold for crack initiation cannot be readily defined. Simpson and Ells
9 (1974) observed DHC with hydrogen concentration as little as 10 ppm in Zr-2.5 percent Nb
10 cladding, although testing was performed at room temperature (i.e., a much lower temperature
11 than those expected during the renewal period). Similarly, Coleman et al. (2009) were able to
12 induce DHC in Zircaloy-4 at 200 wppm of hydrogen. Regarding requisite existing (incipient)
13 crack tips, EPRI (2002) estimated the maximum initial depth of existing crack tips to be 140 μm
14 [5.5 mils] or approximately 28 percent of the remaining wall of a typical 17×17 PWR cladding
15 with 600 μm [23.6 mils] of original cladding thickness, and 100 μm [4 mils] of oxidation during its
16 exposure in the reactor. Conversely, Raynaud and Einziger (2015) estimated the maximum
17 initial depth of existing crack tips to be 120 μm [4.7 mils] for a cladding oxide thickness of
18 100 μm [4 mils]. Regarding requisite hoop stresses for crack initiation, the mechanism requires
19 that the stress intensity factor at the crack tip exceed a threshold value, denoted as K_{IH} .

20 Most DHC studies have been performed under thermal transients representative of reactor
21 operation, primarily on CANDU pressure tubes (Zr-2.5 percent Nb) and Zircaloy-2 cladding.
22 Chan (2013) conducted an extensive literature review of experimentally determined K_{IH} values
23 for DHC crack initiation. In that review, K_{IH} values for Zircaloy-2 are in the range of 5–
24 14 $\text{MPa}\sqrt{\text{m}}$ [4.55–12.74 $\text{ksi}\sqrt{\text{in}}$] at 25–300 degrees C [77–572 degrees F], and in the range of 5–
25 10 $\text{MPa}\sqrt{\text{m}}$ [4.55–9.10 $\text{ksi}\sqrt{\text{in}}$] for Zr-2.5 percent Nb cladding at 75–300 degrees C [167–572
26 degrees F] (Chan, 2013, Figures 2 and 3). Kubo et al. (2012) also compiled K_{IH} values for
27 Zircaloy-2 in the range of 3–13 $\text{MPa}\sqrt{\text{m}}$ [2.73–11.8 $\text{ksi}\sqrt{\text{in}}$]. Kim (2009a) also measured a K_{IH}
28 value of 2.5 $\text{MPa}\sqrt{\text{m}}$ [2.28 $\text{ksi}\sqrt{\text{in}}$] for Zr-2.5 Nb cladding at 160 degrees C [320 degrees F].
29 Based on the available data, the staff considered a reference K_{IH} value of 5.0 $\text{MPa}\sqrt{\text{m}}$
30 [2.73 $\text{ksi}\sqrt{\text{in}}$] to be reasonable for determining the potential for DHC initiation.

31 Raynaud and Einziger (2015) estimated the cladding hoop stresses while conservatively
32 accounting for release of fission gases and decay gases during storage, including stresses due
33 to radiation-induced pellet swelling during storage. Raynaud and Einziger concluded that DHC
34 cannot occur for a K_{IH} of 5 $\text{MPa}\sqrt{\text{m}}$ [4.55 $\text{ksi}\sqrt{\text{in}}$], because the flaw size needed to induce DHC is
35 much larger than the initial depth of potential existing cracks (120 μm [4.7 mils]). The estimated
36 critical flaw size needed to initiate DHC in BWR fuel cladding is larger than 50 percent of the
37 cladding thickness for 300 years of dry storage. For PWR cladding, the critical flaw size is
38 larger than 30 percent of the cladding thickness for the first 5 years of the dry storage and larger
39 than 50 percent of the cladding thickness beyond the first 5 years up to 300 years of dry
40 storage. The calculations in Raynaud and Einziger did not account for the hoop stresses in
41 ZIRLO™-clad IFBA rods with hollow and solid blanket pellets, which are expected to be higher
42 than standard rods (Bratton et al., 2015). Therefore, the staff performed similar calculations to
43 those in Raynaud and Einziger for IFBA rods, assuming a K_{IH} value of 5 $\text{MPa}\sqrt{\text{m}}$ [2.73 $\text{ksi}\sqrt{\text{in}}$]
44 and a conservative IFBA-rod hoop stress of 130 MPa [21.75 ksi]. These calculations show that
45 the critical flaw size for the PWR cladding is still larger than 30 percent of the cladding thickness
46 for the first 5 years of dry storage and larger than approximately 45 percent of the cladding
47 thickness beyond the first 5 years up to 300 years of dry storage. Therefore, the staff concludes
48 that the critical flaw size needed to induce DHC, in both standard and IFBA rods, is much larger

1 than the initial depth of potentially existing cracks (120 μm [4.7 mils]). The staff considers that
2 the hoop stress value assumed for IFBA rods is adequately conservative for this calculation,
3 since a limited (less than 1 percent) population of the rods is expected to experience these
4 pressures (Bratton et al., 2015). In addition, most design-bases peak cladding temperatures are
5 well below the limit defined in ISG-11, Revision 3 (i.e., 400 degrees C [752 degrees F]), which
6 would considerably decrease the cladding hoop stresses. Therefore, the assumptions and
7 analyses discussed above are considered reasonably bounding and indicate that DHC is not a
8 credible aging mechanism during the 60-year timeframe.

9 The staff also considered a DHC model proposed by Kim et al. (2008, 2009b), which evaluated
10 cladding absent thermal cycling, where multiple parameters including creep deformation,
11 cladding burnup, solvus hysteresis, and the δ -to- γ hydride phase transition were analyzed. This
12 model, still under review by the international DHC research community (NRC, 2014a), suggests
13 that K_{IH} may be reduced (i) upon cooling below 180 degrees C [356 degrees F] (due to a
14 hydride phase transformation from the γ to δ phase) and (ii) if there are sufficient stresses and
15 stress risers in the rod (e.g., residual stresses at the end cap weld region, incipient cracks due
16 to fuel-cladding interaction). Thermal gradients may also affect the kinetics of hydride
17 precipitation. The staff reviewed this study, in light of the assumptions made in the previous
18 discussion. However, Kim does not quantify K_{IH} values; therefore, adequate conclusions cannot
19 be made with respect to threshold stresses. The NRC (2014a) and Hanson et al, (2012)
20 summarized Kim's work and proposed additional research for confirmation.

21 Finally, the staff considered the contribution of cladding stresses due to pellet-clad bonding and
22 its potential to facilitate DHC initiation (Wang, 2014a,b). The previously-discussed Raynaud
23 and Einziger (2015) study did not account for potential stress concentration effects due to
24 pellet-pellet interfaces and pellet fragment-to-fragment friction forces that could result in more
25 severe pellet-to-cladding mechanical interaction (PCMI) than for a perfectly cylindrical pellet
26 (as assumed in the paper). Recently, Ahn et al. (2013) estimated stress concentrations from
27 pellet-clad mechanical stresses due to the radiation-induced pellet swelling up to 100 years,
28 independent of hoop stresses due to continued fission and decay gas release. The work
29 estimated that, for HBU fuel, the average pellet-swelling-induced PCMI stress concentration
30 was on the order of 200 MPa [29 ksi] locally.¹ Literature indicates that radiation-induced pellet
31 swelling is expected reach its maximum value beyond the 60-year timeframe (Rondinella et al.,
32 2010a,b; 2012). Therefore, the staff does not have evidence that the potential for high PCMI
33 stress concentrations due to radiation-induced pellet swelling would facilitate DHC crack
34 initiation until past the first renewal period.

35 Based on the above analyses and discussion, the staff concludes that delayed hydride cracking
36 of the zirconium-based cladding is not credible during the 60-year timeframe and therefore,
37 aging management is not required.

38 3.6.1.3 *Thermal creep (high burnup fuel)*

39 Creep is the time-dependent deformation of a material under stress. Creep in zirconium-based
40 cladding is caused by the hoop stresses from the rod internal pressure at a given fuel
41 temperature. Therefore, the mechanism is expected to be self-limiting, due to the decreasing
42 temperatures and creep-induced volume expansion, which results in lower internal rod

¹For low-burnup fuel, pellet expansion stresses will be minimal, because the gap between the cladding and the pellet will accommodate the swelling.

1 pressures with time. Excessive creep of the cladding during dry storage could lead to thinning,
2 hairline cracks, or gross ruptures (Hanson et al., 2012), which may affect the ability to safely
3 retrieve the HBU fuel on a single-assembly basis (if required by the design bases).

4 The main driving force for cladding creep at a given temperature is the hoop stress caused by
5 internal rod pressure, which accounts for the fission and decay gases released to the interspace
6 between the fuel and cladding. Fuel pellet swelling also may result in localized stresses due to
7 the mechanical interaction between the cladding and the fuel. Pellet swelling may occur due to
8 (i) the incorporation of soluble and insoluble solid fission products in the fuel matrix, (ii) the
9 formation of intra- and intergranular fission gas bubbles, particularly in the hot interior region of
10 a fuel pellet, and (iii) the formation of a large number of small gas bubbles in the fine-grained
11 ceramic structure that builds inward from the outer pellet surface for HBU fuel.

12 Raynaud and Einziger (2015) estimated the transient cladding hoop stresses during dry storage
13 for typical 10×10 BWR and 17×17 PWR fuel assemblies. These estimates accounted for a
14 credible release of fission and decay gases to the fuel-cladding interspace, pellet swelling, and
15 fuel and cladding temperature decay with time. The study reported peak cladding hoop
16 stresses less than 50 MPa [7.25 ksi] for BWR and less than 100 MPa [14.5 ksi] for PWR fuel
17 assemblies. Raynaud and Einziger used these hoop stress estimates to calculate cumulative
18 cladding strains for the representative assemblies over a 60-year period of dry storage. The
19 authors reported a maximum cladding strain of 0.54 percent for the representative 10×10 BWR
20 fuel cladding and 1.04 percent for the representative 17×17 PWR fuel cladding. However,
21 these calculations did not account for the hoop stresses in ZIRLO™-clad IFBA rods with hollow
22 and solid blanket pellets, which are expected to be higher than those for standard rods (Bratton
23 et al., 2015). Therefore, the staff performed calculations to estimate the cladding strain for IFBA
24 rods using the Raynaud and Einziger approach. Using a conservatively bounding hoop stress
25 of 150 MPa [21.75 ksi], the maximum cladding strain was estimated to be near 2.1 percent. The
26 elastic strain limit for various zirconium-based cladding alloys with circumferential hydrides is
27 less than 1 percent (Geelhood et al., 2008) and is expected to be lower for cladding containing
28 both circumferential and radial hydrides. Therefore, the staff concludes that the cladding in both
29 standard and IFBA fuel rods is expected to undergo creep strains during the 60-year timeframe.

30 The staff has discussed the potential for creep deformation in ISG-11, Revision 3 (NRC, 2003),
31 which includes acceptance criteria (regarding maximum fuel clad temperature during dry
32 storage operations and adequate thermal cycling limits) to provide reasonable assurance that
33 the spent fuel assemblies will remain in the configuration analyzed in the approved design
34 bases. The references cited in ISG-11, Revision 3, provide experimental evidence that cladding
35 failures are not expected for creep strains below 2 percent. These references provide support
36 that gross ruptures of the cladding are unlikely due to creep during dry storage, because the
37 creep-induced strain is expected to be near or less than 2 percent for the majority of the
38 cladding alloys and close to 2 percent for the ZIRLO™-clad IFBA rods. For example, no failures
39 were observed for creep strains below 2 percent strain for in-creep tests at temperatures
40 between 250 and 400 degrees C [482 and 752 degrees F] for Zircaloy cladding irradiated up to
41 burnup of 64 GWd/MTU (Spilker et al., 1997; Goll et al., 2001; EPRI, 2002). In addition,
42 Bouffioux and Rupa (1998) conducted various cladding creep tests with unirradiated,
43 prehydrided, stress-relief annealed low-Sn Zircaloy-4 PWR cladding tubes, with hydrogen levels
44 in the range of 100–1,100 wppm. The authors observed gross ruptures of the cladding only
45 after creep strains exceeding 8 percent. Tsai and Billone (2003) also tested irradiated
46 stress-relief annealed Zircaloy-4 with varying levels of hydrogen levels at various temperature
47 and hoop stresses, which did not reveal cladding failures at a strain of 5.83 percent. More

1 recent data on optimized ZIRLO™ by Pan et al. (2013) also indicate a plastic strain range in the
2 same range as Zircaloy.

3 The staff concludes that thermal creep of zirconium-based cladding is credible during the
4 60-year timeframe. However, due to the high creep capacity of zirconium-based alloys, thermal
5 creep is not expected to result in cladding failures and reconfiguration of the fuel, if the
6 approved design bases are consistent with the acceptance criteria in ISG-11, Revision 3. The
7 staff recognizes that the experimental evidence used in support of ISG-11, Revision 3, is based
8 on short-term testing. Therefore, the staff issued guidance in Appendix D of NUREG–1927,
9 Revision 1 (NRC, 2016) for the use of a demonstration program to confirm these expected fuel
10 conditions after a substantial storage period (~10 years). The program would provide
11 confirmation for accelerated cladding creep testing used as basis for the guidance
12 recommendation for the maximum temperature in ISG-11 (NRC, 2003), and that sufficient creep
13 capacity exists for the renewal period. For example, non-destructive and destructive
14 examination from the DOE/EPRI cask demonstration project (EPRI, 2014) may be used as
15 confirmation that the design-basis fuel remains in the analyzed configuration and that sufficient
16 creep margin exists for the first renewal period. An example AMP consistent with the guidance
17 in Appendix D of NUREG–1927, Revision 1 is provided in Chapter 5.

18 3.6.1.4 *Low-temperature creep (high burnup fuel)*

19 Low-temperature creep (also called “athermal creep”) may occur when sustained hoop stresses
20 operate on the cladding material at or near ambient temperature (NRC, 2014a). Various
21 athermal creep mechanisms have been proposed at low stresses (e.g., Nabarro-Herring, Coble,
22 and Harper-Dorn creep mechanisms) (Murty, 2000), although there is no evidence or literature
23 information to support that these will be operational on zirconium-based alloys. However, the
24 literature shows that low-temperature creep has been shown to occur in titanium and its alloys,
25 which leads to deformation twinning (Jaworski and Ankem, 2006). Since both titanium and
26 zirconium have the same crystalline structure (hexagonal close packed crystalline), the
27 zirconium-based cladding was reviewed for its susceptibility to low-temperature creep.

28 In materials such as α and α - β titanium alloys, which are comparable to the zirconium-based
29 alloys used for fuel cladding, low-temperature creep has been observed when tensile stresses
30 exceed 25 percent of the yield strength (Ankem and Wilt, 2006). For example, Ankem and Wilt
31 reported a threshold stress in the range of 25–50 percent of the yield stress for Ti Grade 7, and
32 35–60 percent of the yield stress for Ti Grade 24. The yield strength of the irradiated
33 zirconium-based cladding at low temperatures (550–1,000 MPa [79.8–145 ksi]; Geelhood et al,
34 2008; Forgeaud, et al., 2009; Cazalis et al., 2005) is expected to be close to the yield strength of
35 Ti Grade 24 (825 MPa [119.6 ksi]) and well above the yield strength of Ti Grade 7 (275 MPa
36 [39.9 ksi]) (Ibarra et al., 2007). Therefore, the staff considered the results in Ankem and Wilt to
37 provide reasonable acceptance criteria for determining if low-temperature creep is a credible
38 aging mechanism in the 60-year time frame.

39 The main sources of sustained hoop stresses at low temperatures are expected to be the rod
40 internal pressure and pellet-cladding mechanical interaction. Raynaud and Einziger (2015)
41 estimated the cladding hoop stresses after 300 years of storage to be approximately 25 MPa
42 [3.62 ksi] and 35 MPa [5.07 ksi] for representative BWR and PWR fuel cladding, respectively.
43 These estimates accounted for a credible release of fission and decay gases to the
44 fuel-cladding interspace, pellet swelling, and fuel and cladding temperature. The hoop stresses
45 for IFBA rods are conservatively expected to be around or less than 75 MPa [10.87 ksi]
46 (Bratton et al., 2015). These hoop stress estimates are all less than 25 percent of the yield

1 strength of zirconium-based cladding (i.e., below the expected range of 550–1,000 MPa [79.8–
2 145 ksi] near ambient temperature for cladding with circumferential hydrides only
3 (Geelhood et al., 2008; Fourgeaud, et al. 2015; Cazalis et al., 2005)). Further, more recent data
4 (Kim et al., 2015a, 2015b) suggest that, even with the potential decrease in yield strength due to
5 radial hydrides (which conservatively does not account for a potential increase in yield strength
6 due to irradiation), the hoop stresses in the cladding are still maintained below 25 percent of the
7 yield strength of irradiated cladding with both circumferential and radial hydrides.

8 Raynaud and Einziger acknowledged that the low-temperature creep models are not
9 programmed into FRAPCON-DATING, which the authors used to predict the elevated
10 temperature cladding creep (see Section 3.6.1.3). The authors noted that extrapolations of the
11 high-temperature cladding creep model results in immeasurably small values of cladding strains
12 at low temperature. However, the lack of cladding creep beyond 50 years (corresponding to
13 temperatures below approximately 200 degrees C [392 degrees F]) results in smaller strains
14 being predicted in these calculations. Therefore, the calculated cladding hoop stresses are
15 conservative when compared to the 25-percent criteria, as athermal creep-induced strains
16 would reduce these stresses.

17 The staff further considered the contribution of cladding stresses due to pellet-clad bonding and
18 its potential to facilitate athermal creep. The previously discussed Raynaud and Einziger study
19 did not account for potential stress concentration effects due to pellet-pellet interfaces and pellet
20 fragment-to-fragment friction forces that could result in more severe PCMI than for a perfectly
21 cylindrical pellet (as assumed in the paper). Recently, Ahn et al. (2013) estimated stress
22 concentrations from pellet-clad mechanical stresses caused by the radiation-induced pellet
23 swelling up to 100 years, independent of hoop stresses due to fission and decay gas release.
24 The work estimated that, for HBU fuel, the average pellet-swelling-induced PCMI stress
25 concentration was on the order of 200 MPa [29 ksi] locally. Literature indicates that
26 radiation-induced pellet swelling is expected to reach its maximum value beyond the 60-year
27 timeframe (Rondinella et al., 2010a,b; 2012). Therefore, PCMI stress concentrations due to
28 radiation-induced pellet swelling are not expected to exceed a threshold stress of 25 percent of
29 the yield stress (similar to the titanium data in Ankem and Wilt, 2006) during the 60-year
30 timeframe.

31 In summary, literature on the creep strain and creep rate of the zirconium-based cladding
32 materials at room temperature per the hoop stresses expected during extended storage is not
33 available. Therefore, it is not possible to directly assess the low-temperature creep of the
34 zirconium-based cladding materials. However, the staff has reviewed the threshold levels of
35 tensile stresses for low-temperature creep in the similar crystalline-structured (hexagonal close
36 packed crystalline) materials, which indicate that cladding hoop stresses on the cladding must
37 exceed approximately 25 percent of yield strength for athermal creep to be credible. The room
38 temperature hoop stresses on the zirconium-based cladding are expected to be less than
39 25 percent of the yield strength. Therefore, the low-temperature (athermal) creep mechanism
40 is not considered credible, even for the unlikely scenario where fuel reaches room
41 temperature during the 60-year timeframe. Therefore aging management is not required during
42 the 60-year timeframe.

43 3.6.1.5 *Mechanical overload (high burnup fuel)*

44 Mechanical overload is generally associated with PCMI, which could compromise the cladding
45 integrity during storage. PCMI is likely during reactor operations when the reactivity transient
46 during a reactivity-initiated accident (RIA) results in a rapid increase in a fuel rod power, leading

1 to a nearly adiabatic heating of the fuel pellets and potential failure of the fuel cladding. In either
2 commercial BWRs or PWRs, cladding failures have not been attributed to PCMI. However, data
3 generated in experimental reactors conducting ramp testing of heavily hydrided fuel claddings
4 indicate that hydride rims with large hydride number density at the cladding outer surface may
5 lead to crack initiation (Adamson et al., 2006). The cracks could propagate from the outside
6 toward the inner cladding surface, potentially resulting in failures.

7 During dry storage, PCMI stresses could develop due to pellet swelling and release of fission
8 gases to the gap between the fuel and cladding. PCMI could lead to the opening of existing
9 flaws in the cladding, potentially resulting in the release of fission gases and other fission
10 products into the cask environment. The existing flaws in undamaged fuel are likely to be of any
11 of the following: (i) surface (nonthrough-wall) cracks on the inner or outer wall, (ii) hairline
12 cracks, (iii) wall thinning due to oxide spallation on the outer surface, or (iv) wall thinning due to
13 fretting wear on the outer surface (NRC, 2014a).

14 Jernkvist et al. (2004) developed a criterion to determine the likelihood of PCMI during RIA,
15 which relies on estimating a threshold strain as a function of temperature, strain rate, hydrogen
16 concentration in cladding, and neutron fluence. However, this criterion is only applicable when
17 the cladding temperature is increasing, making it inapplicable to dry storage, where
18 temperatures decrease with time, barring any fluctuations from changes in ambient
19 temperature.

20 A method previously used to characterize PCMI failures in the cladding involves measuring the
21 creep strain capacity at a given creep strain rate (Jernkvist et al., 2004). More specifically,
22 PCMI-induced failures are observed when the cladding strain at a given strain rate exceeds a
23 threshold (Jernkvist et al., 2004; Fuketa et al., 2003). The threshold strain is a function of
24 cladding temperature, irradiation, and hydrogen concentration. PCMI-induced failures have
25 been reported at cladding strains exceeding 1 percent for strain rates in the range of
26 10^{-5} to 10^{-3} s⁻¹ at room temperature for various levels of hydrogen concentration (Jernkvist et
27 al., 2004). At higher temperatures, the strain at failure is above 6 percent between 523 and
28 673 K [482 to 752 degrees F] for strain rates in the range of 10^{-5} to 10^{-3} s⁻¹ (Jernkvist et al.,
29 2004). This threshold strain at higher temperature is applicable for cladding hydrogen content
30 up to 1,200 wppm. These results are consistent with those by Fuketa et al. (2003), which
31 exhibited similar threshold strains between 373 and 573 K [212 to 572 degrees F] with hydrogen
32 concentrations up to 1,450 wppm. These results can be compared with data discussed in
33 Section 3.6.1.3, which show that, for comparable strain rates in the order of 10^{-4} s⁻¹ to 10^{-5} s⁻¹,
34 no failures were observed for creep strains below 2 percent for in-creep tests at temperatures
35 between 150 and 400 degrees C [423 and 752 degrees F] for Zircaloy cladding irradiated up to
36 burnup of 64 GWd/MtU (Spilker et al., 1997; Goll et al., 2001; EPRI, 2002).

37 The staff reviewed the aforementioned creep strain and strain rate threshold criteria against the
38 results in Raynaud and Einziger (2015), which estimated the temperature-dependent hoop
39 stresses on the cladding while accounting for credible release of fission and decay gases and
40 pellet swelling. The authors estimated maximum cladding strains of 0.54 percent for the
41 10×10 BWR fuel cladding and 1.04 percent for the 17×17 PWR fuel cladding at a strain rate
42 of 10^{-10} s⁻¹ expected during dry storage. The authors stated that all of the cladding strain is
43 expected to occur during the first 50 years of storage. These calculations did not account for
44 the hoop stresses in ZIRLO™-clad IFBA rods with hollow and solid blanket pellets, which are
45 expected to be higher than standard rods (Bratton et al., 2015). The staff performed
46 calculations to estimate the cladding strain for IFBA rods using the Raynaud and Einziger
47 approach. Using a conservatively bounding hoop stress of 150 MPa [21.75 ksi], the maximum

1 cladding strain was estimated to be near 2.1 percent for IFBA rods. These values indicate
2 sufficient strain capacity per the previously discussed creep strain and strain rate threshold
3 criteria (Jernkvist et al., 2004; Fuketa et al., 2003), which is considered conservatively bounding
4 as the strain rates in dry storage are expected to be approximately five to seven orders of
5 magnitude lower than 10^{-5} to 10^{-3} s⁻¹. Therefore, the staff concludes that cladding failures due
6 to PCMI-induced mechanical overload are not considered credible during the 60-year
7 timeframe, and aging management is not required.

8 3.6.1.6 Oxidation

9 In the presence of residual amounts of water and high enough temperature, zirconium-based
10 cladding can be oxidized according to the following chemical reaction: $Zr + 2H_2O = ZrO_2 + 2H_2$
11 (Jung et al., 2013; Cox, 1976, 1988; Rothman, 1984).

12 Jung et al. (2013) conducted various scoping calculations to determine the extent of cladding
13 oxidation during dry storage in the presence of up to 1 L [0.26 gal] (equivalent to 55.5 moles) of
14 residual water. The amount of residual water considered is significantly higher than the residual
15 water amount of 0.43 moles expected after vacuum drying, as per NUREG–1536 (NRC, 2010).
16 The scoping calculations were based on a representative storage system loaded with the
17 equivalent of 21 Babcock & Wilcox SNF assemblies, each containing 208 fuel rods in a storage
18 canister. Jung et al. discussed temperature-dependent cladding oxidation kinetics for both
19 Zircaloy-2 and Zircaloy-4, concluding that the maximum cladding thickness loss due to oxidation
20 is not expected to exceed 10 μm [0.4 mils], even with complete consumption of the assumed 1 L
21 [0.26 gal] of residual water. The loss of cladding thickness due to oxidation represents less than
22 2 percent of the original cladding thickness. Therefore, cladding oxidation is considered to be
23 insignificant, and aging management is not required during the 60-year timeframe.

24 3.6.1.7 Pitting corrosion

25 Pitting corrosion initiates and propagates when (i) there is an aggressive chemical environment
26 that results in corrosion potential being greater than the repassivation potential and (ii) there is
27 enough cathodic capacity to sustain the propagation of the pitting corrosion (Shukla et al.,
28 2008). Zirconium is a passive material and is protected by a ZrO₂ surface film (Palit and
29 Gadiyar, 1987). The surface oxide readily reforms if broken, but zirconium is not completely
30 immune to pitting. Halides (i.e., anions of fluorine, chlorine, bromine, and iodine) in aqueous or
31 gaseous forms could initiate pitting. For example, pitting of zirconium has been shown to occur
32 in hydrochloric acid solutions containing ferric (Fe³⁺) or cupric (Cu²⁺) ions (Palit and Gadiyar,
33 1987).

34 Inside the cask's or canister's internal environment, a limited amount of residual water is
35 expected to be retained following drying, which will be in the liquid state once temperatures are
36 near or below 100 degrees C [212 degrees F]. The residual water amount is expected to be
37 less than 1 mole per NUREG–1536 (NRC, 2010). During storage, most residual water is
38 expected to decompose into hydrogen and oxidizing species, such as oxygen and hydrogen
39 peroxide, with time (Jung et al., 2013). It is possible for trace amounts of water to remain in the
40 vapor phase but is not expected to be in the liquid phase during dry storage, due to the low
41 relative humidity in the cask or canister cavity. For example, the relative humidity inside a
42 cavity volume of 2.1 m³ [554.8 gal], assuming a residual water content of 0.43 mole
43 [per NUREG–1536] at 25 degrees C [77 degrees F], is estimated to be approximately
44 15 percent using a backfill pressure of 1 atmosphere (atm) [14.7 psi], or 6 percent, using a
45 backfill pressure of 5 atm [73.5 psi]. Further, any residual water in the vapor phase is expected

1 to be spread throughout the cavity and is not expected to be sufficient to provide enough
2 cathodic capacity to initiate and propagate pitting corrosion of the cladding. Confirmation of this
3 expectation is provided in Einziger et al. (2003), which did not observe any evidence of pitting
4 corrosion in cladding after 15 years of dry storage. Therefore, pitting corrosion of the cladding is
5 not considered credible, and aging management is not required during the 60-year timeframe.

6 3.6.1.8 Galvanic corrosion

7 Galvanic corrosion can occur due to a mismatch in corrosion potentials between two metals in
8 an aqueous solution. In fuel assemblies, the mismatch can occur when the cladding is in
9 contact with other metallic components, which could result in the formation in a galvanic cell,
10 provided there is an aqueous solution between the two subcomponents. For example, some of
11 the PWR and BWR fuel assemblies contain spacer grids that are made of Inconel alloys, such
12 as Inconel 718 and Inconel 625. The dominant constituents of these Inconel alloys include
13 nickel, chromium, molybdenum, iron, niobium, and tantalum. A galvanic cell could form if
14 residual water condenses in the gap between the rod and a spacer grid, simultaneously
15 contacting both materials. The cladding could also be covered with a crud layer deposit during
16 reactor operations, which could further facilitate formation of the contact.

17 The standard electrode potential for zirconium and ZrO_2 in aqueous solution at 25 degrees C
18 [77 degrees F] is approximately in the range of -1.5 to $-1.6 V_{SHE}$, where the subscript "SHE"
19 stands for standard hydrogen electrode (Haynes et al., 2013). The standard electrode
20 potentials for chromium, nickel, molybdenum, and iron are approximately equal to -0.74 , -0.20 ,
21 -0.26 , and $-0.44 V_{SHE}$, respectively, at 25 degrees C [77 degrees F] (Bard and Faulkner, 1980;
22 Haynes et al, 2013). The standard electrode potential data indicate that zirconium would be
23 oxidized to zirconium ions during the galvanic reaction, and oxidizing species, such as oxygen
24 and hydrogen peroxide in aqueous solution, would be reduced at the Inconel alloy. The extent
25 of loss of cladding material would depend on the amount of oxidants present in the condensed
26 water. For example, per the stoichiometry of the oxidation and reduction reactions (Jung et al,
27 2013), reduction of 1 mole of hydrogen peroxide would result in oxidation of 0.5 mole of
28 zirconium. Similarly, reduction of 1 mole of oxygen would result in oxidation of 1.0 mole of
29 zirconium. Jung et al. reported scoping calculations to determine the extent of zirconium
30 oxidation with 1 mole of a 5 weight percent H_2O_2 aqueous solution saturated with oxygen at
31 25 degrees C [77 degrees F] and 1 atm [14.7 psi]. Jung et al. concluded that the extent of
32 oxidation would depend on the spread of the condensed water over the large surface area.
33 Therefore, the effect of galvanic corrosion is not expected to be localized.

34 The amount of residual water inside the cask or canister following drying is expected to be less
35 than 1 mole after vacuum drying, as per guidance in NUREG-1536 (NRC, 2010). Most residual
36 water is expected to decompose over time into hydrogen and oxidizing species, such as oxygen
37 and hydrogen peroxide (Jung et al., 2013). It is possible for some trace amount of water to
38 remain in the vapor phase inside the canister after the first renewal period but is not expected to
39 condense into liquid phase during dry storage due to the low relative humidity of the
40 containment cavity. For example, the relative humidity inside a canister with a cavity volume of
41 $2.1 m^3$ [554.8 gal], assuming a residual water content of 0.43 mole (per NUREG-1536) and at
42 25 degrees C [77 degrees F] is estimated to be approximately 15 percent with a backfill
43 pressure of 1 atm, or 6 percent with backfill pressure of 5 atm [73.5 psi]. Further, any residual
44 water in the vapor phase is expected to be spread throughout the containment cavity and is not
45 expected to be sufficient to form a corrosion cell between the cladding and the spacer grids
46 made of Inconel alloys. Therefore, galvanic corrosion of the zirconium-based cladding alloys is
47 not considered credible, and aging management is not required during the 60-year timeframe.

1 3.6.1.9 *Stress corrosion cracking*

2 SCC occurs as a result of a synergistic combination of a susceptible material, an aggressive
3 environment, and sufficiently high tensile stress. The corrosive environment associated with
4 SCC of fuel rods has been attributed to specific fission products, such as iodine, cesium, and
5 cadmium, generated during reactor irradiation (Wisner and Adamson, 1982; Sidky, 1998). SCC
6 of the cladding can occur at the rod's inner surface where the fuel pellet and cladding
7 mechanically interact and is related to PCMI hoop stresses on the cladding. SCC of
8 zirconium-based cladding has been observed in BWRs during power ramp-up (NRC, 1985;
9 Adamson, 2006). PWR cladding is unlikely to undergo similar SCC because of the more
10 gradual power ramp-up. Fuel pellets in PWR cladding are unlikely to undergo sudden
11 expansion and induce high stresses, as in BWR cladding. No cladding failures from SCC are
12 known to have occurred either during pool storage or under dry storage conditions.

13 Prescatore and Cowgill (EPRI, 1997) compiled SCC failure data from Yagee et al. (1979, 1980),
14 Mattas et al.(1982), Shimada and Nagai (1983), Kreyens et al. (1976), and Crescimanno (1984)
15 for the following irradiated cladding materials: recrystallized Zircaloy-2, stress-relieved
16 Zircaloy-2, recrystallized Zircaloy-4, and stress-relieved Zircaloy-4. For Zircaloy-2, the reported
17 data's temperature and tensile stress ranges were 325 to 350degrees C [617 to 662 degrees F],
18 and 119 to 513 MPa [17.3 to 74.4 ksi], respectively. Similarly for Zircaloy-4, the reported SCC
19 data's temperature and tensile stress ranges were 316 to 350 degrees C [601 to
20 662 degrees F], and 164 to 414 MPa [23.8 to 60 ksi], respectively. In the listed data, the
21 SCC-induced failure was reported at 157 MPa [22.8 ksi] and 325 degrees C [617 degrees F] for
22 Zircaloy-2, and at 205 MPa [29.7 ksi] and 360 degrees C [680 degrees F] for Zircaloy-4 (Yagee,
23 1979). Regarding these two failure data points (157 MPa [22.8 ksi] and 325 degrees C
24 [617 degrees F] for Zircaloy-2 and 205-MPa [29.7-ksi] and 360 degrees C [680 degrees F] for
25 Zircaloy-4), Prescatore and Cowgill (EPRI, 1997) argued that failures were misclassified as
26 SCC-induced failures and were more akin to nondetrimental pinhole breaches. Prescatore and
27 Cowgill stated that gross rupture, in the form of axial splitting, was noted in many instances
28 when the stress was greater than about 270 MPa [39.2 ksi], but at lower stresses, pinhole
29 leakage was by far the more common failure mode. If the 157 MPa [22.8 ksi] and
30 325 degrees C [617 degrees F] data point is excluded from the listed data for Zircaloy-2, as
31 argued by Prescatore and Cowgill, the next incident of the SCC-induced failure is noted at
32 247 MPa [35.8 ksi] at 325 degrees C [617 degrees F] for Zircaloy-2. Similarly, if the 205 MPa
33 [29.7 ksi] at 360 degrees C [680 degrees F] data point is excluded for Zircaloy-4, as argued by
34 Prescatore and Cowgill, the next incident of the SCC-induced failure is noted at 273 MPa
35 [39.6 ksi] at 360 degrees C [680 degrees F]. This analysis indicates that at least 240 MPa
36 [34.8 ksi] of hoop stresses are needed to induce SCC for both Zircaloy-2 and Zircaloy-4.

37 Recent work by Raynaud and Einziger (2015) shows that hoop stresses are expected to be
38 below 100 MPa [14.5 ksi], with the most realistic estimate of release of the decay and fission
39 gases from fuel pellets and with the best estimate of fuel swelling during a 300-year dry storage
40 period. However, hoop stresses in ZIRLO™-clad IFBA rods with hollow and solid blanket
41 pellets could be considerably higher. The Raynaud and Einziger study did not account for
42 potential stress concentration effects due to pellet-pellet interfaces and pellet
43 fragment-to-fragment friction forces that could result in more severe PCMI than for a perfectly
44 cylindrical pellet (as assumed in Raynaud and Einziger). Recently, Ahn et al. (2013) estimated
45 stress concentrations from pellet-clad mechanical stresses due to the radiation-induced pellet
46 swelling up to 100 years, independent of hoop stresses due to fission and decay gas release.
47 The work estimated that, for HBU fuel, the average pellet-swelling-induced PCMI stress
48 concentration was on the order of 200 MPa [29 ksi] locally. For low-burnup fuel, pellet

1 expansion stresses will be minimal, because the gap between the cladding and the pellet will
2 accommodate the swelling. Literature indicates that radiation-induced pellet swelling is
3 expected to reach its maximum beyond the first renewal period (Rondinella et al., 2010a,b;
4 2012). Even with the PCMI-induced hoop stresses, the cladding stresses will remain well below
5 the 240 MPa [34.8 ksi] criterion for inducing SCC. Therefore, SCC of the cladding is not
6 considered credible, and aging management is not required during the 60-year timeframe.

7 *3.6.1.10 Radiation embrittlement*

8 Radiation embrittlement of cladding can result in degradation of the mechanical properties of the
9 cladding, such as ductility and strength (PNNL, 2012; NRC, 2014a). This can lead to the
10 reduction in the maximum load that the cladding can withstand, potentially leaving the cladding
11 vulnerable to failure under external loads.

12 Radiation embrittlement of the cladding is mostly observed during reactor operation due to
13 cumulative fast neutron fluence on the order of 10^{22} n/cm² [6.5×10^{22} n/in²] (Hermann et al.,
14 2001) for recrystallized annealed Zircaloy-2 and cold-worked stress-relieved Zircaloy-4 (Morize
15 et al., 1987). During normal operation in the reactor, the cladding material is bombarded with
16 fast neutrons that cause atomic displacement cascades, resulting in the formation of point
17 defects (PNNL, 2012; NRC, 2014a; NWTRB, 2010). This leads to the reduction in the
18 mechanical properties of the cladding material.

19 In dry storage, the cumulative neutron fluence is expected be five orders of magnitude less than
20 in reactor service (Jung et al., 2013). In addition, annealing of irradiation hardening could occur
21 during storage, which would help recover some ductility. It has been shown in literature
22 (Masafumi et al., 2007; Torimaru, et al., 1996) that a post-irradiation heat treatment performed
23 at a temperature above the irradiation temperature can lead to the recovery of the
24 radiation-induced hardening and increased ductility of the cladding. Ito et al. (2004) further
25 showed that hardness also recovers at temperatures lower than an irradiation temperature of
26 360 degrees C [680 degrees F]. More specifically, Ito et al. (2004) showed that hardness
27 continued to recover, albeit quite slowly, at temperatures as low as 330 degrees C
28 [626 degrees F] for 8,000 hours (0.9 year), and nearly 50 percent recovery was observed
29 compared to the annealing over the same time at 360 degrees C [680 degrees F]. Thus, over
30 many years of extended storage, it is possible that thermal annealing could increase cladding
31 ductility, thereby reducing the effects of radiation embrittlement.

32 Because radiation embrittlement is associated with a cumulative fluence of on the order of
33 10^{22} n/cm² [6.5×10^{22} n/in²], which is not expected during storage, radiation embrittlement of
34 cladding is not considered credible, and therefore, aging management is not required during the
35 60-year timeframe.

36 *3.6.1.11 Fatigue*

37 Fatigue occurs when a material is subjected to repeated loading and unloading stresses. If the
38 loads are above a certain threshold, microscopic cracks will begin to form at stress
39 concentrators at the surface, persistent slip bands, and grain interfaces. As a crack reaches a
40 critical size, it will propagate until fracture. Because dry storage is a passive application, purely
41 mechanical cyclic loading is not expected. However, the cladding will experience thermal cycles
42 due to daily and seasonal fluctuations in ambient temperature, as well as extreme weather
43 events within a larger seasonal pattern. These thermal cycles will induce cyclic stresses on the
44 cladding due to either (i) changes in fission and decay gas pressure, as governed by gas laws,

1 which would result in fluctuations in cladding hoop stresses, and (ii) partial restraint on cladding
2 thermal expansion and contraction due to top and bottom nozzles, hold-down springs, and
3 spacer grids. These thermally induced stresses and corresponding strains can produce fatigue
4 damage in the same manner as purely mechanical cyclic loading.

5 Devoe and Robb (2015) conducted steady-state analyses to show that the change in peak
6 cladding temperature is directly proportional to the change in external air temperature of the
7 canister. Although the large thermal mass of the DSS is likely to reduce the amplitude and
8 frequency of the thermal cycles on fuel and cladding temperature, Devoe and Robb assumed a
9 correlation coefficient of unity between the peak cladding and external air temperature. Thus, a
10 1 degree C [1.8 degree F] change in air temperature would result in approximately 1 degree C
11 [1.8 degree F] change in cladding temperature. When evaluating daily temperature fluctuations,
12 the analysis assumed a conservative 25 degrees C maximum daily change [equivalent to
13 45 degrees F change], which is the mean daily temperature change in the United States. The
14 model further assumes a total of 21,900 thermal cycles, corresponding to steady-state
15 temperature cycle every day for 60 years. The staff assumed these conditions to determine if
16 the resulting changes in cladding hoop stresses could lead to fatigue-induced failure of the
17 cladding.

18 Raynaud and Einziger (2015) estimated the cladding hoop stresses while accounting for release
19 of fission gases and decay gases during storage, including pellet swelling stresses due to
20 radiation damage during storage. Raynaud and Einziger estimates included the effect of fuel
21 temperature on cladding hoop stresses. As per the Raynaud and Einziger estimates, a
22 25 degree C variation [45 degree F variation] in cladding temperature will cause up to 10 and
23 30 MPa [1.45 and 4.35 ksi] fluctuations in hoop stress of the BWR and PWR claddings,
24 respectively. Lin and Haicheng (1998) conducted experimental studies to determine fatigue
25 properties of zirconium and Zircaloy-4. Lin and Haicheng (1998) provided a fatigue lifetime
26 curve for zirconium and Zircaloy-4 under reversal bending as a function of the cyclic stress. As
27 per the fatigue lifetime curve in Lin and Haicheng, a cyclic stress amplitude of more than
28 260 MPa [37.7 ksi] is needed for fatigue-induced failure in Zircaloy-4 in 10^7 cycles. The curve
29 also bounds the data for zirconium, and hence, is also assumed to be applicable for other
30 zirconium-based cladding materials, such as Zircaloy-2, ZIRLO™, and M5®. Therefore, using
31 the fatigue lifetime curve in Lin and Haicheng, these fluctuations in hoop stresses (per the
32 assumed conditions in Devoe and Robb, 2015) are not sufficient for fatigue-induced failure in
33 the cladding.

34 The staff also evaluated the effects of extreme seasonal temperature variations, as these are
35 expected to be significantly higher than daily variations and could result in higher cyclic stress
36 amplitudes. Using the off-normal DSS operating conditions of -40 degrees C [-40 degrees F]
37 (winter) and 103 degrees C [217 degrees F] (summer) yields a maximum seasonal temperature
38 variation of 143 degrees C [variation of 257 degrees F]. Similar to the previous analysis, per the
39 Raynaud and Einziger (2015) estimates, a 143 degree C variation [257.4 degree F variation] in
40 cladding temperature will cause up to 10 and 55 MPa [1.45 and 7.8 ksi] fluctuations in hoop
41 stress of the BWR and PWR claddings, respectively. Using the fatigue lifetime curve in Lin and
42 Haicheng (1998), these fluctuations in hoop stresses (per the assumed conditions in Devoe and
43 Robb, 2015) are also not sufficient for fatigue-induced failure in the cladding.

44 As discussed in Section 3.2.1.7, the cyclic stress, σ , induced by the thermal variations also
45 depends on the material's coefficient of thermal expansion (α_0) and Young's modulus of
46 elasticity (E), the actual change in temperature (ΔT), and the degree of constraint on the
47 component. Since the degree of constraint for the cladding is not readily available for cladding,

1 a conservative approach is employed to estimate the cyclic stresses and associated potential
2 impact of thermal fatigue. The coefficient of thermal expansion is estimated to be approximately
3 $4.16 \times 10^{-6}/K$, based on the data in Luscher and Geelhood (2010). The Young's modulus of
4 elasticity of various zirconium-based cladding materials ranges between 32 and 100 GPa
5 [4,641 and 14,504 ksi] (Luscher and Geelhood, 2010); a value of 100 GPa [14,504 ksi] is
6 conservatively used. The assumed values of α_0 and E result in a thermally induced cyclic stress
7 of 10.4 MPa [1.5 ksi] and 59.5 MPa [8.6 ksi] for ΔT equal to 25 and 143 degrees C [45 and
8 257 degrees F], respectively. As per the fatigue lifetime curve in Lin and Haicheng (1998),
9 these fluctuations in hoop stresses are also not sufficient for fatigue-induced failure in
10 the cladding.

11 The staff further considered the cumulative cyclic stresses for all cases described above, which
12 results in stresses ranging from 20 to 70 MPa [2.9 and 10.2 ksi] for BWR and from 65 to
13 115 MPa [9.4 and 16.7 ksi] for PWR claddings. Even the combined conservative values are
14 well below the threshold of 260 MPa [37.7 ksi] needed for fatigue-induced failure in the cladding,
15 per Lin and Haicheng (1998). Therefore, the staff concludes that fatigue-induced failure of the
16 cladding is not credible during the 60-year timeframe, and aging management is not required.

17 **3.6.2 Assembly hardware materials**

18 The assembly hardware considered here includes guide tubes, spacer grids, and lower and
19 upper end fittings. The guide tubes are fabricated using zirconium-based alloys. The other
20 components are fabricated using one of the following materials: zirconium-based alloys,
21 Inconel 718, Inconel 625, Inconel X-750, and stainless steel 304L. These subcomponents are
22 not expected to experience sustained external loads during passive dry storage except for their
23 own weight.

24 **3.6.2.1 Creep**

25 Creep is defined as the time-dependent deformation that takes place at an elevated
26 temperature and constant stress. Because the deformation processes that produce creep are
27 thermally activated, the rate of this time-dependent deformation (i.e., the creep rate) is a strong
28 function of the temperature. The creep rate also depends on the applied stress but does not
29 generally vary with the environment. As a general rule of thumb, at temperatures below $0.4T_m$,
30 where T_m is the melting point of the metal in Kelvin, thermal activation is insufficient to produce
31 significant creep (Cadek, 1988). The melting temperature of various zirconium alloys is above
32 1,800 degrees C [3,272 degrees F]. Similarly, the melting temperature of various Inconel alloys
33 is above 1,260 degrees C [2,300 degrees F]. In addition, the melting temperature of 304L
34 stainless steels is close to 1,400 degrees C [2,552 degrees F].

35 Regarding the zirconium alloys, the $0.4T_m$ criterion yields a creep threshold of 556 degrees C
36 [1,033 degrees F]. The maximum expected temperature of fuel cladding has been estimated to
37 be 400 degrees C [752 degrees F] at the beginning of storage (Jung et. al., 2013). This
38 cladding temperature is expected to decrease to around 266 degrees C [510 degrees F] after
39 20 years and to approximately 127 degrees C [261 degrees F] after 60 years. This indicates
40 that creep of the zirconium alloys is unlikely during the renewal period.

41 Regarding Inconel alloys, the $0.4T_m$ criterion yields a creep threshold of 340 degrees C
42 [644 degrees F]. As stated previously, the peak temperature inside the storage canister is
43 expected to be below 266 degrees C [510 degrees F] after 20 years of storage. This indicates
44 that creep of various Inconel alloys is unlikely during the renewal period.

1 Regarding 304L stainless steel, the $0.4T_m$ criterion yields a creep threshold of 396 degrees C
2 [755 degrees F]. As stated previously, the peak temperature inside the storage canister is
3 expected to be below 300 degrees C [572 degrees F] after 20 years of storage. Further, the
4 $0.4T_m$ rule of thumb underestimates the minimum creep temperature for steels, because
5 temperatures above 500 degrees C [932 degrees F] are required for significant creep in steels
6 (Samuels, 1988). This indicates that creep of 304L stainless steel is unlikely during the
7 renewal period.

8 Therefore, creep of the assembly hardware is not considered credible, and aging management
9 is not required during the 60-year timeframe.

10 3.6.2.2 *Hydriding*

11 Assembly hardware such as guide tubes and spacer grid materials made from zirconium alloys
12 could potentially be subjected to hydriding effects that could reduce the material's ductility and
13 fracture toughness, particularly at lower temperatures (less than 200 degrees C
14 [392 degrees F]), once the fuel has cooled (PNNL, 2012).

15 Hydriding may occur in zirconium alloys that experience hydrogen pickup in reactor service
16 (NRC, 2014a). As the temperature of the assembly hardware decreases, zirconium hydrides
17 precipitate due to the decreasing hydrogen solubility in the zirconium matrix. The hydride
18 precipitation will occur when the hardware cools in the spent fuel pools after reactor discharge.
19 Some of the hydride will dissolve during the drying process and will reprecipitate due to
20 subsequent cooling during storage. Unlike fuel rods with cladding, there is no hoop stress for
21 the zirconium-based assembly hardware to cause hydride reorientation. Any load on the
22 assembly hardware is predominantly expected due to its own weight, which is not sufficient to
23 be equivalent to hoop stresses to cause hydride reorientation. In addition, any additional
24 hydriding of the assembly hardware during extended storage is expected to be negligible
25 (Jung et al., 2013).

26 In summary, the impact of hydriding effects on assembly hardware, especially guide tubes, is far
27 less severe than for cladding with fuel (EPRI, 2011; PNNL, 2012; Hanson et al., 2012).
28 Because there is limited load during storage on assembly hardware, it is unlikely that hydriding
29 will affect the ability of the assembly hardware to ensure that the spent fuel remains in the
30 as-analyzed configuration. Confirmation of this expectation is provided by Einziger et al. (2003),
31 which did not observe any hydriding effects on assembly hardware after 15 years of dry storage.
32 Therefore, hydriding of assembly hardware components is not considered to be significant, and
33 aging management is not required during the 60-year timeframe.

34 3.6.2.3 *General corrosion*

35 Various assembly hardware components made of stainless steel or Inconel may be subjected to
36 general corrosion in the presence of humid air or an aqueous solution. General corrosion of
37 assembly hardware made of zirconium alloys is not considered here; it is excluded per the
38 technical basis discussed in Section 3.6.1.6. The amount of residual water in the canister
39 during the extended storage is expected to be less than 1 mole per the guidance in NUREG–
40 1536 (NRC, 2010). Most residual water is expected to decompose into hydrogen and oxidizing
41 species, such as oxygen and hydrogen peroxide, with time (Jung et al., 2013). However, it is
42 possible for trace amounts of water to remain in the vapor phase in the canister's internal
43 environment for the extended period.

1 The general corrosion rate of the nickel-based Inconel alloys due to humid air is expected to be
2 on the order of 25 nm/yr [10^{-3} mils/yr] (Van Rooyen and Copson, 1968). The general
3 corrosion rate of 304 stainless steel in the presence of humid air has been reported to be
4 negligible (INCO, 1970), and the low-carbon grade 304L is expected to behave similarly.
5 Further, as corrosion proceeds, the residual water would deplete with time. Considering the low
6 general corrosion rate of the Inconel alloy, the negligible corrosion rate of 304 stainless steel
7 under humid air conditions, and the radiolysis of the residual water, it is concluded that the
8 effect of general corrosion in the presence of trace amounts of water is insignificant on
9 assembly hardware components during the renewal period. As such, general corrosion of
10 assembly hardware is considered to be insignificant, and therefore, aging management is not
11 required during the 60-year timeframe.

12 3.6.2.4 *Stress corrosion cracking*

13 Various stainless steel and Inconel assembly hardware components could be susceptible to
14 SCC in the presence of an aggressive environment and sufficient residual tensile stresses.
15 SCC of the structural components may lead to cracking, which can compromise the structural
16 integrity of the component. SCC of assembly hardware made of zirconium alloys is not
17 considered here; it is excluded per the technical basis discussed in Section 3.6.1.9.

18 Residual tensile stresses are expected to be present in the assembly hardware, primarily in
19 welded areas. Regarding the chemical environment, various types of stainless steels are prone
20 to SCC, even in high-purity demineralized water at the temperatures of the BWRs, typically
21 290 degrees C [554 degrees F] (Kain, 2011). This observation is attributed to the presence of
22 dissolved oxygen and other oxidizing species in the primary coolant water (Kain, 2011) of a
23 BWR. Various types of nickel-based alloys, including Inconel, are susceptible to SCC in the
24 presence of hot water, hot caustic solution, hot wet hydrofluoric acid solution, or aqueous
25 solution containing a sufficient amount of chloride at high temperatures (Rebak, 2011).

26 In the canister environment, the water could exist in the liquid state only when the temperature
27 is near or below 100 degrees C [212 degrees F]. The residual water content inside the canister
28 is expected to be less than 1 mole during dry storage, as per guidance in NUREG-1536
29 (NRC, 2010). During storage, most residual water would decompose into hydrogen and
30 oxidizing species, such as oxygen and hydrogen peroxide, due to radiolysis (Jung et al., 2013).
31 However, it is possible for a trace amount of residual water to persist in the vapor phase of the
32 containment cavity. The trace amount of water is unlikely to condense into the liquid phase
33 during dry storage because the relative humidity of the DSS internal environment cannot reach
34 100 percent when the residual amount of water is less than 1 mole. For example, the relative
35 humidity inside a containment cavity volume of 2.1 m³ [554.8 gal] at 25 degrees C
36 [77 degrees F], assuming a residual water amount of 0.43 mole [expected after vacuum drying
37 as per NUREG-1536], is estimated to be approximately 15 percent, using a backfill pressure of
38 1 atm [14.7 psi], or 6 percent using a backfill pressure of 5 atm [73.5 psi] (Green and Perry,
39 2007). Further, SCC of stainless steel and Inconel has not been reported in a nonchloride
40 humid air environment.

41 Because of the lack of halides and the small amount of water in helium and embedded
42 environments, SCC of stainless steel is not considered to be credible. Therefore, aging
43 management of SCC of stainless steel subcomponents exposed to helium is not required during
44 the 60-year timeframe.

1 3.6.2.5 *Radiation embrittlement*

2 Radiation embrittlement of assembly hardware such as guide tubes and spacer grid materials
3 made from zirconium alloys is excluded using the basis provided in Section 3.6.1.10. Similarly,
4 radiation embrittlement of assembly hardware made of stainless steel or Inconel is not
5 considered credible per the technical bases provided in Sections 3.2.1.9, 0, and 0. Therefore,
6 aging management of radiation embrittlement of assembly hardware subcomponents exposed
7 to helium and embedded environments is not required during the 60-year timeframe.

8 3.6.2.6 *Fatigue*

9 Fatigue of assembly hardware such as guide tubes and spacer grid materials made from
10 zirconium alloys is excluded using the basis provided in Section 3.6.1.11. Similarly, fatigue of
11 assembly hardware made of stainless steel or Inconel is not considered credible per the
12 technical bases provided in Sections 3.2.1.7, 3.2.2.7, and 3.2.4.5. Therefore, aging
13 management of fatigue of assembly hardware subcomponents exposed to helium is not
14 required during the 60-year timeframe.

15 **3.6.3 References**

16 Adamson, R., B. Cox, J. Davies, P. Rudling, S. Vidyanathan. "IZNA-6 Special Topical Report:
17 Pellet-Cladding Interaction (PCI and PCMI)," R. Adamson, ed. Skultuna, Sweden:
18 Advanced Nuclear Technology International. 2006.

19 Ahn, T., V. Rondinella, and T. Wiss. "Potential Stress on Cladding Imposed by the Matrix
20 Swelling from Alpha Decay in High Burnup Spent Nuclear Fuel." Paper 6830.
21 *2013 International High-Level Radioactive Waste Management Conference*, April 28–May 2.
22 Albuquerque, New Mexico: American Nuclear Society. 2013.

23 Ankem, R. and T. Wilt. "A Literature Review of Low Temperature (< 0.25 Tmp) Creep Behavior
24 α , α - β , and β Titanium Alloys." ADAMS Accession No. ML072060401. San Antonio, Texas:
25 Center for Nuclear Waste Regulatory Analyses. 2006.

26 Aomi, M., T. Baba, T. Miyashita, K. Kamimura, T. Yasuda, Y. Shinohara, and T. Takeda.
27 "Evaluation of Hydride Reorientation Behavior and Mechanical Property for High-Burnup Fuel-
28 Cladding Tubes in Interim Dry Storage." *Journal of ASTM International*. Vol. 5. pp. 651–673.
29 2008.

30 Bai, J., J. Gilbon, C. Prioul, and D. Francois. "Hydride Embrittlement in Zircaloy-4 Plate, Part I,
31 Influence of Microstructure on the Hydride Embrittlement in Zircaloy-4 at 20°C and 350°C" and
32 Part II, "Interaction Between the Tensile Stress and the Hydride Morphology." *Metallurgical and*
33 *Materials Transactions A*. Vol. 25A, Issue 6. pp. 1,185–1,197. June 1994.

34 Bard, A.J. and L.R. Faulkner. *Electrochemical Methods, Fundamentals and Applications*.
35 New York, New York: John Wiley & Sons, Inc. 1980.

36 Bare, W.C. and L.D. Torgerson. "Dry Cask Storage Characterization Project-Phase 1:
37 CASTOR V/21 Cask Opening and Examination." NUREG/CR-6745, INEEL/EXT-01-00183,
38 September 2001.

- 1 Billone, M.C., T.A. Burtseva, and Y.Y. Liu. "Characterization and Effects of Hydrides in High-
2 Burnup PWR Cladding Alloys." *Proceedings of the International High-Level Radioactive*
3 *Waste Management Conference*, Charleston, South Carolina. Paper No. 12617.
4 American Nuclear Society. April 12–16, 2015.
- 5 Billone, M.C., T.A. Burtseva, Z. Han, and and Y.Y. Liu. "Embrittlement and DBTT of
6 High-Burnup PWR Fuel Cladding Alloys." FCRD-UFD-2013-000401, ANL-13/16. Lemont,
7 Illinois: Argonne National Laboratory. 2013.
- 8 Bratton, R., M. Jessee, and W. Wieselquist. "Rod Internal Pressure Quantification and
9 Distribution Analysis Using FRAPCON." FCRD-UFD-2015-000636, ORNL/TM-2015/557. Oak
10 Ridge, Tennessee: Oak Ridge National Laboratory. 2015.
- 11 Bossis, P., B. Verhaeghe, S. Doriot, D. Gilbon, V. Chabretou, A. Dalmais, J.P. Mardon, M. Blat
12 and A. Miquet. "In PWR Comprehensive Study of High Burn-up Corrosion and Growth
13 Behaviour of M5 and Recrystallised Low-Tin Zircaloy-4." *15th ASTM International Symposium:*
14 *Zirconium in the Nuclear Industry*. Sun River, Oregon. ASTM International. June 20, 2007.
- 15 Bouffioux, P. and N. Rupa. "Impact of Hydrogen on Plasticity and Creep of Unirradiated
16 Zircaloy-4 Cladding Tubes." *12th International Symposium on Zirconium in the Nuclear*
17 *Industry*. ASTM STP 1354. Toronto, Canada. ASTM International. pp. 399–422. 1998.
- 18 Cadek, J. *Creep of Metallic Materials*. Elsevier Science Publishing Company, Inc. 1988.
- 19 Cappelaere, C., R. Limon, T. Bredel, P. Herter, D. Gilbon, S. Allegre, P. Bouffioux and
20 J.P. Mardon. "Long Term Behaviour of the Spent Fuel Cladding in Dry Storage Conditions."
21 *8th International Conference on Radioactive Waste Management and Environmental*
22 *Remediation*. October 2001. Vol. 2. Bruges, Belgium. American Society of Mechanical
23 Engineers. 2001.
- 24 Cazalis, B., C. Bernaudat, P. Yvon, J. Desquines, C. Poussard, and X. Averty. "The
25 PROMETRA program: A Reliable Material Database for Highly Irradiated Zircaloy-4, ZIRLO™
26 and M5™ fuel claddings." Proceeding of the 18th International Conference on Structural
27 Mechanics in Reactor Technology. 18th ed., Paper SMiRT18-C02-1. August 2005.
- 28 Chan, K.S. "A Micromechanical Model for Predicting Hydride Embrittlement in Nuclear Fuel
29 Cladding Material." *Journal of Nuclear Materials*. Vol. 227. pp. 220–236. 1996.
- 30 Chan, K.S. "An Assessment of Delayed Hydride Cracking in Zirconium Alloy Cladding Tubes
31 Under Stress Transients." *International Materials Reviews*. Vol. 58, No. 6. pp. 349–373. 2013.
- 32 Chopra, O., D. Diercks, R. Fabian, Z. Han, and Y. Liu. "Managing Aging Effects on Dry Cask
33 Storage Systems for Extended Long-Term Storage and Transportation of Used Fuel."
34 FCRD–UFD–2014–000476. ANL–13/15, Revision 2. Washington, DC: U.S. Department of
35 Energy. 2014.
- 36 Chu, H.C., S.K. Wu, and R.C. Kuo. "Hydride Reorientation in Zircaloy-4 Cladding." *Journal of*
37 *Nuclear Materials*. Vol. 373. pp. 319–327. 2008.

- 1 Chung, H.M. "Understanding Hydride- and Hydrogen-related Processes in High-Burnup
2 Cladding in Spent-Fuel-Storage and Accident Situations." *2004 International Meeting on LWR
3 Fuel Performance*, Orlando, Florida, September 19–22, 2004. Paper No. 1064. 2004.
- 4 Coleman, C., V. Grigoriev, V. Inozemtsev, V. Markelov, M. Roth, V. Makaevicius, Y.S. Kim,
5 K.L. Ali, J.K. Chakravarty, R. Mizrahi, and R. Lalgud. "Delayed Hydride Cracking in Zircaloy
6 Fuel Cladding—An IAEA Coordinated Research Programme." *Nuclear Engineering and
7 Technology*. Vol. 41, No. 2. pp. 171–177. 2009.
- 8 Cox, B. "Hydrogen Trapping by Oxygen and Dislocations in Zirconium Alloys." *Journal of
9 Alloys and Compositions*. Vol. 256 pp. L4–L7. 1997.
- 10 _____. "Degradation of Zirconium Alloys in Water Cooled Reactors." *Proceedings of the Third
11 International Symposium on Environmental Degradation of Materials in Nuclear Power
12 Systems-Water Reactors*, Warrendale, Pennsylvania: The Metallurgical Society. pp. 65–76.
13 1988.
- 14 _____. "Oxidation of Zirconium and its Alloys." *Advances in Corrosion Science and
15 Technology*. M. Fontana and R.W. Staehle, eds. New York, New York: Plenum Press. 1976.
- 16 Crescimanno, P.J., W.R. Campbell, and I. Goldberg. "A Fracture Mechanics Mode for Iodine
17 Stress Corrosion Crack Propagation in Zircaloy Tubing." In *Environment-Sensitive Fracture
18 Evaluation and Comparison of Test Methods*. ASTM STP 821 (S.W. Dean, E.N. Pugh and
19 O.M. Ugiansky, eds). Philadelphia, Pennsylvania: American Society for Testing and Materials.
20 pp.150–169. 1984.
- 21 Daum, R.S., S. Majumdar, Y. Liu, and M.C. Billone. "Radial-hydride Embrittlement of High-
22 Burnup Zircaloy-4 Fuel Cladding." *Journal of Nuclear Science and Technology*. Vol. 43, No. 9.
23 pp. 1,054–1,067. 2006.
- 24 Devoe, R. and K.R. Robb. "COBRA-SFS Dry Cask Modeling Sensitivities in High-Capacity
25 Canisters." *Proceedings of the International High-Level Radioactive Waste Management
26 Conference*, April 12–16, 2015. Paper No. 12701. Charleston, South Carolina. 2015.
- 27 Einziger, R.E. and R. Kohli. "Low Temperature Rupture Behavior of Zircaloy-Clad Pressurized
28 Water Reactor Spent Fuel Rods Under Dry Storage Conditions." *Nuclear Technology*. Vol. 67.
29 p. 107. 1984.
- 30 Einziger, E. R., H. C. Tsia, M. C. Billone, and B. A. Hilton. "Examination of Spent Fuel Rods
31 After 15 Years in Dry Storage." NUREG/CR-6831. ANL-03/17. September 2003.
- 32 EPRI. "High Burnup Dry Storage Cask Research and Development Project: Final Test Plan."
33 DE-NE-0000593. Palo Alto, California: Electric Power Research Institute. 2014.
- 34 _____. "Extended Storage Collaboration Program (ESCP) Progress Report and Review of Gap
35 Analyses." Report 1022914. Palo Alto, California: Electric Power Research Institute. 2011.
- 36 _____. "Technical Bases for Extended Dry Storage of Spent Nuclear Fuel." Report 1003416.
37 Palo Alto, California: Electric Power Research Institute. 2002.

- 1 _____. "Temperature Limit Determination of the Inert Dry Storage of Spent Nuclear Fuel."
2 Report TR-103949. Palo Alto, California: Electric Power Research Institute. 1997.
- 3 Foregeaud, S., J. Desquines, M. Petit, C. Getrey, and G. Sert. "Mechanical Characteristics of
4 Fuel Rod Claddings in Transport Conditions," *Packaging, Transport, Storage, & Security of*
5 *Radioactive Material*. Vol. 20. pp. 69–76. 2009.
- 6 Fuketa, T., T. Sugiyama, T. Nakamura, H. Sasajima, and F. Nagase. NUREG/CP-01 85,
7 "Effects of Pellet Expansion and Cladding Hydrides on PCMI Failure of High Burnup LWR Fuel
8 During Reactivity Transients." Nuclear Safety Research Conference. Washington, DC. 2003.
- 9 Gilbert, E.R., E.P. Simonen, C.E. Beyer, and P.G. Medvedev. "Update of CSFM Methodology
10 for Determining Temperature Limits for Spent Fuel Dry Storage in Inert Gas." ADAMS
11 Accession No. ML022250067. Washington, DC: U.S. Nuclear Regulatory Commission. 2001.
- 12 Goll, W., H. Spilker and E.H. Toscano. "Short-Term Creep and Rupture Tests on High Burnup
13 Fuel Rod Cladding." *Journal of Nuclear Materials*. Vol. 289. p. 247. 2001.
- 14 Geelhood, K.J., C.E Beyer, and W.G Luscher. "PNNL Stress/Strain Correlation for Zircaloy."
15 Pacific Northwest National Laboratory. PNNL-17700. July 2008.
- 16 Geelhood, K.J. and W.G. Luscher. "FRAPCON-3.5: A Computer Code for the Calculation of
17 Steady-State, Thermal-Mechanical Behavior of Oxide Fuel Rods for High Burnup."
18 Pacific Northwest National Laboratory. PNNL-19418 Vol. 1. Rev. 1. NUREG/CR-7022.
19 Vol. 1, Rev. 1. ADAMS Accession No. ML14295A539. October 2014.
- 20 Green, D. W. and R. Perry. *Perry's Chemical Engineers' Handbook. Eighth Edition.*
21 McGraw-Hill Education, New York, New York, 2007.
- 22 Hanson, B, H. Alsaed, C. Stockman, D. Enos, R. Meyer, and K. Sorenson. "Used Fuel
23 Disposition Campaign: Gap Analysis to Support Extended Storage of Used Nuclear Fuel
24 Rev. 0." Richland, Washington: Pacific Northwest National Laboratory. 2012.
- 25 Hanson, B. D. "High Burnup Fuel, Associated Data Gaps, and Integrated Approach for
26 Addressing the Gaps," Presented to Nuclear Waste Technical Review Board.
27 <<http://www.nwtrb.gov/meetings/2016/feb/hanson.pdf>> February 29, 2016.
- 28 Haynes, W.M., D.R. Lide, and T.J. Bruno. *CRC Handbook of Chemistry and Physics.*
29 *93rd Edition.* CRC Press. Boca Raton, Florida. 2013.
- 30 Hermann, A., M. Martin, P. Porschke, and S. Yagnik. "Ductility Degradation of Irradiated Fuel
31 Cladding." 2001. <https://inis.iaea.org/search/search.aspx?orig_q=RN:32030458>
- 32 International Atomic Energy Agency (IAEA). "Corrosion of zirconium alloys in nuclear power
33 plants." Vienna, Austria: TECDOC-684. January 1993.
- 34 Ibarra, L., T. Wilt, G. Ofoegbu, and A. Chowdhury. "Structural Performance of Drip Shield
35 Subjected to Static and Dynamic Loading." ADAMS Accession No. ML070240131.
36 San Antonio, Texas: Center for Nuclear Waste Regulatory Analyses. 2007.

- 1 INCO. "Corrosion Resistance of the Austenitic Chromium-Nickel Stainless Steels in
2 Atmospheric Environments." The International Nickel Company (INCO), Inc.
3 Suffern, New York. 1970.
4 <<http://www.ohiogratings.com/pdfs/StainlessSteelCorrosionStudy.pdf>>
- 5 Ito, K., K. Kamimura, and Y. Tsukuda. "Evaluation of Irradiation Effect on Spent Fuel Cladding
6 Creep Properties." *2004 International Meeting on LWR Fuel Performance*, Orlando, Florida:
7 September 19–22, 2004. American Nuclear Society. p. 440. 2004.
- 8 Jaworski, A. and S. Ankem. "Influence of the Second Phase on the Room-Temperature Tensile
9 and Creep Deformation Mechanisms of α - β Titanium Alloys: Part I. Tensile Deformation."
10 *Metallurgical and Materials Transactions*. Vol. 37A. pp. 2,739–2,754. 2006.
- 11 Jernkvist, L.O., A. R. Massih, and P. Rudling. "A Strain-Based Clad Failure Criterion for
12 Reactivity Initiated Accidents in Light Water Reactors." SKI Report 2004:32. Uppsala, Sweden:
13 2004.
- 14 Jung, H., P. Shukla, T. Ahn, L. Tipton, K. Das, X. He, and D. Basu. "Extended Storage and
15 Transportation: Evaluation of Drying Adequacy." ADAMS Accession No. ML13169A039.
16 San Antonio, Texas: Center for Nuclear Waste Regulatory Analyses. 2013.
- 17 Kain, V. "Chapter 5: Stress Corrosion Cracking in Stainless Steels." In *Stress Corrosion
18 Cracking: Theory and Practice*. V.S. Raja and T. Shoji, eds. Cambridge, England:
19 Woodhead Publishing. pp. 199–244. 2011.
- 20 Kamimura, K. "Integrity Criteria for Spent Fuel Dry Storage in Japan." *Proceeding of the
21 International Seminar on Interim Storage of Spent Fuel, International Seminar on Spent Fuel
22 Storage (ISSF)*. Tokyo, Japan. p. VI-3-1. 2010.
- 23 Kammenzind, B.F., D.G. Franklin, H.R. Peters, and W.J. Duffin. "Hydrogen Pickup and
24 Redistribution in Alpha-Annealed Zircaloy-4." In *Zirconium in the Nuclear Industry:
25 Eleventh International Symposium*, ASTM STP 1295 (E.R. Bradley and G.P. Sabol, Eds.),
26 West Conshohocken, Pennsylvania: American Society for Testing and Materials, pp. 338–370.
27 1996.
- 28 Kim, Y.S. "Kinetics of Crack Growth in Zirconium Alloys (I): Temperature Dependence of the
29 Crack Growth Rate." *Journal of Applied Physics*. Vol. 106. pp. 123,520-1–123,520-6. 2009a.
- 30 _____. "Hydride Reorientation and Delayed Hydride Cracking of Spent Fuel Rods in Dry
31 Storage." *Metallurgical and Materials Transactions A*. Vol. 40A. pp. 2,867–2,875. 2009b.
- 32 _____. "Delayed Hydride Cracking of Spent Fuel Rods in Dry Storage." *Journal of Nuclear
33 Materials*. Vol. 378. pp. 30–34. 2008.
- 34 Kim, J.-S., Y.-J. Kim, D.-H. Kook, and Y.-S. Kim. "A Study on Hydride Reorientation of Zircaloy-
35 4 Cladding Tube Under Stress," *Journal of Nuclear Materials*, Vol. 456, pp. 246–252, 2015a.
- 36 Kim, J.-S., T.-H. Kim, D.-H. Kook, and Y.-S. Kim. "Effects of Hydride Morphology on the
37 Embrittlement of Zircaloy-4 Cladding," *Journal of Nuclear Materials*, Vol. 456, pp. 235–245.
38 2015b.

- 1 King, S., R. Kesterson, K. Yueh, R. Comstock, W. Herwig, and S. Ferguson. "Impact of
2 Hydrogen on the Dimensional Stability of ZIRLO Fuel Assemblies." In *Zirconium in the Nuclear*
3 *Industry: Thirteenth International Symposium*, ASTM STP 1423. West Conshohocken,
4 Pennsylvania: ASTM International. pp. 471-479. 2002.
- 5 Kubo, T., Y. Kobayashi, and H. Uchikoshi. "Measurements of Delayed Hydride Cracking
6 Propagation Rate in the Radial Direction of Zircaloy-2 Cladding Tubes." *Journal of Nuclear*
7 *Materials*. Vol. 427. pp. 18-29. 2012.
- 8 Kearns, J.J. "Thermal Solubility and Partitioning of Hydrogen in the Alpha Phase of Zirconium,
9 Zircaloy-2 and Zircaloy-4." *Journal of Nuclear Materials*. Vol. 22. pp. 292-303. 1967.
- 10 Kreyns, P.H., G.L. Spahr, and J.E. McCauley. "An Analysis of Iodine Stress Corrosion Cracking
11 of Zircaloy-4 Tubing." *Journal of Nuclear Materials*. Vol. 61. pp. 203-212. 1976.
- 12 Lin, X. and G. Haicheng. "High Cycle Fatigue Properties and Microstructure of Zirconium and
13 Zircaloy-4 Under Reversal Bending." *Materials Science and Engineering A*. Vol. 252.
14 pp. 166-173. 1998.
- 15 Luscher, W.G and K.J. Geelhood. "Material Property Correlations: Comparisons Between
16 FRAPCON-3.4, FRAPTRAN 1.4, and MATPRO." PNNL-19417 (NUREG/CR-7024).
17 Richland, Washington: Pacific Northwest National Laboratory. 2010.
- 18 Mattas, R.F., F.L. Yagee and L.A. Neimark. "Effect of Zirconium Oxide on the Stress corrosion
19 Susceptibility of Irradiated Zircaloy Cladding." In *Zirconium in the Nuclear Industry:*
20 *Fifth International Symposium*. ASTM STP 754, (D.G Franklin, ed. West Conshohocken,
21 Pennsylvania: American Society for Testing and Materials. pp. 158-170. 1982.
- 22 Mardon, J. P., G.L. Garner, and P.B. Hoffmann. "M5[®] A Breakthrough in Zr Alloy."
23 *Proceedings of 2010 LWR Fuel Performance/TopFuel/WRFPM*, Orlando, Florida,
24 September 26-29, 2010. American Nuclear Society. 2010.
- 25 Masafumi, N., K. Uchida, A. Miyazaki, and Y. Ishii. "Annealing Study on Neutron Irradiation
26 Effects in Resonance Frequencies of Zircaloy Plates by EMAR Method." *Journal of Nuclear*
27 *Science and Technology*, Vol. 44, No. 10. pp. 1,285-1,294. 2007.
- 28 Morize P., J. Baicry, and J. P. Mardon. "Effect of Irradiation at 588 K on Mechanical Properties
29 and Deformation Behavior of Zirconium Alloy Strip." *Zirconium in the Nuclear Industry:*
30 *Seventh International Symposium*. ASTM STP 939. R.B. Adamson and L.F.P. Van Swam, eds.
31 ASTM. pp. 101-119. 1987.
- 32 Murty, K.L. "The Internal Pressurization Creep of Zr Alloys for Spent-Fuel Dry Storage
33 Feasibility." *Journal of the Minerals, Metals and Materials Society*. Vol. 52, No. 9. pp. 34-43.
34 2000.
- 35 NRC. NUREG-1927, Rev. 1, "Standard Review Plan for Renewal of Specific Licenses and
36 Certificates of Compliance for Dry Storage of Spent Nuclear Fuel." Revision 1.
37 ADAMS Accession No. ML16179A148. Washington, DC: U.S. Nuclear Regulatory Commission.
38 2016.

1 _____. "Acceptable Fuel Cladding Hydrogen Uptake Models." ADAMS Accession
2 No. ML15133A306. Washington, DC: U.S. Nuclear Regulatory Commission. 2015a.

3 _____. Draft Regulatory Issue Summary 2015-XXX, "Considerations in Licensing High Burnup
4 Spent Fuel in Dry Storage and Transportation." ADAMS Accession No. ML14175A203.
5 Washington, D.C: U.S. Nuclear Regulatory Commission. 2015b.

6 _____. "Identification and Prioritization of the Technical Information Needs Affecting Potential
7 Regulation of Extended Storage and Transportation of Spent Nuclear Fuel." ADAMS Accession
8 No. ML14043A423. Washington, DC: U.S. Nuclear Regulatory Commission. 2014a.

9 _____. NUREG-1536, "Standard Review Plan for Spent Fuel Dry Storage Systems at a
10 General License Facility." Rev 1. Washington, DC: U.S. Nuclear Regulatory Commission.
11 2010.

12 _____. Interim Staff Guidance-1, "Classifying the Condition of Spent Nuclear Fuel for Interim
13 Storage and Transportation Based on Function." Washington, DC: U.S. Nuclear Regulatory
14 Commission. 2007.

15 _____. "Safety Evaluation Report Related to the Topical Report for Castor V/21 Dry Spent Fuel
16 Storage Cask Submitted by General Nuclear Systems, Inc.." NRC-SER-85-9. Washington, DC:
17 U.S. Nuclear Regulatory Commission. 1985.

18 NWTRB. "Evaluation of the Technical Basis for Extended Dry Storage and Transportation of
19 Used Nuclear Fuel." Arlington, Virginia: United States Nuclear Waste Technical Review Board.
20 December 2010.

21 Palit, G.C. and H.S. Gadiyar. "Pitting Corrosion of Zirconium in Chloride Solution."
22 *CORROSION*. Vol. 43, No. 3. pp. 140-148. 1987.

23 Pan, G., A.M. Garde, and A.R. Atwood. "Performance and Property Evaluation of High Burn-up
24 Optimized ZIRLO™ Cladding." In *Proceedings of 17th International ASTM Symposium on*
25 *Zirconium in the Nuclear Industry*. Hyderabad, India. 2013.

26 PNNL. "Used Fuel Disposition Campaign: Gap Analysis to Support Extended Storage of
27 Used Nuclear Fuel Rev. 0." FCRD-USED-2011-000136. Rev. 0. PNNL-20509.
28 Richland, Washington. Pacific Northwest National Laboratory. January 31, 2012.

29 Raynaud, P.A.C. and R.E. Einziger. "Cladding Stress During Extended Storage of High Burnup
30 Spent Nuclear Fuel." *Journal of Nuclear Materials*. Vol. 464. pp. 304-312. 2015.

31 Rebak, R.B. "Chapter 7: Stress Corrosion Cracking (SCC) of Nickel-Based Alloys." In *Stress*
32 *Corrosion Cracking: Theory and Practice*. V.S. Raja and T. Shoji, eds. Cambridge, England:
33 Woodhead Publishing. pp. 273-306. 2011.

34 Rondinella, V.V., T. Wiss, E. Maugeri, J.Y. Colle, D. Wegen, and D. Papaioannou. "Effects of
35 He Build-up on Nuclear Fuel Evolution during Storage." *International Workshop on Spent Fuel*
36 *Integrity in Dry Storage*. Korea Atomic Energy Research Institute. Korea. November 4-5,
37 2010a.

- 1 Rondinella, V.V and T. Wiss. "The High Burnup Structure in Nuclear Fuel," *Materials Today*,
2 Vol. 13, pp. 24–32, 2010b.
- 3 Rondinella, V.V., T. Wiss, D. Papaioannou, and R. Nasyrow. "Studies on Nuclear Fuel
4 Evolution during Storage and Testing of Used Fuel Response to Impact Loadings." PSAM11
5 ESREL2012. Helsinki, June 25–29, 2012.
- 6 Rothman, A.J. "Potential Corrosion and Degradation Mechanisms of Zircaloy Cladding on
7 Spent Nuclear in a Tuff Repository." UCID-20172. Livermore, California: Lawrence Livermore
8 National Laboratory. 1984.
- 9 Samuels, I.E. *Metals Engineering: A Technical Guide*. Metals Park, Ohio: ASM International.
10 p. 116. 1988.
- 11 Sasahara, A. and T. Matsumura. "Post-Irradiation Examinations Focused on Fuel Integrity of
12 Spent BWR-MOX and PWR-UO₂ Fuels Stored for 20 Years." *Nuclear Engineering and Design*.
13 Vol. 238. pp. 1,250–1,259. 2008.
- 14 Scaglione, J.M., G. Radulescu, W.J. Marshall, and K.R. Robb. "A Quantitative Impact
15 Assessment of Hypothetical Spent Fuel Reconfiguration in Spent Fuel Storage Casks and
16 Transportation Packages." NUREG/CR-7203, ORNL/TM-2013/92. Oak Ridge, Tennessee:
17 Oak Ridge National Laboratory. 2015.
- 18 Shimada, S. and M. Nagai. "A Fractographic Study of Iodine-Induced Stress Corrosion
19 Cracking in Irradiated Zircaloy-2 Cladding." *Journal of Nuclear Materials*. Vol. 114. pp. 222–
20 230. 1983.
- 21 Shukla, P.K., R. Pabalan, T. Ahn, L. Yang, X. He, and H. Jung. "Cathodic Capacity of Alloy 22
22 in the Potential Yucca Mountain Repository Environment." *Proceedings of the CORROSION*
23 *2008 Conference, Corrosion in Nuclear Systems Symposium*, New Orleans, Louisiana,
24 March 16–20, 2008. Paper No. 08583. Houston, Texas: NACE International. 2008.
- 25 Sidky, P.S. "Iodine Stress Corrosion Cracking of Zircaloy Reactor Cladding: Iodine Chemistry
26 (A Review)," *Journal of Nuclear Materials*, Vol. 256, pp. 1–17. 1998.
- 27 Simpson, C.J. and C.E. Ells. "Delayed Hydrogen Embrittlement in Zr-2.5 wt % Nb." *Journal of*
28 *Nuclear Materials*. Vol. 52. pp. 289–295. 1974.
- 29 Sindelar, R.L., A.J. Duncan, M.E. Dupont, P.-S. Lam, M.R. Louthan, Jr., and T.E. Skidmore.
30 NUREG/CR-7116, "Materials Aging Issues and Aging Management for Extended Storage and
31 Transportation of Spent Nuclear Fuel." Washington, DC: U.S. Nuclear Regulatory Commission.
32 2011.
- 33 Spilker, H.M., H.-P Dyck Peehs, G. Kaspar, and K. Nissen. "Spent LWR Fuel Dry Storage in
34 Large Transport and Storage Casks after Extended Burnup." *Journal of Nuclear Materials*.
35 Vol. 250. pp. 63–74. 1997.
- 36 Thomazet, J. et al. "The Corrosion of the Alloy M5™: An Overview." IAEA Technical
37 Committee Meeting on Behavior of High Corrosion Zr-Based Alloys. Buenos Aires, Argentina:
38 October 24–28, 2005.

- 1 Torimaru, T., T. Yasuda, and M. Nakatsuka. "Changes in Mechanical Properties of Irradiated
2 Zircaloy-2 Fuel Cladding Due to Short-Term Annealing." *Journal of Nuclear Materials*. Vol. 238.
3 pp. 169–174. 1996.
- 4 Tsai, H. and M.C. Billone. NUREG/CP-0180, "Characterization of High-Burnup PWR and BWR
5 Rods, and PWR Rods After Extended Dry-Cask Storage." Proceedings of the 2002 Nuclear
6 Safety Research Conference, October 28–30, 2002. pp. 157–168. Washington, DC:
7 U.S. Nuclear Regulatory Commission. 2003.
- 8 Van Rooyen, D. and H.R. Copson. "Metal Corrosion in the Atmosphere." Report No. STP 435.
9 West Conshohocken, Pennsylvania: ASTM International. 1968.
- 10 Wang, J.-A. and H. Wang. NUREG/CR-7198, "Mechanical Fatigue Testing of High-Burnup Fuel
11 for Transportation Applications." ADAMS Accession No. ML15139A389. Washington, DC:
12 U.S. Nuclear Regulatory Commission. May 2015.
- 13 Wang, J. J.-A. "Cyclic Integrated Reversible-bending Fatigue Tester (CIRFT) Framework
14 Approaches and Analytical Evaluations." Oak Ridge National Laboratory (ORNL). Presented at
15 Extended Storage Collaboration Program (ESCP) Meeting. Electric Power Research Institute
16 (EPRI). Charlotte, North Carolina, December 2–4, 2014a.
- 17 Wang, J. J.-A. ORNL, 2014 ASTM C26 Committee Meeting, June, 2014b.
- 18 Wisner, S. and R. Adamson. "Combined Effects of Radiation Damage and Hydrides on the
19 Ductility of Zircaloy-2." *Nuclear Engineering and Design*. Vol. 185. pp. 33–49. 1998.
- 20 Wisner, S.B. and R.B. Adamson. "Embrittlement of Irradiated Zircaloy by Cadmium and Iodine."
21 *Embrittlement by Liquid and Solid Metals*. M.H. Kamdar, ed. Metallurgical Society of AIME.
22 pp. 437–456. 1982.
- 23 Yagee, F.L., R.F. Mattas, and L.A. Neimark. "Characterization of Irradiated Zircalloys:
24 Susceptibility to Stress Corrosion Cracking." Interim Report. EPRI NP-1557. Palo Alto,
25 California: Electric Power Research Institute. October 1980.
- 26 _____. "Characterization of Irradiated Zircalloys: Susceptibility to Stress Corrosion Cracking."
27 Interim Report, EPRI NP-1155. Palo Alto, California: Electric Power Research Institute.
28 September 1979.

4 ANALYSIS OF DRY STORAGE SYSTEMS AND SPENT FUEL ASSEMBLIES

4.1 Introduction

This chapter provides (1) a brief description of selected storage system designs and (2) aging management tables for each design that identify the aging mechanisms and effects that must be managed to ensure that the functions of structures, systems, and components (SSCs) are maintained in the period of extended operation. The analyses in Chapter 3 provide the technical bases for those determinations. The aging management tables also identify the use of either a time-limited aging analysis (TLAA), aging management program (AMP), or other analysis to address the aging effects that require management.

The following system descriptions are for general information only. In the review of a renewal application, the technical reviewer should refer to the application, safety analysis report, and drawings to identify the SSCs within the scope of renewal and their functions, materials of construction, and operating environment. Table 4-1 describes the storage system designs that are discussed below and evaluated in the aging management tables.

Table 4-1 Evaluated storage system designs			
MAPS Section No.	Name	NRC Docket No.	Amendments Evaluated
4.2	Standardized NUHOMS®*	72-1004	1–11 and 13
	Standardized Advanced NUHOMS®	72-1029	1 and 3
4.3	HI-STORM 100	72-1014	1–10
	HI-STAR 100	72-1008	1–2
4.4	TN-32	72-1021	1
	TN-68	72-1027	1
4.5	NAC-UMS	72-1015	1–5
	NAC-MPC	72-1025	1–6
	MAGNASTOR	72-1031	1–6
4.6	FuelSolutions	72-1026	1–4
4.7	Concrete Pad (generic)	—	—
4.8	Spent Fuel Assemblies (generic)	—	—

*The staff's review of the Calvert Cliffs specific license renewal application (NRC, 2014) informed the evaluation of the NUHOMS system, and thus the aging management tables for this system may include some unique elements of this site.

1 **4.2 NUHOMS® systems: Standardized and Standardized Advanced**

2 **4.2.1 System description**

3 The NUHOMS family of modular storage systems provide for the horizontal storage of spent
4 nuclear fuel (SNF) in a dry shielded canister (DSC) that is placed in a concrete horizontal
5 storage module (HSM). Each NUHOMS system model type is designated by NUHOMS-XXY.
6 The two digits (XX) refer to the number of fuel assemblies stored in the DSC, and the character
7 (Y) designates the type of fuel being stored—P for pressurized-water reactor (PWR) or B for
8 boiling-water reactor (BWR). For some systems, a fourth character (T) is added to designate
9 that the DSC is also intended for transportation in packages approved under Title 10 of the
10 *Code of Federal Regulations* (10 CFR) Part 71, “Packaging and Transportation of Radioactive
11 Material.” Also, two additional characters, HB, are added for systems that are used to store
12 high-burnup fuels (e.g., NUHOMS-24PHB).

13 The Standardized NUHOMS design is presently licensed for use in the United States under
14 NRC Docket 72-1004, in combination with the 24P, 24PT2, 24PHB, 24PTH, 32PT, 32PTH1,
15 37PTH, 52B, 61BT, 61BTH, and 69BTH DSCs, while the Standardized Advanced NUHOMS
16 design is licensed for use under NRC Docket 72-1029, with the 24PT1, 24PT4, and 32PTH2
17 DSCs. The principal components of the NUHOMS system include (i) a stainless steel DSC with
18 an internal basket to hold SNF assemblies, (ii) a structural steel assemblage that supports the
19 DSC, and (iii) an HSM that is constructed of reinforced concrete (see Figure 4-1). Additional
20 components include an onsite transfer cask (TC) and other fuel transfer and auxiliary equipment
21 used to support DSC loading and transfer operations.

22 The Standardized Advanced system differs from the Standardized system in that it includes
23 modifications to accommodate sites with high seismic levels, limited space, and needs for
24 enhanced radiation shielding performance. To accomplish this, a modified version of the HSM
25 was created, designated as the Advanced Horizontal Storage Modulue (AHSM). A brief
26 summary of the components of the NUHOMS storage systems are provided below.

27 **4.2.2 Dry shielded canister**

28 The NUHOMS DSC is a welded stainless steel canister that uses redundant multipass closure
29 welds. After fuel loading, draining and drying, the canister is backfilled with helium to provide an
30 inert environment. Figure 4-2 and Figure 4-3 show the components of two DSC configurations,
31 which comprise the shell assembly and the internal basket assembly.

32 Shell assembly

33 The DSC shell assembly consists of a stainless steel cylindrical shell that is joined to top and
34 bottom end assemblies with double, redundant seal welds to form the confinement boundary.
35 The bottom end assembly welds are made during fabrication of the DSC, while the top end
36 assembly welds are made after fuel loading. The shell assembly also includes two shielding
37 plugs at both ends for biological shielding. Siphon and vent ports penetrate the top shield plug
38 and are sealed after DSC drying operations are complete. Figure 4-4 shows the pressure and
39 confinement boundaries for the NUHOMS-32PT DSC.

40

41

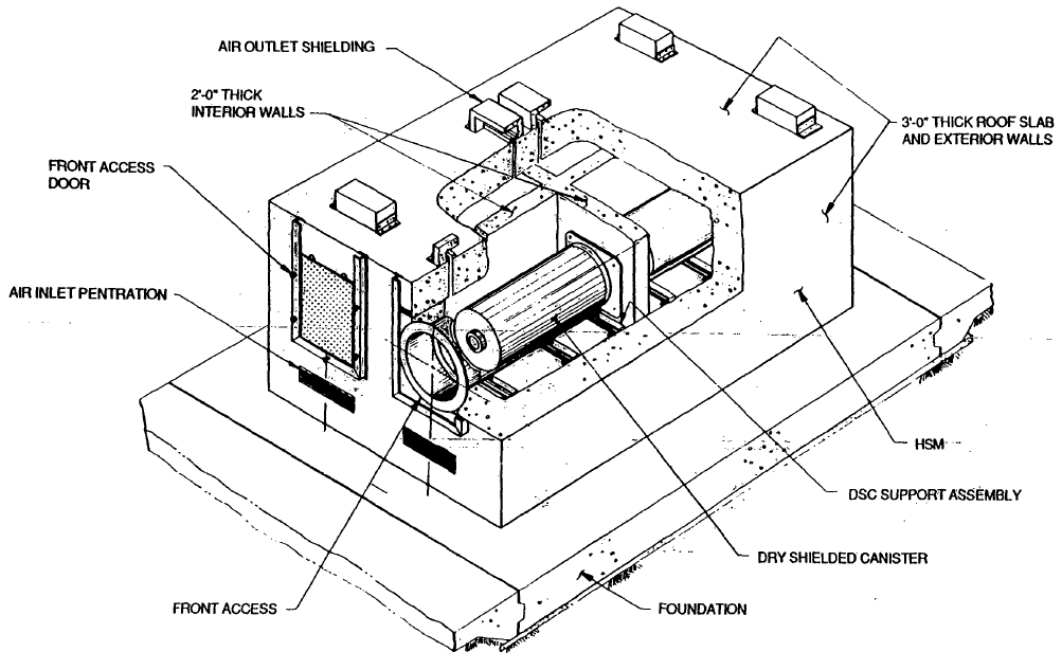


Figure 4-1 NUHOMS dry storage system (Pacific Nuclear Fuel Services, Inc., 1991)

1

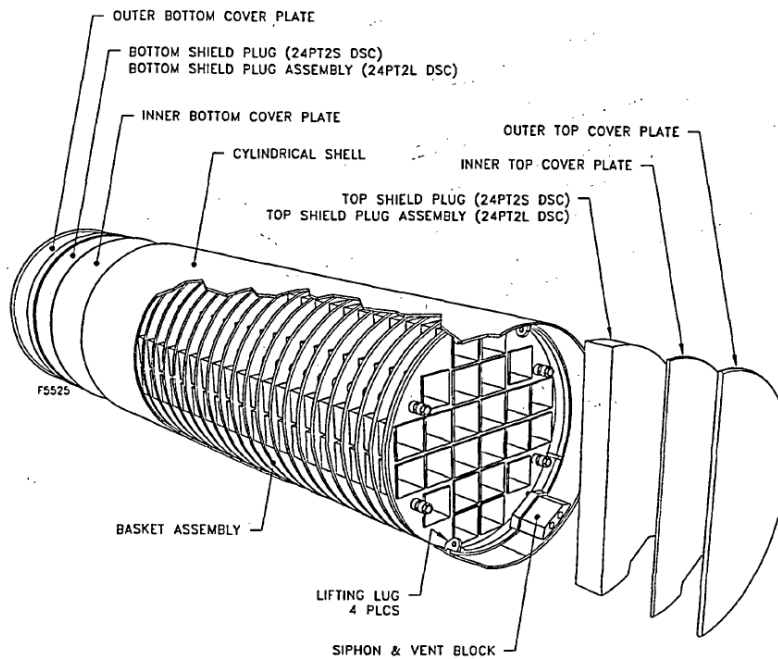


Figure 4-2 NUHOMS-24PT2 DSC assembly—spacer disk design (Transnuclear, 2004)

2

3

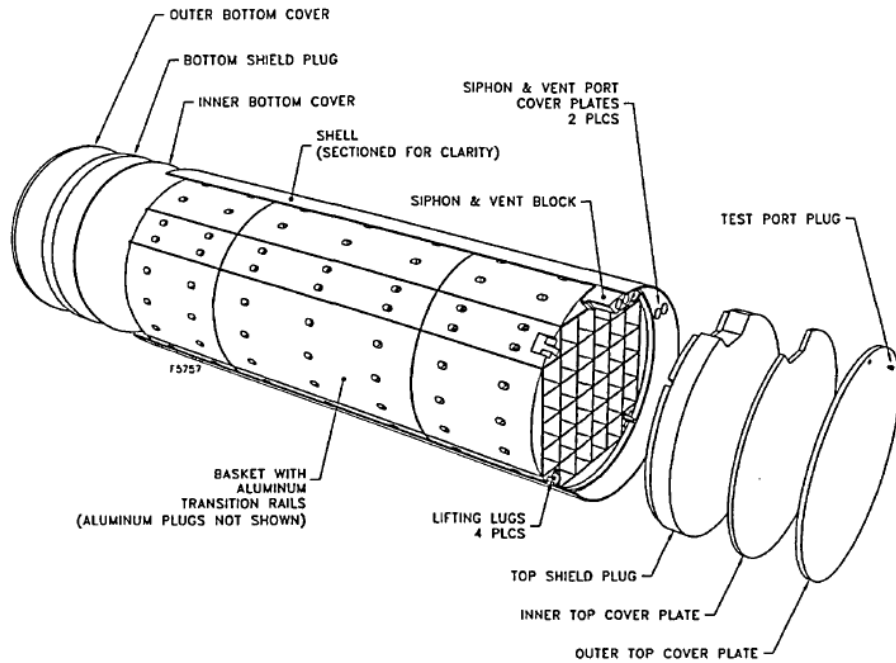


Figure 4-3 NUHOMS-32PT DSC assembly-tube or plate design (Transnuclear, 2004)

1

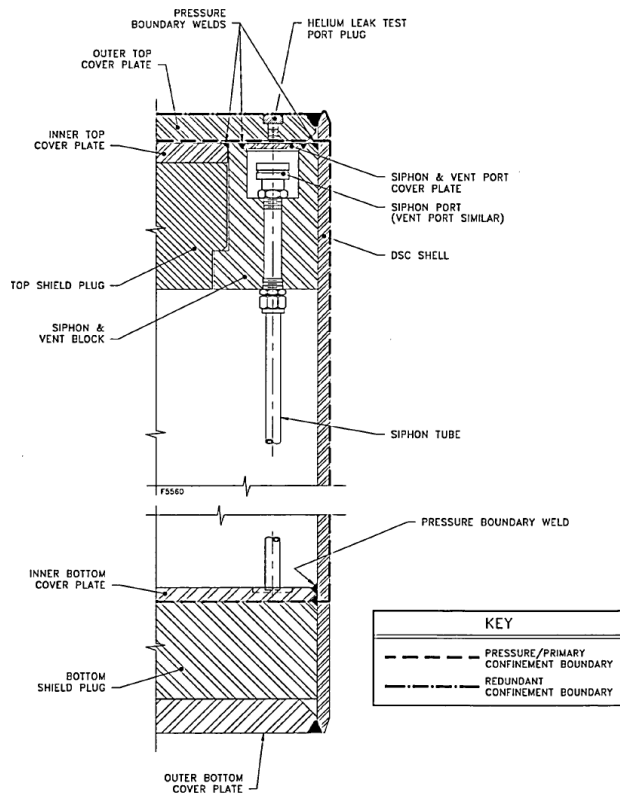


Figure 4-4 Pressure and confinement boundaries for NUHOMS-32PT DSC (Transnuclear, 2004)

2

1 Internal Basket Assembly

2 The internal basket assembly contains a storage position for each fuel assembly. The basket
3 assembly may consist of an assemblage of spacer disc plates supported on vertical rods that
4 extend the length of the DSC cavity (spacer disc design) or individual tubes or plates welded to
5 form a grid-like structure (tube or plate design).

6 The 24P, 24PT1, 24PT2, 24PT4, 24PHB, and 52B DSCs use the spacer disc basket design, as
7 shown in Figure 4-2. Subcriticality is maintained through the geometric separation of the fuel
8 assemblies by the DSC basket assembly and the neutron absorbing capability of the DSC
9 materials of construction. The 52B DSC contains fixed neutron poison material for additional
10 criticality control.

11 The 61BT, 32PT, 24PTH, 61BTH, 32PTH1, 32PTH2, 69BTH, and 37PTH DSCs use the tube or
12 plate grid basket design, as shown in Figure 4-3. Fixed neutron poison material provides the
13 necessary criticality control. Aluminum sheets or plates are used to provide the heat conduction
14 paths from the fuel assemblies to the canister shell. Transition rails, consisting of welded
15 stainless steel plates or aluminum parts, form the transition between the box-like fuel
16 compartment structure and the cylindrical DSC shell.

17 Table 4-2 and Table 4-3 evaluate potential aging mechanisms and effects requiring
18 management for specific components of the Standardized and Standardized Advanced
19 NUHOMS DSC shell and basket designs. The tables also identify AMPs that provide an
20 acceptable approach to managing the aging effects.

21 **4.2.3 Horizontal storage module**

22 Both the HSM and AHSM storage modulus are low-profile structures constructed from
23 reinforced concrete and structural steel that provides a means for passive removal of spent fuel
24 decay heat, structural support and environmental protection of the DSC, and radiation shielding.
25 The AHSM design is similar to the HSM; however, the AHSM contains improved shielding and
26 resistance to high seismic events. The AHSM consists a base storage unit and a top shield
27 block that is tied to the base unit by steel rods in the vertical direction and interlocking concrete
28 keys in the horizontal direction.

29 Heat removal is achieved by a combination of radiation, conduction, and convection. As shown
30 in Figure 4-5 and Figure 4-6, ambient air enters the HSMs through ventilation inlet openings
31 located in the lower region of the front or side walls and circulates around the DSC. Air exits
32 through outlet openings in the top regions of the HSM walls. Thermal monitoring or visual
33 inspections are used to provide indication of HSM performance or a blocked vent condition.
34 Environmental protection and radiation shielding are provided by the thick side walls and roof of
35 the HSM, supplemented by thick wall units attached at the ends of the array and at the rear
36 walls of the HSM if the array is of single row configuration. Each HSM has an access opening
37 or docking flange in the front wall to accommodate transfer of DSCs from and into the shielded
38 TC. The access opening is covered by a thick shielded access door.

39

40

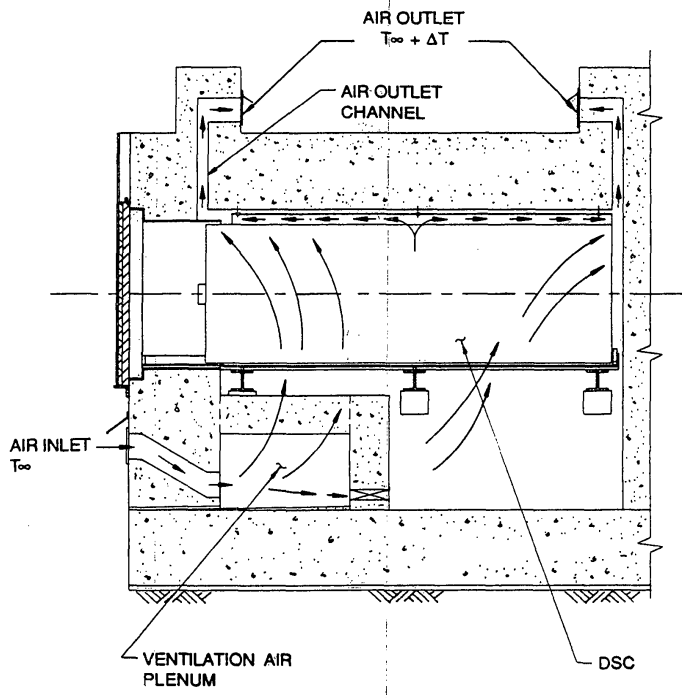


Figure 4-5 Air flow diagram for a typical standardized HSM design (Pacific Nuclear Fuel Services, Inc., 1991)

1

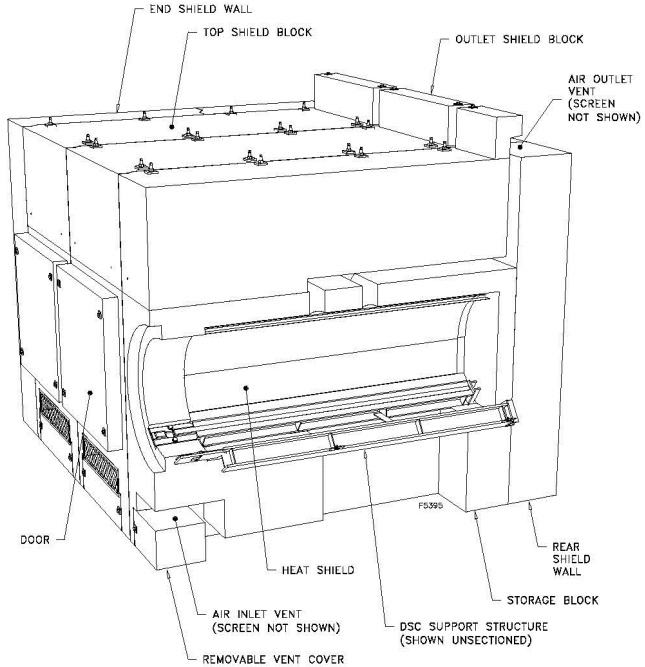


Figure 4-6 Advanced NUHOMS horizontal storage module (AHSM) (Transnuclear, 2003)

2

3

1 Structural support of the loaded DSC is provided by a structural steel frame structure
 2 (HSM model 80 and model 102) anchored to the floor slab and walls of the HSM, or a structural
 3 steel rail assembly (HSM models HSM-H, -152, -202, HSM-HS, AHSM, and AHSM-HS).
 4 Figure 4-7 shows drawings of the side elevation and end view of the DSC rail assembly.
 5 Stainless steel cover plates coated with a dry film lubricant are attached to the rails to provide a
 6 sliding surface for DSC insertion and retrieval. In some designs, Nitronic 60 plates are welded
 7 to the cover plates because of this material's good high-temperature properties and resistance
 8 to oxidation, wear, and galling. Seismic restraints using steel plates or tubes are welded to the
 9 rear and front of the rails for retaining the DSC in place during seismic events.

10 Table 4-4 and Table 4-5 provide a generic evaluation of potential aging mechanisms and effects
 11 requiring management for specific components of the Standardized NUHOMS HSM and
 12 Standardized Advanced NUHOMS AHSM. The tables also identify the AMPs that provide an
 13 acceptable approach to managing the effects of aging.

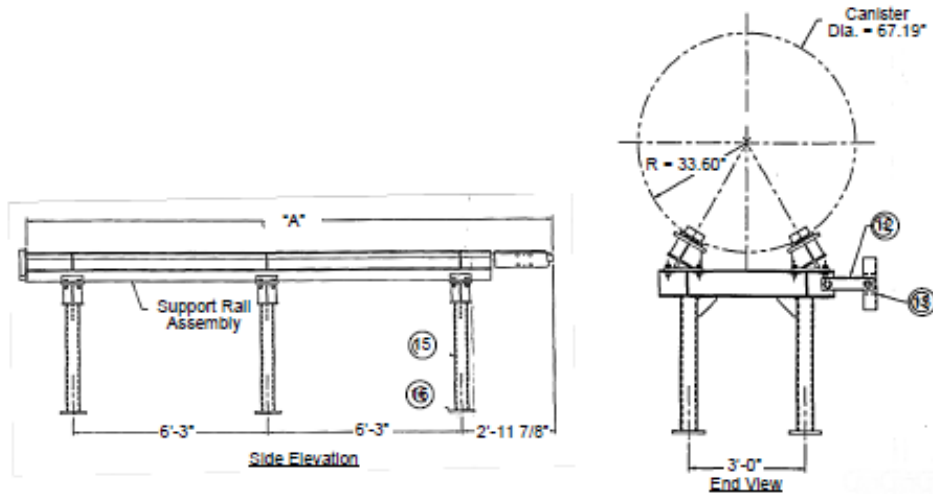


Figure 4-7 Side elevation and end view of the DSC support structure (Transnuclear, 2004)

14 **4.2.4 Transfer cask**

15 The NUHOMS TC is a cylindrical vessel with a bolted top cover plate and a welded bottom end
 16 assembly (Transnuclear, 2014). There are five alternate configurations of the cask.

- 17 • The basic configuration, where the TC is provided with a solid neutron shield, is denoted
 18 as the standardized onsite cask.
- 19 • A second configuration includes the OS197 and OS197H (H: modified for increased
 20 strength), in which water is used to provide neutron shielding.
- 21 • The third configuration, designated as OS197FC, OS197HFC OS197FC-B, or
 22 OS197HFC-B TC, is equipped with a modified top lid to allow air circulation through the
 23 annulus between the DSC and the TC.
- 24 • The fourth configuration, designated as OS197L TC and shown in Figure 4-8, is a
 25 reduced weight version of the OS197 TC.

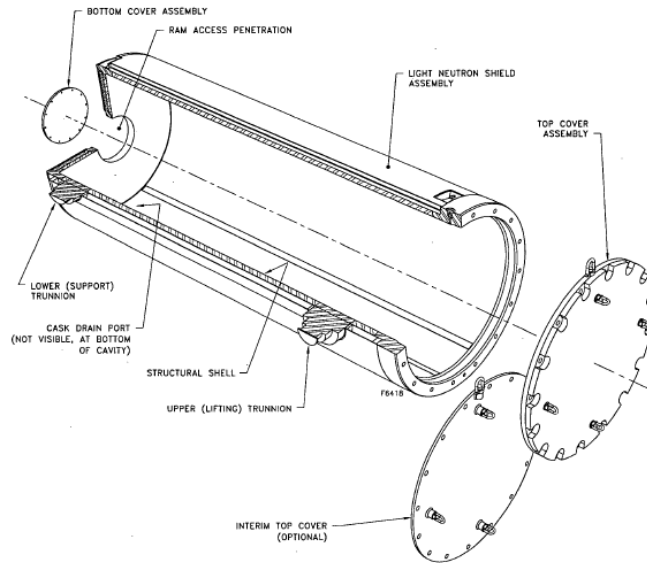


Figure 4-8 OS197L transfer cask (Transnuclear, 2008)

- 1 • The fifth configuration is designated as OS200 or OS200FC TC and has a larger
- 2 diameter to accommodate the larger diameter DSCs with 32PTH1, 37PTH, or 69BTH
- 3 SNF assemblies.

- 4 For all the configurations except the OS197L TC, the TCs are constructed from two concentric
- 5 cylindrical shells: a stainless steel inner shell and a structural shell made of stainless steel or
- 6 carbon steel. The annulus formed by these two shells is filled with cast lead to provide gamma
- 7 shielding. The TC also includes an outer jacket made of stainless steel or carbon steel, which is
- 8 filled with BISCO NS-3 material or water for neutron shielding. The inner and structural shells
- 9 are welded to heavy forged ring assemblies at the top and bottom ends. The bottom end plate
- 10 has a removable stainless steel ram access penetration ring. A stainless steel bottom cover
- 11 plate is provided to seal the hydraulic ram access penetration of the cask during fuel loading.
- 12 Rails fabricated from a nongalling, wear-resistant stainless steel coated with a high contact
- 13 pressure dry film lubricant are provided to facilitate DSC transfer.

- 14 The OS197L TC is constructed from a single, thicker stainless steel structural shell. To
- 15 compensate for the lack of lead shielding, the OS197L TC relies on the use of supplemental
- 16 shielding in conjunction with remote operations during handling in the fuel or reactor building,
- 17 transfer to the ISFSI, and insertion into the HSM operations. The cask support skid
- 18 supplemental shielding consists of a thick carbon steel upper shielding bell and a lower
- 19 shielding sleeve that enclose the TC in the decontamination area, and thick carbon steel plates
- 20 and covers that enclose the TC while on the transfer trailer.

- 21 The NUHOMS TCs have four trunnions made of stainless steel or nickel alloy that are welded to
- 22 the structural shell. Two upper lifting trunnions are located near the top of the cask for lifting the
- 23 cask in the SNF pool building. The lower trunnions, located near the base of the cask, serve as
- 24 the axis of rotation and as supports during transport to the HSM.

- 25 Table 4-6 provides a generic evaluation of potential aging mechanisms and effects requiring
- 26 management for specific components of the NUHOMS transfer casks. The table also identifies
- 27 the AMPs that provide an acceptable approach to managing the aging effects.

Table 4-2 Standardized NUHOMS dry shielded canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Guide sleeves (DSC basket)	CR, SR, TH*	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
Oversleeves (DSC basket)	CR, SR, TH	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
Aluminum plate or sheet, basket plate, compartment plate (DSC basket)	CR, SH, TH	Aluminum	Helium	Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of strength	No	3.2.3.7
				Creep	Change in dimensions	No	3.2.3.5
Spacer disks (DSC basket)	CR, SR	Stainless steel	Helium	General corrosion	Loss of material	No	3.2.3.1
				Radiation embrittlement	Cracking	No	3.2.3.8
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in Dimensions	No	3.2.2.6
Steel	CR, SR	Steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8
Steel	CR, SR	Steel	Helium	Creep	Change in dimensions	No	3.2.1.6
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8

*Safety Functions: Confinement (CO), Subcriticality (CR), Retrievalability (RE), Radiation Shielding (SH), Structural Integrity (SR), Thermal/Heat Removal (TH)

Table 4-2 Standardized NUHOMS dry shielded canister								
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)	
Spacer disks (DSC basket)	CR, SR	Steel	Helium	General corrosion Radiation embrittlement	Loss of material Cracking	No No	3.2.1.1 3.2.1.9	
		Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8	
Support rods (DSC basket)	CR, SR	Stainless steel	Helium	Creep	Change in dimensions Cracking	No No	3.2.2.6 3.2.2.9	
		Stainless steel (welded 17-4 PH)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	TLAA/AMP or a supporting analysis is required	3.2.2.8	
		Stainless steel (17-4 PH)	Helium	Creep	Change in dimensions Cracking	No No	3.2.2.6 3.2.2.9	
		Steel	Helium	Radiation embrittlement Thermal aging	Loss of fracture toughness and loss of ductility	No No	3.2.1.8	
					Creep	Change in dimensions	No	3.2.1.6
					General corrosion Radiation embrittlement	Loss of material Cracking	No No	3.2.1.1 3.2.1.9
Spacer sleeves (DSC basket)	CR, SR	Stainless steel (welded 17-4 PH)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	TLAA/AMP or a supporting analysis is required	3.2.2.8	
		Stainless steel	Helium	Creep	Change in dimensions Cracking	No No	3.2.2.6 3.2.2.9	

Table 4-2 Standardized NUHOMS dry shielded canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Basket rails (DSC basket)	CR, SH, SR, TH	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
Basket rail inserts and shims (DSC basket)	SR, TH	Aluminum	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of strength	TLAA/AMP or a supporting analysis is required	3.2.3.7
		Creep	Change in dimensions	TLAA/AMP or a supporting analysis is required	3.2.3.5		
		General corrosion	Loss of material	No	3.2.3.1		
Basket assembly plates (DSC basket)	CR, SH, SR, TH	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.3.8
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Grid assembly (DSC basket)	SR	Stainless steel	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-2 Standardized NUHOMS dry shielded canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Holdown ring assembly and plates (DSC basket)	SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
		Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
Fuel compartment tubes, wraps, inserts (DSC basket)	CR, SH, SR, TH	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
		Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
Transition rails (DSC basket)	CR, SH, SR, TH	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
		Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
		Aluminum	Helium	Thermal aging	Loss of strength	TLAA/AMP or a supporting analysis is required	3.2.3.7
		Aluminum	Helium	Creep	Change in dimensions	TLAA/AMP or a supporting analysis is required	3.2.3.5
		Aluminum	Helium	General corrosion	Loss of material	No	3.2.3.1
			Radiation embrittlement	Cracking	No	3.2.3.8	

Table 4-2 Standardized NUHOMS dry shielded canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Neutron absorbing plates, poison plates (DSC basket)	CR, TH	Borated stainless steel	Helium	Boron depletion	Loss of criticality control	No; a TLAA may be required	3.4.1.1
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.4.1.3
				Creep	Change in dimensions	No	3.4.1.2
				Radiation embrittlement	Loss of fracture toughness and loss of ductility	No	3.4.1.4
				Boron depletion	Loss of criticality control	No; a TLAA may be required	3.4.2.4
				Thermal aging	Loss of strength	No	3.4.2.6
				General corrosion	Loss of material	No	3.4.2.1
				Creep	Change in dimensions	No	3.4.2.5
				Radiation embrittlement	Loss of fracture toughness and loss of ductility	No	3.4.2.7
				Boron depletion	Loss of criticality control	No; a TLAA may be required	3.4.2.4
Neutron absorbing plates or sheets, poison plates, chevron neutron absorbers (DSC basket)	CR, SH, TH	Boralyn®, Metamic™	Helium	Thermal aging	Loss of strength	No	3.4.2.6
				General corrosion	Loss of material	No	3.4.2.1
				Creep	Change in dimensions	No	3.4.2.5
				Radiation embrittlement	Loss of fracture toughness and loss of ductility	No	3.4.2.7
				Boron depletion	Loss of criticality control	No; a TLAA may be required	3.4.2.4
				Thermal aging	Loss of strength	No	3.4.2.6
				Wet corrosion and blistering	Change in dimensions	No	3.4.2.3
Boral®	Helium	Boral®	Creep	Change in dimensions	No	3.4.2.5	
			Radiation embrittlement	Loss of fracture toughness and loss of ductility	No	3.4.2.7	

Table 4-2 Standardized NUHOMS dry shielded canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Neutron absorbing plates or sheets, poison plates, chevron neutron absorbers (DSC basket)	CR, SH, TH	Borated aluminum	Helium	Boron depletion	Loss of criticality control	No; a TLAA may be required	3.4.2.4
				Thermal aging	Loss of strength	No	3.4.2.6
				General corrosion	Loss of material	No	3.4.2.1
				Creep	Change in dimensions	No	3.4.2.5
				Radiation embrittlement	Loss of fracture toughness and loss of ductility	No	3.4.2.7
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8
				Creep	Change in dimensions	No	3.2.1.6
Support bars (DSC basket)	SR	Steel	Helium	General corrosion	Loss of material	No	3.2.1.1
				Radiation embrittlement	Cracking	No	3.2.1.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8
				Creep	Change in dimensions	No	3.2.1.6
Fastener components	SR	Stainless steel	Helium	General corrosion	Loss of material	No	3.2.1.1
				Radiation embrittlement	Cracking	No	3.2.1.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8
				Creep	Change in dimensions	No	3.2.1.6
Fastener components	SR	Steel	Helium	General corrosion	Loss of material	No	3.2.1.1
				Radiation embrittlement	Cracking	No	3.2.1.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8

Table 4-2 Standardized NUHOMS dry shielded canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Fastener components	SR	Steel	Helium	Radiation embrittlement	Cracking	No	3.2.1.9
Tool socket and closure plate	SR	Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
Components for damaged fuel	CO, SR	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
		Stainless steel	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
Cover plates (inner)	CO, SH, SR	Stainless steel (welded)	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
		Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
		Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
Shield plug (top)	CO, SH, SR, TH	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
		Stainless steel	Helium	Creep	Change in Dimensions	No	3.2.2.6
Shield plug (top)	CO, SH, SR, TH	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
		Stainless steel	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
Shield plug (top)	CO, SH, SR, TH	Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
		Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-2 Standardized NUHOMS dry shielded canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Shield plug (top)	CO, SH, SR, TH	Steel	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Creep	Change in dimensions	No	3.2.1.6
				General corrosion	Loss of material	No	3.2.1.1
				Radiation embrittlement	Cracking	No	3.2.1.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Shield plug (bottom)	CO, SH, SR, TH	Stainless steel	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Creep	Change in dimensions	No	3.2.1.6
				General corrosion	Loss of material	No	3.2.1.1
				Radiation embrittlement	Cracking	No	3.2.1.9
Lead shielding	SH	Lead	Embedded (steel, stainless steel)	None identified	None identified	No	3.2.6
				None identified	None identified	No	3.2.6

Table 4-2 Standardized NUHOMS dry shielded canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Siphon and vent block	CO, SH, SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Siphon and vent port cover plate	CO, SH, SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Test port plug	CO	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Key, shear key	SR	Stainless steel	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Creep	Change in dimensions	No	3.2.2.6		
		Radiation embrittlement	Cracking	No	3.2.2.9		
		Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8		
		Creep	Change in dimensions	No	3.2.2.6		
		Radiation embrittlement	Cracking	No	3.2.2.9		
		Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8		
		Creep	Change in dimensions	No	3.2.2.6		
		Radiation embrittlement	Cracking	No	3.2.2.9		

Table 4-2 Standardized NUHOMS dry shielded canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Pin, anti-rotation pin	SR	Stainless steel	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
DSC support ring	SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in Dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Lifting lugs	SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
DSC shell	CO, SH, SR, TH	Stainless steel (welded)	Sheltered	Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5
				Pitting and crevice corrosion	Loss of material (precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2

Table 4-2 Standardized NUHOMS dry shielded canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
DSC shell	CO, SH, SR, TH	Stainless steel	Sheltered	Galvanic corrosion	Loss of material	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.3
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
Cover plates (outer)	CO, SH, SR, TH	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5
				Pitting and crevice corrosion	Loss of material (precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-2 Standardized NUHOMS dry shielded canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Grapple ring and grapple support	SR	Stainless steel (welded)	Sheltered	Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5
		Stainless steel	Sheltered	Pitting and crevice corrosion	Loss of material (precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2
				Microbiologically influenced corrosion Radiation embrittlement	Loss of material Cracking	No No	3.2.2.4 3.2.2.9

Table 4-3 Standardized Advanced NUHOMS dry shielded canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Guide sleeves (DSC basket)	CR, SR, TH*	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Oversleeves, stop plates (DSC basket)	CR, SR, TH	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Spacer disks (DSC basket)	CR, SR	Steel	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8
				Creep	Change in dimensions	No	3.2.1.6
				General corrosion	Loss of material	No	3.2.1.1
Support rods (DSC basket)	CR, SR	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.1.9
		Stainless steel	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9

*Safety Functions: Confinement (CO), Subcriticality (CR), Retrievalability (RE), Radiation Shielding (SH), Structural Integrity (SR), Thermal/Heat Removal (TH)

Table 4-3 Standardized Advanced NUHOMS dry shielded canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Support rods (DSC basket)	CR, SR	Stainless steel (welded 17-4 PH)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	TAA/AMP or a supporting analysis is required	3.2.2.8
		Stainless steel (17-4 PH)	Helium	Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Spacer sleeves (DSC basket)	CR, SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
		Stainless steel (welded 17-4 PH)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	TAA/AMP or a supporting analysis is required	3.2.2.8
		Stainless steel (17-4 PH)	Helium	Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Shims (DSC basket)	SR, TH	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
		Aluminum	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of strength	TAA/AMP or a supporting analysis is required	3.2.3.7

Table 4-3 Standardized Advanced NUHOMS dry shielded canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Shims (DSC basket)	SR, TH	Aluminum	Helium	Creep	Change in dimensions	TLAA/AMP or a supporting analysis is required	3.2.3.5
				General corrosion	Loss of material	No	3.2.3.1
				Radiation embrittlement	Cracking	No	3.2.3.8
Basket assembly plates (DSC basket)	CR, SH, SR, TH	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Fuel compartment (DSC basket)	CR, SH, SR, TH	Aluminum	Helium	Thermal aging	Loss of strength	TLAA/AMP or a supporting analysis is required	3.2.3.7
				Creep	Change in dimensions	TLAA/AMP or a supporting analysis is required	3.2.3.5
				General corrosion	Loss of material	No	3.2.3.1
				Radiation embrittlement	Cracking	No	3.2.3.8
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
		Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-3 Standardized Advanced NUHOMS dry shielded canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Transition rails (DSC basket)	CR, SH, SR, TH	Aluminum	Helium	Thermal aging	Loss of strength	TLAA/AMP or a supporting analysis is required	3.2.3.7
				Creep	Change in dimensions	TLAA/AMP or a supporting analysis is required	3.2.3.5
				General corrosion	Loss of material	No	3.2.3.1
				Radiation embrittlement	Cracking	No	3.2.3.8
Neutron absorbing poison plate (DSC basket)	CR, SH, TH	Boron carbide/aluminum metal-matrix	Helium	Boron depletion	Loss of criticality control	No; a TLAA may be required	3.4.2.4
				Thermal aging	Loss of strength	No	3.4.2.6
				General corrosion	Loss of material	No	3.4.2.1
				Creep	Change in dimensions	No	3.4.2.5
				Radiation embrittlement	Loss of fracture toughness and loss of ductility	No	3.4.2.7
				Boron depletion	Loss of criticality control	No; a TLAA may be required	3.4.2.4
				Thermal aging	Loss of strength	No	3.4.2.6
Neutron absorbing sheets (DSC basket)	CR, SH, TH	Boral®	Helium	Wet corrosion and blistering	Change in dimensions	No	3.4.2.3

Table 4-3 Standardized Advanced NUHOMS dry shielded canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Neutron absorbing sheets (DSC basket)	CR, SH, TH	Boral®	Helium	Creep	Change in dimensions	No	3.4.2.5
				Radiation embrittlement	Loss of fracture toughness and loss of ductility	No	3.4.2.7
Fastener components	SR	Steel	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8
				Creep	Change in dimensions	No	3.2.1.6
				General corrosion	Loss of material	No	3.2.1.1
				Radiation embrittlement	Cracking	No	3.2.1.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
Components for damaged fuel	CO, SR	Stainless steel (welded)	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Cover plates (inner)	CO, SH, SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
		Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6

Table 4-3 Standardized Advanced NUHOMS dry shielded canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Cover plates (inner)	CO, SH, SR	Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
Shield plug (top)	CO, SH, SR, TH	Stainless steel	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8
Shield plug casing (bottom)	CO, SH, SR, TH	Steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Creep	Change in dimensions	No	3.2.1.6
				General corrosion	Loss of material	No	3.2.1.1
				Radiation embrittlement	Cracking	No	3.2.1.9
				Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5

Table 4-3 Standardized Advanced NUHOMS dry shielded canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Shield plug casing (bottom)	CO, SH, SR, TH	Stainless steel	Sheltered	Pitting and crevice corrosion	Loss of material (Precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
				Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Creep	Change in dimensions	No	3.2.1.6
				General corrosion	Loss of material	No	3.2.1.1
				Radiation embrittlement	Cracking	No	3.2.1.9
Lead shielding	SH	Lead	Embedded (stainless steel)	None identified	None identified	No	3.2.6
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
Siphon and vent block	CO, SH, SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8

Table 4-3 Standardized Advanced NUHOMS dry shielded canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Siphon and vent block	CO, SH, SR	Stainless steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Siphon and vent port cover plate	CO, SH, SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
Test port plug	CO	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
Pin	SR	Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8

Table 4-3 Standardized Advanced NUHOMS dry shielded canister								
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)	
DSC support ring	SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8	
		Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6	
				Radiation embrittlement	Cracking	No	3.2.2.9	
DSC shell	CO, SH, SR, TH	Stainless steel (welded)	Sheltered	Stress corrosion cracking	Cracking	Localized Corrosion and Stress Corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5	
		Stainless steel	Sheltered	Pitting and crevice corrosion	Loss of material (Precursor to stress corrosion cracking)	Localized Corrosion and Stress Corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2	
				Galvanic corrosion	Loss of material	Localized Corrosion and Stress Corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.3	
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4	
				Fatigue	Cracking	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9	
		Radiation embrittlement	Cracking	No	3.2.2.9			

Table 4-3 Standardized Advanced NUHOMS dry shielded canister								
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)	
Cover plates (outer)	CO, SH, SR, TH	Stainless steel (welded)	Sheltered	Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5	
		Stainless steel	Sheltered	Pitting and crevice corrosion	Loss of material (Precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2	
Grapple ring and grapple support	SR			Microbiologically influenced corrosion	Loss of material	No	3.2.2.4	
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7	
				Radiation embrittlement	Cracking	No	3.2.2.9	
		Stainless steel (welded)	Sheltered	Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5	
					Pitting and crevice corrosion	Loss of material (Precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2
					Microbiologically influenced corrosion	Loss of material	No	3.2.2.4

Table 4-3 Standardized Advanced NUHOMS dry shielded canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Grapple ring and grapple support	SR	Stainless steel	Sheltered	Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-4 Standardized NUHOMS horizontal storage module

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)						
Reinforced concrete: base walls, floor slab, roof; basemat; end and rear shield walls, corner shield wall; shielded ventilation air inlet plenum; inlet/outlet vents	SH, SR, TH*	Concrete	Air—outdoor	Aggressive chemical attack	Cracking	Reinforced Concrete Structures AMP	3.5.1.5						
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.5						
					Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.5						
				Reduction of concrete pH (reducing corrosion resistance of steel embedments)	Reinforced Concrete Structures AMP	3.5.1.5	Creep	Cracking	No	3.5.1.2			
								Cracking	No	3.5.1.11			
				Loss of strength	No	3.5.1.11		Dehydration at high temperature	Loss of strength	No	3.5.1.11		
									Loss of material (spalling, scaling)	No	3.5.1.13		
				Loss of strength	No	3.5.1.13			Delayed ettringite formation	Cracking	No	3.5.1.13	
										Cracking	No	3.5.1.13	
				Reinforced Concrete Structures AMP	3.5.1.4	Differential settlement				Cracking	Reinforced Concrete Structures AMP	3.5.1.4	
										Cracking	No	3.5.1.10	
				Reinforced Concrete Structures AMP	3.5.1.1					Fatigue	Cracking	Reinforced Concrete Structures AMP	3.5.1.1
											Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.1
Cracking	No	Radiation damage	Cracking	No	3.5.1.9								
			Loss of strength	No	3.5.1.9								

*Safety Functions: Confinement (CO), Subcriticality (CR), Retrievalability (RE), Radiation Shielding (SH), Structural Integrity (SR), Thermal/Heat Removal (TH)

Table 4-4 Standardized NUHOMS horizontal storage module

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)		
Reinforced concrete: base walls, floor slab, roof, basemat; end and rear shield walls, corner shield wall; shielded ventilation air inlet plenum; inlet/outlet vents	SH, SR, TH	Concrete	Air—outdoor	Reaction with aggregates	Cracking	Reinforced Concrete Structures AMP	3.5.1.3		
				Salt scaling	Loss of strength	Reinforced Concrete Structures AMP	3.5.1.3		
				Shrinkage	Cracking	No	3.5.1.7		
				Leaching of calcium hydroxide	Loss of strength	Reinforced Concrete Structures AMP	3.5.1.8		
				Aggressive chemical attack	Increase in porosity and permeability	Reinforced Concrete Structures AMP	3.5.1.8		
					Reduction of concrete pH (reducing corrosion resistance of steel embedments)	Reinforced Concrete Structures AMP	3.5.1.8		
				Sheltered	Aggressive chemical attack	Loss of strength	No	3.5.1.5	
						Cracking	No	3.5.1.5	
						Loss of material (spalling, scaling)	No	3.5.1.5	
						Creep	No	3.5.1.2	
						Dehydration at high temperature	No	3.5.1.11	
						Delayed ettringite formation	No	3.5.1.11	
						Differential settlement	Loss of strength	No	3.5.1.13
							Cracking	No	3.5.1.13
Fatigue	Fatigue	Reinforced Concrete Structures AMP	3.5.1.4						
		No	3.5.1.10						

Table 4-4 Standardized NUHOMS horizontal storage module

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)	
Reinforced concrete: base walls, floor slab, roof; basemat; end and rear shield walls, corner shield wall; shielded ventilation air inlet plenum; inlet/outlet vents	SH, SR, TH	Concrete	Sheltered	Freeze and thaw	Cracking	No	3.5.1.1	
				Radiation damage	Loss of material (spalling, scaling)	No	No	3.5.1.1
					Cracking	No	No	3.5.1.9
				Reaction with aggregates	Loss of strength	No	No	3.5.1.9
			Cracking		Reinforced Concrete Structures AMP	Reinforced Concrete Structures AMP	3.5.1.3	
			Salt scaling	Loss of strength	Reinforced Concrete Structures AMP	Reinforced Concrete Structures AMP	3.5.1.3	
				Loss of material (spalling, scaling)	No	No	3.5.1.14	
			Shrinkage	Cracking	No	No	3.5.1.7	
				Aggressive chemical attack	Reinforced Concrete Structures AMP	Reinforced Concrete Structures AMP	3.5.1.5	
			Groundwater/soil	Aggressive chemical attack	Groundwater/soil	Aggressive chemical attack	Cracking	Reinforced Concrete Structures AMP
Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	Reinforced Concrete Structures AMP					3.5.1.5	
Groundwater/soil	Aggressive chemical attack	Groundwater/soil	Aggressive chemical attack	Loss of strength	Reinforced Concrete Structures AMP	Reinforced Concrete Structures AMP	3.5.1.5	
				Reduction of concrete pH (reducing corrosion resistance of steel embedments)	Reinforced Concrete Structures AMP	Reinforced Concrete Structures AMP	3.5.1.5	

Table 4-4 Standardized NUHOMS horizontal storage module

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Reinforced concrete: base walls, floor slab, roof; basemat; end and rear shield walls, corner shield wall; shielded ventilation air inlet plenum; inlet/outlet vents	SH, SR, TH	Concrete	Groundwater/soil	Creep	Cracking	No	3.5.1.2
				Dehydration at high temperature	Cracking	No	3.5.1.11
				Delayed ettringite formation	Loss of strength	No	3.5.1.11
					Loss of material (spalling, scaling)	No	3.5.1.13
				Differential settlement	Loss of strength	No	3.5.1.13
					Cracking	No	3.5.1.13
				Fatigue	Cracking	Reinforced Concrete Structures AMP	3.5.1.4
				Freeze and thaw	Cracking	No	3.5.1.10
				Microbiological degradation	Cracking	Reinforced Concrete Structures AMP	3.5.1.1
					Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.1
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.12
				Radiation damage	Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.12
					Increase in porosity and permeability	Reinforced Concrete Structures AMP	3.5.1.12
Reduction of concrete pH (reducing corrosion resistance of steel embeddings)	Reinforced Concrete Structures AMP	3.5.1.12					
Cracking	No	3.5.1.9					
Loss of strength	No	3.5.1.9					

Table 4-4 Standardized NUHOMS horizontal storage module

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Reinforced concrete: base walls, floor slab, roof, basement; end and rear shield walls, corner shield wall; shielded ventilation air inlet plenum; inlet/outlet vents	SH, SR, TH	Concrete	Groundwater/soil	Reaction with aggregates	Cracking	Reinforced Concrete Structures AMP	3.5.1.3
				Salt scaling	Loss of strength	Reinforced Concrete Structures AMP	3.5.1.3
				Shrinkage	Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.14
				Leaching of calcium hydroxide	Cracking	No	3.5.1.7
				Leaching of calcium hydroxide	Loss of strength	Reinforced Concrete Structures AMP	3.5.1.8
				Leaching of calcium hydroxide	Increase in porosity and permeability	Reinforced Concrete Structures AMP	3.5.1.8
				Corrosion of reinforcing steel	Reduction of concrete pH (reducing corrosion resistance of steel embedments)	Reinforced Concrete Structures AMP	3.5.1.8
				Air—outdoor; groundwater	Loss of concrete/steel bond	Reinforced Concrete Structures AMP	3.5.1.6
				Reinforcing steel	Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.6
				Cracking	Cracking	Reinforced Concrete Structures AMP	3.5.1.6
DSC support structure assembly hardware, base unit assembly hardware, module accessories	SR	Steel	Sheltered	Stress corrosion cracking	Loss of strength	Reinforced Concrete Structures AMP	3.5.1.6
				General corrosion	Cracking	No	3.2.1.5
					Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1

Table 4-4 Standardized NUHOMS horizontal storage module

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
DSC support structure assembly hardware, base unit assembly hardware, module accessories	SR	Steel	Sheltered	Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
DSC support structure assembly: support rail, rail extension plate and rail baseplate, plates, crossbeam, DSC stop plate extension	SR, TH	Stainless steel (welded)	Sheltered	Stress relaxation	Loss of preload	External Surfaces Monitoring of Metallic Components AMP	3.2.1.10
				Stress corrosion cracking	Cracking	External Surfaces Monitoring of Metallic Components AMP	3.2.2.5
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
DSC support structure assembly: support rail, rail extension plate and rail baseplate, plates, crossbeam, DSC stop plate extension	SR, TH	Stainless steel	Sheltered	Pitting and crevice corrosion	Loss of material (precursor to stress corrosion cracking)	External Surfaces Monitoring of Metallic Components AMP	3.2.2.2
				Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-4 Standardized NUHOMS horizontal storage module

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
DSC support structure assembly: support rail beams, support structure miscellaneous steel, plates, attachment/ installation hardware, DSC stop plate assembly, rail extension embedment, tube steel leg column	SR, TH	Steel	Sheltered	Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
DSC support structure assembly: support rail plate	SR, TH	Stainless steel	Sheltered	Radiation embrittlement	Cracking	No	3.2.1.9
				Stress corrosion cracking	Cracking	External Surfaces Monitoring of Metallic Components AMP	3.2.2.5
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Pitting and crevice corrosion	Loss of material (precursor to stress corrosion cracking)	External Surfaces Monitoring of Metallic Components AMP	3.2.2.2
				Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-4 Standardized NUHOMS horizontal storage module							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
DSC axial retainer assembly: axial retainer, plate	SR	Steel	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Microbiologically Influenced corrosion	Loss of material	No	3.2.2.4
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress corrosion cracking	Cracking	No	3.2.1.5
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
Cask restrain assembly: embedment assembly (rods, hex nuts, sleeve nuts), cask restraint embedment	SR	Steel	Sheltered	Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4

Table 4-4 Standardized NUHOMS horizontal storage module							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Cask restraint assembly: embedment assembly (rods, hex nuts, sleeve nuts), cask restraint embedment	SR	Steel	Sheltered	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
Heat shield assemblies: attachment hardware	SR	Steel	Sheltered	Stress relaxation	Loss of preload	External Surfaces Monitoring of Metallic Components AMP	3.2.1.10
				Stress corrosion cracking	Cracking	No	3.2.1.5
				Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				Stress relaxation	Loss of preload	External Surfaces Monitoring of Metallic Components AMP	3.2.1.10

Table 4-4 Standardized NUHOMS horizontal storage module							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Heat shield assemblies: support structure, Z bracket, screw	SR	Stainless Steel	Sheltered	Stress corrosion cracking	Cracking	No	3.2.2.5
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10
Heat shield assemblies: roof and side wall mounted heat shields, Z bracket	TH	Steel (galvanized)	Sheltered	Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9

Table 4-4 Standardized NUHOMS horizontal storage module

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Heat shield assemblies: roof and side wall mounted heat shields/Z bracket, side heat shield fins, backing sheet, top louvered heat shield	TH	Aluminum	Sheltered	Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.3.3
				Microbiologically influenced corrosion	Loss of material	No	3.2.3.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.3.2
				Radiation embrittlement	Cracking	No	3.2.3.8
Heat shield assemblies: side heat shield, top heat shield	TH	Stainless steel	Sheltered	Stress corrosion cracking	Cracking	No	3.2.2.5
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Radiation embrittlement	Cracking	No	3.2.2.9
Shielded door assembly: door attachment hardware	SR	Steel	Air—outdoor	Stress corrosion cracking	Cracking	No	3.2.1.5
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4

Table 4-4 Standardized NUHOMS horizontal storage module

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Shielded door assembly: door attachment hardware	SR	Steel	Air—outdoor	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				Stress relaxation	Loss of preload	No	3.2.1.10
Shielded door assembly: steel plates	SH, SR	Steel	Air—outdoor	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
Shielded door assembly: steel plates	SH, SR	Steel	Sheltered	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	No	3.2.1.4

Table 4-4 Standardized NUHOMS horizontal storage module

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)				
Shielded door assembly: concrete core	SH, SR	Reinforced concrete, nonshrink grout or pea gravel or mortar mix	Fully encased (steel)	Delayed ettringite formation	Loss of material (spalling, scaling)	No	3.5.1.13				
					Cracking	No	3.5.1.13				
					Loss of strength	No	3.5.1.13				
				Radiation damage	Cracking	TLAA/AMP or a supporting analysis is required	3.5.1.9				
					Loss of strength	TLAA/AMP or a supporting analysis is required	3.5.1.9				
				Reaction with aggregates	Cracking	Reinforced Concrete Structures AMP	3.5.1.3				
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.3				
				Inlet/outlet vents: outlet vent attachments	SR	Steel	Air—outdoor	Stress corrosion cracking	Cracking	No	3.2.1.5
								General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
								Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2								
Radiation embrittlement	Cracking	No	3.2.1.9								
Stress relaxation	Loss of preload	No	3.2.1.10								

Table 4-4 Standardized NUHOMS horizontal storage module

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Inlet/outlet vents: liner plates	SH, TH	Steel	Air—outdoor	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
Inlet/outlet vents: liner plates	SH, TH	Steel	Air—outdoor	Microbiologically influenced corrosion Pitting and crevice corrosion	Loss of material	No	3.2.1.4
Shielded ventilation air inlet plenum	TH	Stainless steel (welded)	Air—outdoor	Stress corrosion cracking	Cracking	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
		Stainless steel	Air—outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
		Stainless steel	Air—outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
Ventilation air outlet shielding blocks	TH	Stainless steel (welded)	Air—outdoor	Pitting and crevice corrosion	Loss of material (precursor to stress corrosion cracking)	External Surfaces Monitoring of Metallic Components AMP	3.2.2.2
		Stainless Steel	Air—outdoor	Radiation embrittlement	Cracking	No	3.2.2.9
		Stainless Steel	Air—outdoor	Stress corrosion cracking	Cracking	External Surfaces Monitoring of Metallic Components AMP	No
		Stainless Steel	Air—outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4

Table 4-4 Standardized NUHOMS horizontal storage module

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Ventilation air outlet shielding blocks	TH	Stainless Steel	Air—outdoor	Pitting and crevice corrosion	Loss of material (precursor to stress corrosion cracking)	External Surfaces Monitoring of Metallic Components AMP	3.2.2.2
				Radiation embrittlement	Cracking	No	3.2.2.9
Roof attachment assembly: angles, plates, dowel bar splicer	SR	Steel	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
Roof attachment assembly: roof attachment hardware	SR	Steel	Sheltered	Stress corrosion cracking	Cracking	No	3.2.1.5
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9

Table 4-4 Standardized NUHOMS horizontal storage module

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Roof attachment assembly: roof attachment hardware	SR	Steel	Sheltered	Stress relaxation	Loss of preload	External Surfaces Monitoring of Metallic Components AMP No	3.2.1.10 3.2.1.5
End and rear shield walls attachment hardware	SR	Steel	Sheltered	Stress corrosion cracking General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP No	3.2.1.1 3.2.1.4
HSM-to-HSM spacer channels	SR	Steel	Air—outdoor	Microbiologically influenced corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP No	3.2.1.2
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP No	3.2.1.9
				Radiation embrittlement	Cracking	External Surfaces Monitoring of Metallic Components AMP No	3.2.1.10
HSM-to-HSM spacer channels	SR	Steel	Air—outdoor	Stress relaxation	Loss of preload	External Surfaces Monitoring of Metallic Components AMP No	3.2.1.10
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP No	3.2.1.1 3.2.1.4
HSM-to-HSM spacer channels	SR	Steel	Air—outdoor	Microbiologically influenced corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP No	3.2.1.2
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP No	3.2.1.2

Table 4-4 Standardized NUHOMS horizontal storage module							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
HSM-to-HSM spacer channels	SR	Steel	Air—outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
Dose reduction hardware: dose reduction assembly	SH	Steel	Air—outdoor	Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
Module-to-module connections	SR	Stainless steel	Air—outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Radiation embrittlement	Cracking	No	3.2.2.9
		Steel	Air—outdoor	Stress corrosion cracking	Cracking	No	3.2.1.5
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1

Table 4-4 Standardized NUHOMS horizontal storage module

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Module-to-module connections	SR	Steel	Air—outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
Lightning protection system	SR	Copper	Air—outdoor	Stress relaxation	Loss of preload	No	3.2.1.10
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.5.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.5.3
Threaded fasteners and expansion anchors	SH, TH	Stainless Steel	Air—outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.5.2
				Radiation embrittlement	Cracking	No	3.2.5.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-4 Standardized NUHOMS horizontal storage module

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Threaded fasteners and expansion anchors	SH, TH	Stainless Steel	Air—outdoor	Stress relaxation	Loss of preload	No	3.2.2.10
Handrail	SR	Steel	Air—outdoor	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9

Table 4-5 Standardized Advanced NUHOMS horizontal storage module								
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Technical Basis (Section)		
Reinforced concrete: base unit walls, floor slab, roof, top shield block, basemat, end and rear shield walls, corner shield wall, inlet/outlet vents	SH, SR, TH*	Concrete	Air—outdoor	Aggressive chemical attack	Cracking	Reinforced Concrete Structures AMP	3.5.1.5	
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.5	
					Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.5	
				Creep	Reduction of concrete pH (reducing corrosion resistance of steel embedments)	Cracking	No	3.5.1.2
						Cracking	No	3.5.1.11
						Loss of strength	No	3.5.1.11
						Loss of material (spalling, scaling)	No	3.5.1.13
				Delayed ettringite formation	Loss of strength	Cracking	No	3.5.1.13
						Cracking	No	3.5.1.13
						Cracking	No	3.5.1.13
Differential settlement	Fatigue	Cracking	Reinforced Concrete Structures AMP	3.5.1.4				
		Cracking	No	3.5.1.10				
Freeze and thaw	Freeze and thaw	Cracking	Reinforced Concrete Structures AMP	3.5.1.1				
		Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.1				

*Safety Functions: Confinement (CO), Subcriticality (CR), Retrievability (RE), Radiation Shielding (SH), Structural Integrity (SR), Thermal/Heat Removal (TH)

Table 4-5 Standardized Advanced NUHOMS horizontal storage module							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Reinforced concrete: base unit walls, floor slab, roof, top shield block, basemat, end and rear shield walls, corner shield wall, inlet/outlet vents	SH, SR, TH	Concrete	Air—outdoor	Radiation damage	Cracking	No	3.5.1.9
					Loss of strength	No	3.5.1.9
			Reaction with aggregates	Cracking	Reinforced Concrete Structures AMP	3.5.1.3	
				Loss of strength	Reinforced Concrete Structures AMP	3.5.1.3	
			Salt scaling	Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.14	
				Shrinkage	No	3.5.1.7	
			Leaching of calcium hydroxide	Loss of strength	Reinforced Concrete Structures AMP	3.5.1.8	
				Increase in porosity and permeability	Reinforced Concrete Structures AMP	3.5.1.8	
			Aggressive chemical attack	Reduction of concrete pH (reducing corrosion resistance of steel embedments)	Reinforced Concrete Structures AMP	3.5.1.8	
				Loss of strength	No	3.5.1.5	
			Creep	Cracking	No	3.5.1.5	
				Loss of material (spalling, scaling)	No	3.5.1.5	
Cracking	No	3.5.1.2					

Table 4-5 Standardized Advanced NUHOMS horizontal storage module							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Technical Basis (Section)	
Reinforced concrete: base unit walls, floor slab, roof, top shield block, basemat, end and rear shield walls, corner shield wall, inlet/outlet vents	SH, SR, TH	Concrete	Sheltered	Dehydration at high temperatures	Cracking	No	3.5.1.11
					Loss of strength	No	3.5.1.11
				Delayed ettringite formation	Loss of material (spalling, scaling)	No	3.5.1.13
					Loss of strength	No	3.5.1.13
					Cracking	No	3.5.1.13
				Differential settlement	Cracking	Reinforced Concrete Structures AMP	3.5.1.4
				Fatigue	Cracking	No	3.5.1.10
				Freeze and thaw	Cracking	No	3.5.1.1
					Loss of material (spalling, scaling)	No	3.5.1.1
				Radiation damage	Cracking	No	3.5.1.9
					Loss of strength	No	3.5.1.9
				Reaction with aggregates	Cracking	Reinforced Concrete Structures AMP	3.5.1.3
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.3
				Salt scaling	Loss of material (spalling, scaling)	No	3.5.1.14
Shrinkage	Cracking	No	3.5.1.7				

Table 4-5 Standardized Advanced NUHOMS horizontal storage module									
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)		
Reinforced concrete: base unit walls, floor slab, roof, top shield block, basemat, end and rear shield walls, corner shield wall, inlet/outlet vents	SH, SR, TH	Concrete	Groundwater/soil	Aggressive chemical attack	Cracking	Reinforced Concrete Structures AMP	3.5.1.5		
					Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.5		
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.5		
				Creep	Dehydration at high temperatures	Delayed ettringite formation	Reduction of concrete pH (reducing corrosion resistance of steel embedments)	Reinforced Concrete Structures AMP	3.5.1.5
							Cracking	No	3.5.1.2
							Cracking	No	3.5.1.11
				Fatigue	Freeze and thaw		Loss of strength	No	3.5.1.11
							Loss of material (spalling, scaling)	No	3.5.1.13
							Loss of strength	No	3.5.1.13
				Differential settlement			Cracking	No	3.5.1.13
							Cracking	No	3.5.1.13
							Cracking	No	3.5.1.13
				Fatigue			Cracking	Reinforced Concrete Structures AMP	3.5.1.4
Cracking	No	3.5.1.10							
Cracking	Reinforced Concrete Structures AMP	3.5.1.1							
Freeze and thaw			Cracking	Reinforced Concrete Structures AMP	3.5.1.1				
			Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.1				

Table 4-5 Standardized Advanced NUHOMS horizontal storage module							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Reinforced concrete: base unit walls, floor slab, roof, top shield block, basemat, end and rear shield walls, corner shield wall, inlet/outlet vents	SH, SR, TH	Concrete	Groundwater/ soil	Microbiological degradation	Loss of strength	Reinforced Concrete Structures AMP	3.5.1.12
					Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.12
				Increase in porosity and permeability	Reinforced Concrete Structures AMP	3.5.1.12	
					Reduction of concrete pH (reducing corrosion resistance of steel embeddings)	Reinforced Concrete Structures AMP	3.5.1.12
				Cracking	No	3.5.1.9	
				Loss of strength	No	3.5.1.9	
				Reaction with aggregates	Reinforced Concrete Structures AMP	3.5.1.3	
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.3
				Salt scaling	Reinforced Concrete Structures AMP	3.5.1.14	
					Cracking	No	3.5.1.7
				Leaching of calcium hydroxide	Reinforced Concrete Structures AMP	3.5.1.8	
					Increase in porosity and permeability	Reinforced Concrete Structures AMP	3.5.1.8

Table 4-5 Standardized Advanced NUHOMS horizontal storage module							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Reinforced concrete: base unit walls, floor slab, roof, top shield block, basemat, end and rear shield walls, corner shield wall, inlet/outlet vents	SH, SR, TH	Reinforcing steel	Air—outdoor, groundwater	Leaching of calcium hydroxide	Reduction of concrete pH (reducing corrosion resistance of steel embedments)	Reinforced Concrete Structures AMP	3.5.1.8
					Loss of concrete/steel bond	Reinforced Concrete Structures AMP	3.5.1.6
					Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.6
					Cracking	Reinforced Concrete Structures AMP	3.5.1.6
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.6
					Cracking	No	3.2.2.5
					Loss of material	No	3.2.2.4
					Loss of material	No	3.2.2.2
					Cracking	No	3.2.2.9
					Loss of preload	No	3.2.2.10
DSC support structure assembly hardware	SR	Stainless steel	Sheltered	Stress corrosion cracking	Cracking	No	3.2.2.5
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10

Table 4-5 Standardized Advanced NUHOMS horizontal storage module								
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)	
DSC support structure assembly: support rail, rail extension plate, rail baseplate, stiffener plate, gusset plate, crossbeam, DSC stop plate extension	SR, TH	Stainless steel (welded)	Sheltered	Stress corrosion cracking	Cracking	External Surfaces Monitoring of Metallic Components AMP	3.2.2.5	
		Stainless steel	Sheltered	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4	
				Pitting and crevice corrosion	Loss of material (precursor to stress corrosion cracking)	External Surfaces Monitoring of Metallic Components AMP	No	3.2.2.2
				Radiation embrittlement	Cracking	No	No	3.2.2.9
DSC axial retainer assembly: axial retainer	SR	Stainless steel	Sheltered	Stress corrosion cracking	Cracking	No	3.2.2.5	
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4	
				Pitting and crevice corrosion	Loss of material	No	No	3.2.2.2
				Radiation embrittlement	Cracking	No	No	3.2.2.9
DSC axial retainer assembly: axial retainer, plates	SR	Steel	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1	
				Microbiologically influenced corrosion	Loss of material	No	No	3.2.1.4

Table 4-5 Standardized Advanced NUHOMS horizontal storage module							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
DSC axial retainer assembly: axial retainer, plates	SR	Steel	Sheltered	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
Heat shield assemblies: attachment hardware	SR	Stainless steel	Sheltered	Stress corrosion cracking	Cracking	No	3.2.2.5
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10
				Stress corrosion cracking	Cracking	No	3.2.1.5
		Steel	Sheltered	Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4

Table 4-5 Standardized Advanced NUHOMS horizontal storage module							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Heat shield assemblies: attachment hardware	SR	Steel	Sheltered	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				Stress relaxation	Loss of preload	External Surfaces Monitoring of Metallic Components AMP	3.2.1.10
Heat shield assemblies: side heat shield, top heat shield	TH	Stainless steel	Sheltered	Stress corrosion cracking	Cracking	No	3.2.2.5
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
Shielded door assembly: door attachment hardware	SR	Stainless steel	Air—outdoor	Radiation embrittlement	Cracking	No	3.2.2.9
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10

Table 4-5 Standardized Advanced NUHOMS horizontal storage module							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Shielded door assembly: door attachment hardware	SR	Steel	Air—outdoor	Stress corrosion cracking	Cracking	No	3.2.1.5
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
Shielded door assembly: backing plates	SR	Stainless steel	Air—outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
				Stress relaxation	Loss of preload	No	3.2.1.10
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
			Sheltered	Radiation embrittlement	Cracking	No	3.2.2.9
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4

Table 4-5 Standardized Advanced NUHOMS horizontal storage module							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Shielded door assembly: backing plates	SR	Stainless steel	Sheltered	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Radiation embrittlement	Cracking	No	3.2.2.9
Shielded door assembly: plates	SH, SR	Steel	Air—outdoor	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
			Sheltered	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1

Table 4-5 Standardized Advanced NUHOMS horizontal storage module							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Shielded door assembly: concrete core	SH, SR	Reinforced concrete, non-shrink grout or pea gravel or mortar mix	Embedded (steel or stainless steel)	Delayed ettringite formation	Loss of material (spalling, scaling)	No	3.5.1.13
					Cracking	No	3.5.1.13
					Loss of strength	No	3.5.1.13
Inlet/outlet vents: outlet vent attachment hardware	SR	Stainless steel	Air—outdoor	Radiation damage	Cracking	TLAA/AMP or a supporting analysis is required	3.5.1.9
					Loss of strength	TLAA/AMP or a supporting analysis is required	3.5.1.9
				Reaction with aggregates	Cracking	Reinforced Concrete Structures AMP	3.5.1.3
				Stress corrosion cracking	Loss of strength	Reinforced Concrete Structures AMP	3.5.1.3
					Cracking	No	3.2.2.5
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
					Pitting and crevice corrosion	Loss of material	No
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10

Table 4-5 Standardized Advanced NUHOMS horizontal storage module							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Inlet/outlet vents: outlet vent attachment hardware	SR	Steel	Air—outdoor	Stress corrosion cracking	Cracking	No	3.2.1.5
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
Inlet/outlet vents: liner plates	SH, TH	Stainless steel	Air—outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
				Stress relaxation	Loss of preload	No	3.2.1.10
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Pitting and crevice corrosion	Loss of material (precursor to stress corrosion cracking)	No	3.2.2.2
Roof attachment assembly: angles, plates, dowel bar splicer	SR	Stainless steel	Sheltered	Radiation embrittlement	Cracking	No	3.2.2.9
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4

Table 4-5 Standardized Advanced NUHOMS horizontal storage module									
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)		
Roof attachment assembly: angles, plates, dowel bar splicer	SR	Stainless steel	Sheltered	Pitting and crevice corrosion	Loss of material (precursor to stress corrosion cracking)	No	3.2.2.2		
				Radiation embrittlement	Cracking	No	3.2.2.9		
		Steel	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1		
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4		
		Roof attachment assembly: roof attachment hardware	SR	Stainless steel	Sheltered	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
						Radiation embrittlement	Cracking	No	3.2.1.9
						Stress corrosion cracking	Cracking	No	3.2.2.5
						Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
						Pitting and crevice corrosion	Loss of material	No	3.2.2.2
						Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-5 Standardized Advanced NUHOMS horizontal storage module								
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)	
Roof attachment assembly: roof attachment hardware	SR	Stainless steel	Sheltered	Stress relaxation	Loss of preload	No	3.2.2.10	
		Steel	Sheltered	Stress corrosion cracking	Cracking	No	3.2.1.5	
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	No	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	No	3.2.1.2
				Radiation embrittlement	Cracking	No	No	3.2.1.9
				Stress relaxation	Loss of preload	External Surfaces Monitoring of Metallic Components AMP	No	3.2.1.10
				Stress corrosion cracking	Cracking	No	No	3.2.2.5
				Microbiologically influenced corrosion	Loss of material	No	No	3.2.2.4
				Pitting and crevice corrosion	Loss of material	No	No	3.2.2.2
End and rear shield walls attachment hardware	SR	Stainless steel	Sheltered					

Table 4-5 Standardized Advanced NUHOMS horizontal storage module									
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)		
End and rear shield walls attachment hardware	SR	Stainless steel	Sheltered	Radiation embrittlement	Cracking	No	3.2.2.9		
				Stress relaxation	Loss of preload	No	3.2.2.10		
		Steel	Sheltered	Stress corrosion cracking	Cracking	No	3.2.1.5		
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1		
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4		
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2		
					Radiation embrittlement	Cracking	No	3.2.1.9	
					Stress relaxation	Loss of preload	External Surfaces Monitoring of Metallic Components AMP	3.2.1.10	
		Module-to-module connection hardware	SR	Stainless steel	Air—outdoor	Stress corrosion cracking	Cracking	No	3.2.2.5
						Microbiologically influenced corrosion	Loss of material	No	3.2.2.4

Table 4-5 Standardized Advanced NUHOMS horizontal storage module							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Module-to-module connection hardware	SR	Stainless steel	Air—outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10

Table 4-6 NUHOMS transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Structural shell (Cask body)	SH, SR, TH*	Steel	Embedded (neutron shielding)	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Radiation embrittlement	Cracking	No	3.2.1.9
		Stainless steel (welded)	Demineralized water	Stress corrosion cracking	Cracking	No	3.2.2.5
				Stress corrosion cracking	Cracking	No	3.2.2.5
		Stainless steel	Embedded (neutron shielding)	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
		Demineralized water		Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
Air—indoor/outdoor		Pitting and crevice corrosion	Loss of material	No	3.2.2.2		

*Safety Functions: Confinement (CO), Subcriticality (CR), Retrievability (RE), Radiation Shielding (SH), Structural Integrity (SR), Thermal/Heat Removal (TH)

Table 4-6 NUHOMS transfer cask								
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)	
Structural shell (Cask body)	SH, SR, TH	Stainless steel	Air— indoor/outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4	
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7	
				Radiation embrittlement	Cracking	No	3.2.2.9	
Inner shell (Cask body)	SH, SR, TH	Stainless steel (welded) Stainless steel	Air— indoor/outdoor	Stress corrosion cracking	Cracking	No	3.2.2.5	
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2	
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4	
Top flange (Cask body)	SH, SR	Stainless steel	Air— indoor/outdoor	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7	
				Radiation embrittlement	Cracking	No	3.2.2.9	
				Embedded (Lead)	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7	
Top flange (Cask body)	SH, SR	Stainless steel	Air— indoor/outdoor	Radiation embrittlement	Cracking	No	3.2.2.9	
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2	
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4	

Table 4-6 NUHOMS transfer cask								
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)	
Top flange (Cask body)	SH, SR	Stainless steel	Air— indoor/outdoor	Stress corrosion cracking	Cracking	No	3.2.2.5	
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7	
				Radiation embrittlement	Cracking	No	3.2.2.9	
			Embedded (neutron shielding)		Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
					Radiation embrittlement	Cracking	No	3.2.2.9
			Demineralized water		Pitting and crevice corrosion	Loss of material	No	3.2.2.2
					Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
					Stress corrosion cracking	Cracking	No	3.2.2.5
					Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
					Radiation embrittlement	Cracking	No	3.2.2.9
Bottom support ring and bottom end forging (Cask body)	SH, SR	Stainless steel	Air— indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2	

Table 4-6 NUHOMS transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Bottom support ring and bottom end forging (Cask body)	SH, SR	Stainless steel	Air— indoor/outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
Bottom end plate (Cask body)	SH, SR	Stainless steel (welded) Stainless steel	Air— indoor/outdoor Air— indoor/outdoor	Radiation embrittlement	Cracking	No	3.2.2.9
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-6 NUHOMS transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Bottom end plate (Cask body)	SH, SR	Stainless steel	Embedded (stainless steel)	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
Lead gamma shielding (Cask body)	SH, TH	Lead	Embedded (steel, stainless steel)	None identified	None identified	No	3.2.6
Rails (Cask attachments)	SR	Stainless steel	Air— indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
Screw thread insert (Cask attachments)	SH, SR	Stainless steel	Embedded (stainless steel)	Wear	Loss of material	Transfer Casks AMP	3.2.2.11
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-6 NUHOMS transfer cask								
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)	
Upper trunnions (Cask attachments)	SH, SR	Stainless steel (welded)	Air— indoor/outdoor	Stress corrosion cracking	Cracking	No	3.2.2.5	
			Demineralized water	Stress corrosion cracking	Cracking	No	3.2.2.5	
		Stainless steel	Air— indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2	
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4	
				Radiation embrittlement	Cracking	No	3.2.2.9	
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7	
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2	
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4	
		Steel	SH, SR	Air— indoor/outdoor	Radiation embrittlement	Cracking	No	3.2.2.9
					Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
Upper trunnion sleeves (Cask attachments)	SH, SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1	

Table 4-6 NUHOMS transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Upper trunnion sleeves (Cask attachments)	SH, SR	Steel	Air— indoor/outdoor	Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Wear	Loss of material	Transfer Casks AMP	3.2.1.11
			DeminerIALIZED water	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Wear	Loss of material	Transfer Casks AMP	3.2.1.11

Table 4-6 NUHOMS transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Upper trunnion sleeves (Cask attachments)	SH, SR	Stainless steel	Air— indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Wear	Loss of material	Transfer Casks AMP	3.2.2.11
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
Wear	Loss of material	Transfer Casks AMP	3.2.2.11				

Table 4-6 NUHOMS transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Upper trunnion cover plate and pad (Cask attachments)	SH, SR	Stainless steel	Air— indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Pitting and crevice corrosion	Loss of material	No	3.2.4.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.4.3
				Stress corrosion cracking	Cracking	No	3.2.4.4
				Radiation embrittlement	Cracking	No	3.2.4.6
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.4.5
Upper and lower trunnion neutron shielding (Cask attachments)	SH, TH	Bisco NS-3	Embedded (steel, stainless steel)	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.3.1.2

Table 4-6 NUHOMS transfer cask								
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)	
Upper and lower trunnion neutron shielding (Cask attachments)	SH, TH	Bisco NS-3	Embedded (steel, stainless steel)	Radiation embrittlement	Cracking	No	3.3.1.3	
				Boron depletion	Loss of shielding	TLAA/AMP or a supporting analysis is required	3.3.1.1	
Lower trunnions (Cask attachments)	SH, SR	Stainless steel (welded)	Air—indoor/outdoor	Stress corrosion cracking	Cracking	No	3.2.2.5	
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2	
		Stainless steel	Air—indoor/outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4	
				Radiation embrittlement	Cracking	No	3.2.2.9	
Lower trunnions sleeves (Cask attachments)	SH, SR	Steel	Air—indoor/outdoor	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7	
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1	
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2	
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4	
				Radiation embrittlement	Cracking	No	3.2.1.9	

Table 4-6 NUHOMS transfer cask								
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)	
Lower trunnions sleeves (Cask attachments)	SH, SR	Steel	Air— indoor/outdoor	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7	
				Wear	Loss of material	Transfer Casks AMP	3.2.1.11	
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1	
			Demineralized water	Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2	
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4	
				Radiation embrittlement	Cracking	No	3.2.1.9	
		Stainless steel	Air— indoor/outdoor	Fatigue	Cracking	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
					Wear	Loss of material	Transfer Casks AMP	3.2.1.11
					Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4	
					Stress corrosion cracking	Cracking	No	3.2.2.5
					Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-6 NUHOMS transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Lower trunnions sleeves (Cask attachments)	SH, SR	Stainless steel	Air— indoor/outdoor	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Wear	Loss of material	Transfer Casks AMP	3.2.2.11
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Wear	Loss of material	Transfer Casks AMP	3.2.2.11
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
Lower trunnion sleeve nickel alloy weld overlay (Cask attachments)	SR	Stainless steel	Air— indoor/outdoor	Stress corrosion cracking	Cracking	No	3.2.2.5
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-6 NUHOMS transfer cask										
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)			
Lower trunnion sleeve nickel alloy weld overlay (Cask attachments)	SR	Stainless steel	Air— indoor/outdoor	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7			
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2			
			Microbiologically influenced corrosion	Loss of material	No	3.2.2.4				
			Stress corrosion cracking	Cracking	No	3.2.2.5				
			Radiation embrittlement	Cracking	No	3.2.2.9				
			Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7				
			Internal sleeve components (Cask attachments)	SR	Aluminum	Embedded (stainless steel)	Radiation embrittlement	Cracking	No	3.2.3.8
							Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.3.6
							Radiation embrittlement	Cracking	No	3.2.1.9
			Bottom head cap screw for internal sleeve (Cask attachments)	SR	Steel	Embedded (stainless steel, aluminum)	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
Stress relaxation	Loss of preload	No					3.2.1.10			

Table 4-6 NUHOMS transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Washer for internal sleeve (Cask attachments)	SR	Stainless steel	Embedded (steel, stainless steel)	Radiation embrittlement	Cracking	No	3.2.2.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
Spacer assembly (Cask attachments)	SR	Stainless steel	Air— indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
Ram access penetration ring (Cask penetration)	SH, SR	Stainless steel	Air— indoor/outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7

Table 4-6 NUHOMS transfer cask								
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)	
Neutron shield panel support angles (Cask neutron shield)	SH, SR, TH	Stainless steel	Embedded (neutron shielding)	Radiation embrittlement	Cracking	No	3.2.2.9	
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7	
			Demineralized water	Pitting and crevice corrosion	Loss of material	No	3.2.2.2	
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4	
				Stress corrosion cracking	Cracking	No	3.2.2.5	
				Radiation embrittlement	Cracking	No	3.2.2.9	
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7	
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2	
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4	
				Stress corrosion cracking	Cracking	No	3.2.2.5	
Neutron shield panels and plates (Cask neutron shield)	SH, SR, TH	Stainless steel	Air— indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2	
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4	
				Stress corrosion cracking	Cracking	No	3.2.2.5	
				Radiation embrittlement	Cracking	No.	3.2.2.9	

Table 4-6 NUHOMS transfer cask										
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)			
Neutron shield panels and plates (Cask neutron shield)	SH, SR, TH	Stainless steel	Air— indoor/outdoor	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7			
				Radiation embrittlement	Cracking	No	3.2.2.9			
			Embedded (neutron shielding)	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7			
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2			
			Demineralized water	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4			
				Stress corrosion cracking	Cracking	No	3.2.2.5			
			Embedded (steel, stainless steel)	Radiation embrittlement	Cracking	No	3.2.2.9			
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7			
			Castable neutron shielding material (Cask neutron shield)	SH, TH	Bisco NS-3	Embedded (steel, stainless steel)	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.3.1.2
							Radiation embrittlement	Cracking	No	3.3.1.3

Table 4-6 NUHOMS transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Castable neutron shielding material (Cask neutron shield)	SH, TH	Bisco NS-3	Embedded (steel, stainless steel)	Boron depletion	Loss of shielding	TLAA/AMP or a supporting analysis is required	3.3.1.1
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
Inner, outer, and side top cover plates (Cask cover assembly)	SH, SR	Steel	Air—indoor/outdoor	Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
Radiation embrittlement	Cracking	No	3.2.2.9				
Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7				

Table 4-6 NUHOMS transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Bottom cover plate (Cask cover assembly)	SH, SR, TH	Stainless steel	Air— indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
Top and bottom cover neutron shielding (Cask cover assembly)	SH	Bisco NS-3	Embedded (stainless steel)	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.3.1.2
				Radiation embrittlement	Cracking	No	3.3.1.3
				Boron depletion	Loss of shielding	TLAA/AMP or a supporting analysis is required	3.3.1.1
Bolts, screws, and washers for top and bottom cover plates (Cask cover assembly)	SH, SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2

Table 4-6 NUHOMS transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Bolts, screws, and washers for top and bottom cover plates (Cask cover assembly)	SH, SR	Steel	Air— indoor/outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Stress relaxation	Loss of preload	No	3.2.1.10
Socket head cap screws for bottom cover plate (Cask cover assembly)	SH, SR	Stainless steel	Air— indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
Airflow wedge plates (Cask cover assembly)	SH, SR, TH	Stainless steel	Air— indoor/outdoor	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Stress relaxation	Loss of preload	No	3.2.2.10
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2

Table 4-6 NUHOMS transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Airflow wedge plates (Cask cover assembly)	SH, SR, TH	Stainless steel	Air— indoor/outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
Support skid supplemental shielding	SH, SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
Bolts and washers for support skid supplemental shielding	SR	Steel	Air— indoor/outdoor	Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4

Table 4-6 NUHOMS transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Bolts and washers for support skid supplemental shielding	SR	Steel	Air— indoor/outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Stress relaxation	Loss of preload	No	3.2.1.10
Upper and lower decon area cask shielding	SH, SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7

1 **4.3 HI-STORM 100 and HI-STAR 100 systems**

2 **4.3.1 System description**

3 Holtec International developed the HI-STORM (Holtec International–Storage and Transfer
4 Operation Reinforced Module) 100 system and the HI-STAR (Holtec International–Storage,
5 Transport, and Repository) 100 system. The HI-STORM 100 system consists of a metallic
6 multipurpose canister (MPC) that contains the SNF assemblies, a vertical concrete storage
7 overpack that contains the MPC during storage, and a HI-TRAC (Holtec International–Transfer
8 Cask) TC that contains the MPC during loading, unloading, and transfer operations. The
9 HI-STAR 100 system consists of an MPC and a vertical metal overpack, which is used to load,
10 unload, transfer, and store the SNF assemblies contained in the MPC. The HI-STORM 100
11 system is certified only for storage, while the HI-STAR 100 system (including its metal overpack)
12 is certified for both storage and transportation. Figure 4-9 presents schematics of the
13 HI-STORM 100 and HI-STAR 100 systems.

14 The HI-STORM design is presently licensed for use in the United States under NRC
15 Docket 72-1014, in combination with the MPC-24, MPC-32, and MPC-68 canisters, while the
16 HI-STAR design is licensed for use under NRC Docket 72-1008, with the MPC-24 and MPC-68
17 canisters. As in the case for the NUHOMS DSCs, the names of the Holtec MPCs reflect the
18 number of fuel assemblies each MPC can hold. In addition, a variant design of the HI-STAR
19 overpack, designated HI-STAR HB, is being used in conjunction with the MPC-HB canister
20 under a site-specific license at the Humboldt Bay ISFSI. The details of the components of the
21 two storage systems are provided below.

22 **4.3.2 Multipurpose canister**

23 The MPCs are welded cylindrical structures with an identical outer diameter, so that any MPC
24 will fit into either the HI-STORM or HI-STAR overpacks. However, only certain MPC and
25 overpack combinations are currently licensed for use. Each MPC is an assembly consisting of a
26 honeycombed fuel basket, baseplate, canister shell, lid, and closure ring. A cross sectional
27 elevation view of a fuel basket for the MPC-68 series is shown in Figure 4-10. The number of
28 spent fuel storage locations in each of the MPCs depends on the SNF assembly characteristics.

29 Ten MPC models, distinguished by the type and number of SNF assemblies authorized for
30 loading, are presently certified by the NRC for use in the United States. These are the MPC-24
31 series (including the MPC-24E and MPC-24EF), the MPC-32 series (including the MPC-32F),
32 and the MPC-68 series (including the MPC-68F, MPC-68FF, MPC-68M, and MPC-HB), shown
33 in cross sectional views in Figure 4-11. The numerical suffix for each canister series denotes
34 the maximum number of fuel elements that it can accommodate. Those canisters with “E” and
35 “F” designations are designed for the storage of damaged fuel rods and fuel debris. The MPC-
36 68M design contains a fuel basket constructed of Metamic-HT™, a neutron absorbing material
37 that also has a structural function.

38 The fuel storage cells in the MPC-24 series are physically separated from one another by a
39 water gap, also called a flux trap, for criticality control. Flux traps are not used in the MPC-32
40 and MPC-68 series. The MPC-32 design includes credit for soluble boron in the MPC water
41 during wet fuel loading and unloading operations for criticality control. The MPC fuel basket is
42 positioned and supported within the MPC shell by a set of basket supports welded to the inside
43 of the MPC shell. In the early-vintage MPCs that were loaded under the original HI-TORM 100
44 design, optional heat conduction elements (fabricated from thin aluminum Alloy 1100) may have

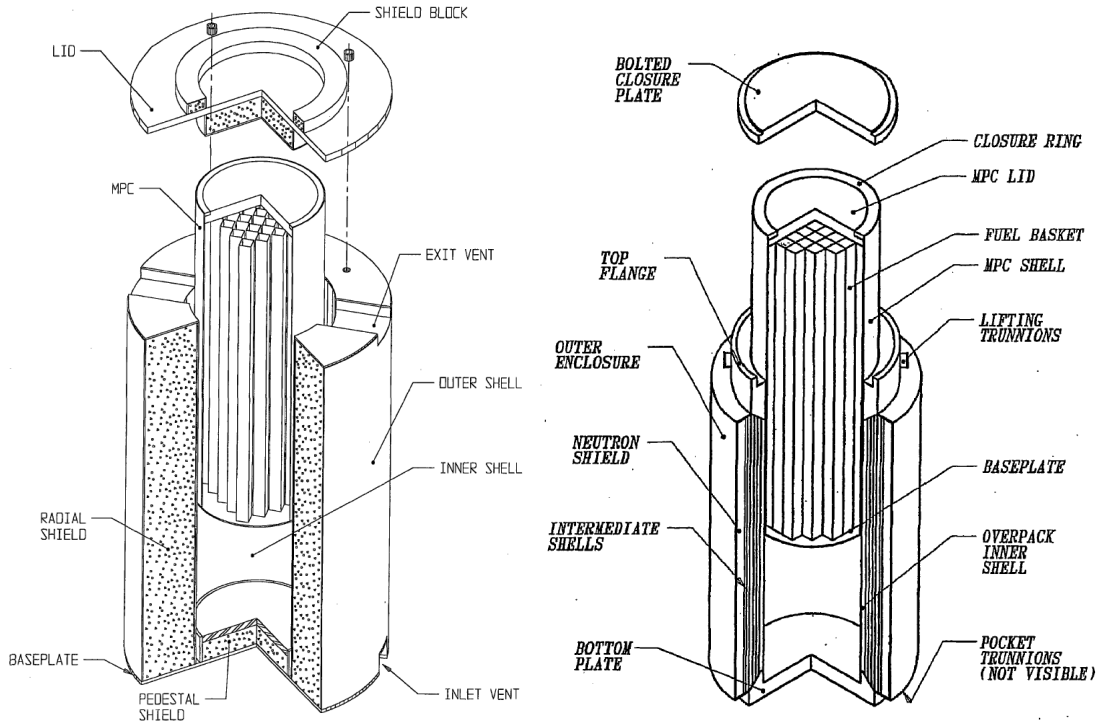


Figure 4-9 HI-STORM 100 (left) (Holtec International, 2013) and HI-STAR 100 (right) (Holtec International, 2001) systems

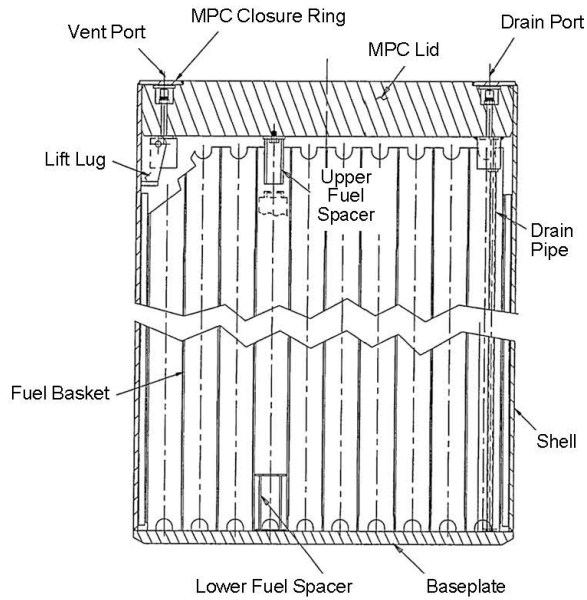


Figure 4-10 Cross section elevation view of MPC (Holtec International, 2013)

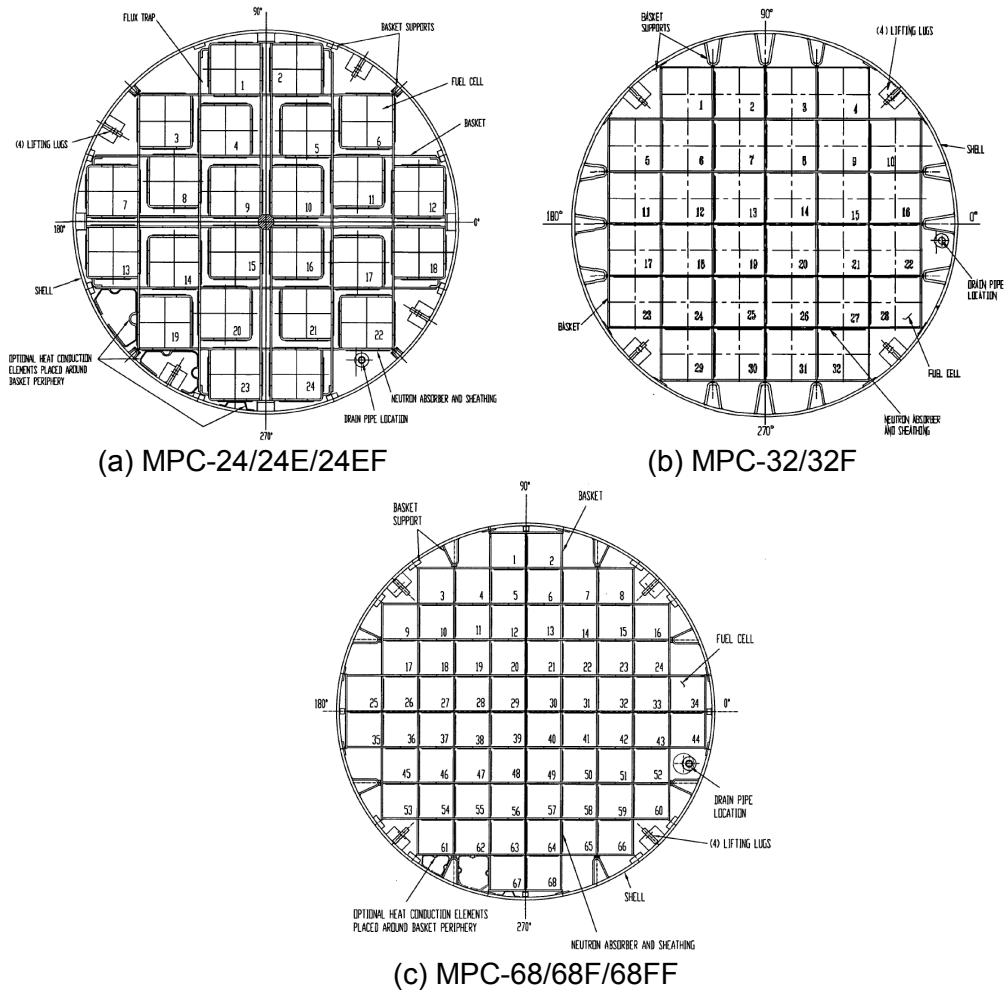


Figure 4-11 Cross sectional views of different MPC designs (Holtec International, 2013)

- 1 been installed between the periphery of the basket, the MPC shell, and the basket supports.
- 2 For shorter SNF assemblies, upper and lower fuel spacers, as appropriate, maintain the axial
- 3 position of the SNF assembly within the MPC basket.
- 4 All structural components in MPCs are made of a material designated by the manufacturer as
- 5 Alloy X. Candidate Alloy X materials include Types 304, 304LN, 316, and 316LN austenitic
- 6 stainless steels. All MPC components that are likely to come in contact with spent fuel pool
- 7 water or the ambient environment are constructed from stainless steel, with the exception of
- 8 neutron poison plates, aluminum seals on vent and drain port caps, and optional aluminum heat
- 9 conduction elements.
- 10 Lifting lugs attached to the inside surface of the MPC canister shell (shown in Figure 4-10)
- 11 permit placement of the empty MPC into the HI-TRAC transfer cask and also serve to axially
- 12 locate the MPC lid before welding. They are not used to handle a loaded MPC, because the
- 13 MPC lid is installed before any handling of a loaded canister.
- 14 The top end of the MPC incorporates a redundant closure system. The MPC lid is a circular
- 15 plate (fabricated from one piece or two pieces—split top and bottom) that is welded to the MPC

1 outer shell. In the two-piece lid design, only the top piece comprises a part of the enclosure
2 vessel's pressure boundary; the bottom piece is attached to the top piece with a nonstructural,
3 nonpressure-retaining weld and acts as a radiation shield. The lid is equipped with vent and
4 drain ports that are used to remove moisture and air from the MPC and backfill the MPC with
5 helium. The vent and drain ports are covered and seal-welded before the closure ring is
6 installed. The closure ring is a circular ring edge-welded to the MPC shell and lid. The MPC lid
7 provides sufficient rigidity to allow the entire MPC, loaded with spent nuclear fuel, to be lifted by
8 the threaded holes in the MPC lid.

9 Boral[®] and METAMIC[™] neutron poison panels are used in the basket and are enclosed in
10 Alloy X stainless steel sheathing that is stitch-welded to the MPC basket cell walls along their
11 entire periphery. Thus, the neutron poison material is contained in a tight, welded pocket
12 enclosure. The shear strength of the pocket weld joint, which is an order of magnitude greater
13 than the weight of a fuel assembly, ensures that the neutron poison and its enveloping
14 sheathing pocket will maintain their as-installed position under all loading, storage, and transport
15 conditions. In addition, the pocket joint detail ensures that fuel assembly insertion or withdrawal
16 into or out of the MPC basket will not lead to a disconnection of the sheathing from the cell wall.

17 The MPC does not require any valves, gaskets or mechanical seals for confinement.
18 Figure 4-12 shows the MPC confinement boundary. All components of the confinement
19 boundary are safety significant and are fabricated entirely of stainless steel. The primary
20 confinement boundary is defined by the outline formed by the sealed, cylindrical enclosure of
21 the MPC shell (including any associated axial or circumferential welds) welded to the baseplate
22 at the bottom, the MPC lid welded around the top circumference to the shell wall, and the port
23 cover plates welded to the lid. A shield lid is bolted to the top of the MPC lid and provides
24 radiation shielding.

25 The helium backfill gas plays an important role in the MPC thermal performance. It fills all the
26 spaces between solid components and provides an improved conduction medium relative to air
27 for dissipating decay heat in the MPC. Furthermore, the pressurized helium environment within
28 the MPC sustains a closed-loop thermo-siphon action, removing SNF decay heat by upward
29 flow of helium through the storage cells.

30 Table 4-7 provides a generic evaluation of potential aging mechanisms and effects requiring
31 management for specific components of the MPC. The AMPs that provide an acceptable
32 approach to managing the aging effects are also identified in the table.

33 **4.3.3 HI-STORM concrete overpack**

34 The HI-STORM overpacks are thick-walled concrete cylindrical vessels enclosed in a steel
35 shell. There are three base HI-STORM overpack designs: (i) HI-STORM 100,
36 (ii) HI-STORM 100S, and (iii) HI-STORM 100S Version B. *The significant differences among*
37 *the three are overpack height, MPC pedestal height, location of the air outlet ducts, and vertical*
38 *alignment of the inlet and outlet air ducts. The HI-STORM 100S Version B overpack design*
39 *does not include a concrete-filled pedestal to support the MPC. Instead, the MPC rests upon a*
40 *steel plate that maintains the MPC sufficiently above the inlet air ducts to prevent direct*
41 *radiation shine through the ducts. Figure 4-13 shows cross sectional views of the*
42 *HI-STORM 100 and 100S overpacks. The HI-STORM 100A and 100SA overpack designs are*
43 *the anchored variant of the HI-STORM 100 and 100S designs.*

44

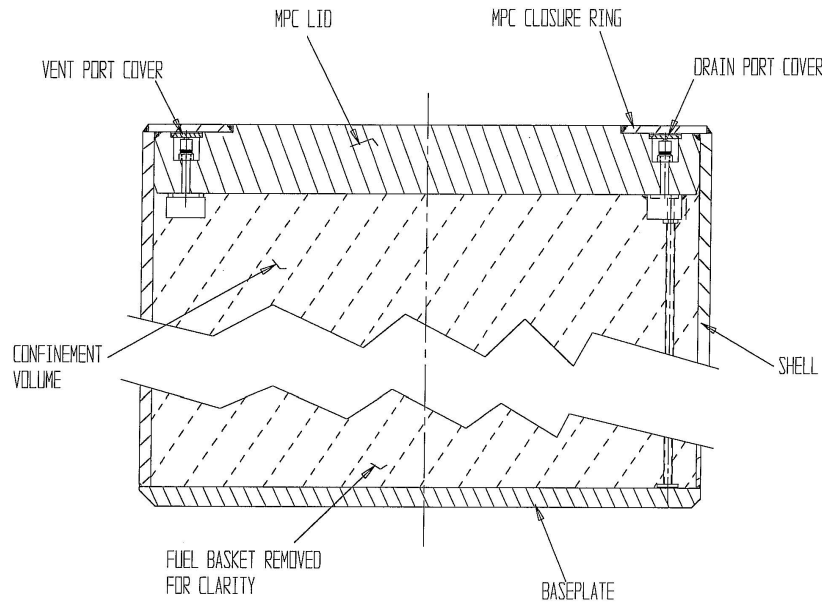


Figure 4-12 MPC confinement boundary (Holtec International, 2013)

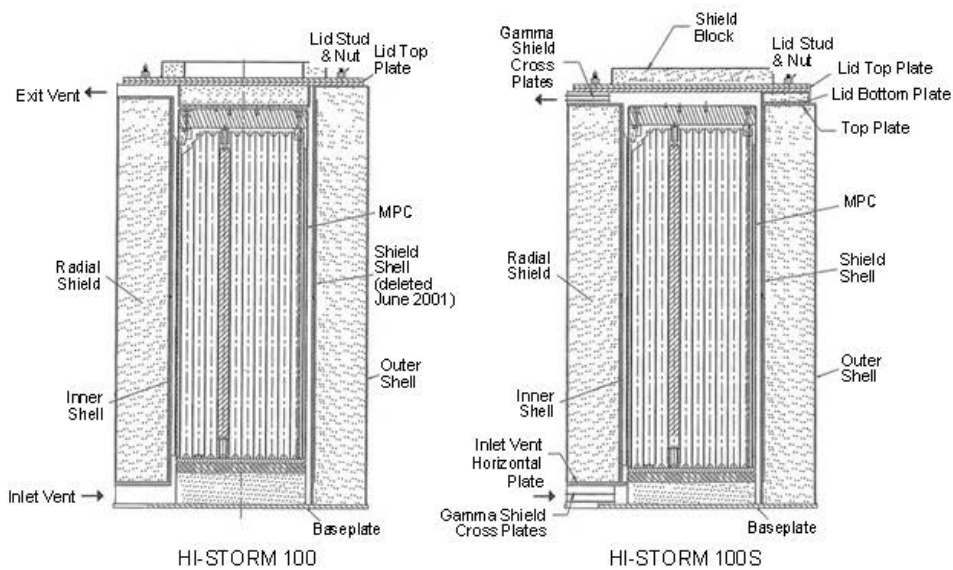


Figure 4-13 Cross sectional views of the HI-STORM 100 and 100S overpacks with an MPC inserted (Holtec International, 2013)

- 1 A base HI-STORM overpack design is capable of storing each type of MPC. The overpack
- 2 inner shell is provided with channels distributed around the inner cavity that provide guidance
- 3 for MPC insertion and removal, and a flexible medium to absorb some of the impact during a
- 4 tipover. They also allow the flow of cooling air through the overpack. The main structural
- 5 function of the HI-STORM overpack is provided by carbon steel, and the main shielding function
- 6 is provided by concrete. The concrete, enclosed by cylindrical inner and outer steel shells, a
- 7 thick baseplate, and a top plate, is specified to provide the necessary shielding properties and
- 8 compressive strength. The overpack lid has appropriate concrete shielding to provide neutron
- 9 and gamma attenuation in the vertical direction.

1 The HI-STORM overpack has air ducts to allow for passive natural convection cooling of the
2 contained MPC. A minimum of four air inlets and four air outlets are located at the lower and
3 upper extremities of the storage system, respectively. The vertical annulus between the MPC
4 and the inner shell of the overpack facilitates an upward flow of air by buoyancy forces, drawing
5 ambient air from the inlet vents and releasing it from the outlet vents at the top of the HI-STORM
6 storage system. The annulus ventilation flow cools the hot MPC surfaces and transfers decay
7 heat to the outside environment.

8 The principal function of the concrete is to provide shielding against gamma and neutron
9 radiation. However, it also imparts a large thermal inertia to the HI-STORM overpack, allowing
10 it to moderate the rise in temperature of the system under hypothetical conditions when all
11 ventilation passages are assumed to be blocked. The high thermal inertia characteristics of the
12 HI-STORM concrete also control the temperature of the MPC in the event of a postulated fire
13 accident at the ISFSI. Although the annular concrete mass in the overpack shell is not a
14 structural member, it does act as an elastic/plastic filler of the intershell space.

15 Four threaded anchor blocks, located at 90-degree intervals around the circumference of the top
16 of the overpack lid, are provided for lifting. The anchor blocks are integrally welded to the radial
17 plates, which in turn are full-length welded to the overpack inner shell, outer shell, and
18 baseplate (HI-STORM 100) or the inlet air duct horizontal plates (HI-STORM 100S).

19 The HI-STORM 100S Version B overpack design incorporates partial-length radial plates at the
20 top of the overpack to secure the anchor blocks and uses both gussets and partial-length radial
21 plates at the bottom of the overpack for structural stability. The overpack may also be lifted
22 from the bottom using specially designed lifting transport devices, including hydraulic jacks, air
23 pads, Hillman rollers, or other designs based on site-specific needs and capabilities.

24 For anchoring, the HI-STORM 100A overpack baseplate is extended to allow it to be attached to
25 the reinforced concrete structure of the ISFSI. Sector lugs are bolted to the ISFSI pad using
26 anchor studs. The lateral load-bearing capacity of the HI-STORM/pad interface is many times
27 greater than the horizontal sliding force exerted on the cask under the postulated design-basis
28 earthquake seismic event. Thus, the potential for lateral sliding of the HI-STORM 100A system
29 during a seismic event is precluded, as is the potential for any bending action on the
30 anchor studs.

31 The HI-STORM 100 system also includes a variant 100U underground module design. The
32 HI-STORM 100U design provides storage of an MPC inside a cylindrical cavity located entirely
33 below the top of the grade of the ISFSI. HI-STORM 100U comprises the cavity enclosure
34 container, consisting of the container shell welded to the bottom plate and the container flange,
35 and the closure lid, divider shell, insulation, and bearing pads, as well as the interfacing and
36 proximate structures, such as interface pad, support foundation pad, subgrade surrounding the
37 module, top surface pad, and retaining wall.

38 Table 4-8 provides a generic evaluation of potential aging mechanisms and effects requiring
39 management for specific components of the HI-STORM overpack, respectively. The AMPs that
40 provide an acceptable approach to managing the aging effects are also identified in the tables.

41 **4.3.4 HI-STAR metal overpack**

42 The HI-STAR 100 overpack is a sealed, thick-walled carbon and low-alloy steel cylindrical
43 vessel. The overpack containment boundary is formed by an inner shell welded at the bottom to

1 a cylindrical main flange and bolted to a top closure plate. The HI-STAR 100 overpack with the
2 MPC partially inserted is shown in Figure 4-9. The overpack consists of one inner shell, five
3 intermediate shells, and one enclosure shell, which form the body of the overpack. Figure 4-14
4 and Figure 4-15 provide an elevation and cross section view, respectively, of the overpack.

5 Two concentric grooves are machined into the closure plate to accept the metallic seals. The
6 bolted closure plate is recessed into the top flange, and the bolted joint is configured to provide
7 maximum protection to the closure bolts and seals in the event of a drop accident. The closure
8 plate has test and vent ports, which are sealed by a threaded port plug with a metallic seal. The
9 bottom plate has a drain port that is sealed by a threaded port plug with a metallic seal. The
10 inner surfaces of the HI-STAR overpack form an internal cylindrical cavity for housing the MPC.

11 The outer surface of the overpack inner shell is buttressed with the five layers of intermediate
12 shells of gamma shielding in the form of layers of carbon steel plate installed so as to ensure a
13 permanent state of contact between adjacent layers. Besides serving as an effective gamma
14 shield, these intermediate layers provide additional strength to the overpack to resist potential
15 punctures or penetrations from external missiles. Radial channels are vertically welded to the
16 outside surface of the outermost intermediate shell at equal intervals around the circumference
17 (see Figure 4-15). The radial channels act as fins for improved heat conduction to the overpack
18 outer enclosure shell surface and as cavities for retaining and protecting the Holtite-A™ neutron
19 shield described below.

20 The outer enclosure shell is formed by welding enclosure shell panels between each pair of
21 radial channels to form the neutron shielding cavities, as shown in Figure 4-15. At the top of the
22 outer enclosure shell, rupture disks are positioned in a recessed area. These rupture disks
23 relieve internal pressure that may develop as a result of a fire accident and subsequent off
24 gassing of the neutron shield material. Within each radial channel, a layer of silicone sponge is
25 positioned to act as a thermal expansion foam to compress as the neutron shield expands.

26 The exposed steel surfaces of the overpack are painted to prevent corrosion. Lifting trunnions
27 are attached to the overpack top flange forging for lifting and for rotating the cask body between
28 vertical and horizontal positions. The lifting trunnions are located 180 degrees apart in the sides
29 of the top flange. Pocket trunnions are welded to the lower side of the overpack to provide a
30 pivoting axis for rotation. The lifting trunnions do not protrude beyond the cylindrical envelope of
31 the overpack enclosure shell. This feature reduces the potential for a direct impact on a
32 trunnion in the event of an overpack side impact. The overpack is provided with aluminum
33 honeycomb impact limiters, one at each end, to ensure that the impact loadings during accident
34 conditions are maintained below the design levels. The neutron shielding material used in the
35 HI-STAR 100 overpack is Holtite-A™, a poured-in-place solid borated synthetic
36 neutron-absorbing polymer.

37 Table 4-9 provides a generic evaluation of potential aging mechanisms and effects requiring
38 management for specific components of the HI-STAR overpack. The AMPs that provide an
39 acceptable approach to managing the aging effects are also identified in the table.

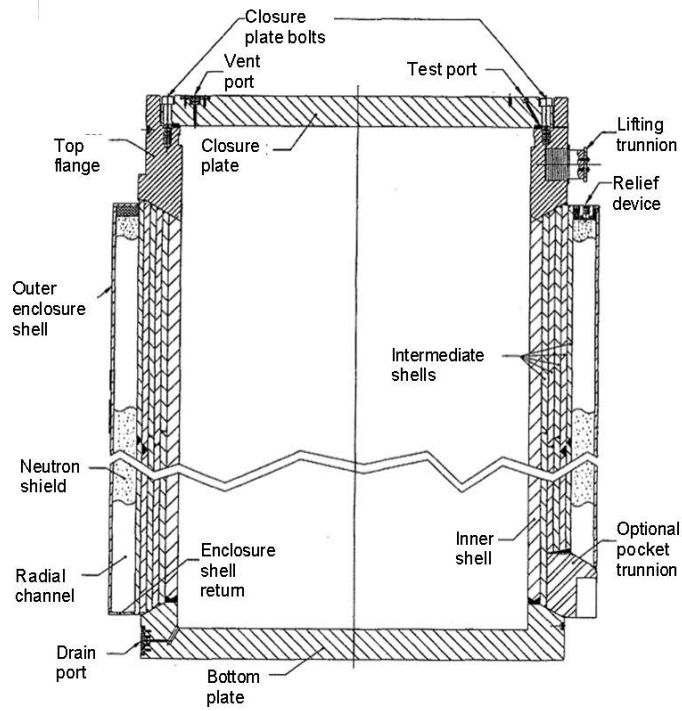


Figure 4-14 HI-STAR 100 overpack elevation view (Holtec International, 2001)

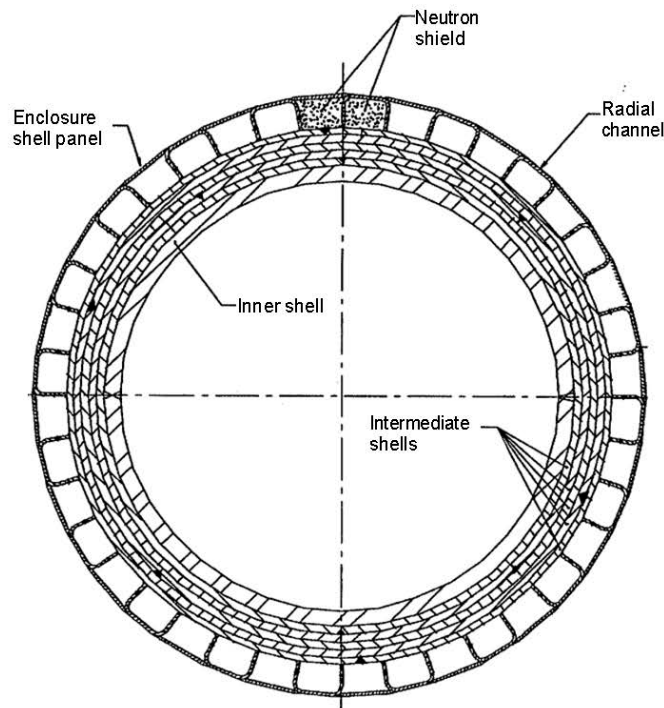


Figure 4-15 HI-STAR 100 overpack cross sectional view (Holtec International, 2001)

1 **4.3.5 Transfer cask**

2 The HI-TRAC TC is a heavy-walled carbon steel cylindrical vessel composed of an inner shell
3 and an outer shell with lead in between to provide gamma shielding (Holtec International, 2013).
4 The TC also includes an exterior carbon steel water jacket for neutron shielding. There are four
5 basic HI-TRAC TC designs: two standard designs, which are a 125-ton HI-TRAC 125 and a
6 100-ton HI-TRAC 100, and two optional designs with a dual-purpose lid for loading and transfer
7 operations, which are the 125-ton HI-TRAC 125D and the 100-ton HI-TRAC 100D. Figure 4-16
8 shows the cross section of a standard HI-TRAC 125 TC with both a pool lid and a transfer lid
9 attached. Since all the MPCs have the same outer diameter, the inner diameter of all HI-TRAC
10 TCs is the same. However, the external dimensions of the HI-TRAC TCs are different, because
11 the 100-ton TCs have a reduced thickness of lead and water shielding.

12 The main structural function of the HI-TRAC TCs is provided by carbon steel, and the main
13 neutron and gamma shielding functions are provided by water and lead, respectively. The top
14 lid of the HI-TRAC 125 and HI-TRAC 125D TCs contains additional Holtite-A™ neutron
15 shielding material. The MPC access hole through the HI-TRAC top lid allows the lowering or
16 raising of the MPC between the TC and the overpack.

17 The standard design HI-TRAC TCs (including HI-TRAC 100 and HI-TRAC 125) include two
18 bottom lids (pool lid and transfer lid). The pool lid is bolted to the bottom flange of the HI-TRAC
19 and is used during MPC fuel loading and sealing operations. In addition to providing shielding
20 in the axial direction, the pool lid incorporates two gasket seals, one between the pool lid top
21 and the bottom flange and the other between the MPC outer wall and the TC inner wall close to
22 the top lid of the TC. These seals provide a barrier from contamination of the exterior of the
23 MPC by the spent fuel pool water. After the MPC has been drained, dried, and sealed, the pool
24 lid is removed and the transfer lid is attached. The transfer lid incorporates two sliding doors
25 that allow the opening of the HI-TRAC bottom for the MPC to be raised or lowered. Unlike the
26 standard designs, the HI-TRAC 100D and HI-TRAC 125D TCs do not require swapping the pool
27 lid for a transfer lid to facilitate transfer of the MPC. The HI-STORM mating device is used to
28 remove the pool lid during MPC transfer operations.

29 In the standard designs, the HI-TRAC TC is equipped with two lifting trunnions located below
30 the top flange and two pocket trunnions located above the bottom flange. The lifting trunnions,
31 made of nickel alloy or stainless steel, are used to vertically handle the HI-TRAC TC. The
32 carbon steel pocket trunnions provide a pivot point for the rotation of the HI-TRAC TC for
33 downending or upending the HI-TRAC TC with a loaded MPC. The HI-TRAC 100D and
34 HI-TRAC 125D TCs are equipped with only lifting trunnions.

35 Table 4-10 provides a generic evaluation of potential aging mechanisms and effects requiring
36 management for specific components of the HI-TRAC TC. The AMPs that provide an
37 acceptable approach to managing the aging effects are also identified in the table.

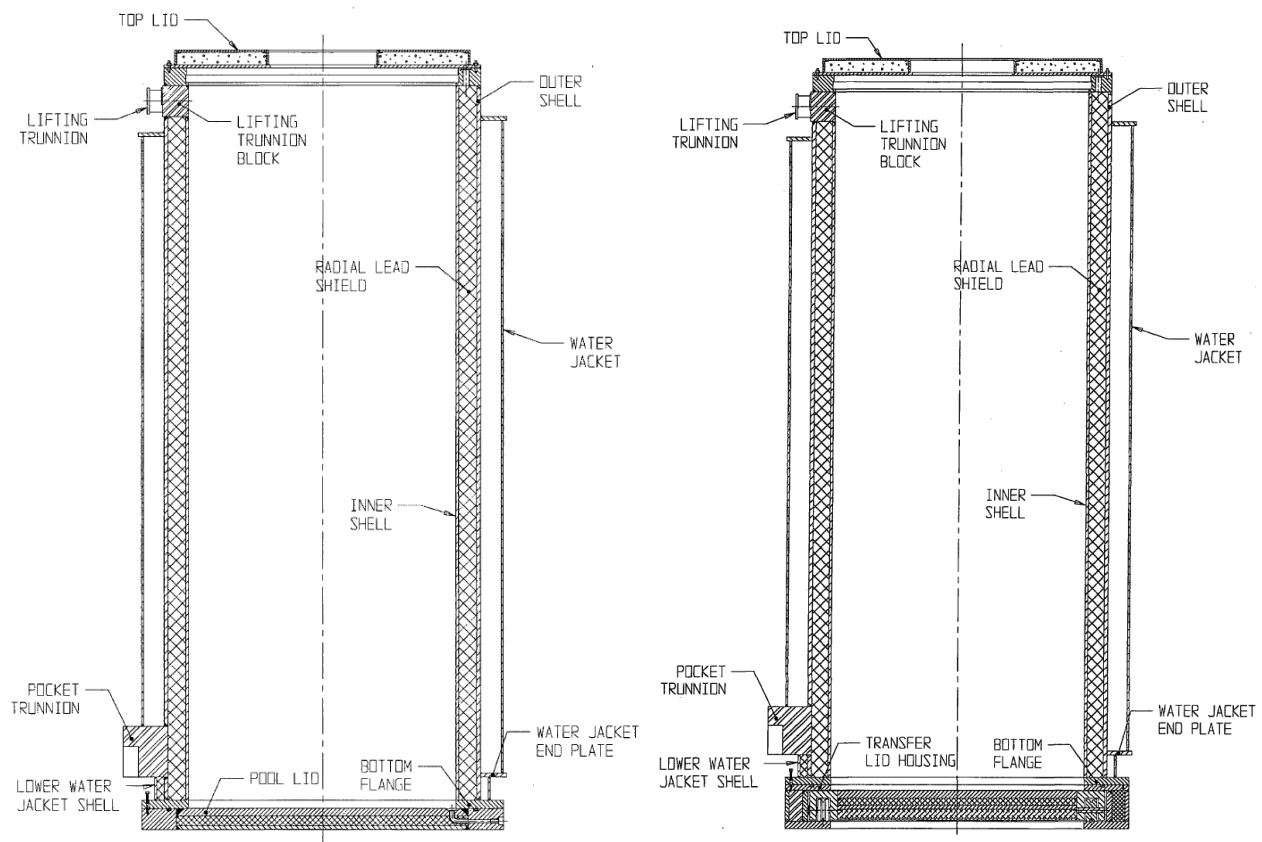


Figure 4-16 Cross sectional views of the HI-TRAC 125 transfer cask with pool lid (left) and transfer lid (right) (Holtec International, 2013)

1

2

Table 4-7 HI-STORM / HI-STAR multipurpose canister												
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)					
Shell	CO, SH, SR, TH*	Stainless steel (welded)	Sheltered	Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5					
								Stainless steel	Sheltered	Pitting and crevice corrosion	Loss of material (precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP
		Microbiologically influenced corrosion	Loss of material	No	3.2.2.4							
						Fatigue	Cracking					
		Radiation embrittlement	Cracking	No	3.2.2.9							
						Helium	Fatigue		Cracking	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
		Creep	Change in dimensions	No	3.2.2.6							
		Baseplate	CO, SH, SR, TH	Stainless steel (welded)	Sheltered	Stress corrosion cracking	Cracking		Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5		

*Safety Functions: Confinement (CO), Subcriticality (CR), Retrievalability (RE), Radiation Shielding (SH), Structural Integrity (SR), Thermal/Heat Removal (TH)

Table 4-7 HI-STORM / HI-STAR multipurpose canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Baseplate	CO, SH, SR, TH	Stainless steel	Sheltered	Pitting and crevice corrosion	Loss of material (precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5
				Pitting and crevice corrosion	Loss of material (precursor to stress corrosion cracking)	Loss of material	3.2.2.4
Microbiologically influenced corrosion	Loss of material	No	3.2.2.4				
Lid	CO, SH, SR, TH	Stainless steel (welded)	Sheltered	Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4

Table 4-7 HI-STORM / HI-STAR multipurpose canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Lid	CO, SH, SR, TH	Stainless steel	Sheltered	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
			Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Closure ring	CO	Stainless steel (welded)	Sheltered	Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5
				Pitting and crevice corrosion	Loss of material (precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2
		Stainless steel	Sheltered	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
Port cover plates	CO	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8

Table 4-7 HI-STORM / HI-STAR multipurpose canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Port cover plates	CO	Stainless steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Basket cell plates	CR, SH, SR, TH	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
BWR fuel basket	CR, SH, SR, TH	Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
				Boron depletion	Loss of criticality control	No; a TLAA may be required	3.4.2.4
				Thermal aging	Loss of strength	No	3.4.2.6
				General corrosion	Loss of material	No	3.4.2.1
				Creep	Change in dimensions	No	3.4.2.5
				Radiation embrittlement	Loss of fracture toughness and loss of ductility	No	3.4.2.7
				Boron depletion	Loss of criticality control	No; a TLAA may be required	3.4.2.4
				Thermal aging	Loss of strength	No	3.4.2.6
				Neutron absorber	CR, SH, TH	Boral®	Helium
				Thermal aging	Loss of strength	No	3.4.2.6

Table 4-7 HI-STORM / HI-STAR multipurpose canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Neutron absorber	CR, SH, TH	Boral [®]	Helium	Wet corrosion and blistering	Change in dimensions	No	3.4.2.3
				Creep	Change in dimensions	No	3.4.2.5
				Radiation embrittlement	Loss of fracture toughness and loss of ductility	No	3.4.2.7
		Metamic [™]	Helium	Boron depletion	Loss of criticality control	No; a TLAA may be required	3.4.2.4
				Thermal aging	Loss of strength	No	3.4.2.6
				General corrosion	Loss of material	No	3.4.2.1
				Creep	Change in dimensions	No	3.4.2.5
Drain and vent shield blocks	SH	Stainless steel (welded)	Helium	Radiation embrittlement	Loss of fracture toughness and loss of ductility	No	3.4.2.7
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
Bottom portion of two-piece lid	SH	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
Stainless steel	Helium	Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
				Thermal aging	Change in dimensions	No	3.2.2.6

Table 4-7 HI-STORM / HI-STAR multipurpose canister									
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)		
Bottom portion of two-piece lid	SH	Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9		
		Steel coated with stainless steel	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8		
Sheathing	SR			Creep	Change in dimensions	No	3.2.1.6		
				Radiation embrittlement	Cracking	No	3.2.1.9		
				General corrosion	Loss of material	No	3.2.2.1		
			Helium	Stainless steel (welded)	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8	
				Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
					Helium	Radiation embrittlement	Cracking	No	3.2.2.9
Basket supports	SR, CR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8		
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7		
				Creep	Change in dimensions	No	3.2.2.6		
		Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9		

Table 4-7 HI-STORM / HI-STAR multipurpose canister

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Lifting lugs	SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
		Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
Lifting lug base plate	SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
		Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
Upper fuel spacer column	SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
		Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
Upper fuel spacer end plate	SR	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
		Stainless steel	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
		Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-7 HI-STORM / HI-STAR multipurpose canister

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Lower fuel spacer column	SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
Lower fuel spacer end plate	SR	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
Vent shield block spacer	SR	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
Vent and drain tubes	SR	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
		Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
		Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6

Table 4-7 HI-STORM / HI-STAR multipurpose canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Damaged fuel container	SR, CO	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
Threaded disc, plug adjustment	CO	Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Creep	Change in dimensions	No	3.2.2.6
Vent and drain plugs	CO	Aluminum	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of strength	No	3.2.3.7
				Creep	Change in dimensions	No	3.2.3.5
				General corrosion	Loss of material	No	3.2.3.1
				Radiation embrittlement	Cracking	No	3.2.3.8

Table 4-8 HI-STORM 100 overpack											
Structure, or System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)				
Concrete shield: radial shield, shield block, pedestal shield, lid shield	SH*	Concrete	Fully encased (steel)	Delayed ettringite formation	Loss of material (spalling, scaling)	No	3.5.1.13				
					Cracking	No	3.5.1.13				
					Loss of strength	No	3.5.1.13				
								Radiation damage	Cracking	No	3.5.1.9
								Loss of strength	No	3.5.1.9	
								Reaction with aggregates	Cracking	TLAA/AMP or a supporting analysis is required	3.5.1.3
Shield block (base, ring, shell)	SH	Steel	Air—outdoor	General corrosion	Loss of strength	TLAA/AMP or a supporting analysis is required	3.5.1.3				
					Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1				
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4				
					Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2			
				Radiation embrittlement	Cracking	No	3.2.1.9				
					Radiation embrittlement	Cracking	No	3.2.1.9			

*Safety Functions: Confinement (CO), Subcriticality (CR), Retrievalability (RE), Radiation Shielding (SH), Structural Integrity (SR), Thermal/Heat Removal (TH)

Table 4-8 HI-STORM 100 overpack							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Lid inner ring	SR	Steel	Embedded (concrete)	Radiation embrittlement	Cracking	No	3.2.1.9
Lid outer ring	SH	Steel	Air—outdoor	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
Shield shell	SH	Steel	Embedded (concrete)	Radiation embrittlement	Cracking	No	3.2.1.9
Gamma shield cross plates	SH	Stainless steel (welded)	Sheltered	Stress corrosion cracking	Cracking	External Surfaces Monitoring of Metallic Components AMP	3.2.2.5
			Air—outdoor	Stress corrosion cracking	Cracking	External Surfaces Monitoring of Metallic Components AMP	3.2.2.5
		Stainless steel	Sheltered	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
			Air—outdoor	Pitting and crevice corrosion	Loss of material (precursor to stress corrosion cracking)	External Surfaces Monitoring of Metallic Components AMP	No
				Radiation embrittlement	Cracking	No	3.2.2.9
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4

Table 4-8 HI-STORM 100 overpack							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Gamma shield cross plates	SH	Stainless steel	Air—outdoor	Pitting and crevice corrosion	Loss of material (precursor to stress corrosion cracking)	External Surfaces Monitoring of Metallic Components AMP	3.2.2.2
				Radiation embrittlement	Cracking	No	3.2.2.9
Baseplate, base spacer block	SR	Steel	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
Outer shell	SR	Steel	Air—outdoor	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9

Table 4-8 HI-STORM 100 overpack							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Outer shell	SR	Steel	Air—outdoor	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
			Embedded (concrete)	Radiation embrittlement	Cracking	No	3.2.1.9
			Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
Microbiologically influenced corrosion	Loss of material	No		3.2.1.4			
Inner shell, lid bottom plate, and lid shell	SR	Steel	Sheltered	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
			Embedded (concrete)	Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
Pedestal shell	SR	Steel	Sheltered	Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
			Embedded (concrete)	Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
Outer shell	SR	Steel	Air—outdoor	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
			Embedded (concrete)	Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
Inner shell, lid bottom plate, and lid shell	SR	Steel	Sheltered	Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
			Embedded (concrete)	Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1

Table 4-8 HI-STORM 100 overpack							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Pedestal platform, MPC support	SH	Steel	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				Radiation embrittlement	Cracking	No	3.2.1.9
Inlet/outlet vent, vertical and horizontal plates, top plate, lid top plate, shear ring	SR	Steel	Air—outdoor	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				Radiation embrittlement	Cracking	No	3.2.1.9
Heat shield, heat/lid shield ring	TH	Steel	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4

Table 4-8 HI-STORM 100 overpack							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Heat shield, heat/lid shield ring	TH	Steel	Sheltered	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				Radiation embrittlement	Cracking	No	3.2.1.9
Radial plate, radial gusset	SR	Steel	Embedded (concrete)	Stress corrosion cracking	Cracking	No	3.2.1.5
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
Lid stud and nut, lid closure bolt	SR	Steel	Air—outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				Stress relaxation	Loss of preload	No	3.2.1.10
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10
				Lid stud	SR	Stainless steel	Air—outdoor

Table 4-8 HI-STORM 100 overpack							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Lid washer	SR	Stainless steel	Air—outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Radiation embrittlement	Cracking	No	3.2.2.9
Bolt anchor block	SR	Steel	Embedded (concrete)	Radiation embrittlement	Cracking	No	3.2.1.9
				Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3
Channel	SR	Steel (galvanized)	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
		Stainless steel (welded)	Sheltered	Atmospheric stress corrosion cracking	Cracking	External Surfaces Monitoring of Metallic Components AMP	3.2.2.5
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
		Stainless steel	Sheltered	Pitting and crevice corrosion	Loss of material (precursor to stress corrosion cracking)	External Surfaces Monitoring of Metallic Components AMP	3.2.2.2

Table 4-8 HI-STORM 100 overpack							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Channel	SR	Stainless steel	Sheltered	Radiation embrittlement	Cracking	No	3.2.2.9
Channel mounts	SR	Steel	Sheltered	Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
Lid lift block	SR	Steel	Air—outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
Lug support ring, gusset	SR	Steel	Air—outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4

Table 4-8 HI-STORM 100 overpack							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Lug support ring, gusset	SR	Steel	Air—outdoor	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
Stud with nut	SR	Steel	Air—outdoor	Stress corrosion cracking	Cracking	No	3.2.1.5
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
Closure lid concrete (HI-STORM 100U)	SH	Concrete	Fully-encased (steel)	Radiation embrittlement	Cracking	No	3.2.1.9
				Stress relaxation	Loss of preload	No	3.2.1.10
				Dehydration at high temperatures	Cracking	No	3.5.1.11
				Delayed ettringite formation	Loss of strength	No	3.5.1.11
				Radiation damage	Loss of material (spalling, scaling)	No	3.5.1.13
					Loss of strength	No	3.5.1.13
					Cracking	No	3.5.1.13
				Cracking	Loss of strength	TLAA/AMP or a supporting analysis is required	3.5.1.9
				Loss of strength	Loss of strength	TLAA/AMP or a supporting analysis is required	3.5.1.9

Table 4-8 HI-STORM 100 overpack							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Closure lid concrete (HI-STORM 100U)	SH	Concrete	Fully-encased (steel)	Reaction with aggregates	Cracking	TLAA/AMP or a supporting analysis is required	3.5.1.3
				Reaction with aggregates	Loss of strength	TLAA/AMP or a supporting analysis is required	3.5.1.3
Closure lid steel (HI-STORM 100U)	SH	Steel	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
Closure lid steel (HI-STORM 100U)	SH	Steel	Air—outdoor	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				Radiation embrittlement	Cracking	No	3.2.1.9
Closure lid steel (HI-STORM 100U)	SH	Steel	Embedded (concrete)	Radiation embrittlement	Cracking	No	3.2.1.9
				Radiation embrittlement	Cracking	No	3.2.1.9

Table 4-8 HI-STORM 100 overpack							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Container shell, bottom plate (HI-STORM 100U)	SR	Steel	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.4
			Groundwater/soil	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	TLAA/AMP or a supporting analysis is required	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	TLAA/AMP or a supporting analysis is required	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
Embedded (concrete)			Embedded (concrete)	General corrosion	Loss of material	TLAA/AMP or a supporting analysis is required	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
Embedded (steel)			Embedded (steel)	Pitting and crevice corrosion	Loss of material	TLAA/AMP or a supporting analysis is required	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				Radiation embrittlement	Cracking	No	3.2.1.9

Table 4-8 HI-STORM 100 overpack							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Container flange (HI-STORM 100U)	SR	Steel	Air—outdoor	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
Divider shell and divider shell restraints (HI-STORM 100U)	TH	Steel	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
Upper and lower MPC guides (HI-STORM 100U)	SR	Steel	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9

Table 4-8 HI-STORM 100 overpack							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
MPC bearing pads (HI-STORM 100U)	SR	Steel (with stainless steel liners)	Sheltered	Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
Insulation (HI-STORM 100U)	TH	Kaowool (ceramic fiber) or equivalent	Fully encased (steel)	Radiation embrittlement	Cracking	No	3.2.1.9
				Moisture absorption	Loss of insulation efficiency (increasing thermal conductivity)	No	3.5.2
Reinforced concrete: VVM interface pad, top surface pad (HI-STORM 100U)	SR, SH	Concrete	Air—outdoor	Radiation embrittlement	Cracking	TLAA/AMP or a supporting analysis is required	3.5.2
				Aggressive chemical attack	Cracking	Reinforced Concrete Structures AMP	3.5.1.5
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.5
					Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.5
					Reduction of concrete pH (reducing corrosion resistance of steel embedments)	Reinforced Concrete Structures AMP	3.5.1.5
	Creep	Cracking	No	3.5.1.2			

Table 4-8 HI-STORM 100 overpack							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Reinforced concrete: VWM interface pad, top surface pad (HI-STORM 100U)	SR, SH	Concrete	Air—outdoor	Dehydration at high temperatures	Cracking	No	3.5.1.11
				Delayed ettringite formation	Loss of strength Loss of material (spalling, scaling) Loss of strength	No No No	3.5.1.11 3.5.1.13 3.5.1.13
				Differential settlement	Cracking	No	3.5.1.13
				Fatigue	Cracking	Reinforced Concrete Structures AMP	3.5.1.4
				Freeze and thaw	Cracking	No	3.5.1.10
				Radiation damage	Cracking	Reinforced Concrete Structures AMP	3.5.1.1
				Reaction with aggregates	Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.1
				Salt scaling	Cracking	TLAA/AMP or a supporting analysis is required	3.5.1.9
				Shrinkage	Loss of strength	TLAA/AMP or a supporting analysis is required	3.5.1.9
					Cracking	Reinforced Concrete Structures AMP	3.5.1.3
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.3
					Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.14
					Cracking	No	3.5.1.7

Table 4-8 HI-STORM 100 overpack									
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)		
Reinforced concrete: VVM interface pad, top surface pad (HI-STORM 100U)	SR, SH	Concrete	Air—outdoor	Leaching of calcium hydroxide	Loss of strength	Reinforced Concrete Structures AMP	3.5.1.8		
					Increase in porosity and permeability	Reinforced Concrete Structures AMP	3.5.1.8		
					Reduction of concrete pH (reducing corrosion resistance of steel embedments)	Reinforced Concrete Structures AMP	3.5.1.8		
			Groundwater/soil			Aggressive chemical attack	Cracking	Reinforced Concrete Structures AMP	3.5.1.5
							Loss of strength	Reinforced Concrete Structures AMP	3.5.1.5
							Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.5
							Reduction of concrete pH (reducing corrosion resistance of steel embedments)	Reinforced Concrete Structures AMP	3.5.1.5
							Cracking	No	3.5.1.2
							Cracking	No	3.5.1.11
							Loss of strength	No	3.5.1.11
Delayed ettringite formation				Loss of material (spalling, scaling)	No	3.5.1.13			
				Loss of strength	No	3.5.1.13			
				Cracking	No	3.5.1.13			

Table 4-8 HI-STORM 100 overpack							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Reinforced concrete: VWM interface pad, top surface pad (HI-STORM 100U)	SR, SH	Concrete	Groundwater/soil	Differential settlement	Cracking	Reinforced Concrete Structures AMP	3.5.1.4
				Fatigue	Cracking	No	3.5.1.10
				Freeze and thaw	Cracking	Reinforced Concrete Structures AMP	3.5.1.1
				Microbiological degradation	Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.1
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.12
				Microbiological degradation	Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.12
					Increase in porosity and permeability	Reinforced Concrete Structures AMP	3.5.1.12
					Reduction of concrete pH (reducing corrosion resistance of steel embedments)	Reinforced Concrete Structures AMP	3.5.1.12
				Radiation damage	Cracking	TLAA/AMP or a supporting analysis is required	3.5.1.9
				Reaction with aggregates	Loss of strength	TLAA/AMP or a supporting analysis is required	3.5.1.9
					Cracking	Reinforced Concrete Structures AMP	3.5.1.3
				Salt scaling	Loss of strength	Reinforced Concrete Structures AMP	3.5.1.3
					Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.14

Table 4-8 HI-STORM 100 overpack							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Reinforced concrete: VVM interface pad, top surface pad (HI-STORM 100U)	SR, SH	Concrete	Groundwater/soil	Shrinkage	Cracking	No	3.5.1.7
				Leaching of calcium hydroxide	Loss of strength	Reinforced Concrete Structures AMP	3.5.1.8
					Increase in porosity and permeability	Reinforced Concrete Structures AMP	3.5.1.8
					Reduction of concrete pH (reducing corrosion resistance of steel embedments)	Reinforced Concrete Structures AMP	3.5.1.8
Retaining wall, support foundation pad (HI-STORM 100U)	SR, SH	Reinforcing steel	Air - outdoor; groundwater	Corrosion of reinforcing steel	Loss of concrete/steel bond Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.6
					Cracking	Reinforced Concrete Structures AMP	3.5.1.6
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.6
					Cracking	Reinforced Concrete Structures AMP	3.5.1.5
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.5
					Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.5
		Reduction of concrete pH (reducing corrosion resistance of steel embedments)	Reinforced Concrete Structures AMP	3.5.1.5			

Table 4-8 HI-STORM 100 overpack							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Retaining wall, support foundation pad (HI-STORM 100U)	SR, SH	Concrete	Groundwater/soil	Creep	Cracking	No	3.5.1.2
				Dehydration at high temperatures	Cracking	No	3.5.1.11
				Delayed ettringite formation	Loss of strength	No	3.5.1.11
				Differential settlement	Loss of material (spalling, scaling)	No	3.5.1.13
					Loss of strength	No	3.5.1.13
				Fatigue	Cracking	No	3.5.1.13
				Freeze and thaw	Cracking	No	3.5.1.10
				Microbiological degradation	Cracking	No	3.5.1.1
					Loss of material (spalling, scaling)	No	3.5.1.1
				Microbiological degradation	Loss of strength	Reinforced Concrete Structures AMP	3.5.1.12
					Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.12
				Microbiological degradation	Increase in porosity and permeability	Reinforced Concrete Structures AMP	3.5.1.12
					Reduction of concrete pH (reducing corrosion resistance of steel embedments)	Reinforced Concrete Structures AMP	3.5.1.12
				Radiation damage	Cracking	TLAA/AMP or a supporting analysis is required	3.5.1.9
Loss of strength	TLAA/AMP or a supporting analysis is required	3.5.1.9					

Table 4-8 HI-STORM 100 overpack							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Retaining wall, support foundation pad (HI-STORM 100U)	SR, SH	Concrete	Groundwater/soil	Reaction with aggregates	Cracking	Reinforced Concrete Structures AMP	3.5.1.3
				Salt scaling	Loss of strength	Reinforced Concrete Structures AMP	3.5.1.3
				Leaching of calcium hydroxide	Loss of material (spalling, scaling)	No	3.5.1.14
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.8
				Increase in porosity and permeability	Reinforced Concrete Structures AMP	3.5.1.8	
				Reduction of concrete pH (reducing corrosion resistance of steel embedments)	Reinforced Concrete Structures AMP	3.5.1.8	
				Loss of concrete/steel bond	Reinforced Concrete Structures AMP	3.5.1.6	
				Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.6	
				Cracking	Reinforced Concrete Structures AMP	3.5.1.6	
				Loss of strength	Reinforced Concrete Structures AMP	3.5.1.6	

Table 4-9 HI-STAR 100 overpack							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Inner shell	CO, SH*	Steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Radiation embrittlement	Cracking	No	3.2.1.9
Bottom plate	CO, SH, SR	Steel	Air—outdoor	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
Top flange	CO, SH, SR	Steel	Helium	Radiation embrittlement	Cracking	No	3.2.1.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Radiation embrittlement	Cracking	No	3.2.1.9
Top flange	CO, SH, SR	Steel	Air—outdoor	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7

*Safety Functions: Confinement (CO), Subcriticality (CR), Retrievability (RE), Radiation Shielding (SH), Structural Integrity (SR), Thermal/Heat Removal (TH)

Table 4-9 HI-STAR 100 overpack							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Top flange	CO, SH, SR	Steel	Air—outdoor	Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Radiation embrittlement	Cracking	No	3.2.1.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
Closure plate	CO, SH, SR	Steel	Air—outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.1.4

Table 4-9 HI-STAR 100 overpack

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Closure plate	CO, SH, SR	Steel	Air—outdoor	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
			Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Radiation embrittlement	Cracking	No	3.2.1.9
				Stress corrosion cracking	Cracking	No	3.2.4.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.4.5
				Microbiologically influenced corrosion	Loss of material	No	3.2.4.3
				Pitting and crevice corrosion	Loss of material	No	3.2.4.2
				Radiation embrittlement	Cracking	No	3.2.4.6
				Stress relaxation	Loss of preload	No	3.2.4.6
Port plug	CO	Stainless steel	Air—outdoor	Stress corrosion cracking	Cracking	No	3.2.2.5
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
			Pitting and crevice corrosion	Loss of material	No	3.2.2.2	
			Radiation embrittlement	Cracking	No	3.2.2.9	
			Stress relaxation	Loss of preload	No	3.2.2.10	

Table 4-9 HI-STAR 100 overpack

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Port plug seal and port cover seal	CO	Nickel alloy	Air—outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.4.3
				Pitting and crevice corrosion	Loss of material	No	3.2.4.2
				Radiation embrittlement	Cracking	No	3.2.4.6
Closure plate seals	CO	Nickel alloy	Air—outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.4.3
				Pitting and crevice corrosion	Loss of material	No	3.2.4.2
				Radiation embrittlement	Cracking	No	3.2.4.6
Intermediate shells	SH, SR	Steel	Embedded (Holtite-A™)	Radiation embrittlement	Cracking	No	3.2.1.9
				Radiation embrittlement	Cracking	No	3.3.1.3
Neutron shield	SH	Holtite-A™	Embedded (steel)	Thermal aging	Loss of fracture toughness and loss of ductility	TLAA/AMP or a supporting analysis is required	3.3.1.2
				Boron depletion	Loss of shielding	TLAA/AMP or a supporting analysis is required	3.3.1.1
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
Removable shear ring	SH	Steel	Air—outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9

Table 4-9 HI-STAR 100 overpack								
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)	
Pocket trunnion plug plate	SH	Stainless steel (welded)	Air—outdoor	Stress corrosion cracking	Cracking	External Surfaces Monitoring of Metallic Components AMP	3.2.2.5	
		Stainless steel	Air—outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4	
				Pitting and crevice corrosion	Loss of material (precursor to stress corrosion cracking)	External Surfaces Monitoring of Metallic Components AMP	No	3.2.2.2
				Radiation embrittlement	Cracking	No	No	3.2.2.9
Radial channels	SR, TH	Steel	Air—outdoor	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1	
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4	
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	No	3.2.1.2
				Radiation embrittlement	Cracking	No	No	3.2.1.9
Pocket trunnion	SH		Embedded (Holtite-A™)	Radiation embrittlement	Cracking	No	3.2.1.9	
		Stainless steel (welded)	Air—outdoor	Stress corrosion cracking	Cracking	External Surfaces Monitoring of Metallic Components AMP	3.2.2.5	
		Stainless steel	Air—outdoor	Microbiologically influenced corrosion	Loss of material	No	No	3.2.2.4

Table 4-9 HI-STAR 100 overpack

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Pocket trunnion	SH	Stainless steel	Air—outdoor	Pitting and crevice corrosion	Loss of material (precursor to stress corrosion cracking)	External Surfaces Monitoring of Metallic Components AMP	3.2.2.2
				Radiation embrittlement	Cracking	No	3.2.2.9
Lifting trunnion	SR	Nickel alloy	Air—outdoor	Atmospheric stress-corrosion cracking	Cracking	No	3.2.4.4
				Microbiologically influenced corrosion	Loss of material	No	3.2.4.3
				Pitting and crevice corrosion	Loss of material	No	3.2.4.2
Rupture disk	SR	Brass	Air—outdoor	Radiation embrittlement	Cracking	No	3.2.4.6
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.5.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.5.3
				Pitting and crevice corrosion	Loss of material	No	3.2.5.2
				Radiation embrittlement	Cracking	No	3.2.5.4
				Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3
Rupture disk plate	SR	Steel	Air—outdoor	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1

Table 4-9 HI-STAR 100 overpack

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Rupture disk plate	SR	Steel	Air—outdoor	Microbiologically influenced corrosion Pitting and crevice corrosion	Loss of material	No	3.2.1.4
Removable shear ring bolt, pocket trunnion plug screw, and alignment pin	SR	Steel	Air—outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
				Stress corrosion cracking	Cracking	No	3.2.1.5
Enclosure shell panels and enclosure shell return	SR	Steel	Air—outdoor	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				Stress relaxation	Loss of preload	No	3.2.1.10
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4

Table 4-9 HI-STAR 100 overpack

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Enclosure shell panels and enclosure shell return	SR	Steel	Air—outdoor	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				Radiation embrittlement	Cracking	No	3.2.1.9
Port cover	SR	Steel	Air—outdoor	Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
Port cover bolt	SR	Steel	Air—outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
				Stress corrosion cracking	Cracking	No	3.2.1.5
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4

Table 4-9 HI-STAR 100 overpack

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Port cover bolt	SR	Steel	Air—outdoor	Pitting and crevice corrosion Radiation embrittlement Stress relaxation	Loss of material Cracking	External Surfaces Monitoring of Metallic Components AMP No	3.2.1.2 3.2.1.9
Trunnion locking pad and end cap bolts	SR	Steel	Air—outdoor	Stress corrosion cracking Galvanic corrosion General corrosion	Loss of preload Cracking Loss of material	No No External Surfaces Monitoring of Metallic Components AMP	3.2.1.10 3.2.1.5 3.2.1.3
Lifting trunnion end cap and locking pad	SR	Steel	Air—outdoor	Microbiologically influenced corrosion Pitting and crevice corrosion Radiation embrittlement Stress relaxation Galvanic corrosion General corrosion	Loss of material Cracking Loss of preload Loss of material Loss of material	No External Surfaces Monitoring of Metallic Components AMP No External Surfaces Monitoring of Metallic Components AMP	3.2.1.4 3.2.1.2 3.2.1.9 3.2.1.10 3.2.1.3

Table 4-9 HI-STAR 100 overpack

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Lifting trunnion end cap and locking pad	SR	Steel	Air—outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9

Table 4-10 HI-TRAC transfer cask										
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)			
Outer shell	SH, SR, TH*	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1			
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4			
				Microbiologically Influenced corrosion	Loss of material	No	3.2.1.2			
				Radiation embrittlement	Cracking	No	3.2.1.9			
			Embedded (lead)	Radiation embrittlement	Cracking	No	3.2.1.9			
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1			
			Inner shell	SH, SR, TH	Steel	DeminerIALIZED water or 25% ethylene glycol solution	Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
							Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
							Radiation embrittlement	Cracking	No	3.2.1.9
							General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
Inner shell	SH, SR, TH	Steel	Air— indoor/outdoor	Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4			
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.2			
				Radiation embrittlement	Cracking	No	3.2.1.9			
				General corrosion	Cracking	No	3.2.1.9			
Inner shell	SH, SR, TH	Steel	Embedded (lead)	Radiation embrittlement	Cracking	No	3.2.1.9			
				Radiation embrittlement	Cracking	No	3.2.1.9			

*Safety Functions: Confinement (CO), Subcriticality (CR), Retrievability (RE), Radiation Shielding (SH), Structural Integrity (SR), Thermal/Heat Removal (TH)

Table 4-10 HI-TRAC transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Radial ribs	SH, SR, TH	Steel	Demineralized water or 25% ethylene glycol solution	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
Radial lead shield	SH, TH	Lead—ASTM B29	Embedded (steel)	None identified	None identified	No	3.2.5.4
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
Water jacket enclosure shell panels	SH, SR, TH	Steel	Air—indoor/outdoor	Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
Lower water jacket shell	SH, SR, TH	Steel	Demineralized water or 25% ethylene glycol solution	Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
Lower water jacket shell	SH, SR, TH	Steel	Air—indoor/outdoor	Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4

Table 4-10 HI-TRAC transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Lower water jacket shell	SH, SR, TH	Steel	Air—indoor/outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
			Demineralized water or 25% ethylene glycol solution	Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
Water jacket end plate, short rib	SH, SR, TH	Steel	Air—indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
			Demineralized water or 25% ethylene glycol solution	Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice Corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
Pool lid outer ring	SH, SR, TH	Steel	Air—indoor/outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1

Table 4-10 HI-TRAC transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Pool lid outer ring	SH, SR, TH	Steel	Air— indoor/outdoor	Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
Pool lid top and bottom plates	SH, SR, TH	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
Pool lid bolt	SR	Steel	Air— indoor/outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
Pool lid lead shield	SH, TH	Lead	Embedded (steel)	Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				Stress relaxation	Loss of preload	No	3.2.1.10
Top flange	SR, SH	Steel	Air— indoor/outdoor	None identified	None identified	No	3.2.5.4
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4

Table 4-10 HI-TRAC transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Top flange	SR, SH	Steel	Air— indoor/outdoor	Microbiologically influenced corrosion Radiation embrittlement	Loss of material Cracking	No No	3.2.1.2 3.2.1.9
Top lid outer and inner rings, top and bottom plates, lifting block	SR, SH	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
			Microbiologically influenced corrosion	Loss of material	No	3.2.1.2	
			Radiation embrittlement	Cracking	No	3.2.1.9	
Top lid stud or bolt	SR	Steel	Embedded (Holtite-A™) Air— indoor/outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
			Air— indoor/outdoor	Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
			Air— indoor/outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
				Stress relaxation	Loss of preload	No	3.2.1.10
Top lid nut	SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.2

Table 4-10 HI-TRAC transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Top lid nut	SR	Steel	Air— indoor/outdoor	Radiation embrittlement Stress relaxation	Cracking Loss of preload	No No	3.2.1.9 3.2.1.10
Top lid shielding	SH, TH	Holtite-A™	Embedded (steel)	Radiation embrittlement	Cracking	TLAA/AMP or a supporting analysis is required	3.3.1.3
				Thermal aging	Loss of fracture toughness and loss of ductility	TLAA/AMP or a supporting analysis is required	3.3.1.2
				Boron depletion	Loss of shielding	TLAA/AMP or a supporting analysis is required	3.3.1.1
Fill port plugs	SR, SH	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
Lifting trunnion block	SR	Steel	Embedded (lead) Air— indoor/outdoor	Radiation embrittlement Stress relaxation	Cracking Loss of preload	No No	3.2.1.9 3.2.1.10
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.2

Table 4-10 HI-TRAC transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Lifting trunnion block	SR	Steel	Air— indoor/outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
Lifting trunnion	SR	Nickel alloy	Air— indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.4.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.4.3
				Stress corrosion cracking	Cracking	No	3.2.4.4
				Radiation embrittlement	Cracking	No	3.2.4.6
				Wear	Loss of material	Transfer Casks AMP	3.2.4.8
		Stainless steel	Air— indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Wear	Loss of material	Transfer Casks AMP	3.2.2.11
Lifting trunnion end cap	SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.2

Table 4-10 HI-TRAC transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Lifting trunnion end cap	SR	Steel	Air— indoor/outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
Pocket trunion, removable pocket trunion, pocket trunion base	SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
Dowel pins, pocket trunnion bolts	SR	Stainless steel	Air— indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10
Bottom flange	SR, SH	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
				Radiation Embrittlement	Cracking	No	3.2.1.9

Table 4-10 HI-TRAC transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Transfer lid top, bottom, intermediate cover, and cover side plates	SR, SH	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
Transfer lid door top, middle, bottom, interface, side, and end plates	SR, SH	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
Transfer lid door top, middle, and side plates	SR, SH	Steel	Embedded (Holtite-A™)	Radiation embrittlement	Cracking	No	3.2.1.9
				Radiation embrittlement	Cracking	No	3.2.1.9
Transfer lid door wheel housing	SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9

Table 4-10 HI-TRAC transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Transfer lid door wheel housing	SR	Steel	Embedded (lead)	Radiation embrittlement	Cracking	No	3.2.1.9
			Embedded (Holtite-A™)	Radiation embrittlement	Cracking	No	3.2.1.9
Transfer lid wheel shaft	SR	Steel	Air—indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
Transfer lid housing stiffener	SR	Steel	Air—indoor/outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
Transfer lid door lock bolt	SR	Steel	Air—indoor/outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9

Table 4-10 HI-TRAC transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Transfer lid door lock bolt	SR	Steel	Air— indoor/outdoor	Stress relaxation	Loss of preload	No	3.2.1.10
Transfer lid lifting lug, lug pad	SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
Transfer lid wheel track	SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
Transfer lid door stop block	SR	Steel	Air— indoor/outdoor	Wear	Loss of material	Transfer Casks AMP	3.2.1.11
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
				Radiation Embrittlement	Cracking	No	3.2.1.9

Table 4-10 HI-TRAC transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Transfer lid door stop block bolt	SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.4
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.2
				Radiation embrittlement	Cracking	No	3.2.1.9
Transfer lid door shielding	SH, TH	Holtite-A™	Embedded (steel)	Stress relaxation	Loss of preload	No	3.2.1.10
				Radiation embrittlement	Cracking	TLAA/AMP or a supporting analysis is required	3.3.1.3
				Thermal aging	Loss of fracture toughness and loss of ductility	TLAA/AMP or a supporting analysis is required	3.3.1.2
Transfer lid door lead shield	SH, TH	Lead	Embedded (steel)	Boron depletion	Loss of shielding	TLAA/AMP or a supporting analysis is required	3.3.1.1
Transfer lid side lead shield	SH, TH	Lead	Embedded (steel)	None identified	None identified	No	3.2.5.4
				None identified	None identified	No	3.2.5.4

1 **4.4 TN-32 and TN-68 systems**

2 **4.4.1 System description**

3 The Transnuclear Inc. (TN) spent-fuel storage cask is a vertical metal cask with a bolted lid
4 closure and two metallic O-rings forming the seal. As a storage cask, it provides confinement,
5 shielding, criticality control, and passive heat removal. There are three types of TN metal
6 storage casks: TN-32, TN-40 (TN-40HT), and TN-68. Only the TN-32 (NRC Docket 72-1021)
7 and TN-68 (NRC Docket 72-1027) casks are evaluated here. The TN-32 cask accommodates
8 32 PWR fuel assemblies. The TN-68 cask accommodates up to 68 BWR fuel assemblies and is
9 also licensed for transportation. Damaged fuel that can be handled by normal means may be
10 stored in eight peripheral compartments of the TN-68 cask that are fitted with damaged-fuel end
11 caps designed to retain gross fragments of fuel.

12 **4.4.2 Bolted metal cask**

13 The TN-32 and TN-68 cask body is a right circular cylinder composed of the following
14 components: (i) confinement vessel with bolted lid closure, (ii) basket for fuel assemblies,
15 (iii) gamma and neutron shield, (iv) pressure/leak-tightness monitoring system, (v) weather
16 cover, and (vi) and trunnions. Figure 4-17 shows the components of the TN-32 cask, and
17 Figure 4-18 shows the confinement-boundary components of the TN-68 cask. The details of the
18 components of the TN-32 cask are provided below as an example of both TN metal casks.

19 Confinement boundary, closure lid, and pressure-monitoring system

20 The TN-32 cask confinement boundary consists of a welded cylindrical low-alloy steel inner
21 shell with an integrally welded low-alloy steel bottom closure. A flange forging is welded to the
22 top of the inner shell to accommodate a bolted low-alloy steel lid closure. The inner shell has a
23 sprayed metallic aluminum coating for corrosion protection. The confinement vessel is
24 surrounded by a carbon steel gamma shield wall and bottom. The cask is sealed with a carbon
25 steel closure lid, which is secured to the top flange of the containment vessel by 48 bolts.

26 The closure lid uses a double-barrier seal system with two metallic O-rings (Helicoflex seals)
27 forming the seal. The annular space between the metallic O-rings is connected to a pressure
28 monitoring system placed between the lid and the protective cover, also called the weather
29 cover, shown in Figure 4-19. Pressure in the tank of the pressure-monitoring system is
30 maintained above the pressure in the cask cavity to prevent either flow of fission gases out of or
31 air into the cask cavity, which, under normal storage conditions, is pressurized above
32 atmospheric pressure with helium. The transducers/switches monitor the pressure in the
33 annular space between the metallic O-rings to provide an indication of seal failure before any
34 release is possible. Two identical transducers/switches are provided to ensure a functional
35 system through redundancy.

36 The TN-32 cask body has four carbon steel trunnions that are welded to the gamma shield.
37 Two of these are located near the top of the cylindrical steel forging, spaced 180 degrees apart,
38 and are used for lifting the cask. The remaining two trunnions are 180 degrees apart and
39 located near the bottom of the cask. The lower trunnions are used to rotate the unloaded cask
40 between vertical and horizontal positions. The lifting trunnions are hollow to permit installation
41 of neutron-shielding material and eliminate a path for neutron streaming. The TN-68 design
42 differs from the TN-32 design in that its two top trunnions are bolted to the gamma shield.

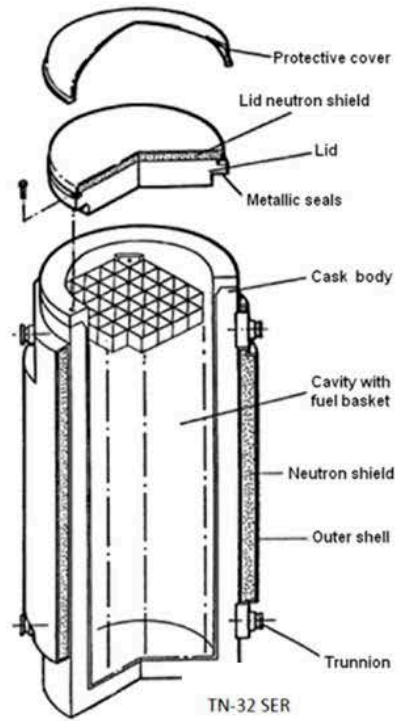


Figure 4-17 Components of the TN-32 storage cask (NRC, 1996)

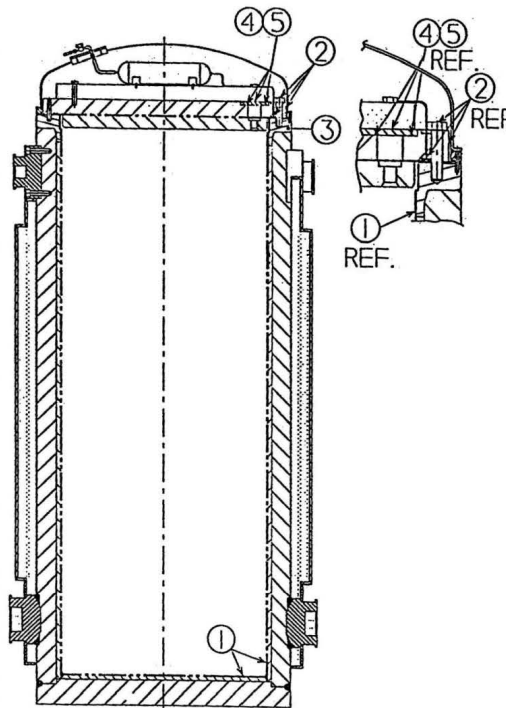


Figure 4-18 TN-68 cask confinement boundary components (Transnuclear Inc., 2005)

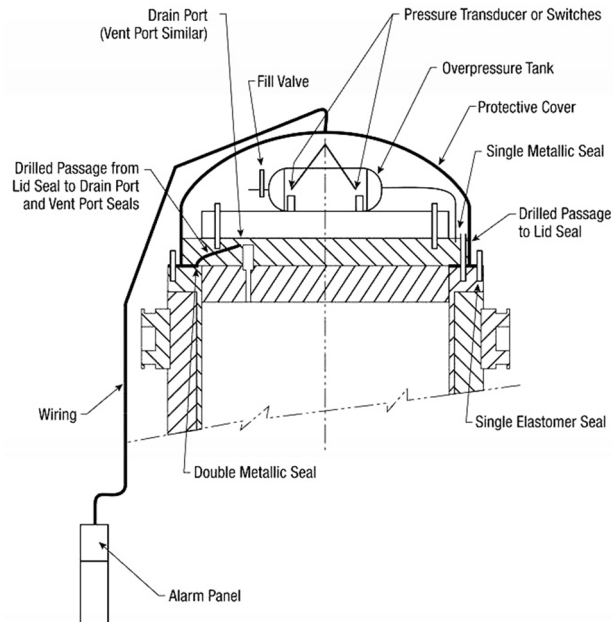


Figure 4-19 TN-32 cask seal pressure-monitoring system (NRC, 1996)

1 The TN-32 cask lid has three confinement access ports—a drain port, a vent port, and an
 2 overpressure system port. The drain and vent ports are covered by a bolted stainless steel
 3 closure plate having a double-barrier seal system with two metallic O-rings forming the seal,
 4 similar to the one used for the lid closure. The overpressure port is also covered by a bolted
 5 stainless steel closure plate but has a single metallic O-ring forming the seal. The closure lid
 6 has drilled interseal passageways connecting the annular space between the seals at each port
 7 to the annular space between the closure-lid seals, as shown in Figure 4-19. The cavity drain
 8 line penetrates the closure lid and terminates in the bottom of the cask cavity. This line is used
 9 to drain water from the cask cavity after underwater fuel loading. It is also used during the
 10 drying and helium backfilling of the cask cavity.

11 The all-metal Helicoflex seal used in the TN metal casks has a central helical energizing spring
 12 with inner and outer liners. Sealing is accomplished by plastic flow of the outer liner against the
 13 mating sealing surfaces. The helical spring aids in keeping a sufficient load against the outer
 14 liner to follow temperature fluctuations and small deformations.

15 The TN-32 confinement vessel has a cylindrical cavity that holds a fuel basket with
 16 32 compartments to locate and support the PWR fuel assemblies. The basket assembly also
 17 transfers heat from the fuel assembly to the cask body wall and provides neutron absorption to
 18 satisfy nuclear criticality requirements, especially during loading and unloading operations that
 19 occur underwater. During storage, with the cavity dry, filled with inert gas, and sealed from the
 20 environment, criticality control measures within the cask are not necessary because of the low
 21 reactivity of the fuel in the dry cask and the assurance that no water can enter the cask during
 22 storage.

23 *Fuel basket assemblies and shielding*

24 The fuel cavities in the basket are formed by a sandwich of aluminum plates, Boral® plates, and
 25 stainless steel boxes. The stainless steel fuel-compartment box sections are attached by a

1 series of stainless steel plugs that pass through the aluminum plates and the poison plates and
2 are fusion-welded to both adjacent stainless steel box sections. The aluminum provides the
3 heat-conduction paths from the fuel assemblies to the cask cavity wall. The poison material
4 provides the necessary criticality control. The basket is held in place by aluminum rails that run
5 the axial length of the cask body, as shown in Figure 4-20.

6 Surrounding the outside of the confinement vessel wall is a steel gamma shield, as shown in
7 Figure 4-21. The bolted closure lid provides the gamma shielding at the upper end of the cask
8 body. Neutron emissions from the stored fuel are attenuated by a neutron shield, consisting of
9 a borated polyester resin compound, enclosed in long aluminum boxes that surround the
10 gamma shield. These aluminum containers are held in place by a steel shell. Neutron
11 emissions from the top of the cask are attenuated by a polypropylene disc, encased in a steel
12 shell and placed on the top of the closure lid. There is no neutron shielding provided on the
13 bottom of the cask.

14 The inside surfaces of the inner shell and bottom have a sprayed metallic coating of aluminum
15 for corrosion protection. The external surfaces of the cask are metal-sprayed with aluminum
16 and/or painted for ease of decontamination and corrosion protection. The neutron shield,
17 pressure-monitoring system, and shield cap are placed on top of the cask after fuel is loaded
18 into the cask. A stainless steel overlay is applied to the O-ring seating surfaces on the body for
19 corrosion protection. A protective cover is bolted to the top of the cask body to provide weather
20 protection for the lid penetrations and other components.

21 Table 4-11 provides a generic evaluation of potential aging mechanisms and effects requiring
22 management for specific components of the TN-32 and TN-68 casks. The AMPs that provide
23 an acceptable approach to managing the aging effects are also identified in the table.

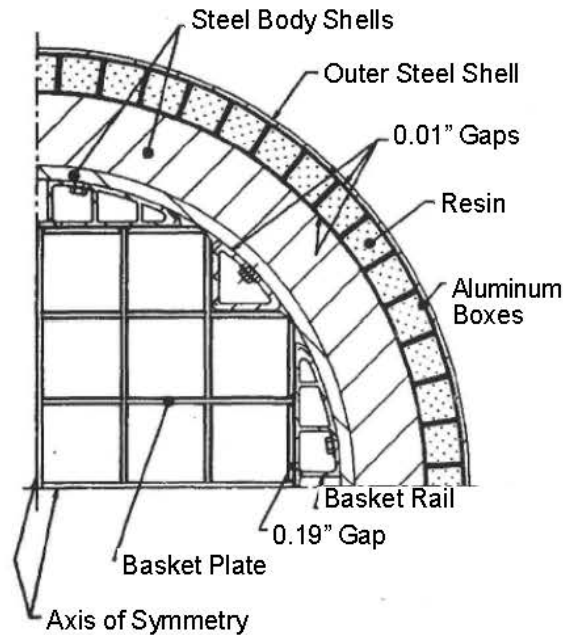


Figure 4-20 Radial cross section of TN-32 cask showing basket, basket rails, and gamma and neutron shields (NRC, 1996)

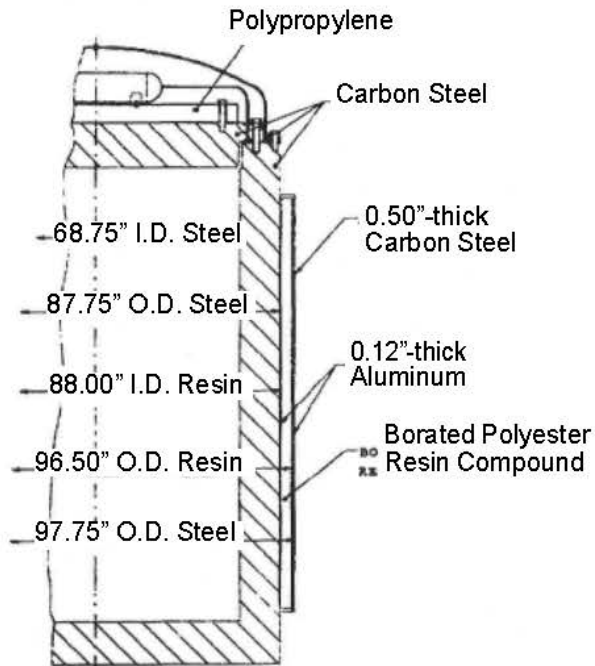


Figure 4-21 TN-32 cask shielding configuration (NRC, 1996)

1

Table 4-11 TN bolted metal casks							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Outer shell	SH, SR, TH*	Steel	Air—outdoor	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
Radial neutron shield	SH, TH	Borated polyester resin	Embedded (aluminum)	Radiation embrittlement	Cracking	TLAA/AMP or a supporting analysis is required	3.3.1.3
				Thermal aging	Loss of fracture toughness and loss of ductility	TLAA/AMP or a supporting analysis is required	3.3.1.2
				Boron depletion	Loss of shielding	TLAA/AMP or a supporting analysis is required	3.3.1.1
Radial neutron shield box	TH	Aluminum	Embedded (borated polyester resin)	Thermal aging	Loss of strength	No	3.2.3.7
				Creep	Change in dimensions	No	3.2.3.5
				Radiation embrittlement	Cracking	No	3.2.3.8
Gamma shield	SH, SR, TH	Steel	Air—outdoor	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2

*Safety Functions: Confinement (CO), Subcriticality (CR), Retrievalability (RE), Radiation Shielding (SH), Structural Integrity (SR), Thermal/Heat Removal (TH)

Table 4-11 TN bolted metal casks							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Gamma shield	SH, SR, TH	Steel	Air—outdoor	Microbiologically influenced corrosion Radiation embrittlement	Loss of material Cracking	No No	3.2.1.4 3.2.1.9
Cask body bottom	SH, SR, TH	Steel	Air—outdoor	General corrosion Pitting and crevice corrosion Microbiologically influenced corrosion Radiation embrittlement	Loss of material Loss of material Loss of material Cracking	External Surfaces Monitoring of Metallic Components AMP External Surfaces Monitoring of Metallic Components AMP No No	3.2.1.1 3.2.1.2 3.2.1.4 3.2.1.9
Upper and lower trunnions	SR	Steel	Air—outdoor	General corrosion Pitting and crevice corrosion Microbiologically influenced corrosion Radiation embrittlement Wear	Loss of material Loss of material Loss of material Cracking Loss of material	External Surfaces Monitoring of Metallic Components AMP External Surfaces Monitoring of Metallic Components AMP No No	3.2.1.1 3.2.1.2 3.2.1.4 3.2.1.9
Upper trunnion	SR	Stainless steel	Air—outdoor	Stress corrosion cracking Pitting and crevice corrosion	Cracking Loss of material	No No	3.2.1.11 3.2.2.5 3.2.2.2

Table 4-11 TN bolted metal casks							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Upper trunnion	SR	Stainless steel	Air—outdoor	Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.2.3
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Radiation embrittlement Wear	Cracking Loss of material	No External Surfaces Monitoring of Metallic Components AMP	3.2.2.9 3.2.2.11
Trunnion bolts	SR	Steel	Air—outdoor	Stress relaxation	Loss of preload	No	3.2.1.10
				Stress corrosion cracking	Cracking	No	3.2.1.5
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
Trunnion cover screw	SH, SR	Stainless steel	Air—outdoor	Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Stress relaxation	Loss of preload	No	3.2.2.10
				Stress corrosion cracking	Cracking	No	3.2.2.5

Table 4-11 TN bolted metal casks							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Trunnion cover screw	SH, SR	Stainless steel	Air—outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Radiation embrittlement	Cracking	No	3.2.2.9
Top neutron shield	SH, TH	Polypropylene	Embedded (steel)	Thermal aging	Loss of fracture toughness and loss of ductility	TLAA/AMP or a supporting analysis is required	3.3.1.2
				Radiation embrittlement	Cracking	No	3.3.1.3
Top neutron shield bolt, vent & drain port cover bolts	SR	Stainless steel	Sheltered	Stress relaxation	Loss of preload	No	3.2.2.10
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
Steel			Sheltered	Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	Bolted Cask Seal Leakage Monitoring AMP	3.2.1.10
				Stress corrosion cracking	Cracking	No	3.2.1.5
				General corrosion	Loss of material	Bolted Cask Seal Leakage Monitoring AMP	3.2.1.1

Table 4-11 TN bolted metal casks							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Top neutron shield bolt, vent & drain port cover bolts	SR	Steel	Sheltered	Galvanic corrosion	Loss of material	Bolted Cask Seal Leakage Monitoring AMP	3.2.1.3
				Pitting and crevice corrosion	Loss of material	Bolted Cask Seal Leakage Monitoring AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Radiation embrittlement	Cracking	No	3.2.1.9
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Radiation embrittlement	Cracking	No	3.2.2.9
Lid	CO, SH, SR, TH	Steel	Sheltered	General corrosion	Loss of material	Bolted Cask Seal Leakage Monitoring AMP	3.2.1.1
				Galvanic corrosion	Loss of material	Bolted Cask Seal Leakage Monitoring AMP	3.2.1.3
				Pitting and crevice corrosion	Loss of material	Bolted Cask Seal Leakage Monitoring AMP	3.2.1.2

Table 4-11 TN bolted metal casks							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Lid	CO, SH, SR, TH	Steel	Sheltered	Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
			Radiation embrittlement	Cracking	No	3.2.1.9	
			General corrosion	Loss of material	No	3.2.1.1	
Lid assembly shim Flange	SH, SR, TH CO, SH, SR, TH	Steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Radiation embrittlement	Cracking	No	3.2.1.9
			Radiation embrittlement	Cracking	No	3.2.1.9	
			General corrosion	Loss of material	No	3.2.1.1	
Lid assembly shim Flange	SH, SR, TH CO, SH, SR, TH	Steel	Embedded (steel)	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3
			Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2	
			Microbiologically influenced corrosion	Loss of material	No	3.2.1.4	
Lid assembly shim Flange	SH, SR, TH CO, SH, SR, TH	Steel	Sheltered	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Radiation embrittlement	Cracking	No	3.2.1.9
			General corrosion	Loss of material	Bolted Cask Seal Leakage Monitoring AMP	3.2.1.1	

Table 4-11 TN bolted metal casks							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Flange	CO, SH, SR, TH	Steel	Sheltered	Galvanic corrosion	Loss of material	Bolted Cask Seal Leakage Monitoring AMP	3.2.1.3
				Pitting and crevice corrosion	Loss of material	Bolted Cask Seal Leakage Monitoring AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	No	3.2.1.1
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Radiation embrittlement	Cracking	No	3.2.1.9
				Stress relaxation	Loss of preload	Bolted Cask Seal Leakage Monitoring AMP	3.2.1.10
				Stress corrosion cracking	Cracking	No	3.2.1.5
Lid bolts	CO, SH, SR, TH	Steel	Sheltered	General corrosion	Loss of material	Bolted Cask Seal Leakage Monitoring AMP	3.2.1.1
				Galvanic corrosion	Loss of material	Bolted Cask Seal Leakage Monitoring AMP	3.2.1.3
				Pitting and crevice corrosion	Loss of material	Bolted Cask Seal Leakage Monitoring AMP	3.2.1.2

Table 4-11 TN bolted metal casks							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Lid bolts	CO, SH, SR, TH	Steel	Sheltered	Microbiologically influenced corrosion Fatigue Radiation embrittlement	Loss of material Cracking Cracking	No Evaluate design code TLAA, if applicable No	3.2.1.4 3.2.1.7 3.2.1.9
Lid threaded insert	SR	Stainless steel	Sheltered	Stress corrosion cracking Pitting and crevice corrosion Microbiologically influenced corrosion Radiation embrittlement	Cracking Loss of material Loss of material Cracking	No No No No	3.2.2.5 3.2.2.2 3.2.2.4 3.2.2.9
Lid seal, vent & drain port cover seal	CO, SH, SR, TH	Aluminum	Sheltered	General corrosion Galvanic corrosion Pitting and crevice corrosion Microbiologically influenced corrosion Radiation embrittlement General corrosion	Loss of material Loss of material Loss of material Loss of material Cracking Loss of material	Bolted Cask Seal Leakage Monitoring AMP Bolted Cask Seal Leakage Monitoring AMP Bolted Cask Seal Leakage Monitoring AMP No No No	3.2.3.1 3.2.3.3 3.2.3.2 3.2.3.4 3.2.3.8 3.2.3.1

Table 4-11 TN bolted metal casks							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Lid seal, vent & drain port cover seal	CO, SH, SR, TH	Aluminum	Helium	Thermal aging	Loss of strength	Bolted Cask Seal Leakage Monitoring AMP	3.2.3.7
				Creep	Change in dimensions	Bolted Cask Seal Leakage Monitoring AMP	3.2.3.5
				Radiation embrittlement	Cracking	No	3.2.3.8
Drain port cover, vent port cover	CO, SH, SR, TH	Stainless steel	Sheltered	Stress corrosion cracking	Cracking	No	3.2.2.5
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Radiation embrittlement	Cracking	No	3.2.2.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
Lid shield plate	SH, SR, TH	Steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				General corrosion	Loss of material	No	3.2.1.1
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8
				Creep	Change in dimensions	No	3.2.1.6
				Radiation embrittlement	Cracking	No	3.2.1.9

Table 4-11 TN bolted metal casks							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Inner confinement shell, bottom confinement plate	CO, SH, SR, TH	Steel	Helium	General corrosion	Loss of material	No	3.2.1.1
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	No	3.2.3.1
Basket rails	SR, TH	Aluminum	Helium	Thermal aging	Loss of strength	TLAA/AMP or a supporting analysis is required	3.2.3.7
				Creep	Change in dimensions	TLAA/AMP or a supporting analysis is required	3.2.3.5
				Radiation embrittlement	Cracking	No	3.2.3.8
Basket rail shim	TH	Aluminum	Helium	General corrosion	Loss of material	No	3.2.3.1
				Creep	Change in dimensions	TLAA/AMP or a supporting analysis is required	3.2.3.5
				Radiation embrittlement	Cracking	No	3.2.3.8
Basket shim	SR, TH	Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Aluminum plate	TH	Aluminum	Helium	General corrosion	Loss of material	No	3.2.3.1
				Galvanic corrosion	Loss of material	No	3.2.3.3

Table 4-11 TN bolted metal casks							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Aluminum plate	TH	Aluminum	Helium	Creep	Change in dimensions	No	3.2.3.5
				Radiation embrittlement	Cracking	No	3.2.3.8
Poison plate	CR, TH	Borated aluminum	Helium	General corrosion	Loss of material	No	3.4.2.1
				Galvanic corrosion	Loss of material	No	3.4.2.2
				Thermal aging	Loss of strength	No	3.4.2.6
				Creep	Change in dimensions	No	3.4.2.5
				Radiation embrittlement	Cracking	No	3.4.2.7
				Boron depletion	Loss of criticality control	No; a TLAA may be required	3.4.2.4
				General corrosion	Loss of material	No	3.4.2.1
				Galvanic corrosion	Loss of material	No	3.4.2.2
				Thermal aging	Loss of strength	No	3.4.2.6
				Creep	Change in dimensions	No	3.4.2.5
Boral®			Helium	Radiation embrittlement	Cracking	No	3.4.2.7
				Boron depletion	Loss of criticality control	No; a TLAA may be required.	3.4.2.4
				General corrosion	Loss of material	No	3.4.2.1
				Galvanic corrosion	Loss of material	No	3.4.2.2
				Thermal aging	Loss of strength	No	3.4.2.6
				Wet corrosion and blistering	Change in dimensions	No	3.4.2.3
				Creep	Change in dimensions	No	3.4.2.5

Table 4-11 TN bolted metal casks							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Poison plate	CR, TH	Boral®	Helium	Radiation embrittlement Boron depletion	Cracking Loss of criticality control	No No; a TLAA may be required	3.4.2.7 3.4.2.4
Fuel compartment, structural plates, basket hold down	CR, SR, TH	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
Basket shear key	SR	Steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				General corrosion	Loss of material	No	3.2.1.1
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8
				Creep	Change in dimensions	No	3.2.1.6
				Radiation embrittlement	Cracking	No	3.2.1.9

1 **4.5 NAC International Systems**

2 **4.5.1 System description**

3 NAC International Inc. (NAC) has three dry storage systems (DSSs) approved for use under a
4 general license: the Universal Storage System (NAC-UMS), the Multi-Purpose Canister
5 (NAC-MPC) system, and the Modular Advanced Generation Nuclear All-Purpose Storage
6 (MAGNASTOR) system. These systems are canister-plus-overpack designs that use a vertical
7 concrete storage cask to house a stainless steel storage canister with a welded closure. The
8 sections below describe the details of the three storage systems.

9 **4.5.2 NAC-UMS**

10 The principal components of NAC-UMS are the transportable storage canister (TSC), vertical
11 concrete cask (VCC), and transfer cask (TC), as shown in Figure 4-22 (NAC International,
12 2004). There are five TSC configurations of different lengths for storage of different classes of
13 PWR and BWR fuel assemblies. The TSC assembly consists of a right circular cylindrical shell
14 with a welded bottom plate, fuel basket, shield lid, two penetration port covers, and a structural
15 lid. The cylindrical shell, bottom plate, and lids constitute the confinement boundary.
16 Figure 4-23 shows the various components of the NAC-UMS TSC for PWR fuel. All TSC
17 components are made of stainless steel, with the exception of neutron poison plates, heat
18 transfer disks, and support disks (BWR TSC fuel baskets only). The fuel basket is designed to
19 accommodate up to 24 PWR or 56 BWR fuel assemblies. The fuel tubes are laterally supported
20 by a series of stainless steel support disks in the PWR basket or carbon steel support disks in
21 the BWR basket, which are retained by spacers on radially located tie rods. The carbon steel
22 support disks are coated with electroless nickel. The square fuel tubes in the PWR basket
23 include stainless-steel encased Boral[®] sheets on all four sides for criticality control. The square
24 fuel tubes in the BWR basket may include stainless-steel encased Boral[®] sheets on up to two
25 sides for criticality control. Aluminum heat transfer disks are spaced midway between the
26 support disks and are the primary path for conducting heat from the fuel assemblies to the TSC
27 wall.

28 The VCC is the storage overpack for the TSC and provides structural support, shielding,
29 protection from environmental conditions, and natural convection cooling of the TSC during
30 storage. Five concrete casks of different lengths are designed to accommodate different TSC
31 configurations. The VCC side walls consist of reinforced concrete and a carbon steel inner
32 liner. The VCC has an annular air passage to allow the natural circulation of air around the
33 canister to remove the decay heat from the spent fuel stored in the TSC. The steel-lined air
34 inlet and outlet vents take nonplanar paths to the cask cavity to minimize radiation streaming.
35 The base plate assembly contains the air inlets and the pedestal that supports the TSC inside
36 the VCC. The top of the VCC is closed by a shield plug consisting of a carbon steel plate for
37 gamma shielding and solid neutron shielding of Bisco NS-3 or NS-4-FR, and a carbon steel lid.
38 The carbon steel lid is installed above the shield plug and is bolted in place. The VCC is lifted
39 from the bottom using an air-pad system. In an alternative design, a set of four carbon steel
40 lifting lugs at the top of the VCC allows for lifting the cask with a loaded TSC from the top end.

41 The transfer cask is used for the vertical transfer of the TSC between workstations and the
42 VCC. Five TCs of different lengths are designed to handle the five TSC configurations. The TC
43 incorporates a multiwall (steel/lead/neutron shield/steel) design and a top retaining ring that is
44 bolted in place to prevent a loaded canister from being inadvertently removed through the top of
45 the TC. All transfer cask structural components are fabricated with high-strength, low-alloy

1 steel, with the exception of stainless steel retaining-ring bolts and shield door lock bolts. The
2 TC contains retractable bottom shield doors for transfer of the TSC from the transfer cask into
3 the VCC, as shown in Figure 4-24. Shield door rails are welded to the bottom plate of the TC to
4 facilitate TSC transfer. The TC has two trunnions near the top of the cask. The trunnions are
5 welded to the inner and outer shells for vertical cask-handling operations. All of the exposed
6 surfaces of the TC, other than the load-bearing surfaces of the trunnions and the bottom door
7 rails, are coated with an epoxy enamel coating to protect the carbon steel and to provide a
8 smooth surface to facilitate decontamination.

9 Table 4-12 through Table 4-14 provide a generic evaluation of potential aging mechanisms and
10 effects requiring management for specific components of the NAC-UMS. The tables also
11 identify AMPs that provide an acceptable approach to managing the aging effects.

12 **4.5.3 NAC-MPC**

13 The NAC-MPC system is similar to the NAC-UMS but is designed for fuel from specific, older
14 power plants. Like the UMS, the principal components of the NAC-MPC include a TSC, VCC,
15 and a TC (NAC International, 2000).

16 The TSC contains a fuel basket that is designed to accommodate up to 36 PWR spent fuel
17 assemblies and reconfigured fuel assemblies with up to 4 damaged fuel cans from the Yankee
18 Rowe Nuclear Power Station, 24 or 26 PWR spent fuel assemblies and reconfigured fuel
19 assemblies with up to 4 damaged fuel cans from the Connecticut Yankee Nuclear Power Plant,
20 or 68 BWR spent fuel assemblies and reconfigured fuel assemblies with up to 32 damaged fuel
21 cans from the LaCrosse Nuclear Generating Station.

22 The canister assembly for the Yankee Rowe and Connecticut Yankee configurations consists of
23 a right circular cylindrical shell with a welded bottom plate, a fuel basket, a shield lid, two
24 penetration port covers, and a structural lid. The cylindrical shell, bottom plate, and lids
25 constitute the confinement boundary. The canister assembly for the La Crosse configuration
26 consists of a right circular cylindrical shell with a welded bottom plate, a fuel basket, a closure
27 lid, a closure ring, and two sets of redundant penetration port covers. The cylindrical shell,
28 bottom plate, closure lid, and inner port covers constitute the confinement boundary.

29 All TSC components are made of stainless steel, with the exception of neutron poison plates,
30 aluminum heat-transfer disks, and an aluminum spacer plate attached to the underside of the
31 closure lid (La Crosse BWR-MPC only). The fuel tubes are laterally supported by a series of
32 support disks that are retained by spacers on radially located tie rods. Aluminum heat-transfer
33 disks are spaced midway between the support disks and are the primary path for conducting
34 heat from the fuel assemblies in the TSC wall. The fuel assemblies are contained in square
35 stainless steel fuel tubes. The fuel tubes are covered with stainless steel-encased Boral[®]
36 sheets on all four sides for criticality control. An alternative fuel basket design has enlarged fuel
37 tubes in the four corner locations. In this alternative configuration, the Boral[®] sheet and
38 stainless steel cover are removed from each side of the fuel tube in the four corner locations.

39 The VCC serves as the storage overpack for the TSC and provides structural support, shielding,
40 protection from environmental conditions, and natural convection cooling of the TSC during
41 storage. The VCC is fabricated from reinforced concrete with a carbon steel liner and base.
42 The VCC has an annular air passage to allow the natural circulation of air around the TSC. The
43 air inlet and outlet vents are steel-lined penetrations that take nonplanar paths to the cask cavity
44 to minimize radiation streaming. The base-plate assembly contains the air inlets and the

1 pedestal that supports the TSC inside the VCC. The top of the VCCs for the Yankee Rowe and
2 Connecticut Yankee configurations is closed by a shield plug and a carbon steel lid bolted in
3 place. The shield plug incorporates a carbon steel plate for gamma shielding and Bisco NS-3 or
4 NS-4-FR for neutron shielding. For the La Crosse configuration, the top of the VCC is closed by
5 a carbon steel and concrete lid bolted in place. The steel-enclosed concrete extends into the
6 cask cavity from the bottom surface of the carbon steel lid.

7 The TC is similar in design and construction to that of the UMS described in Section 4.5.2. It is
8 a multiwall (steel/lead/neutron shield/steel) design with retractable bottom shield doors to allow
9 the TSC to be lowered into the VCC.

10 Table 4-15 through Table 4-17 provide a generic evaluation of potential aging mechanisms and
11 effects requiring management for specific components of the NAC-MPC system. The tables
12 also identify AMPs that provide an acceptable approach to managing the effects of aging.

13 **4.5.4 MAGNASTOR**

14 NAC developed the MAGNASTOR system to improve upon its previous designs in terms of
15 storage capacity, thermal performance, and operations. The principal components of the
16 MAGNASTOR system include a TSC with a welded closure, a concrete cask, and a TC
17 (NAC International, 2015). Figure 4-25 presents schematics of the MAGNASTOR system.

18 The TSC consists of a right circular cylindrical shell with a welded bottom plate, a fuel basket, a
19 closure lid, a closure ring, and two sets of redundant penetration port covers. There are two
20 TSC lengths to accommodate fuel of different lengths. The cylindrical shell plus the bottom
21 plate, closure lid, and welded inner port covers are constructed of stainless steel and constitute
22 the confinement boundary. The closure ring and the outer redundant port covers are also
23 stainless steel and provide the required redundant closure for a welded canister system. There
24 is an alternative closure lid design with a two-piece composite lid assembly that consists of a
25 stainless steel closure lid and a carbon steel shield plate. The shield plate is coated with
26 electroless nickel and bolted to the closure lid. The fuel basket, fabricated from carbon steel
27 and coated with electroless nickel, is designed to accommodate up to 87 BWR fuel assemblies
28 or 37 PWR fuel assemblies, including up to four damaged fuel can locations. The fuel basket
29 design is an arrangement of square fuel tubes held in a right circular cylinder configuration using
30 support weldments that are bolted to the outer fuel tubes. The fuel assembly cells in the fuel
31 baskets include neutron absorber panels on up to four sides for criticality control. The materials
32 of construction for the neutron absorber panels include Boral[®], borated aluminum, and borated
33 metal matrix composite. Each neutron absorber panel is covered by a stainless steel sheet to
34 protect the material during fuel loading and unloading and to maintain it in position. The neutron
35 absorber and stainless steel cover are secured to the fuel tube using weld posts located across
36 the width and along the length of the fuel tube.

37 The concrete cask is the storage overpack for the TSC and provides structural support,
38 shielding, protection from environmental conditions, and natural convection cooling of the TSC
39 during storage. The concrete cask is a reinforced concrete structure with a carbon steel inner
40 liner and base, shown in Figure 4-26 (NAC International, 2014). There are four concrete cask
41 configurations of different lengths and design variations. The concrete cask provides an annular
42 air passage to allow the natural circulation of air around the TSC to remove the decay heat from
43 the stored spent fuel. The lower air inlets and upper air outlets are steel-lined penetrations in
44 the concrete cask body. The base plate assembly contains the air inlets and the pedestal that
45 supports the TSC. Carbon steel channels that are attached to the inner liner assist in centering

1 the TSC in the overpack. The top of the concrete cask is closed by a carbon steel and concrete
2 lid bolted in place. The concrete cask is lifted by means of carbon steel lift anchor and lug
3 assemblies embedded in the top of the concrete cask body. Alternatively, the concrete cask
4 can be lifted and moved by means of air pads inserted into the four lower inlets.

5 The transfer cask provides shielding during TSC movements between workstations and the
6 concrete cask. The TC is provided in two different configurations (referred to as MTC1 and
7 MTC2) that differ primarily in the structural materials and overall length. The materials of
8 construction for the TC structural components are carbon steel for the MTC1 configuration and
9 stainless steel for the MTC2 configuration. The TC is a multiwall (steel/lead/NS-4-FR/steel)
10 design. It incorporates stainless steel retaining blocks or a bolted retaining ring to prevent a
11 loaded TSC from being inadvertently removed through the top of the TC. The TC contains
12 retractable bottom shield doors that are used during TSC loading and unloading operations.
13 Shield door rails are welded to the bottom ring of the TC to facilitate TSC transfer. The TC has
14 two trunnions near the top of the cask that are welded to the top ring for vertical cask-handling
15 operations. The exposed carbon steel surfaces of the MTC1 transfer cask, except for the wear
16 surfaces of the shield doors and rails, are coated with an epoxy enamel coating to protect the
17 components from corrosion and adverse interactions with the operating environments.

18 Table 4-18 through Table 4-20 provide a generic evaluation of potential aging mechanisms and
19 effects requiring management for specific components of the MAGNASTOR system. The tables
20 also identify AMPs that provide an acceptable approach to managing the aging effects.

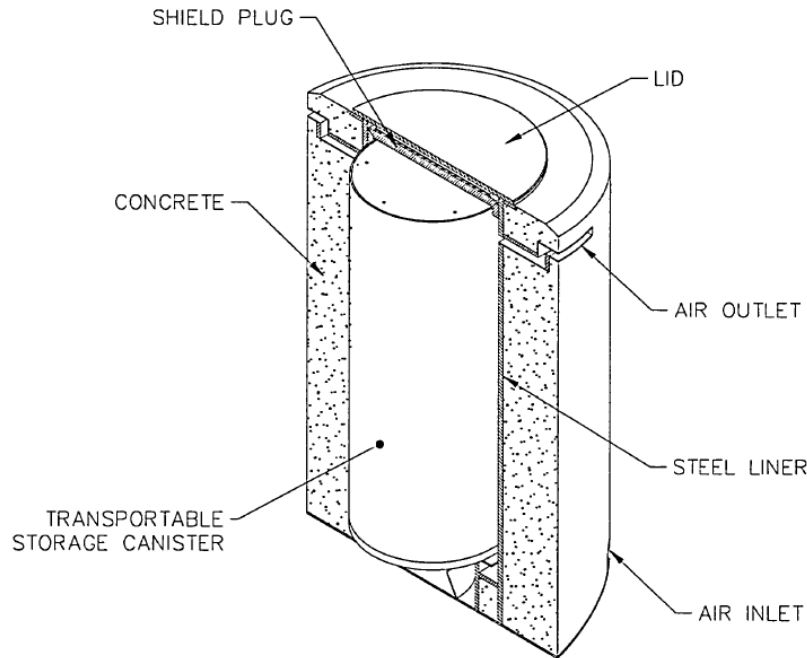


Figure 4-22 NAC-UMS (NAC International, 2004)

21

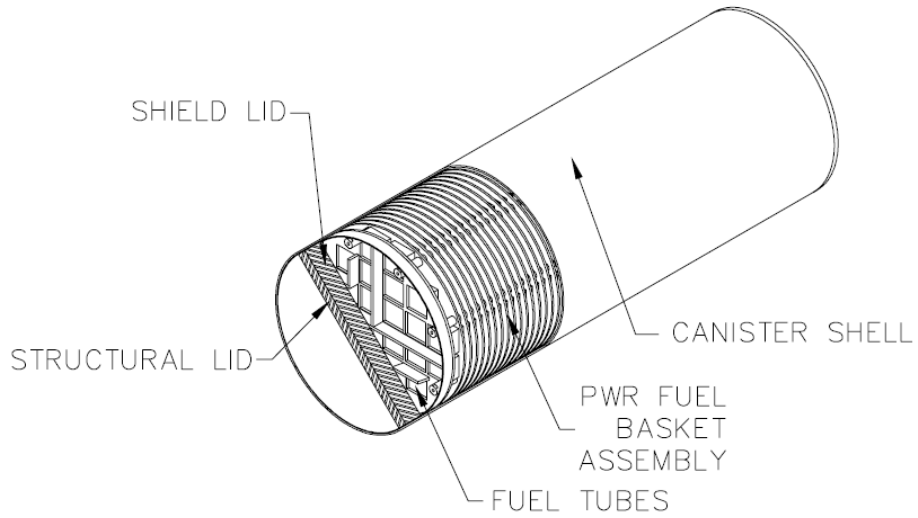


Figure 4-23 NAC-UMS transportable storage canister for PWR fuel (NAC International, 2004)

1

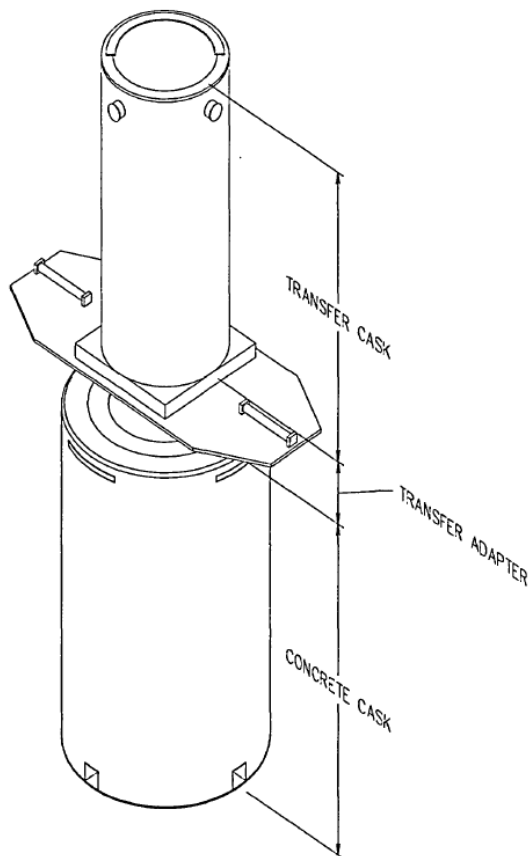


Figure 4-24 NAC-UMS VCC and transfer cask arrangement (NAC International, 2004)

2

3

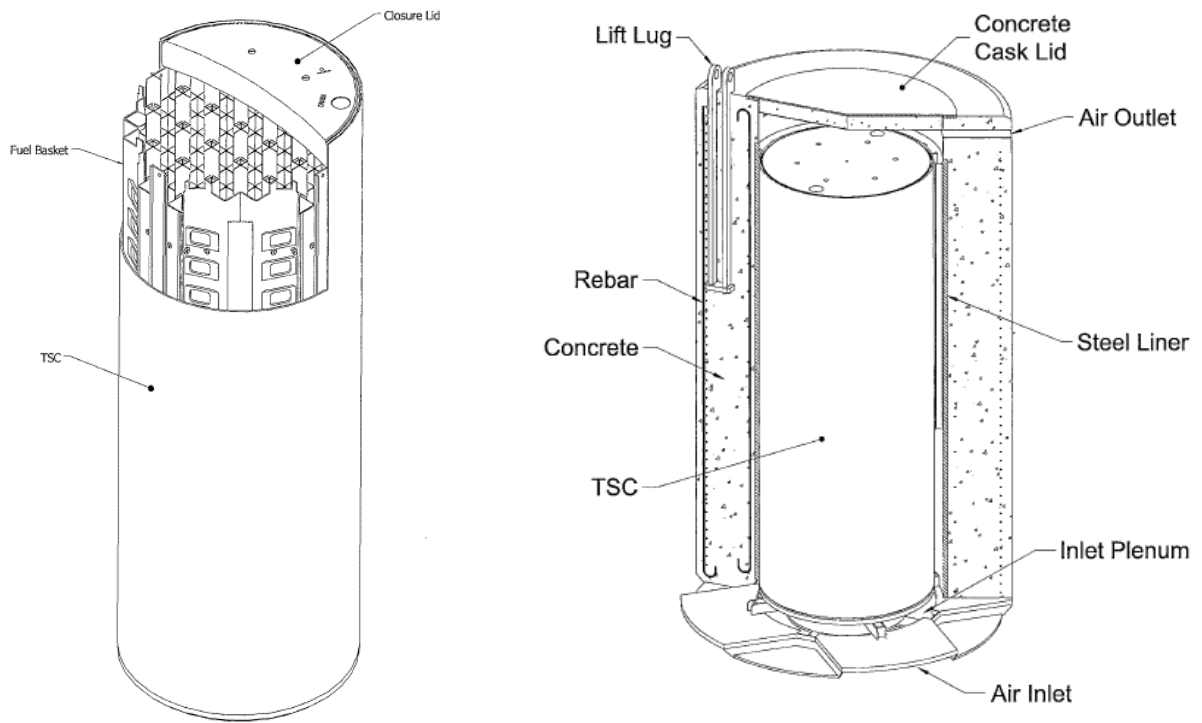


Figure 4-25 NAC MAGNASTOR TSC and concrete cask (NAC International, 2015)

1



Figure 4-26 NAC MAGNASTOR concrete cask (NAC International, 2014)

2

Table 4-12 NAC-UMS transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Shell	CO, SR*	Stainless steel (welded)	Sheltered	Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5
		Stainless steel	Sheltered	Pitting and crevice corrosion	Loss of material (Precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2
Bottom	CO, SR	Stainless steel (welded)	Sheltered	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5
Stainless steel	Sheltered	Pitting and crevice corrosion	Loss of material (Precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2		

*Safety Functions: Confinement (CO), Subcriticality (CR), Retrievability (RE), Radiation Shielding (SH), Structural Integrity (SR), Thermal/Heat Removal (TH)

Table 4-12 NAC-UMS transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Bottom	CO, SR	Stainless steel	Sheltered	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
				Radiation embrittlement	Cracking	No	3.2.2.9
Structural lid	SR	Stainless steel (welded)	Sheltered	Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5
				Pitting and crevice corrosion	Loss of material (Precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
			Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-12 NAC-UMS transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Spacer ring	SR	Stainless steel (welded)	Sheltered	Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5
		Stainless steel	Sheltered	Pitting and crevice corrosion	Loss of material (Precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2
Shield lid, support ring	SH, SR	Stainless steel (welded)	Helium	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Radiation embrittlement	Cracking	No	3.2.2.9
		Stainless steel	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
Port cover	CO	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6

Table 4-12 NAC-UMS transportable storage canister

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Port cover	CO	Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
Fuel tube, cladding, CR, SR flange		Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
Neutron absorber	CR	Boral	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				General corrosion	Loss of material	No	3.4.2.1
				Galvanic corrosion	Loss of material	No	3.4.2.2
				Thermal aging	Loss of strength	No	3.4.2.6
				Wet corrosion and blistering	Change in dimensions	No	3.4.2.3
				Creep	Change in dimensions	No	3.4.2.5
				Radiation embrittlement	Cracking	No	3.4.2.7
Boron depletion	Loss of criticality control	No; a TLAA may be required.	3.4.2.4				

Table 4-12 NAC-UMS transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Fuel basket bottom weldment	SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
Fuel basket top weldment	SR	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
Fuel basket tie rod, spacer, washer	SR	Stainless steel (welded)	Helium	Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-12 NAC-UMS transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Fuel basket support disk	SR	Stainless steel (17-4 PH)	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
				General corrosion	Loss of material	No	3.2.1.1
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8
		Steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Creep	Change in dimensions	No	3.2.1.6
				Radiation embrittlement	Cracking	No	3.2.1.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
Fuel basket top nut	SR	Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7

Table 4-12 NAC-UMS transportable storage canister

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Fuel basket top nut	SR	Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10
Fuel basket heat transfer disk	TH	Aluminum	Helium	General corrosion	Loss of material	No	3.2.3.1
				Thermal aging	Loss of strength	No	3.2.3.7
				Creep	Change in dimensions	No	3.2.3.5
				Radiation embrittlement	Cracking	No	3.2.3.8
Maine Yankee fuel can tube body	CR, SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
Maine Yankee fuel can bottom and side plates	CR, SR	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7

Table 4-12 NAC-UMS transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Maine Yankee fuel can bottom and side plates	CR, SR	Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Maine Yankee fuel can lid plate, lid bottom	CR, SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
Maine Yankee fuel can lift tee, support ring	SR	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
Maine Yankee fuel can lid collar, screen	CO	Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6

Table 4-12 NAC-UMS transportable storage canister

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Maine Yankee fuel can lid collar, screen	CO	Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-13 NAC-UMS vertical concrete cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Technical Basis (Section)	
Concrete shell	SH, SR*	Concrete	Air—outdoor	Aggressive chemical attack	Cracking	Reinforced Concrete Structures AMP	3.5.1.5
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.5
					Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.5
				Creep	Cracking	No	3.5.1.2
					Cracking	No	3.5.1.11
					Loss of strength	No	3.5.1.11
				Delayed ettringite formation	Loss of material (spalling, scaling)	No	3.5.1.13
					Loss of strength	No	3.5.1.13
					Cracking	No	3.5.1.13
				Fatigue	Cracking	No	3.5.1.10
					Cracking	Reinforced Concrete Structures AMP	3.5.1.1
					Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.1
				Radiation damage	Cracking	No	3.5.1.9
Loss of strength	No	3.5.1.9					

*Safety Functions: Confinement (CO), Subcriticality (CR), Retrievalability (RE), Radiation Shielding (SH), Structural Integrity (SR), Thermal/Heat Removal (TH)

Table 4-13 NAC-UMS vertical concrete cask								
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)	
Concrete shell	SH, SR	Concrete	Air—outdoor	Reaction with aggregates	Cracking	Reinforced Concrete Structures AMP	3.5.1.3	
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.3	
				Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.14		
				Shrinkage	No	3.5.1.7		
		Leaching of calcium hydroxide	Loss of strength	Reinforced Concrete Structures AMP	3.5.1.8			
			Increase in porosity and permeability	Reinforced Concrete Structures AMP	3.5.1.8			
		Reinforcing steel	Air—outdoor, groundwater	Reinforcing steel	Corrosion of reinforcing steel	Reduction of concrete pH (reducing corrosion resistance of steel embedments)	Reinforced Concrete Structures AMP	3.5.1.8
						Loss of concrete/steel bond	Reinforced Concrete Structures AMP	3.5.1.6
						Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.6
						Cracking	Reinforced Concrete Structures AMP	3.5.1.6
Loss of strength	Reinforced Concrete Structures AMP	3.5.1.6						

Table 4-13 NAC-UMS vertical concrete cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Inner shell	SH, SR	Steel	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
Top flange, support ring	SR	Steel	Embedded (concrete)	Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
Pedestal plate	SR	Steel	Sheltered	Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3

Table 4-13 NAC-UMS vertical concrete cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Pedestal plate	SR	Steel	Sheltered	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
Pedestal cover	SR	Stainless steel	Sheltered	Stress corrosion cracking	Cracking	No	3.2.2.5
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
Base plate assembly	SH, SR, TH	Steel	Sheltered	Radiation embrittlement	Cracking	No	3.2.2.9
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9

Table 4-13 NAC-UMS vertical concrete cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Base plate nelson studs	SR	Steel	Embedded (concrete)	General corrosion	Loss of material	TLAA/AMP or a supporting analysis is required	3.2.1.1
				Pitting and crevice corrosion	Loss of material	TLAA/AMP or a supporting analysis is required	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
Outlet vent assembly	SH, SR, TH	Steel	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
Inlet and outlet vent hardware	SR	Stainless steel	Sheltered	Radiation embrittlement	Cracking	No	3.2.1.9
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2

Table 4-13 NAC-UMS vertical concrete cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Inlet and outlet vent hardware	SR	Stainless steel	Sheltered	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10
Inlet vent supplemental shielding assembly	SH	Steel	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
Lid	SR	Steel	Air—outdoor	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4

Table 4-13 NAC-UJS vertical concrete cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Lid	SR	Steel	Air—outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
			Sheltered	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
			Air—outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
				Stress corrosion cracking	Cracking	No	3.2.2.5
			Sheltered	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
			Air—outdoor	Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10
Shield plug assembly	SH, SR	Steel	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1

Table 4-13 NAC-UMS vertical concrete cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Shield plug assembly	SH, SR	Steel	Sheltered	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Thermal aging	Loss of fracture toughness and loss of ductility	TLAA/AMP or a supporting analysis is required	3.3.1.2
Neutron shield (Shield plug)	SH	NS-4-FR	Embedded (steel)	Radiation embrittlement	Cracking	TLAA/AMP or a supporting analysis is required	3.3.1.3
				Boron depletion	Loss of shielding	TLAA/AMP or a supporting analysis is required	3.3.1.1
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.3.1.2
				Radiation embrittlement	Cracking	No	3.3.1.3
Lift anchor	SR	Steel	Air—outdoor	Boron depletion	Loss of shielding	TLAA/AMP or a supporting analysis is required	3.3.1.1
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2

Table 4-13 NAC-UMS vertical concrete cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Lift anchor	SR	Steel	Air—outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
			Embedded (concrete)	General corrosion	Loss of material	TLAA/AMP or a supporting analysis is required	3.2.1.1
				Pitting and crevice corrosion	Loss of material	TLAA/AMP or a supporting analysis is required	3.2.1.2
Lift lug	SR	Steel	Sheltered	Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
			Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
			Sheltered	Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
Lift anchor hardware	SR	Steel	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1

Table 4-13 NAC-UMS vertical concrete cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Lift anchor hardware	SR	Steel	Sheltered	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Stress relaxation	Loss of preload	External Surfaces Monitoring of Metallic Components AMP	3.2.1.10

Table 4-14 NAC-UMS transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Outer shell	SR*	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Galvanic corrosion	Loss of material	Transfer Casks AMP	3.2.1.3
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
Inner shell	SR	Steel	Air— indoor/outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
Gamma shield (Cask body)	SH	Lead	Embedded (steel, NS-4-FR)	Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Radiation embrittlement	Cracking	No	3.2.1.9
				None identified	None identified	No	3.2.6

*Safety Functions: Confinement (CO), Subcriticality (CR), Retrievability (RE), Radiation Shielding (SH), Structural Integrity (SR), Thermal/Heat Removal (TH)

Table 4-14 NAC-UMS transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Neutron shield (Cask body)	SH	NS-4-FR	Embedded (steel, lead)	Thermal aging	Loss of fracture toughness and loss of ductility	TLAA/AMP or a supporting analysis is required	3.3.1.2
				Radiation embrittlement	Cracking	TLAA/AMP or a supporting analysis is required	3.3.1.3
				Boron depletion	Loss of shielding	TLAA/AMP or a supporting analysis is required	3.3.1.1
Top plate	SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
Bottom plate	SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9

Table 4-14 NAC-UMS transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Retaining ring	SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Galvanic corrosion	Loss of material	Transfer Casks AMP	3.2.1.3
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10
Trunnion	SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2

Table 4-14 NAC-UMS transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Trunnion	SR	Steel	Air— indoor/outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Wear	Loss of material	Transfer Casks AMP	3.2.1.11
Shield door plates	SH, SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.3
Shield door rails	SH, SR	Steel	Air— indoor/outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Galvanic corrosion	Loss of material	Transfer Casks AMP	3.2.1.3
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Wear	Loss of material	Transfer Casks AMP	3.2.1.11

Table 4-14 NAC-UMS transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Shield door lock bolts	SR	Stainless steel	Air— indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10
Transfer adapter	SH	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
Wear strip	SR	Stainless steel	Air— indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5

Table 4-14 NAC-UMS transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Wear strip	SR	Stainless steel	Air— indoor/outdoor	Radiation embrittlement Wear	Cracking Loss of material	No Transfer Casks AMP	3.2.2.9 3.2.2.11

Table 4-15 NAC-MPC transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Shell	CO, SR*	Stainless steel (welded)	Sheltered	Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5
					Loss of material (Precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2
		Stainless steel	Sheltered	Pitting and crevice corrosion	Loss of material	No	3.2.2.4
					Microbiologically influenced corrosion	Evaluate design code TLAA, if applicable	3.2.2.7
Bottom	CO, SR	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
					Cracking	TLAA/AMP or a supporting analysis is required	3.2.2.9
		Stainless steel	Sheltered	Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5
					Loss of material (Precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2

*Safety Functions: Confinement (CO), Subcriticality (CR), Retrievability (RE), Radiation Shielding (SH), Structural Integrity (SR), Thermal/Heat Removal (TH)

Table 4-15 NAC-MPC transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Bottom	CO, SR	Stainless steel	Sheltered	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
				Radiation embrittlement	Cracking	No	3.2.2.9
Closure lid; structural lid	SR	Stainless steel (welded)	Sheltered	Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5
				Pitting and crevice corrosion	Loss of material (Precursor to stress corrosion cracking)	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
			Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-15 NAC-MPC transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Closure ring; spacer ring	SR	Stainless steel (welded)	Sheltered	Atmospheric stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5
				Pitting and crevice corrosion	Loss of material (Precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
Closure lid assembly bolt, washer	SR	Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10
Closure lid assembly spacer	SR	Aluminum	Helium	General corrosion	Loss of material	No	3.2.3.1
				Galvanic corrosion	Loss of material	No	3.2.3.3
				Thermal aging	Loss of strength	TLLAA/AMP or a supporting analysis is required	3.2.3.7
				Creep	Change in dimensions	TLLAA/AMP or a supporting analysis is required	3.2.3.5

Table 4-15 NAC-MPC transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Closure lid assembly spacer	SR	Aluminum	Helium	Radiation embrittlement	Cracking	No	3.2.3.8
Shield lid, support ring	SH, SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Port cover	CO	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
			Sheltered	Atmospheric stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5
		Stainless steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
			Sheltered	Pitting and crevice corrosion	Loss of material (Precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2

Table 4-15 NAC-MPC transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Port cover	CO	Stainless steel	Sheltered	Microbiologically influenced corrosion Radiation embrittlement	Loss of material Cracking	No No	3.2.2.4 3.2.2.9
Fuel tube, cladding, flange	CR, SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Neutron absorber	CR	Boral	Helium	General corrosion	Loss of material	No	3.4.2.1
				Galvanic corrosion	Loss of material	No	3.4.2.2
				Thermal aging	Loss of strength	No	3.4.2.6
				Wet corrosion and blistering	Change in dimensions	No	3.4.2.3
				Creep	Change in dimensions	No	3.4.2.5
				Radiation embrittlement	Cracking	No	3.4.2.7

Table 4-15 NAC-MPC transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Neutron absorber	CR	Boral	Helium	Boron depletion	Loss of criticality control	No; a TLAA may be required	3.4.2.4
Plate in lieu of neutron absorber	TH	Aluminum	Helium	General corrosion	Loss of material	No	3.2.3.1
				Galvanic corrosion	Loss of material	No	3.2.3.3
				Thermal aging	Loss of strength	No	3.2.3.7
				Creep	Change in dimensions	No	3.2.3.5
				Radiation embrittlement	Cracking	No	3.2.3.8
Fuel basket bottom weldment	SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
Fuel basket top weldment	SR	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
Fuel basket top weldment		Stainless steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7

Table 4-15 NAC-MPC transportable storage canister								
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)	
Fuel basket top weldment	SR	Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6	
				Radiation embrittlement	Cracking	No	3.2.2.9	
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8	
Fuel basket tie rod, spacer, washer	SR	Stainless steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7	
				Creep	Change in dimensions	No	3.2.2.6	
				Radiation embrittlement	Cracking	No	3.2.2.9	
				Thermal aging	Loss of fracture toughness and loss of ductility	TLAA/AMP or a supporting analysis is required	3.2.2.8	
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7	
				Creep	Change in dimensions	No	3.2.2.6	
Fuel basket top nut	SR	Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9	
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8	
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7	
Fuel basket top nut	SR	Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6	
				Radiation embrittlement	Cracking	No	3.2.2.9	
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8	
Fuel basket top nut	SR	Stainless steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7	
				Creep	Change in dimensions	No	3.2.2.6	
				Radiation embrittlement	Cracking	No	3.2.2.9	

Table 4-15 NAC-MPC transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Fuel basket top nut	SR	Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
Fuel basket heat transfer disk	TH	Aluminum	Helium	Stress relaxation	Loss of preload	No	3.2.2.10
				General corrosion	Loss of material	No	3.2.3.1
				Thermal aging	Loss of strength	No	3.2.3.7
				Creep	Change in dimensions	No	3.2.3.5
				Radiation embrittlement	Cracking	No	3.2.3.8
Damaged fuel can tube body	CR, SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
Damaged fuel can bottom and side plates, screen cover plate	CR, SR	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
				Creep	Change in dimensions	No	3.2.2.6

Table 4-15 NAC-MPC transportable storage canister									
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)		
Damaged fuel can bottom and side plates, screen cover plate	CR, SR	Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9		
Damaged fuel can lid plate, lid bottom	CR, SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8		
Damaged fuel can lift tee, support ring	SR	Stainless steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7		
				Creep	Change in dimensions	No	3.2.2.6		
				Radiation embrittlement	Cracking	No	3.2.2.9		
Damaged fuel can screen	CO	Stainless steel	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8		
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7		
				Creep	Change in dimensions	No	3.2.2.6		
Damaged fuel can screen	CO	Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9		
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8		

Table 4-15 NAC-MPC transportable storage canister								
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)	
Damaged fuel can screen	CO	Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6	
Removable fuel rod retainer assembly (Yankee-MPC)	SR	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9	
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8	
Reconfigured fuel assembly shell casing, top ring (Yankee-MPC)	SR	Stainless steel (welded)	Helium	Creep	Change in dimensions	No	3.2.2.6	
				Radiation embrittlement	Cracking	No	3.2.2.9	
Reconfigured fuel assembly top end fitting assembly (Yankee-MPC)	CR, SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8	
				Creep	Change in dimensions	No	3.2.2.6	
Reconfigured fuel assembly top end fitting assembly (Yankee-MPC)	CR, SR	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9	
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8	
Reconfigured fuel assembly top end fitting assembly (Yankee-MPC)	CR, SR	Stainless steel (welded)	Helium	Creep	Change in dimensions	No	3.2.2.6	
				Radiation embrittlement	Cracking	No	3.2.2.9	
Reconfigured fuel assembly top end fitting assembly (Yankee-MPC)	CR, SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8	
				Creep	Change in dimensions	No	3.2.2.6	

Table 4-15 NAC-MPC transportable storage canister								
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)	
Reconfigured fuel assembly top end fitting assembly (Yankee-MPC)	CR, SR	Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9	
Reconfigured fuel assembly bottom end fitting assembly (Yankee-MPC)	CR, SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8	
				Creep	Change in dimensions	No		3.2.2.6
Reconfigured fuel assembly top nozzle bolt (Yankee-MPC)	CR, SR	Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9	
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8	
				Creep	Change in dimensions	No	3.2.2.6	
				Radiation embrittlement	Cracking	No	3.2.2.9	
Reconfigured fuel assembly basket corner angle, tie plate (Yankee-MPC)	CR, SR	Stainless steel (welded)	Helium	Stress relaxation	Loss of preload	No	3.2.2.10	
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8	
		Stainless steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7	

Table 4-15 NAC-MPC transportable storage canister								
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)	
Reconfigured fuel assembly fuel basket corner angle, tie plate (Yankee-MPC)	CR, SR	Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6	
				Radiation embrittlement	Cracking	No	3.2.2.9	
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8	
Reconfigured fuel assembly fuel tube, top and bottom caps (Yankee-MPC)	CR, SR	Stainless steel (welded)	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7	
				Creep	Change in dimensions	No	3.2.2.6	
				Radiation embrittlement	Cracking	No	3.2.2.9	
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8	
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7	
				Creep	Change in dimensions	No	3.2.2.6	
Reconfigured fuel assembly fuel tube, corner angle, support grid (CY-MPC)	SR	Stainless steel (welded)	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7	
				Creep	Change in dimensions	No	3.2.2.6	
				Radiation embrittlement	Cracking	No	3.2.2.9	

Table 4-15 NAC-MPC transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Reconfigured fuel assembly screens (CY-MPC)	CO	Stainless steel	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Reconfigured fuel assembly top and bottom housing, retaining plate and ring, guide plate, rod retaining plate, screen ring and housing (CY-MPC)	SR	Stainless steel	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-16 NAC-MPC vertical concrete cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Concrete shell	SH, SR*	Concrete	Air—outdoor	Aggressive chemical attack	Cracking	Reinforced Concrete Structures AMP	3.5.1.5
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.5
					Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.5
				Creep	Cracking	No	3.5.1.2
					Cracking	No	3.5.1.11
					Loss of strength	No	3.5.1.11
				Delayed ettringite formation	Loss of material (spalling, scaling)	No	3.5.1.13
					Loss of strength	No	3.5.1.13
					Cracking	No	3.5.1.13
				Fatigue	Cracking	No	3.5.1.10
					Cracking	No	3.5.1.10
				Freeze and thaw	Cracking	Reinforced Concrete Structures AMP	3.5.1.1
					Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.1
Radiation damage	Cracking	No	3.5.1.9				
	Loss of strength	No	3.5.1.9				

*Safety Functions: Confinement (CO), Subcriticality (CR), Retrievability (RE), Radiation Shielding (SH), Structural Integrity (SR), Thermal/Heat Removal (TH)

Table 4-16 NAC-MPC vertical concrete cask									
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)		
Concrete shell	SH, SR	Concrete	Air—outdoor	Reaction with aggregates	Cracking	Reinforced Concrete Structures AMP	3.5.1.3		
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.3		
					Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.14		
				Salt scaling	Cracking	No	3.5.1.7		
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.8		
				Shrinkage	Leaching of calcium hydroxide	Increase in porosity and permeability	Reinforced Concrete Structures AMP	3.5.1.8	
						Reduction of concrete pH (reducing corrosion resistance of steel embedments)	Reinforced Concrete Structures AMP	3.5.1.8	
						Loss of concrete/steel bond	Reinforced Concrete Structures AMP	3.5.1.6	
				Reinforcing steel	Air—outdoor, groundwater	Corrosion of reinforcing steel	Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.6
							Cracking	Reinforced Concrete Structures AMP	3.5.1.6
Loss of strength	Reinforced Concrete Structures AMP	3.5.1.6							

Table 4-16 NAC-MPC vertical concrete cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Inner shell	SR	Steel	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
Top flange, support ring	SR	Steel	Embedded (concrete)	Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9

Table 4-16 NAC-MPC vertical concrete cask								
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)	
Pedestal plate	SR	Steel	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1	
				Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3	
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2	
Pedestal cover	SR	Stainless steel	Sheltered	Microbiologically influenced corrosion	Loss of material	No	3.2.1.4	
				Radiation embrittlement	Cracking	No	3.2.1.9	
				Stress corrosion cracking	Cracking	No	3.2.2.5	
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2	
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4	
				Radiation embrittlement	Cracking	No	3.2.2.9	

Table 4-16 NAC-MPC vertical concrete cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Base plate assembly	SH, SR, TH	Steel	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
Base plate nelson studs	SR	Steel	Embedded (concrete)	Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	TLAA/AMP or a supporting analysis is required	3.2.1.1
				Pitting and crevice corrosion	Loss of material	TLAA/AMP or a supporting analysis is required	3.2.1.2
Outlet vent assembly	SH, SR, TH	Steel	Sheltered	Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1

Table 4-16 NAC-MPC vertical concrete cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Outlet vent assembly	SH, SR, TH	Steel	Sheltered	Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
Inlet vent supplemental shielding assembly	SH	Steel	Sheltered	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2

Table 4-16 NAC-MPC vertical concrete cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Inlet vent supplemental shielding assembly	SH	Steel	Sheltered	Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
Lid assembly	SH, SR, TH	Steel	Air—outdoor	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
			Sheltered	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4

Table 4-16 NAC-MPC vertical concrete cask								
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)	
Lid assembly	SH, SR, TH	Steel	Sheltered	Radiation embrittlement	Cracking	No	3.2.1.9	
		Concrete	Embedded (steel)	Delayed ettringite formation	Loss of material (spalling, scaling)	No	3.5.1.13	
					Cracking	No	3.5.1.13	
					Loss of strength	No	3.5.1.13	
Lid center support, nelson studs	SR	Steel	Embedded (concrete)	Radiation damage	Cracking	TLAA/AMP or a supporting analysis is required	3.5.1.9	
					Loss of strength	TLAA/AMP or a supporting analysis is required	3.5.1.9	
				Reaction with aggregates	Cracking	TLAA/AMP or a supporting analysis is required	3.5.1.3	
					Loss of strength	TLAA/AMP or a supporting analysis is required	3.5.1.3	
				General corrosion	Loss of material	TLAA/AMP or a supporting analysis is required	3.2.1.1	
					Pitting and crevice corrosion	Loss of material	TLAA/AMP or a supporting analysis is required	3.2.1.2
						Microbiologically influenced corrosion	Loss of material	No
					Radiation embrittlement	Cracking	No	3.2.1.9

Table 4-16 NAC-MPC vertical concrete cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Lid hardware	SR	Stainless steel	Air—outdoor	Stress corrosion cracking	Cracking	No	3.2.2.5
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10
Shield plug assembly	SH	Steel	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
Neutron shield (Shield plug)	SH	NS-4-FR	Embedded (steel)	Thermal aging	Loss of fracture toughness and loss of ductility	TAA/AMP or a supporting analysis is required	3.3.1.2

Table 4-16 NAC-MPC vertical concrete cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Neutron shield (Shield plug)	SH	NS-4-FR	Embedded (steel)	Radiation embrittlement	Cracking	TLAA/AMP or a supporting analysis is required	3.3.1.3
				Boron depletion	Loss of shielding	TLAA/AMP or a supporting analysis is required	3.3.1.1
			Thermal aging	Loss of fracture toughness and loss of ductility	No	3.3.1.2	
		NS-3	Embedded (steel)	Radiation embrittlement	Cracking	No	3.3.1.3
				Boron depletion	Loss of shielding	TLAA/AMP or a supporting analysis is required	3.3.1.1

Table 4-17 NAC-MPC transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Outer shell	SR*	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Galvanic corrosion	Loss of material	Transfer Casks AMP	3.2.1.3
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
Inner shell	SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Radiation embrittlement	Cracking	No	3.2.1.9
Gamma shield (Cask body)	SH	Lead	Embedded (steel, NS-4-FR)	None identified	None identified	No	3.2.6
				None identified	None identified	No	3.2.6
*Safety Functions: Confinement (CO), Subcriticality (CR), Retrievalability (RE), Radiation Shielding (SH), Structural Integrity (SR), Thermal/Heat Removal (TH)							

Table 4-17 NAC-MPC transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Neutron shield (Cask body)	SH	NS-4-FR	Embedded (steel, lead)	Thermal aging	Loss of fracture toughness and loss of ductility	TLAA/AMP or a supporting analysis is required	3.3.1.2
				Radiation embrittlement	Cracking	TLAA/AMP or a supporting analysis is required	3.3.1.3
				Boron depletion	Loss of shielding	TLAA/AMP or a supporting analysis is required	3.3.1.1
Top plate	SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
Bottom plate	SR	Steel	Air— indoor/outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9

Table 4-17 NAC-MPC transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Retaining ring	SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Galvanic corrosion	Loss of material	Transfer Casks AMP	3.2.1.3
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
Retaining ring bolts	SR	Stainless steel	Air— indoor/outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10
Trunnion	SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4

Table 4-17 NAC-MPC transfer cask								
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)	
Trunnion	SR	Steel	Air— indoor/outdoor	Radiation embrittlement	Cracking	No	3.2.1.9	
				Wear	Loss of material	Transfer Casks AMP	3.2.1.11	
Shield door plates	SH	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1	
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2	
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4	
				Radiation embrittlement	Cracking	No	3.2.1.9	
Shield door rails	SH, SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1	
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2	
				Galvanic corrosion	Loss of material	Transfer Casks AMP	3.2.1.3	
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4	
Shield door lock bolts	SR	Stainless steel	Air— indoor/outdoor	Radiation embrittlement	Cracking	No	3.2.1.9	
				Wear	Loss of material	Transfer Casks AMP	3.2.1.11	
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2	

Table 4-17 NAC-MPC transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Shield door lock bolts	SR	Stainless steel	Air— indoor/outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10

Table 4-18 MAGNASTOR transportable storage canister								
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)	
Shell	CO, SR*	Stainless steel (welded)	Sheltered	Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5	
		Stainless steel	Sheltered	Pitting and crevice corrosion	Loss of material (Precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2	
Bottom	CO, SR	Stainless steel (welded)	Sheltered	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4	
					Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
					Radiation embrittlement	Cracking	No	3.2.2.9
					Radiation embrittlement	Cracking	No	3.2.2.9
					Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5
					Pitting and crevice corrosion	Loss of material (Precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2

*Safety Functions: Confinement (CO), Subcriticality (CR), Retrievability (RE), Radiation Shielding (SH), Structural Integrity (SR), Thermal/Heat Removal (TH)

Table 4-18 MAGNASTOR transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Bottom	CO, SR	Stainless steel	Sheltered	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
				Radiation embrittlement	Cracking	No	3.2.2.9
Closure lid	SR	Stainless steel (welded)	Sheltered	Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5
				Pitting and crevice corrosion	Loss of material (Precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
			Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-18 MAGNASTOR transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Closure ring	SR	Stainless steel (welded)	Sheltered	Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5
		Stainless steel	Sheltered	Pitting and crevice corrosion	Loss of material (Precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2
Closure lid assembly bolt, washer	SR	Stainless steel	Helium	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Radiation embrittlement	Cracking	No	3.2.2.9
Stress relaxation	Loss of preload	No	3.2.2.10				

Table 4-18 MAGNASTOR transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Closure lid shield plate Port cover	SH, SR CO	Steel	Helium	General corrosion	Loss of material	No	3.2.1.1
				Galvanic corrosion	Loss of material	No	3.2.1.3
				Pitting and crevice corrosion	Loss of material	No	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5
				Pitting and crevice corrosion	Loss of material (Precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
Radiation embrittlement	Cracking	No	3.2.2.9				

Table 4-18 MAGNASTOR transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Port cover	CO	Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
Lifting lug, anti-rotation lug	SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
Fuel tube	CR, SR	Steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				General corrosion	Loss of material	No	3.2.1.1
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
Neutron absorber	CR	Borated aluminum	Helium	Creep	Change in dimensions	No	3.2.1.6
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	No	3.4.2.1
				Galvanic corrosion	Loss of material	No	3.4.2.2
				Thermal aging	Loss of strength	No	3.4.2.6
				Creep	Change in dimensions	No	3.4.2.5

Table 4-18 MAGNASTOR transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Neutron absorber	CR	Borated aluminum	Helium	Radiation embrittlement	Cracking	No	3.4.2.7
				Boron depletion	Loss of criticality control	No; a TLAA may be required	3.4.2.4
		Boral	Helium	General corrosion	Loss of material	No	3.4.2.1
				Galvanic corrosion	Loss of material	No	3.4.2.2
				Thermal aging	Loss of strength	No	3.4.2.6
				Wet corrosion and blistering	Change in dimensions	No	3.4.2.3
				Creep	Change in dimensions	No	3.4.2.5
				Radiation embrittlement	Cracking	No	3.4.2.7
				Boron depletion	Loss of criticality control	No; a TLAA may be required	3.4.2.4
				General corrosion	Loss of material	No	3.4.2.1
				Galvanic corrosion	Loss of material	No	3.4.2.2
				Thermal aging	Loss of strength	No	3.4.2.6
		Creep	Change in dimensions	No	3.4.2.5		
		Radiation embrittlement	Cracking	No	3.4.2.7		
Borated metal matrix composite	Helium	General corrosion	Loss of material	No	3.4.2.1		
		Galvanic corrosion	Loss of material	No	3.4.2.2		
		Thermal aging	Loss of strength	No	3.4.2.6		
		Creep	Change in dimensions	No	3.4.2.5		
		Radiation embrittlement	Cracking	No	3.4.2.7		

Table 4-18 MAGNASTOR transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Neutron absorber	CR	Borated metal matrix composite	Helium	Boron depletion	Loss of criticality control	No; a TLAA may be required	3.4.2.4
Plate in lieu of neutron absorber	TH	Aluminum	Helium	General corrosion	Loss of material	No	3.2.3.1
				Galvanic corrosion	Loss of material	No	3.2.3.3
				Thermal aging	Loss of strength	No	3.2.3.7
				Creep	Change in dimensions	No	3.2.3.5
				Radiation embrittlement	Cracking	No	3.2.3.8
Neutron absorber retainer	SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
Weld post	SR	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
		Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6

Table 4-18 MAGNASTOR transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Weld post	SR	Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
Fuel basket support plates & gussets, connector pin washer	SR	Steel	Helium	General corrosion	Loss of material	No	3.2.1.1
Neutron absorber retainer clip, fuel basket support tube, vent and drain tube restrictor plate, fuel basket pins, spacer	SR	Steel	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Creep	Change in dimensions	No	3.2.1.6
				Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	No	3.2.1.1
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Creep	Change in dimensions	No	3.2.1.6
				Radiation embrittlement	Cracking	No	3.2.1.9

Table 4-18 MAGNASTOR transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Fuel basket corner support bar, support pin, fuel tube pin, connector pin	SR	Steel	Helium	General corrosion	Loss of material	No	3.2.1.1
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Creep	Change in dimensions	No	3.2.1.6
				Radiation embrittlement	Cracking	No	3.2.1.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
Basket restraining block	SR	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
Basket support mounting bolt	SR	Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Basket support mounting bolt	SR	Stainless steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8

Table 4-18 MAGNASTOR transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Basket support mounting bolt	SR	Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10
Basket support washer, blocking strap	SR	Stainless steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Basket shim	SR	Stainless steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
Damaged fuel can tube body	CR, SR	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8

Table 4-18 MAGNASTOR transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Damaged fuel can tube body	CR, SR	Stainless steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Damaged fuel can bottom, side plates	CR, SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
Damaged fuel can lid plate, lid guide, lid bottom	CR, SR	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
Damaged fuel can lid plate, lid guide, lid bottom	CR, SR	Stainless steel	Helium	Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
Damaged fuel can lid plate, lid guide, lid bottom	CR, SR	Stainless steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-18 MAGNASTOR transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Damaged fuel can collar, lift tee, support ring, tid tab	SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
Damaged fuel can screens	CO	Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
Damaged fuel spacer plate	SR	Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-18 MAGNASTOR transportable storage canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Damaged fuel closure lid shield plate	SH	Stainless steel	Sheltered	Stress corrosion cracking	Cracking	No	3.2.2.5
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-19 MAGNASTOR concrete cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Concrete shell	SH, SR*	Concrete	Air—outdoor	Aggressive chemical attack	Cracking	Reinforced Concrete Structures AMP	3.5.1.5
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.5
					Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.5
				Creep	Cracking	No	3.5.1.2
					Cracking	No	3.5.1.11
					Loss of strength	No	3.5.1.11
				Delayed ettringite formation	Loss of material (spalling, scaling)	No	3.5.1.13
					Loss of strength	No	3.5.1.13
					Cracking	No	3.5.1.13
				Fatigue	Cracking	No	3.5.1.10
					Cracking	Reinforced Concrete Structures AMP	3.5.1.1
					Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.1
Radiation damage	Cracking	No	3.5.1.9				
	Loss of strength	No	3.5.1.9				

*Safety Functions: Confinement (CO), Subcriticality (CR), Retrievability (RE), Radiation Shielding (SH), Structural Integrity (SR), Thermal/Heat Removal (TH)

Table 4-19 MAGNASTOR concrete cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Concrete shell	SH, SR	Concrete	Air—outdoor	Reaction with aggregates	Cracking	Reinforced Concrete Structures AMP	3.5.1.3
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.3
					Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.14
					Cracking	No	3.5.1.7
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.8
					Increase in porosity and permeability	Reinforced Concrete Structures AMP	3.5.1.8
					Reduction of concrete pH (reducing corrosion resistance of steel embedments)	Reinforced Concrete Structures AMP	3.5.1.8
					Loss of concrete/steel bond	Reinforced Concrete Structures AMP	3.5.1.6
					Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.6
					Cracking	Reinforced Concrete Structures AMP	3.5.1.6
Inner shell	SR, SH, TH	Steel	Sheltered	General corrosion	Loss of strength	Reinforced Concrete Structures AMP	3.5.1.6
					Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1

Table 4-19 MAGNASTOR concrete cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Inner shell	SR, SH, TH	Steel	Sheltered	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Radiation embrittlement	Cracking	No	3.2.1.9
Top flange	SR	Steel	Air—outdoor	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
			Sheltered	Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9

Table 4-19 MAGNASTOR concrete cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Pedestal plate	SR	Steel	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
Pedestal cover	SR	Stainless steel	Sheltered	Radiation embrittlement	Cracking	No	3.2.1.9
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
Base plate assembly (including nelson studs)	SH, SR, TH	Steel	Air—outdoor	Radiation embrittlement	Cracking	No	3.2.2.9
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4

Table 4-19 MAGNASTOR concrete cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Base plate assembly (including nelson studs)	SH, SR, TH	Steel	Air—outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
			Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
Channels	SR	Steel		Radiation embrittlement	Cracking	No	3.2.1.9
			Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
Lid assembly	SH, SR, TH	Steel	Air—outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3

Table 4-19 MAGNASTOR concrete cask									
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)		
Lid assembly	SH, SR, TH	Steel	Air—outdoor	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2		
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4		
				Radiation embrittlement	Cracking	No	3.2.1.9		
			Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1		
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2		
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4		
		Concrete	Embedded (steel)	Concrete	Radiation embrittlement	Cracking	No	3.2.1.9	
					Delayed ettringite formation	Loss of material (spalling, scaling)	No	3.5.1.13	
						Cracking	No	3.5.1.13	
				Concrete	Embedded (steel)		Loss of strength	No	3.5.1.13
							Cracking	TLAA/AMP or a supporting analysis is required	3.5.1.9
							Loss of strength	TLAA/AMP or a supporting analysis is required	3.5.1.9
			Reaction with aggregates	Cracking	TLAA/AMP or a supporting analysis is required	3.5.1.3			

Table 4-19 MAGNASTOR concrete cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Lid assembly	SH, SR, TH	Concrete	Embedded (steel)	Reaction with aggregates	Loss of strength	TAA/AMP or a supporting analysis is required	3.5.1.3
Lid hardware	SR	Stainless steel	Air—outdoor	Stress corrosion cracking	Cracking	No	3.2.2.5
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10
Lift anchor (standard configuration)	SR	Steel	Air—outdoor	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
Lift anchor (standard and alternative configurations)	SR	Steel	Embedded (concrete)	General corrosion	Loss of material	TAA/AMP or a supporting analysis is required	3.2.1.1
				Pitting and crevice corrosion	Loss of material	TAA/AMP or a supporting analysis is required	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4

Table 4-19 MAGNASTOR concrete cask									
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)		
Lift anchor (standard and alternative configurations)	SR	Steel	Embedded (concrete)	Radiation embrittlement	Cracking	No	3.2.1.9		
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1		
Lift lug	SR	Steel	Sheltered	Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2		
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4		
				Radiation embrittlement	Cracking	No	3.2.1.9		
				Stress corrosion cracking	Cracking	No	3.2.4.4		
Lift lug bolt	SR	Nickel alloy	Sheltered	Pitting and crevice corrosion	Loss of material	No	3.2.4.2		
				Microbiologically influenced corrosion	Loss of material	No	3.2.4.3		
				Radiation embrittlement	Cracking	No	3.2.4.6		
				Stress relaxation	Loss of preload	No	3.2.4.7		
				Stress corrosion cracking	Cracking	No	3.2.2.5		
Lift lug washer, base plate dowel pin	SR	Stainless steel	Sheltered	Pitting and crevice corrosion	Loss of material	No	3.2.2.2		
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4		

Table 4-19 MAGNASTOR concrete cask									
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)		
Lift lug washer, base plate dowel pin	SR	Stainless steel	Sheltered	Radiation embrittlement	Cracking	No	3.2.2.9		
Cover plate	SR	Steel	Air—outdoor	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1		
				Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3		
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2		
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4		
Cover plate hardware	SR	Stainless steel	Air—outdoor	Radiation embrittlement	Cracking	No	3.2.1.9		
				Stress-corrosion cracking	Cracking	No	3.2.2.5		
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2		
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4		
				Radiation embrittlement	Cracking	No	3.2.2.9		
				Stress relaxation	Loss of preload	No	3.2.2.10		

Table 4-20 MAGNASTOR transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Outer shell	SR*	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Radiation embrittlement	Cracking	No	3.2.1.9
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Radiation embrittlement	Cracking	No	3.2.2.9
				Radiation embrittlement	Cracking	No	3.2.2.9
Inner shell	SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2

*Safety Functions: Confinement (CO), Subcriticality (CR), Retrievability (RE), Radiation Shielding (SH), Structural Integrity (SR), Thermal/Heat Removal (TH)

Table 4-20 MAGNASTOR transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Inner shell	SR	Steel	Air— indoor/outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
			Embedded (lead)	Radiation embrittlement	Cracking	No	3.2.1.9
		Stainless steel (welded)	Air— indoor/outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
		Stainless steel	Air— indoor/outdoor	Stress corrosion cracking	Cracking	No	3.2.2.5
Gamma shield (Cask body)	SH	Lead	Air— indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
			Embedded (lead)	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
			Embedded (steel, NS-4-FR)	Radiation embrittlement	Cracking	No	3.2.2.9
		NS-4-FR	Embedded (stainless steel, NS-4-FR)	Radiation embrittlement	Cracking	No	3.2.2.9
			Embedded (steel, NS-4-FR)	None identified	None identified	No	3.2.6
			Embedded (stainless steel, NS-4-FR)	None identified	None identified	No	3.2.6
Neutron shield (Cask body)	SH	NS-4-FR	Embedded (steel, lead)	Thermal aging	Loss of fracture toughness and loss of ductility	TLAA/AMP or a supporting analysis is required	3.3.1.2
			Embedded (steel, lead)	Radiation embrittlement	Cracking	TLAA/AMP or a supporting analysis is required	3.3.1.3

Table 4-20 MAGNASTOR transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Neutron shield (Cask body)	SH	NS-4-FR	Embedded (steel, lead)	Boron depletion	Loss of shielding	TLAA/AMP or a supporting analysis is required	3.3.1.1
				Thermal aging	Loss of fracture toughness and loss of ductility	TLAA/AMP or a supporting analysis is required	3.3.1.2
				Radiation embrittlement	Cracking	TLAA/AMP or a supporting analysis is required	3.3.1.3
Top ring	SR	Steel	Air—indoor/outdoor	Boron depletion	Loss of shielding	TLAA/AMP or a supporting analysis is required	3.3.1.1
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-20 MAGNASTOR transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Bottom ring	SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
Trunnion	SR	Stainless steel	Air— indoor/outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5

Table 4-20 MAGNASTOR transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Trunnion	SR	Stainless steel	Air— indoor/outdoor	Radiation embrittlement	Cracking	No	3.2.2.9
Trunnion bushing, rotating bushing	SR	Stainless steel	Air— indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Wear	Loss of material	Transfer Casks AMP	3.2.2.11
Shield door plates	SH	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
		Stainless steel	Air— indoor/outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-20 MAGNASTOR transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Shield door rails	SH, SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
				Wear	Loss of material	Transfer Casks AMP	3.2.1.11
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Wear	Loss of material	Transfer Casks AMP	3.2.2.11
Shield door tab	SR	Steel	Air— indoor/outdoor	General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9

Table 4-20 MAGNASTOR transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Shield door tab	SR	Stainless steel	Air— indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
Shield door lock pin	SR	Stainless steel	Air— indoor/outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
Retaining block and ring	SR	Stainless steel	Air— indoor/outdoor	Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
Retaining pin	SR	Steel	Air— indoor/outdoor	Radiation embrittlement	Cracking	No	3.2.2.9
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4

Table 4-20 MAGNASTOR transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Retaining pin	SR	Steel	Air—indoor/outdoor	Radiation embrittlement	Cracking	No	3.2.1.9
		Stainless steel	Air—indoor/outdoor	Pitting and crevice corrosion Microbiologically influenced corrosion Stress corrosion cracking Radiation embrittlement	Loss of material Loss of material Cracking Cracking	No No No No	3.2.2.2 3.2.2.4 3.2.2.5 3.2.2.9
Retainer assembly spring plunger, guide bolt, handle bolt	SR	Stainless steel	Air—indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
Retainer assembly handle	SR	Stainless steel	Air—indoor/outdoor	Stress relaxation	Loss of preload	No	3.2.2.10
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-20 MAGNASTOR transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Retaining ring bolt and screw thread	SR	Stainless steel	Air—indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
				Wear	Loss of material	Transfer Casks AMP	3.2.2.11
Wear strip (cask attachment)	SR	Stainless steel	Air—indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9
Fill/drain assembly tube and bar	SR	Stainless steel	Air—indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress-corrosion cracking	Cracking	No	3.2.2.5
				Radiation embrittlement	Cracking	No	3.2.2.9

1 **4.6 FuelSolutions™ storage system**

2 **4.6.1 System description**

3 The FuelSolutions™ storage system uses a stainless steel storage canister stored within a
4 vertical cylindrical concrete storage cask. The principal components of the storage system are
5 the W21 and W74 canisters, the W150 concrete storage cask, and the W100 transfer cask. The
6 W21 canister is designed to accommodate nearly all domestic commercial spent nuclear fuel
7 with a capacity of up to 21 PWR fuel assemblies. The W74 canister is designed to
8 accommodate the three assembly types used at the Big Rock Point Nuclear Plant, including
9 mixed oxide, partial, and damaged fuel assemblies, with a capacity of up to 64 BWR fuel
10 assemblies. The W150 concrete storage cask provides radiation shielding and contains internal
11 air flowpaths that allow decay heat from the canister spent fuel contents to be removed by
12 natural air circulation around the canister wall. The W100 TC is used to move the loaded
13 canisters to and from the storage cask. The sections below summarize the components of the
14 FuelSolutions™ storage system.

15 **4.6.2 W21 and W74 canisters**

16 The W21 and W74 canisters, shown in Figure 4-27, each have several design configurations
17 consisting of different materials of construction and dimensions. A typical W21 or W74 canister
18 consists of a shell assembly, top and bottom inner closure plates, vent and drain port covers,
19 internal basket assembly, top and bottom shield plugs, and top and bottom outer closure plates.
20 All structural components of the canister are constructed of high-strength carbon or stainless
21 steel. Any carbon steel used in the canister is coated with electroless nickel for corrosion
22 protection. The canister shell, top and bottom inner closure plates, and vent and drain port
23 covers form the confinement boundary and are fabricated from stainless steel.

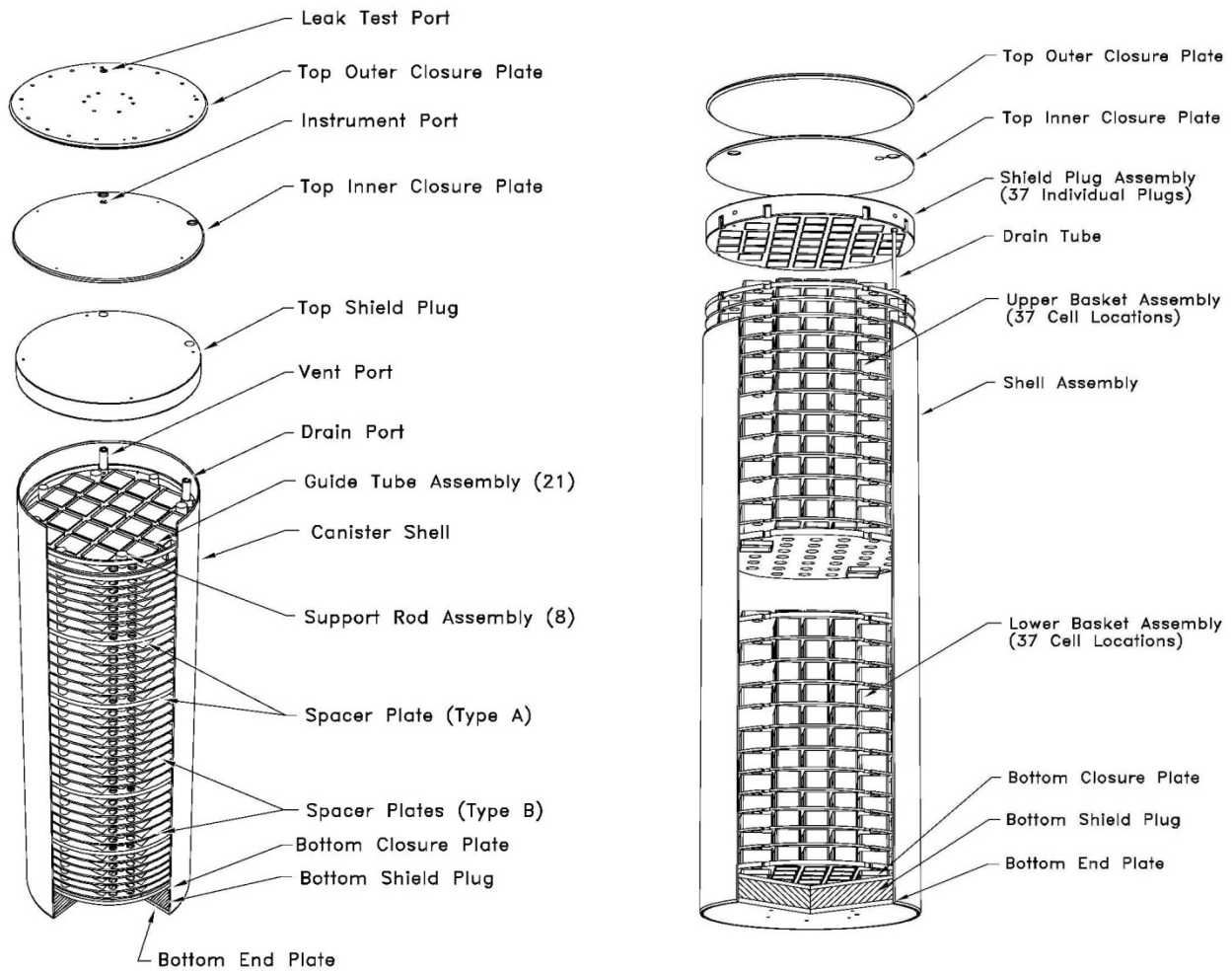
24 W21 basket assembly

25 The W21 PWR fuel basket assembly is a right circular cylinder configuration with 21 stainless
26 steel guide tubes for PWR contents (FuelSolutions, 2007a). The guide tubes are laterally
27 supported by a series of spacer plates, held in position by support rods that run through support
28 rod sleeves placed between the spacer plates. The square guide tubes include Boral® neutron
29 poison sheets on all four sides for criticality control. There are two classes of canister for the
30 W21 canister based on different materials of construction: W21M and W21T. Each class of
31 canister has four different types, which differ in dimension (exterior canister length and internal
32 cavity length) and the material used for end plug shielding (steel, lead, or depleted uranium
33 (W21M only)).

34 W74 basket assembly

35 The W74 BWR fuel basket assembly consists of two right circular cylindrical baskets, with a
36 total of 56 guide tubes and a capacity of up to 64 assemblies (FuelSolutions, 2007b). The guide
37 tubes are supported by a series of spacer plates held in position by support rods that run
38 through support rod sleeves placed between the spacer plates. The square guide tubes include
39 neutron poison sheets made of borated stainless steel, either on one side or on two opposite
40 sides, in an arrangement within the basket that assures that there is a poison sheet between all
41 of the assemblies. There are two classes of canister for the W74 canister based on different
42 materials of construction (W74M and W74T). Unlike the W21 design, each W74 canister class

- 1 has only one canister length and one cavity size, and carbon steel is used for end plug shielding
- 2 material.
- 3 Table 4-21 provides a generic evaluation of potential aging mechanisms and effects requiring
- 4 management for specific components of the W21 and W74 canisters. The table also identifies
- 5 AMPs that provide an acceptable approach to managing the aging effects.



(a) (b)
Figure 4-27 Typical FuelSolutions™ (a) W21 and (b) W74 canisters (FuelSolutions, 2007a,b)

6 4.6.3 W150 storage cask

7 The W150 storage cask, shown in Figure 4-28, is the overpack for storing both the long and
 8 short versions of the W21 and W74 canisters by varying the length of the middle concrete
 9 segment (FuelSolutions, 2007c). The overpack consists of a standard reinforced concrete
 10 structure with three precast segments (top, middle, and bottom) and a top cover made of steel
 11 and concrete. Stainless steel tie rods are used to tie the concrete segments together. A shear
 12 key between each two concrete segments provides positive lateral engagement and alignment
 13 and serves to minimize radiation streaming. Grout is installed between the keyed joints of the

1 concrete segments to provide a weather barrier. The exterior surfaces of the concrete are
2 exposed to the outdoor environment.

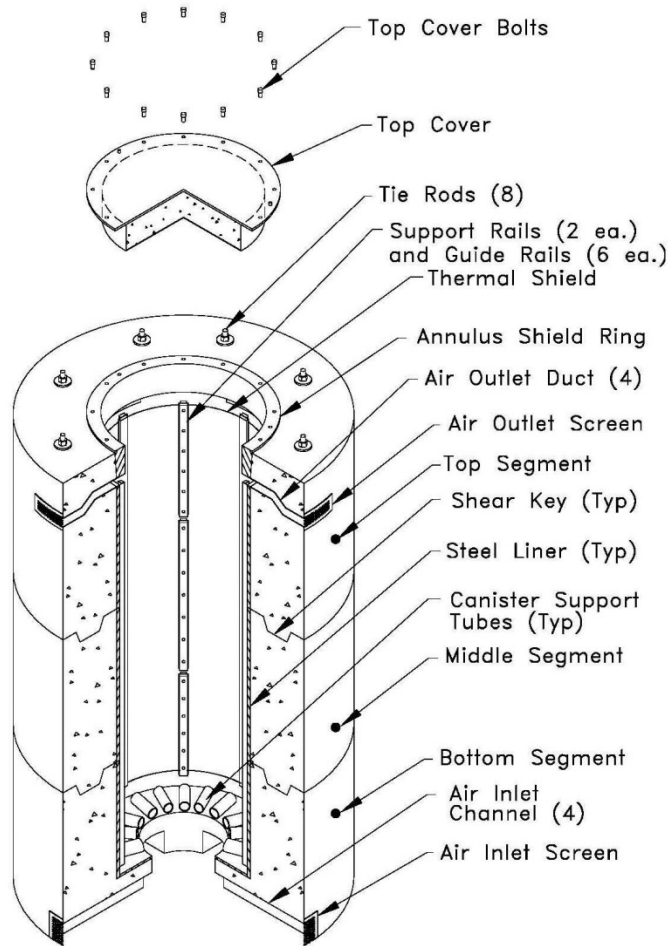


Figure 4-28 FuelSolutions™ W150 storage cask (FuelSolutions, 2007c)

3 The top cover of the overpack is bolted to the overpack top end segment shielding ring and is
4 sealed with weather sealant. Inside the cavity of the overpack are a steel liner, an aluminum
5 thermal shield, steel support and guide rails, and stainless steel canister support tubes. Guide
6 rails are welded to the steel liner for centering the canister radially in the cavity. Canister
7 support tubes are welded to the bottom plate of the steel liner plate to provide vertical support of
8 the canister and to limit the g-load on the canister in a postulated accident. All steel
9 components, such as the liner, top cover, and support and guide rails, are coated with
10 temperature- and radiation-resistant coatings.

11 The overpack concrete bottom segment includes four inlet vents that converge into a single
12 cylindrical inlet duct at the bottom center of the cask cavity. The center inlet duct also provides
13 hydraulic ram access during horizontal canister transfer operations. The inlet and outlet vents
14 have protective screens to prevent debris or wildlife from entering the ventilation ducts.

15 Table 4-22 provides a generic evaluation of potential aging mechanisms and effects requiring
16 management for specific components of the W150 storage cask. The table also identifies AMPs
17 that provide an acceptable approach to managing the aging effects.

1 **4.6.4 W100 transfer cask**

2 The W100 TC, shown in Figure 4-29, is a multiwall, stainless steel cylindrical vessel with covers
3 on both ends (FuelSolutions, 2007c). The TC is composed of a structural shell and a stainless
4 steel inner liner, with lead in the annular space to provide gamma shielding. The TC also
5 includes an outer stainless steel jacket filled with demineralized water for neutron shielding.
6 The penetrations in the neutron shield cavity consist of two quick-connect fittings that are used
7 to drain and fill the neutron shield cavity and to prevent intrusion of contaminated spent fuel pool
8 water. A pressure relief device is used to provide over-pressure protection for the neutron
9 shield.

10 The structural shell and inner liner are welded to stainless steel flanges at the top and bottom
11 ends. Both the top and bottom covers are made of stainless steel plates and an encased solid
12 neutron shielding of RX-277 or BISCO NS-3. The top cover includes a secondary central cover
13 for ram access during horizontal loading and unloading operations. The bottom cover has
14 O-rings to prevent spent fuel pool water from entering the cask during loading operations.
15 Nitronic 60 stainless steel guide rails are welded to the inner shell cavity to facilitate horizontal
16 canister transfer.

17 The W100 TC has four stainless steel trunnions. Two upper lifting trunnions located near the
18 top of the cask for vertical cask handling operations are welded to the structural shell and inner
19 liner. The lower trunnions used for upending and downending the TC are welded to the
20 structural shell. Heat removal from the TC is primarily by conduction through the cask wall. A
21 high emissivity, low absorptivity coating is applied to the exterior of the liquid neutron shield
22 jacket to facilitate radiative heat transfer to the environment.

23 Table 4-23 provides a generic evaluation of potential aging mechanisms and effects requiring
24 management for specific components of the W100 TC. The table also identifies AMPs that
25 provide an acceptable approach to managing the aging effects.

26

27

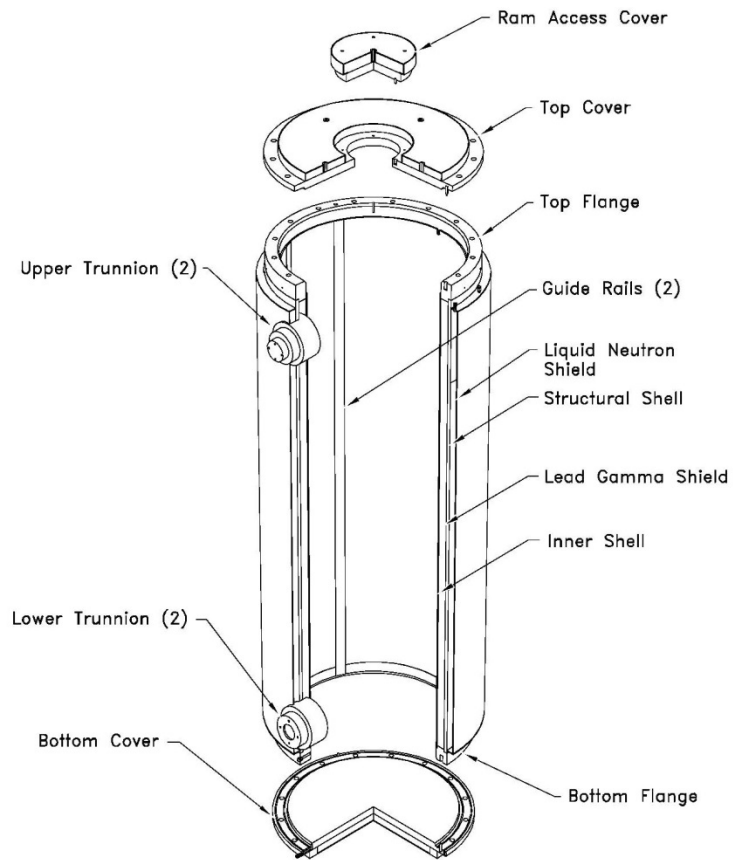


Figure 4-29 FuelSolutions™ W100 transfer cask (FuelSolutions, 2007c)

1

Table 4-21 FuelSolutions™ canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Shell	CO, SR*	Stainless steel (welded)	Sheltered	Stress-corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5
		Stainless steel	Sheltered	Pitting and crevice corrosion	Loss of material (Precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2
Bottom closure plate	CO, SR	Stainless steel (welded)	Helium	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
				Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
Creep	Change in dimensions	No	3.2.2.6				

*Safety Functions: Confinement (CO), Subcriticality (CR), Retrievability (RE), Radiation Shielding (SH), Structural Integrity (SR), Thermal/Heat Removal (TH)

Table 4-21 FuelSolutions™ canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Bottom closure plate	CO, SR	Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
			Embedded (steel, depleted uranium)	Radiation embrittlement	Cracking	No	3.2.2.9
Bottom end plate	SR	Stainless steel (welded)	Sheltered	Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5
			Sheltered	Pitting and crevice corrosion	Loss of material (Precursor to stress-corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2
Top outer closure plate	CO, SR*	Stainless steel (welded)	Embedded (steel, depleted uranium)	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Radiation embrittlement	Cracking	No	3.2.2.9
			Sheltered	Radiation embrittlement	Cracking	No	3.2.2.9
Top outer closure plate	CO, SR*	Stainless steel (welded)	Sheltered	Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5

Table 4-21 FuelSolutions™ canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Top outer closure plate	CO, SR	Stainless steel	Sheltered	Pitting and crevice corrosion	Loss of material (Precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
Top inner closure plate	CO, SR	Stainless steel (welded)	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
Alignment bar, adapter	SR	Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-21 FuelSolutions™ canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Shield plug	SH	Steel	Helium	General corrosion	Loss of material	No	3.2.1.1
				Galvanic corrosion	Loss of material	No	3.2.1.3
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8
				Creep	Change in dimensions	No	3.2.1.6
				Radiation embrittlement	Cracking	No	3.2.1.9
				Radiation embrittlement	Cracking	No	3.2.1.9
				Radiation embrittlement	Cracking	No	3.2.1.9
				None identified	None identified	No	3.2.6
				None identified	None identified	No	3.2.7
				None identified	None identified	No	3.2.7
Shield plug support assembly	SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
				None identified	None identified	No	

Table 4-21 FuelSolutions™ canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Leak test port cover	CO	Stainless steel (welded)	Sheltered	Stress corrosion cracking	Cracking	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.5
		Stainless steel	Sheltered	Pitting and crevice corrosion	Loss of material (Precursor to stress corrosion cracking)	Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP	3.2.2.2
Instrument port cover, vent/drain port cover	CO	Stainless steel (welded)	Helium	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
				Radiation embrittlement	Cracking	No	3.2.2.9
		Stainless steel	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Vent and drain port	SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Creep	Change in dimensions	No	3.2.2.6

Table 4-21 FuelSolutions™ canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Vent and drain port	SR	Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
Guide tube assembly	CR, SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
		Stainless steel	Helium	Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
Neutron absorber	CR	Boral	Helium	Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
				General corrosion	Loss of material	No	3.4.2.1
				Galvanic corrosion	Loss of material	No	3.4.2.2
				Thermal aging	Loss of strength	No	3.4.2.6
				Wet corrosion and blistering	Change in dimensions	No	3.4.2.3
				Creep	Change in dimensions	No	3.4.2.5
				Radiation embrittlement	Cracking	No	3.4.2.7
				Boron depletion	Loss of criticality control	No; a TLAA may be required	3.4.2.4
				Boron depletion	Loss of criticality control	No; a TLAA may be required	3.4.1.1
Borated stainless steel	Helium	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.4.1.3	
			Creep	Change in dimensions	No	3.4.1.2	
				Radiation embrittlement	Loss of fracture toughness and loss of ductility	No	3.4.1.4

Table 4-21 FuelSolutions™ canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Fuel basket support rod, support sleeve	SR	Stainless steel	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Fuel basket support rod	SR	Stainless steel (17-4 PH)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	TLAA/AMP or a supporting analysis is required	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Fuel basket support sleeve	SR	Steel	Helium	General corrosion	Loss of material	No	3.2.1.1
				Galvanic corrosion	Loss of material	No	3.2.1.3
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7
				Creep	Change in dimensions	No	3.2.1.6
				Radiation embrittlement	Cracking	No	3.2.1.9

Table 4-21 FuelSolutions™ canister											
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)				
Fuel basket bolt	SR	Steel	Helium	General corrosion	Loss of material	No	3.2.1.1				
				Galvanic corrosion	Loss of material	No	3.2.1.3				
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8				
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7				
				Creep	Change in dimensions	No	3.2.1.6				
				Radiation embrittlement	Cracking	No	3.2.1.9				
				Stress relaxation	Loss of preload	TLAA/AMP or a supporting analysis is required	3.2.1.10				
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8				
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7				
				Creep	Change in dimensions	No	3.2.2.6				
Fuel basket spacer assembly	SR	Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9				
				General corrosion	Loss of material	No	3.2.1.1				
				Galvanic corrosion	Loss of material	No	3.2.1.3				
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8				
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7				
				Fuel basket spacer assembly	SR	Steel	Helium	General corrosion	Loss of material	No	3.2.1.1
								Galvanic corrosion	Loss of material	No	3.2.1.3
								Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.1.8
								Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.1.7

Table 4-21 FuelSolutions™ canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Fuel basket spacer assembly	SR	Steel	Helium	Creep	Change in dimensions	No	3.2.1.6
				Radiation embrittlement	Cracking	No	3.2.1.9
Damaged fuel can top lid assembly (W74 Canister)	SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
Damaged fuel can top lid assembly hardware (W74 Canister)	SR	Stainless steel	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10
Damaged fuel can guide tube assembly (W74 Canister)	SR	Stainless steel (welded)	Helium	Thermal aging	Loss of fracture toughness and loss of ductility	No	3.2.2.8
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Creep	Change in dimensions	No	3.2.2.6
		Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
				Creep	Change in dimensions	No	3.2.2.6

Table 4-21 FuelSolutions™ canister							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Damaged fuel can guide tube assembly (W74 Canister)	SR	Stainless steel	Helium	Radiation embrittlement	Cracking	No	3.2.2.9
Damaged fuel can neutron absorber (W74 Canister)	CR	Borated stainless steel	Helium	Boron depletion	Loss of criticality control	No; a TLAA may be required	3.4.1.1
				Thermal aging	Loss of fracture toughness and loss of ductility	No	3.4.1.3
				Creep	Change in dimensions	No	3.4.1.2
				Radiation embrittlement	Loss of fracture toughness and loss of ductility	No	3.4.1.4

Table 4-22 FuelSolutions™ storage cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Concrete shell, shear key	SH, SR*	Concrete, grout	Air—outdoor	Aggressive chemical attack	Cracking	Reinforced Concrete Structures AMP	3.5.1.5
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.5
				Creep	Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.5
					Cracking	No	3.5.1.2
				Dehydration at high temperature	Cracking	No	3.5.1.11
					Loss of strength	No	3.5.1.11
				Delayed ettringite formation	Loss of material (spalling, scaling)	No	3.5.1.13
					Loss of strength	No	3.5.1.13
				Fatigue	Cracking	No	3.5.1.13
					Cracking	No	3.5.1.10
Freeze and thaw	Cracking	Reinforced Concrete Structures AMP	3.5.1.1				
	Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.1				
Radiation damage	Cracking	TLAA/AMP or a supporting analysis is required	3.5.1.9				
	Loss of strength	TLAA/AMP or a supporting analysis is required	3.5.1.9				

*Safety Functions: Confinement (CO), Subcriticality (CR), Retrievability (RE), Radiation Shielding (SH), Structural Integrity (SR), Thermal/Heat Removal (TH)

Table 4-22 FuelSolutions™ storage cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Concrete shell, shear key	SH, SR	Concrete	Air—outdoor	Reaction with aggregates	Cracking	Reinforced Concrete Structures AMP	3.5.1.3
				Salt scaling	Loss of strength	Reinforced Concrete Structures AMP	3.5.1.3
				Shrinkage	Cracking	No	3.5.1.7
				Leaching of calcium hydroxide	Loss of strength	Reinforced Concrete Structures AMP	3.5.1.8
Concrete shell	SH, SR	Reinforcing steel	Air—outdoor, groundwater	Increase in porosity and permeability	Loss of strength	Reinforced Concrete Structures AMP	3.5.1.8
				Reduction of concrete pH (reducing corrosion resistance of steel embedments)	Cracking	Reinforced Concrete Structures AMP	3.5.1.8
				Corrosion of reinforcing steel	Loss of concrete/steel bond	Reinforced Concrete Structures AMP	3.5.1.6
				Loss of material (spalling, scaling)	Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.6
Concrete shell	SH, SR	Reinforcing steel	Air—outdoor, groundwater	Cracking	Loss of strength	Reinforced Concrete Structures AMP	3.5.1.6
				Loss of strength	Loss of strength	Reinforced Concrete Structures AMP	3.5.1.6

Table 4-22 FuelSolutions™ storage cask											
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)				
Steel liner, shield ring	SH, SR	Steel	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1				
				Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3				
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2				
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4				
				Radiation embrittlement	Cracking	No	3.2.1.9				
				Radiation embrittlement	Cracking	No	3.2.1.9				
				Thermal shield	TH	Aluminum	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.3.1
								Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.3.3
								Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.3.2
								Microbiologically influenced corrosion	Loss of material	No	3.2.3.4
								Thermal aging	Loss of strength	No	3.2.3.7
								Creep	Change in dimensions	No	3.2.3.5
								Radiation embrittlement	Cracking	No	3.2.3.8

Table 4-22 FuelSolutions™ storage cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Shear lug, thermal shield support lug	SR	Steel	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
Support rail, guide rail	SR	Stainless steel	Sheltered	Stress corrosion cracking	Cracking	No	3.2.2.5
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Radiation embrittlement	Cracking	No	3.2.2.9
Canister support tube	SR	Stainless steel	Sheltered	Wear	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.2.11
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Radiation embrittlement	Cracking	No	3.2.2.9
				Wear	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.2.11

Table 4-22 FuelSolutions™ storage cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Tie rod, tie rod plate	SR	Stainless steel	Sheltered	Stress corrosion cracking	Cracking	No	3.2.2.5
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10
				General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
Tie rod hardware	SR	Steel	Sheltered	Galvanic corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.3
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.4
				Loss of material	Loss of material	No	

Table 4-22 FuelSolutions™ storage cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Tie rod hardware	SR	Steel	Sheltered	Radiation embrittlement	Cracking	No	3.2.1.9
				Stress relaxation	Loss of preload	External Surfaces Monitoring of Metallic Components AMP	3.2.1.10
Ram anchor	SR	Steel	Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
Top cover assembly	SR	Steel	Air—outdoor	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Radiation embrittlement	Cracking	No	3.2.1.9
			Sheltered	General corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.2

Table 4-22 FuelSolutions™ storage cask									
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)		
Top cover assembly	SR	Steel	Sheltered	Microbiologically influenced corrosion	Loss of material	No	3.2.1.4		
				Radiation embrittlement	Cracking	No	3.2.1.9		
		Concrete	Embedded (steel)	Delayed ettringite formation	Loss of material (spalling, scaling)	No	3.5.1.13		
					Cracking	No	3.5.1.13		
				Radiation damage	Loss of strength	No	3.5.1.13		
					Cracking	TLAA/AMP or a supporting analysis is required	3.5.1.9		
				Reaction with aggregates	Loss of strength	TLAA/AMP or a supporting analysis is required	3.5.1.9		
					Cracking	TLAA/AMP or a supporting analysis is required	3.5.1.3		
		Top cover bolt	SR	Steel	Sheltered	General corrosion	Loss of strength	TLAA/AMP or a supporting analysis is required	3.5.1.3
							Loss of material	External Surfaces Monitoring of Metallic Components AMP	3.2.1.1
Pitting and crevice corrosion	Loss of material					External Surfaces Monitoring of Metallic Components AMP	3.2.1.2		
	Microbiologically influenced corrosion					Loss of material	No	3.2.1.4	
				Radiation embrittlement	Cracking	No	3.2.1.9		

Table 4-22 FuelSolutions™ storage cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Top cover bolt	SR	Steel	Sheltered	Stress relaxation	Loss of preload	External Surfaces Monitoring of Metallic Components AMP	3.2.1.10
Coating on carbon steel components	SR	Coating	Air—outdoor	Radiation embrittlement	Coating degradation	TLAA/AMP or a supporting analysis is required	3.2.8
				Thermal aging	Coating degradation	TLAA/AMP or a supporting analysis is required	3.2.8
			Sheltered	Radiation embrittlement	Coating degradation	TLAA/AMP or a supporting analysis is required	3.2.8
				Thermal aging	Coating degradation	TLAA/AMP or a supporting analysis is required	3.2.8

Table 4-23 FuelSolutions™ transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Structural shell	SR*	Stainless steel (welded)	Air—indoor/outdoor	Stress corrosion cracking	Cracking	No	3.2.2.5
		Stainless steel	Air—indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
Inner liner	SR	Stainless steel (welded)	Air—indoor/outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
		Stainless steel	Air—indoor/outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Stress corrosion cracking	Cracking	No	3.2.2.5
Neutron shield jacket, trunnion support plate, thermowell	SR	Stainless steel	Air—indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2

*Safety Functions: Confinement (CO), Subcriticality (CR), Retrievalability (RE), Radiation Shielding (SH), Structural Integrity (SR), Thermal/Heat Removal (TH)

Table 4-23 FuelSolutions™ transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Neutron shield jacket, trunnion support plate, thermowell	SR	Stainless steel	Air—indoor/outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
Neutron shield jacket support rib	SR	Stainless steel	DeminerIALIZED water	Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
				None identified	None identified	No	3.2.6
Gamma shield	SH	Lead	Embedded (stainless steel)				
Guide rail	SR	Stainless steel	Air—indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
				Wear	Loss of material	Transfer Casks AMP	3.2.2.11

Table 4-23 FuelSolutions™ transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Top flange, bottom flange	SR	Stainless steel (welded)	Air—indoor/outdoor	Stress corrosion cracking	Cracking	No	3.2.2.5
		Stainless steel	Air—indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
Screw thread insert	SR	Stainless steel	Embedded (stainless steel)	Radiation embrittlement	Cracking	No	3.2.2.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
Block	SR	Stainless steel	Embedded (stainless steel, lead)	Radiation embrittlement	Cracking	No	3.2.2.9
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
Swagelok quick connect body, coupling, fitting, cap	SR	Stainless steel	Air—indoor/outdoor	Radiation embrittlement	Cracking	No	3.2.2.9
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-23 FuelSolutions™ transfer cask									
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)		
Upper trunnion, lower trunnion	SR	Stainless steel (welded)	Air—indoor/outdoor	Stress corrosion cracking	Cracking	No	3.2.2.5		
			Demineralized water	Stress corrosion cracking	Cracking	No	3.2.2.5		
		Stainless steel		Air—indoor/outdoor	Pitting and crevice corrosion	Loss of material	No		3.2.2.2
					Microbiologically influenced corrosion	Loss of material	No		3.2.2.4
					Fatigue	Cracking	Evaluate design code TLAA, if applicable		3.2.2.7
					Radiation embrittlement	Cracking	No		3.2.2.9
					Pitting and crevice corrosion	Loss of material	No		3.2.2.2
					Microbiologically influenced corrosion	Loss of material	No		3.2.2.4
					Fatigue	Cracking	Evaluate design code TLAA, if applicable		3.2.2.7
					Radiation embrittlement	Cracking	No		3.2.2.9
Trunnion retainer, trunnion sleeve	SR	Stainless steel	Air—indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2		
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4		
				Stress corrosion cracking	Cracking	No	3.2.2.5		
				Fatigue	Cracking	Evaluate design code TLAA, if applicable		3.2.2.7	
				Radiation embrittlement	Cracking	No		3.2.2.9	
				Wear	Loss of material	Transfer Casks AMP		3.2.2.11	

Table 4-23 FuelSolutions™ transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Trunnion retainer, trunnion sleeve	SR	Stainless steel	Demineralized water	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
				Wear	Loss of material	Transfer Casks AMP	3.2.2.11
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
Bolt for top cover, bottom cover, ram access cover	SR	Steel	Air—indoor/outdoor	Radiation embrittlement	Cracking	No	3.2.2.9
				Stress relaxation	Loss of preload	No	3.2.2.10
				General corrosion	Loss of material	Transfer Casks AMP	3.2.1.1
				Pitting and crevice corrosion	Loss of material	Transfer Casks AMP	3.2.1.2
				Galvanic corrosion	Loss of material	Transfer Casks AMP	3.2.1.3

Table 4-23 FuelSolutions™ transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Bolt for top cover, bottom cover, ram access cover	SR	Steel	Air—indoor/outdoor	Microbiologically influenced corrosion	Loss of material	No	3.2.1.4
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.1.9
Washer for trunnion, top cover, bottom cover, ram access cover	SR	Stainless steel	Air—indoor/outdoor	Stress relaxation	Loss of preload	No	3.2.1.10
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
Top cover, ram access cover	SR	Stainless steel	Air—indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
				Radiation embrittlement	Cracking	No	3.2.2.9

Table 4-23 FuelSolutions™ transfer cask							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Bottom cover	SR	Stainless steel	Air— indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
Top lifting insert, bottom support ring	SR	Stainless steel	Air— indoor/outdoor	Radiation embrittlement	Cracking	No	3.2.2.9
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
Neutron shield plate	SR	Stainless steel	Air— indoor/outdoor	Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
				Radiation embrittlement	Cracking	No	3.2.2.9
				Pitting and crevice corrosion	Loss of material	No	3.2.2.2
				Microbiologically influenced corrosion	Loss of material	No	3.2.2.4
				Stress corrosion cracking	Cracking	No	3.2.2.5
				Fatigue	Cracking	Evaluate design code TLAA, if applicable	3.2.2.7
Radiation embrittlement	Cracking	No	3.2.2.9				

Table 4-23 FuelSolutions™ transfer cask									
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)		
Neutron shield plate	SR	Stainless steel	Embedded (RX-277, NS-3)	Radiation embrittlement	Cracking	No	3.2.2.9		
Neutron shield	SH	RX-277, NS-3	Embedded (stainless steel)	Thermal aging	Loss of fracture Toughness and loss of ductility	No	3.3.1.2		
				Radiation embrittlement	Cracking	No	3.3.1.3		
Coating on neutron shield jacket	TH	Coating	Air—indoor/outdoor	Boron depletion	Loss of shielding	TLAA/AMP or a supporting analysis is required	3.3.1.1		
				Radiation embrittlement	Coating degradation	TLAA/AMP or a supporting analysis is required	3.2.8		
Pressure relief device	SR	Brass	Air—indoor/outdoor	Thermal aging	Coating degradation	TLAA/AMP or a supporting analysis is required	3.2.8		
				General corrosion	Loss of material	Transfer Casks AMP	3.2.5.1		
				Pitting and crevice corrosion	Loss of material	No	3.2.5.2		
				Microbiologically influenced corrosion	Loss of material	No	3.2.5.3		
				Radiation embrittlement	Cracking	No	3.2.5.4		

1 **4.7 Concrete pad**

2 The support pad of an ISFSI is a reinforced concrete structure that provides a stable foundation
3 for the DSSs and transfer equipment. Depending on the design basis of the system or site, the
4 pad may be within the scope of renewal as an important-to-safety component or as a not-
5 important-to-safety component, the failure of which could prevent the fulfillment of a function that
6 is important-to-safety. Typically, the concrete pad is exposed to outdoor air and groundwater or
7 soil environments and is designed and constructed in accordance with ACI codes and
8 standards.

9 Table 4-24 provides a generic evaluation of potential aging mechanisms and effects requiring
10 management for the concrete pad. The AMPs that provide an acceptable approach to
11 managing the aging effects are also identified in the table.

Table 4-24 Concrete pad

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Reinforced concrete: ISFSI pad	SR*	Concrete	Air—outdoor	Aggressive chemical attack	Cracking	Reinforced Concrete Structures AMP	3.5.1.5
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.5
					Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.5
				Reduction of concrete pH (reducing corrosion resistance of steel embedments)		Reinforced Concrete Structures AMP	3.5.1.5
				Creep		No	3.5.1.2
				Dehydration at high temperatures		No	3.5.1.11
				Loss of strength		No	3.5.1.11
				Loss of material (spalling, scaling)		No	3.5.1.13
				Loss of strength		No	3.5.1.13
				Cracking		No	3.5.1.13
				Cracking		Reinforced Concrete Structures AMP	3.5.1.4
				Differential settlement		No	3.5.1.10
				Fatigue		Reinforced Concrete Structures AMP	3.5.1.1
Freeze and thaw		Reinforced Concrete Structures AMP	3.5.1.1				
Loss of material (spalling, scaling)		Reinforced Concrete Structures AMP	3.5.1.1				

*Safety Functions: Confinement (CO), Subcriticality (CR), Retrievability (RE), Radiation Shielding (SH), Structural Integrity (SR), Thermal/Heat Removal (TH)

Table 4-24 Concrete pad

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Reinforced concrete: ISFSI pad	SR	Concrete	Air—outdoor	Radiation damage	Cracking	No	3.5.1.9
					Loss of strength	No	3.5.1.9
				Reaction with aggregates	Cracking	Reinforced Concrete Structures AMP	3.5.1.3
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.3
				Salt scaling	Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.14
					Cracking	No	3.5.1.7
				Leaching of calcium hydroxide	Loss of strength	Reinforced Concrete Structures AMP	3.5.1.8
					Increase in porosity and permeability	Reinforced Concrete Structures AMP	3.5.1.8
				Aggressive chemical attack	Reduction of concrete pH (reducing corrosion resistance of steel embedments)	Reinforced Concrete Structures AMP	3.5.1.8
					Cracking	Reinforced Concrete Structures AMP	3.5.1.5
Loss of strength	Reinforced Concrete Structures AMP	3.5.1.5					

Table 4-24 Concrete pad

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Reinforced concrete: ISFSI pad	SR	Concrete	Groundwater/soil	Aggressive chemical attack	Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.5
					Reduction of concrete pH (reducing corrosion resistance of steel embedments)	Reinforced Concrete Structures AMP	3.5.1.5
				Creep	Cracking	No	3.5.1.2
					Cracking	No	3.5.1.11
					Loss of strength	No	3.5.1.11
				Delayed ettringite formation	Loss of material (spalling, scaling)	No	3.5.1.13
					Loss of strength	No	3.5.1.13
					Cracking	No	3.5.1.13
				Differential settlement	Cracking	Reinforced Concrete Structures AMP	3.5.1.4
					Cracking	No	3.5.1.10
				Fatigue	Cracking	Reinforced Concrete Structures AMP	3.5.1.1
					Cracking	Reinforced Concrete Structures AMP	3.5.1.1
Freeze and thaw	Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.1				
	Loss of strength	Reinforced Concrete Structures AMP	3.5.1.12				
Microbiological degradation	Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.12				
	Loss of strength	Reinforced Concrete Structures AMP	3.5.1.12				

Table 4-24 Concrete pad

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)	
Reinforced concrete: ISFSI pad	SR	Concrete	Groundwater/soil	Microbiological degradation	Increase in porosity and permeability	Reinforced Concrete Structures AMP	3.5.1.12	
					Reduction of concrete pH (reducing corrosion resistance of steel embedments)	Reinforced Concrete Structures AMP	3.5.1.12	
				Radiation damage	Cracking	No	No	3.5.1.9
					Loss of strength	No	No	3.5.1.9
				Reaction with aggregates	Cracking	Reinforced Concrete Structures AMP	3.5.1.3	
					Loss of strength	Reinforced Concrete Structures AMP	3.5.1.3	
				Salt scaling	Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.14	
				Shrinkage	Cracking	No	No	3.5.1.7
				Leaching of calcium hydroxide	Loss of strength	Reinforced Concrete Structures AMP	3.5.1.8	
					Increase in porosity and permeability	Reinforced Concrete Structures AMP	3.5.1.8	

Table 4-24 Concrete pad

Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Reinforced concrete: ISFSI pad	SR	Concrete	Groundwater/soil	Leaching of calcium hydroxide	Reduction of concrete pH (reducing corrosion resistance of steel embedments)	Reinforced Concrete Structures AMP	3.5.1.8
		Reinforcing steel	Air—outdoor; groundwater	Corrosion of reinforcing steel	Loss of concrete/steel bond	Reinforced Concrete Structures AMP	3.5.1.6
					Loss of material (spalling, scaling)	Reinforced Concrete Structures AMP	3.5.1.6
					Cracking	Reinforced Concrete Structures AMP	3.5.1.6
			Loss of strength	Reinforced Concrete Structures AMP	3.5.1.6		

1 **4.8 Spent fuel assemblies**

2 **4.8.1 Spent fuel assembly description**

3 Dry storage systems are designed to store a wide range of SNF assemblies in a dried and
4 inerted (helium) atmosphere. This section provides a general description of the PWR and BWR
5 spent fuel assembly components.

6 **4.8.2 Fuel cladding and assembly hardware**

7 Pressurized-water reactor fuel assemblies

8 While there are a number of fuel assembly design variants for PWRs, the assemblies mainly
9 consist of the top nozzle, fuel rods, spacer grids, guide thimble tubes, and bottom nozzle. The
10 various components of a typical 17 × 17 PWR fuel assembly are shown in Figure 4-30. Each
11 fuel rod consists of enriched uranium dioxide pellets inserted into a cladding tube. The cladding
12 tube is then capped with Zircaloy end plugs and seal welded at both ends to confine the fuel
13 pellets and fission gases. The fuel cladding, fabricated from zirconium-based alloys, including
14 Zircaloy-4, ZIRLO™, and M5®, provides a confinement barrier.

15 The structural support of the fuel assembly is provided by the top and bottom nozzles, the
16 spacer grid assemblies, and the guide thimbles. Guide tubes, fabricated from zirconium-based
17 alloys, are the main structural members of the fuel assembly. They also provide channels for
18 neutron absorber rods and burnable poison rods. The bottom of the guide tube is fitted with an
19 end plug with a flow port, which is then fastened into the bottom nozzle. Both the top and
20 bottom nozzles are made of either stainless steel or Inconel, which also serve as structural
21 members of the fuel assembly. The spacer grid assemblies provide support for the fuel
22 cladding tubes. Two types of grid assemblies, fabricated from zirconium-based alloys or
23 Inconel, are used in the fuel assemblies.

24 Boiling-water reactor fuel assemblies

25 Similar to the case for PWRs, there are a number of fuel assembly design variants for BWRs.
26 The main components include the (i) upper tie plate, (ii) fuel rods, (iii) spacer grids, (iv) water
27 rods, (v) channel, and (vi) lower tie plate, as shown in Figure 4-31 for the GE14 BWR fuel
28 assembly in a 10 × 10 fuel rod array. Two types of fuel rods are used in the GE14 fuel bundles:
29 standard rods and tie rods. The fuel rods are hollow cladding tubes fabricated from Zircaloy-2.
30 Zircaloy end plugs are welded into place to seal the ends of the fuel rods. The tie rods differ
31 from the standard fuel rods in that the end plugs are threaded into the tie plates. They hold the
32 fuel bundle together and support the weight of the fuel bundle during fuel handling operations.

33 In the BWR fuel assembly, fuel bundles are enclosed in open-ended, square tubes (also called
34 channels) and are supported at the ends of the fuel bundles by the upper and lower tie plates.
35 The channels made of zirconium-based alloys are ducts for coolant flow that prevent lateral flow
36 of coolant among the fuel assemblies operating at different power levels. Both the upper tie and
37 lower tie plates are fabricated from stainless steel. The upper tie plate provides alignment and
38 support for the fuel rods at the top of the fuel bundle, while the lower tie plate positions the fuel
39 rods laterally. The spacer grids, fabricated from zirconium-based alloys or Inconel, hold the fuel
40 rods in the proper location so that optimum fuel spacing is maintained.

41

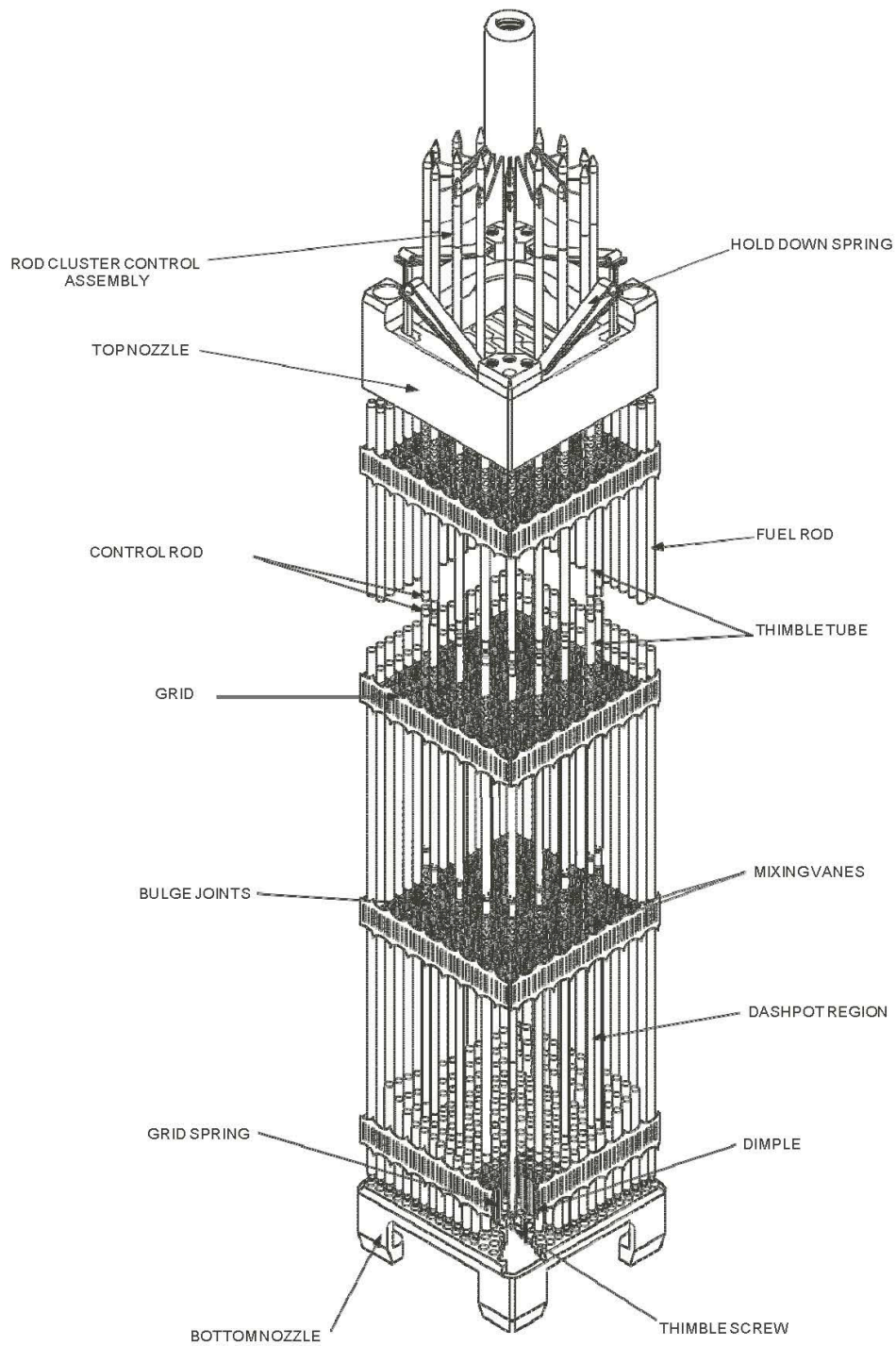


Figure 4-30 Typical pressurized-water reactor fuel assembly (NRC, 2002)

1

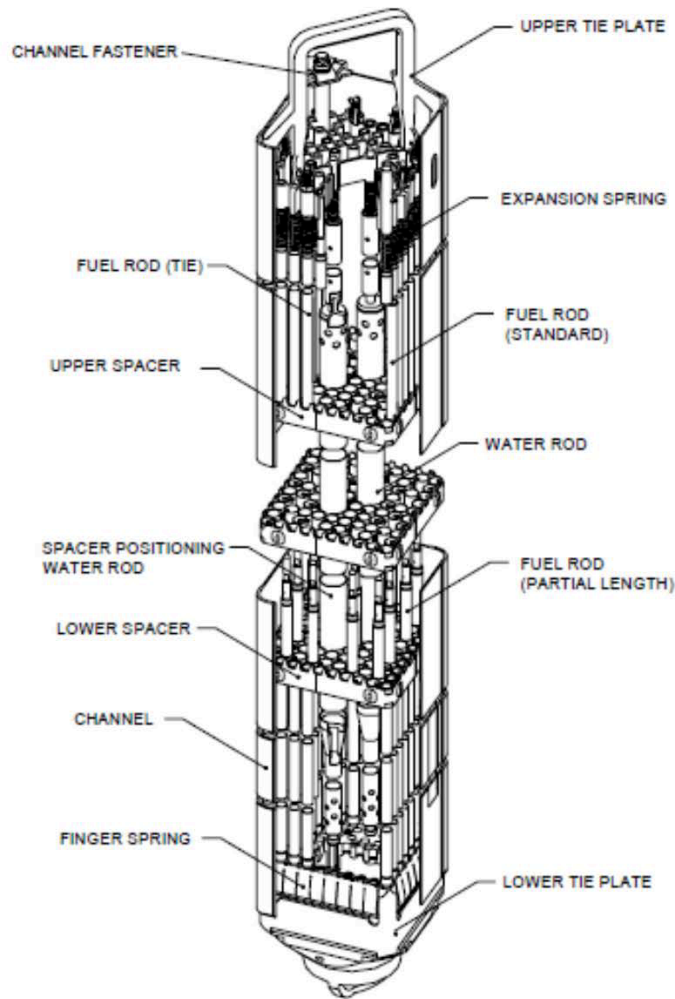


Figure 4-31 Boiling-water reactor GE14 fuel assembly (GNF, 2005)

- 1 Table 4-25 provides a generic evaluation of potential aging mechanisms and effects requiring
- 2 management for specific components of the SNF assemblies. The AMPs that provide an
- 3 acceptable approach to managing the aging effects are also identified in the table.
- 4

Table 4-25 Spent fuel assemblies							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Technical Basis (Section)	
Fuel rod cladding	CO, CR, RE, SH, SR, TH*	Zirconium-based alloy (Zircaloy-2, Zircaloy-4, ZIRLO™, or M5®)	Helium	Oxidation	Loss of load bearing capacity	No	3.6.1.6
				Pitting corrosion	Loss of material	No	3.6.1.7
				Galvanic corrosion	Loss of material	No	3.6.1.8
				Stress corrosion cracking	Cracking	No	3.6.1.9
				Hydride-induced embrittlement	Loss of ductility	No	3.6.1.1
				Delayed hydride cracking	Cracking	No	3.6.1.2
				Thermal Creep	Changes in dimensions	High-Burnup Fuel Monitoring and Assessment AMP	3.6.1.3
				Low-temperature creep	Changes in dimensions	No	3.6.1.4
				Radiation embrittlement	Loss of strength	No	3.6.1.10
				Fatigue	Cracking	No	3.6.1.11
				Mechanical overload	Cracking	No	3.6.1.5
				Creep	Changes in dimensions	No	3.6.2.1
				Hydriding	Changes in dimensions	No	3.6.2.2
Guide tubes (PWR) or water channels (BWR)	RE, SR	Zirconium-based alloy	Helium	Radiation embrittlement	Loss of strength	No	3.6.1.10
				Fatigue	Cracking	No	3.6.1.11
Spacer grids	CR, RE, SR, TH	Zirconium-based alloy	Helium	Mechanical overload	Cracking	No	3.6.1.5
				Creep	Changes in dimensions	No	3.6.2.1
*Safety Functions: Confinement (CO), Subcriticality (CR), Retrievalability (RE), Radiation Shielding (SH), Structural Integrity (SR), Thermal/Heat Removal (TH)							

Table 4-25 Spent fuel assemblies							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Spacer grids	CR, RE, SR, TH	Zirconium-based alloy	Helium	Hydriding	Changes in dimensions	No	3.6.2.2
				Radiation embrittlement	Loss of strength	No	3.6.1.10
				Fatigue	Cracking	No	3.6.1.11
				Creep	Change in dimensions	No	3.6.2.1
				General corrosion	Loss of material	No	3.6.2.3
				Stress corrosion cracking	Cracking	No	3.6.2.4
				Radiation embrittlement	Loss of strength	No	3.6.1.10
		Inconel	Helium	Fatigue	Cracking	No	3.6.1.11
				Creep	Change in dimensions	No	3.6.2.1
				General corrosion	Loss of material	No	3.6.2.3
				Stress corrosion cracking	Cracking	No	3.6.2.4
				Radiation embrittlement	Loss of strength	No	3.6.1.10
				Fatigue	Cracking	No	3.6.1.11
				Creep	Change in dimensions	No	3.6.2.1
Lower and upper end fittings	CR, RE, SR	Stainless steel	Helium	General corrosion	Loss of material	No	3.6.2.3
				Stress corrosion cracking	Cracking	No	3.6.2.4
				Radiation embrittlement	Loss of strength	No	3.6.1.10
				Fatigue	Cracking	No	3.6.1.11
				Creep	Change in dimensions	No	3.6.2.1
				General corrosion	Loss of material	No	3.6.2.3
				Stress corrosion cracking	Cracking	No	3.6.2.4
Inconel	Helium	Inconel	Helium	General corrosion	Loss of material	No	3.6.2.3
				Stress corrosion cracking	Cracking	No	3.6.2.4
				Radiation embrittlement	Loss of strength	No	3.6.1.10
				Fatigue	Cracking	No	3.6.1.11
				Creep	Change in dimensions	No	3.6.2.1
				General corrosion	Loss of material	No	3.6.2.3
				Stress corrosion cracking	Cracking	No	3.6.2.4
Inconel	Helium	Inconel	Helium	General corrosion	Loss of material	No	3.6.2.3
				Stress corrosion cracking	Cracking	No	3.6.2.4
				Radiation embrittlement	Loss of strength	No	3.6.1.10
				Fatigue	Cracking	No	3.6.1.11
				Creep	Change in dimensions	No	3.6.2.1
				General corrosion	Loss of material	No	3.6.2.3
				Stress corrosion cracking	Cracking	No	3.6.2.4

Table 4-25 Spent fuel assemblies							
Structure, System, or Component	Intended Safety Function	Material	Environment	Aging Mechanism	Aging Effect	Aging Management	Technical Basis (Section)
Fuel channel (BWR)	CR, TH	Zirconium-based alloy	Helium	Creep	Change in dimensions	No	3.6.2.1
				Hydridding	Change in dimensions	No	3.6.2.2
Poison rod assemblies (PWR)	CR	Stainless steel	Helium	Radiation embrittlement	Loss of strength	No	3.6.1.10
				Fatigue	Cracking	No	3.6.1.11
				Creep	Change in dimensions	No	3.6.2.1
				General corrosion	Loss of material	No	3.6.2.3
				Stress corrosion cracking	Cracking	No	3.6.2.4
				Radiation embrittlement	Loss of strength	No	3.6.1.10
				Fatigue	Cracking	No	3.6.1.11

1 **4.9** References

- 2 EPRI. "Industry Spent Fuel Storage Handbook." Report 1021048. Palo Alto, California:
3 Electric Power Research Institute. July 2010.
- 4 FuelSoutions. "FuelSoutions W21 Canister Storage Final Safety Analysis Report."
5 Docket No. 72-1026, Revision 5. ADAMS Accession No. ML071510213. April 2007a.
- 6 _____. "FuelSoutions W74 Canister Storage Final Safety Analysis Report."
7 Docket No. 72-1026, Revision 6. ADAMS Accession No. ML071510213. April 2007b.
- 8 _____. "FuelSoutions Storage System Final Safety Analysis Report." Docket No. 72-1026.
9 Revision 4. ADAMS Accession No. ML071510207. April 2007c.
- 10 GNF. "GE14 Fuel Assembly Mechanical Design Report." NEDC-33236.
11 Wilmington, North Carolina: Global Nuclear Fuel. ADAMS Accession No. ML053540338.
12 2005.
- 13 Holtec International. "Final Safety Analysis Report for the HI-STORM 100 Cask System."
14 Non-Proprietary. Docket No. 72-1014. HI-20024444. Rev. 11. ADAMS Accession
15 No. ML13246A042. August 1, 2013.
- 16 _____. "Final Safety Analysis Report for the Holtec International Storage, Transport, and
17 Repository Cask System (HI-STAR 100 Cask System)." HI-2012610, Rev. 0, Vol I of II.
18 Holtec International. ADAMS Accession No. ML072410190. March 2001.
- 19 NAC International. "MAGNASTOR, Final Safety Analysis Report," Non-Proprietary Version.
20 Docket No. 72-1025. Rev. 15A. ADAMS Accession No. ML15225A469. July 2015.
- 21 _____. Retrieved from <http://www.nacintl.com/magnastor>. 2014.
- 22 _____. "Final Safety Analysis Report for the UMS Universal Storage System."
23 Docket No. 72-1015, Rev 3. ADAMS Accession No. ML051290397. March 2004.
- 24 _____. "Final Safety Analysis Report for the NAC Multi-Purpose Canister System."
25 Non-Proprietary. Docket No. 72-1025. Rev. 0. ADAMS Accession No. ML073551117.
26 April 2000.
- 27 NRC. "Safety Evaluation Report for License Renewal: Calvert Cliffs Nuclear Power Plant
28 Independent Spent Fuel Storage Installation." ADAMS Accession No. ML14274A038
29 Washington, DC: U.S. Nuclear Regulatory Commission. October, 2014.
- 30 _____. "Westinghouse Systems Course R-304P." ADAMS Accession No. ML023030412.
31 Washington, DC: U.S. Nuclear Regulatory Commission. 2002.
- 32 _____. "Safety Evaluation Report for the TN-32 Dry Storage Cask." Washington, DC:
33 U.S. Nuclear Regulatory Commission. November 1996.

34

- 1 Pacific Nuclear Fuel Services, Inc. "Topical Report for the NUTECH Horizontal Modular
2 Storage System for Irradiated Nuclear Fuel NUHOMS-24P." NUH-002.0103, Rev. 2A,
3 Volume I. San Jose, California: Pacific Nuclear Fuel Services, Inc. ADAM Accession No.
4 ML110730769. April 1991.
- 5 Transnuclear Inc. "Certificate of Compliance Renewal Application for the Standardized
6 NUHOMS System." Rev. 0. ADAMS Accession No. ML14309A343. Columbia, Maryland:
7 Transnuclear, Inc. November 4, 2014.
- 8 _____. "Updated Final Safety Analysis Report for the Standardized NUHOMS Modular Storage
9 System for Irradiated Nuclear Fuel." NUH-003.0103, Rev. 10, Volume 1 and 3 (Appendix M).
10 Columbia, Maryland: Transnuclear, Inc. February 2008.
- 11 _____. "Final Safety Analysis Report for the TN-68 Dry Storage Cask." Amendment 1, Rev. 0.
12 Hawthorne, New York: Transnuclear, Inc. January 2005.
- 13 _____. "Final Safety Analysis Report for the Standardized NUHOMS Modular Storage System
14 for Irradiated Nuclear Fuel." NUH-003.0103, Rev. 8, Volume 1 and 3 (Appendix M). ADAMS
15 Accession Nos. ML042110421 and ML051040570. Hawthorne, New York: Transnuclear, Inc.
16 June 2004.
- 17 _____. "Final Safety Analysis Report for the Standardized Advanced NUHOMS Modular
18 Storage System for Irradiated Nuclear Fuel." ANUH-01.0150, Rev. 0, Volume 1. ADAMS
19 Accession No. ML050410252 (Cover through Section 3.6.3). Fremont, CA: Transnuclear, Inc.
20 February 2003.

5 TIME-LIMITED AGING ANALYSES

5.1 Introduction

Renewal applicants are required to reevaluate all aging-related calculations or analyses involving time-limited assumptions that were contained in the original design basis (e.g., fatigue analyses, corrosion wastage calculations). These evaluations are designated as time-limited aging analyses (TLAAs), and Title 10 of the *Code of Federal Regulations* (10 CFR) 72.3, "Definitions," defines them as those calculations and analyses meeting all six of the following criteria:

- (1) Involve SSCs important to safety within the scope of the specific-license renewal, as delineated in Subpart F of 10 CFR Part 72, or within the scope of the spent fuel storage CoC renewal, as delineated in Subpart L of 10 CFR Part 72, respectively.
- (2) Consider the effects of aging.
- (3) Involve time-limited assumptions defined by the current operating term.
- (4) Were determined to be relevant by the specific licensee or certificate holder in making a safety determination.
- (5) Involve conclusions or provide the basis of conclusions related to the capability of SSCs to perform their intended safety functions.
- (6) Are contained or incorporated by reference in the design bases.

5.2 Review

NUREG-1927, Revision 1, "Standard Review Plan for Renewal of Specific Licenses and Certificates of Compliance for Dry Storage of Spent Nuclear Fuel" (NRC, 2016), provides detailed staff guidance for the review of TLAAAs.

The NRC reviewer should use the final safety analysis report (FSAR) and other documents that detail the design bases and confirm that the renewal applicant did not omit any TLAAAs submitted as part of the approved design bases. In some cases, the original analyses may have been performed as part of a code design but not explicitly discussed in the FSAR. Thus, the reviewer must identify and review any design codes and standards associated with a storage system to ensure that any required analyses are captured in the applicant's TLAAAs. Table 5-1 identifies some examples of fatigue analyses that are incorporated into the design codes for the dry storage system designs this report evaluates.

The reviewer also should ensure that the applicant addresses any design basis calculations that use materials properties that may be time dependent. For example, aluminum alloys used in some fuel baskets can lose strength over time at elevated temperatures (see Section 3.2.3.7), and this may affect the performance of the fuel basket in a cask tipover analysis. If the original design basis calculations did not adequately account for such material property changes through the period of extended operation, the analyses should be updated.

The reviewer should ensure that the applicant has appropriately dispositioned an identified TLAA by using one of the following methods:

- 1 • Demonstrate that the existing analysis remains valid for the period of extended
2 operation, has already considered the requested period of extended operation, and
3 concludes that the structure, system, or component (SSC) will continue to perform its
4 intended function through the end of the requested period of extended operation.
- 5 • Revise or update the existing analysis to demonstrate that it has been projected to the
6 end of the requested period of extended operation and concludes that the SSC will
7 continue to perform its intended function through the end of the requested period of
8 extended operation.
- 9 • Manage the effects of aging on the SSC for the requested period of extended operation
10 through an aging management program.

Table 5-1 Examples of fatigue analyses contained within storage system design bases		
System	SSC	Fatigue Evaluation Criteria (ASME Code Section III, Division 1 (ASME, 2007))
Standardized and Advanced NUHOMS®*	DSC Confinement	NB-3222.4
	Transfer Cask	NC-3219
HI-STORM 100, HI-STAR 100	MPC Confinement	NB-3222.4
	Fuel Basket	NG-3222.4
	HI-STAR Overpack Helium Boundary	NB-3222.4
TN-32 & 68	Confinement Boundary	NB-3222.4
	Fuel Basket	NB-3222.4
NAC-UMS, MPC, and MAGNASTOR	Canister Confinement	NB-3222.4
	Fuel Basket	NG-3222.4
FuelSolutions™	Canister Confinement	NB-3222.4
	Fuel Basket	NG-3222.4
	Transfer Cask	NC-3219

11

12 **5.3 References**

13 NRC. NUREG–1927, “Standard Review Plan for Renewal of Specific Licenses and Certificates
14 of Compliance for Dry Storage of Spent Nuclear Fuel.” Revision 1. Washington, DC:
15 U.S. Nuclear Regulatory Commission. ADAMS Accession No. ML16179A148. 2016.

16 ASME. Boiler and Pressure Vessel Code, Section III, Division 1, “Rules for Construction of
17 Nuclear Facility Components,” Division 1, Subsection NB, “Class 1 Components,” Subsection
18 NC, “Class 2 Components,” and Subsection NG, “Core Support Structures”; American Society
19 of Mechanical Engineers. 2007.

6 EXAMPLE AGING MANAGEMENT PROGRAMS

6.1 Introduction

The example aging management programs (AMPs) presented in this chapter and listed in Table 6-1 below describe a generically acceptable approach to managing the credible aging effects that were identified in the technical bases discussions in Chapter 3 and the aging management review tables in Chapter 4. AMPs monitor and control the degradation of structures, systems, and components (SSCs) within the scope of renewal, so that aging effects will not result in a loss of intended functions during the period of extended operation. An AMP includes all activities that are credited for managing aging mechanisms or effects for specific SSCs. An effective AMP prevents, mitigates, or detects the aging effects and provides for the prediction of the extent of the effects of aging and timely corrective actions before there is a loss of intended function.

If an applicant credits these generic AMPs in the renewal application, the NRC staff should ensure that the applicant demonstrates that the design features, environmental conditions, and operating experience for the subject independent spent fuel storage installation (ISFSI) or dry storage system (DSS) are bounded by those evaluated in this report. Otherwise, the staff should ensure that the applicant augments the AMPs as appropriate to address the impact of unique design or operating parameters.

Table 6-1 Example aging management programs

Section	AMP
6.5	Localized Corrosion and Stress Corrosion Cracking of Welded Stainless Steel Dry Storage Canisters
6.6	Reinforced Concrete Structures
6.7	External Surfaces Monitoring of Metallic Components
6.8	Ventilation Systems
6.9	Bolted Cask Seal Leakage Monitoring
6.10	Transfer Casks
6.11	High-Burnup Fuel Monitoring and Assessment

6.2 Alternative approaches

An applicant may propose alternative approaches to manage the effects of aging. In its review of alternative AMPs, the staff should use the guidance in NUREG-1927, Revision 1, "Standard Review Plan for Renewal of Specific Licenses and Certificates of Compliance for Dry Storage of Spent Nuclear Fuel" (NRC, 2016). As described in greater detail in NUREG-1927, an AMP generally should contain the following 10 elements:

- (1) Scope of program: the specific SSCs and subcomponents covered by the AMP and the intended functions to be maintained, in addition to stating the specific materials, environments, and aging mechanisms and effects to be managed

- 1 (2) Preventive actions: actions to prevent aging or mitigate the rates of aging for SSCs
- 2 (3) Parameters monitored or inspected: the specific parameters that will be monitored or
3 inspected and a description of how those parameters will be capable of identifying
4 degradation before a loss of intended function
- 5 (4) Detection of aging effects: the inspection and monitoring details, including method or
6 technique, frequency, sample size, data collection, and timing of inspections
- 7 (5) Monitoring and trending: how data will be evaluated and trended to ensure timely
8 corrective actions
- 9 (6) Acceptance criteria: the criteria against which the need for corrective action will
10 be evaluated
- 11 (7) Corrective actions: The measures to be taken when the acceptance criteria are not met,
12 including root cause determination and prevention of recurrence, as appropriate
- 13 (8) Confirmation process: processes in place to verify that preventive actions are adequate
14 and that appropriate corrective actions have been completed and are effective
- 15 (9) Administrative controls: processes in place that provide a formal review and approval
16 process for activities related to the AMP (e.g., inspector requirements, instrument
17 calibration)
- 18 (10) Operating experience: a review of operational experience that supports the
19 determination that the AMP is capable of maintaining SSC functions in the period of
20 extended operation

21 The reviewer should examine the applicant's proposed 10 elements to verify that the program is
22 capable of managing the specific aging mechanisms and effects identified by the aging
23 management review (AMR). The reviewer should recognize that an applicant may develop
24 AMPs following a different format or style. For such reviews, the NRC staff should ensure that
25 sufficient detail (i.e., supporting technical bases) is provided in the alternative format in
26 comparison with the 10 AMP elements of this guidance.

27 An applicant may credit existing site maintenance and inspection activities to manage the
28 effects of aging. In such cases, the reviewer should ensure that the design basis
29 documentation describes those activities with sufficient detail ensure that the 10 AMP elements
30 are fully addressed.

31 **6.3 Learning aging management**

32 As described in NUREG-1927, the reviewer should ensure that the application includes
33 provisions to conduct periodic future reviews of operating experience to confirm the
34 effectiveness of the AMPs or identify a need to enhance or modify an AMP. The reviewer also
35 should verify that the applicant: (1) references a specific system to be used to obtain,
36 aggregate, and enter site-specific, design-specific, and industrywide operating experience, and
37 (2) discusses how it intends to provide timely reporting of operating experience to this system.

1 If an applicant follows this approach, the reviewer should ensure that the description of the
2 periodic assessments includes specific performance criteria (e.g., program-specific performance
3 indicators for each of the 10 AMP elements) and proposed actions based on the assessment
4 findings. The reviewer should also ensure that the timing of the assessments appropriately
5 considers the rate of aging degradation and the anticipated availability of data from industry
6 initiatives.

7 Nuclear Energy Institute (NEI) 14-03, “Format, Content, and Implementation Guidance for Dry
8 Cask Storage Operations-Based Aging Management,” Revision 2, provides a proposed
9 framework for learning AMPs through the use of “tollgates” (NEI, 2016). NEI 14-03 defines
10 “tollgates” as periodic points within the period of extended operation when licensees would be
11 required to evaluate aggregate feedback and perform and document a safety assessment that
12 confirms the safe storage of spent fuel. At the time of publication of this report, the NRC staff
13 was continuing its review of NEI 14-03, Revision 2, for proposed NRC endorsement. However,
14 until a time when NEI 14-03 may be endorsed by the NRC, Section 3.6.1.10 of NUREG–1927,
15 Revision 1, provides guidance to reviewers on ensuring AMP effectiveness.

16 **6.4 References**

17 NEI. “Format, Content and Implementation Guidance for Dry Cask Storage Operations-Based
18 Aging Management for Dry Cask Storage.” NEI 14-03, Rev. 2. ADAMS Accession
19 No. ML16356A210. Nuclear Energy Institute. 2016.

20 NRC. NUREG–1927, “Standard Review Plan for Renewal of Specific Licenses and Certificates
21 of Compliance for Dry Storage of Spent Nuclear Fuel.” Revision 1. Washington, DC:
22 U.S. Nuclear Regulatory Commission. 2016.

1 **6.5 Localized Corrosion and Stress Corrosion Cracking of Welded Stainless**
2 **Steel Dry Storage Canisters**

3 Welded stainless steel canisters are used in the majority of the DSSs in the United States for
4 spent nuclear fuel (SNF) from commercial power reactors at both specific-licensed and
5 general-licensed ISFSIs. The welded stainless steel canisters are the primary confinement
6 boundary during storage. While there are no known operational occurrences of aging or
7 localized corrosion of welded stainless steel canisters, operational experience with nuclear
8 reactors that were located close to an open ocean or bay has shown that pitting corrosion,
9 crevice corrosion, and chloride-induced stress corrosion cracking (CISCC) can occur in welded
10 stainless steel components as a result of atmospheric deposition and deliquescence of
11 chloride-containing salts. Laboratory and natural exposure tests suggest that CISCC can occur
12 with sufficient surface chloride concentrations and that, with those concentrations of chloride,
13 crack propagation rates can be of engineering significance for welded stainless steel canisters
14 during the period of extended operation.

15 Based on reactor operating experience, as well as laboratory and field testing, localized
16 corrosion and CISCC are potential aging mechanisms for welded stainless steel canisters.
17 Environments where chloride-containing salts may be deposited on welded stainless steel
18 canisters include coastal locations near salt water and locations that are close to cooling towers
19 or roads that are salted. The Electric Power Research Institute (EPRI) has developed aging
20 management guidance to address the potential for CISCC of welded stainless steel canisters
21 (Fuhr et al., 2017). In addition, the American Society of Mechanical Engineers Boiler and
22 Pressure Vessel Code (ASME Code), Section XI, has formed a task group to develop a code
23 case to establish the requirements for inservice inspection and acceptance criteria for DSS
24 canisters (Code Case N-860) that may follow the recommendations of the EPRI aging
25 management guidance. However, the development of a consensus-based code case for
26 inservice inspection of DSS canisters may take several years to complete. To address potential
27 aging effects as a result of localized corrosion cracking and stress corrosion cracking (SCC) in
28 the absence of an acceptable code case, the NRC has provided an example AMP for welded
29 stainless steel canisters used in DSSs that relies on guidance from consensus codes for
30 inservice inspection of nuclear power plant components. Elements of an NRC staff-developed
31 example AMP are described in Table 6-2.

32

33

Table 6-2 Example aging management program for Localized Corrosion And Stress Corrosion Cracking Of Welded Stainless Steel Dry Storage Canisters

Element	Description
1. Scope of Program	<p>Inspection of welded stainless steel dry storage canister confinement boundary external surfaces for atmospheric deposits, localized corrosion, and SCC.</p> <p>Examinations should be focused on areas with the following attributes:</p> <ul style="list-style-type: none"> • canister fabrication welds and weld heat affected zones • closure welds and weld heat affected zones • areas of the canister to which temporary supports or attachments were attached by welding and subsequently removed • locations where a crevice is formed on the canister surface • horizontal ($\pm 30^\circ$) surfaces where deposit accumulation may accumulate at a faster rate compared to vertical surfaces • canister surfaces that are cold relative to the average surface temperature • canister surfaces with higher amounts of atmospheric deposits <p>Effort should be made to identify and prioritize examinations of areas on canisters that have two or more of the above attributes (e.g., canister surface that is cold relative to average surface temperature and also has a weld or weld heat affected zone).</p>
2. Preventive Actions	<p>None; AMP is for condition monitoring. However, DSS canister designs may include preventive actions such as fabrication procedures and surface modification methods to impart compressive residual stresses on the canister welds and weld heat-affected zones to reduce the potential for SCC. Preventive actions may also include the use of DSS canister confinement boundary materials that are resistant to localized corrosion and SCC. For such cases the preventive actions described should be supported with an analysis and data demonstrating the preventive actions are effective.</p>
3. Parameters Monitored/ Inspected	<p>Parameters monitored or inspected should include:</p> <ul style="list-style-type: none"> • visual evidence of discontinuities and imperfections such as localized corrosion, including pitting corrosion, crevice corrosion and SCC of the canister welds and weld heat-affected zones • size and location of localized corrosion and SCC • appearance and location of deposits on the canister surfaces
4. Detection of Aging Effects	<p>Visually examine deposits on the canister surfaces and identify corrosion products that may be indicators of localized corrosion and SCC in the welds and weld heat-affected zones. Visual examination instrumentation with demonstrated sizing and depth measurement</p>

Table 6-2 Example aging management program for Localized Corrosion And Stress Corrosion Cracking Of Welded Stainless Steel Dry Storage Canisters

Element	Description
	<p>capability may be useful in the determination of the size and depth of pits open to the surface. Visual examination may also detect the presence of cracks originating from pits. However, the ability to detect cracks on clean metal surfaces using visual examination methods is dependent on several factors and can be difficult for tight crack opening displacements (Cumblidge et al., 2004, 2007). The presence of significant corrosion product accumulation may also interfere with the identification of SCC using visual examination methods.</p> <p>Volumetric examination is necessary to characterize SCC. Volumetric examination of pits and areas immediately adjacent to pits is necessary when pits are located within 25 mm [1 in] of a through thickness weld or within 25 mm [1 in] of an area where an temporary attachment was known to be located.</p> <p><u>Visual Examination</u></p> <p>Pitting and crevice corrosion that is open to the surface can potentially be detected by visual testing (ASME Code Section V, Table A-110). Because of the high neutron and gamma radiation fields near the surface of the stainless steel dry storage canisters, direct visual examination is not possible. Procedures for remote visual examination should be performance demonstrated; procedure attributes, for example, equipment resolution and lighting requirements, should reference applicable standards, such as ASME Code Section XI, Article IWA-2200 for VT-1 and VT-3 examinations (ASME, 2007) and BWRVIP-03 (Selby, 2005) for EVT-1 examinations.</p> <p><u>Volumetric Examination</u></p> <p>Additional assessment is necessary for suspected areas of localized corrosion and SCC. In these cases, the severity of degradation must be assessed, including the dimensions of the affected area and the depth of penetration with respect to the thickness of the canister. For accessible areas where adequate cleaning can be performed, remote visual examination meeting the requirements for VT-1 Examination (ASME Code Section XI, IWA-2211) may be used to determine the type of degradation present (e.g., pitting corrosion or SCC) and the location of degradation. Examinations to characterize the extent and severity of localized corrosion and SCC should be conducted using surface or volumetric examination methods consistent with the requirements of ASME Code Section XI, IWB-2500, for category B-J components (ASME, 2007).</p>

Table 6-2 Example aging management program for Localized Corrosion And Stress Corrosion Cracking Of Welded Stainless Steel Dry Storage Canisters

Element	Description
	<p><u>Sample Size</u></p> <p>For sites where inspections are necessary, there should be a minimum of one canister at each site. Preference should be given to the canisters with the greatest susceptibility for localized corrosion or SCC. Factors to be considered include older and colder canisters with the greatest potential for the accumulation and deliquescence of deposited salts that may promote localized corrosion and SCC, types of systems used at the site, canister location with respect to potential sources of atmospheric deposits, system design, and operational experience. Industry guidance on evaluating susceptibility has been published by the EPRI (Fuhr et al., 2015).</p> <p>Justification for not conducting inspections for localized corrosion or SCC should be provided on a case-by-case basis for each ISFSI site where welded stainless steel canisters are in use. Acceptable justification may be based on a comparison of susceptibility for the ISFSI location versus at least two other ISFSI sites determined to have greater susceptibility but that showed no evidence of localized corrosion or SCC in inspections completed within 5 years of the time of the assessment. The justification must consider the full range of available ISFSI susceptibility assessments and welded stainless steel canister examination results.</p> <p><u>Data Collection</u></p> <p>Canister Examination: documentation of the examination of the canister, location, and appearance of deposits and an assessment of the suspect areas where corrosion products were observed as described in corrective actions</p> <p>Bounding Analysis: a complete listing of other sites considered, susceptibility assessments for those sites, and results of examinations conducted at those sites, as well as a justification for not including other sites where examinations showed evidence of localized corrosion or SCC</p> <p><u>Frequency</u></p> <p>The frequency of inspection should be determined based on the localized corrosion and SCC susceptibility of both the site and the canisters in service, aggregated operational experience of similar storage system canisters and previous site specific examination results</p>

Table 6-2 Example aging management program for Localized Corrosion And Stress Corrosion Cracking Of Welded Stainless Steel Dry Storage Canisters

Element	Description
	<p><u>Timing of Inspections</u></p> <p>The timing of the inspections includes the preapplication inspection or general-licensee baseline inspection, performed per Sections 3.4.1.2 and 3.6.1.10 of NUREG–1927, Revision 1, and at the frequency specified by the AMP.</p> <p>Alternative detection methods or techniques may be provided. For these cases:</p> <ul style="list-style-type: none"> • The method or technique should be adequate and proven to be capable of evaluating the condition of the external surface of the canister against the acceptance criteria for the detection of localized corrosion and SCC. • The proposed intervals for inspection or monitoring are consistent with applicable site-specific, design-specific, or industrywide operating experience and should have sufficient frequency to ensure that the confinement function will be maintained until the next scheduled inspection. • The data collection methods should be sufficient for evaluating localized corrosion and SCC and should reference specific methods to be used for data acquisition, including any applicable consensus codes and standards.
<p>5. Monitoring and Trending</p>	<p>Monitoring and trending methods are in accordance with ASME Code Section XI evaluation criteria.</p> <p>Monitoring and trending methods reference plans/procedures are used to do the following:</p> <ul style="list-style-type: none"> • Establish a baseline before or at the beginning of the period of extended operation • Track trending of parameters or effects not corrected following a previous inspection including <ul style="list-style-type: none"> — the locations and size of any areas of localized corrosion or SCC — the disposition of canisters with identified aging effects and the results of supplemental canister inspections <p>Monitoring and trending should also include:</p> <ul style="list-style-type: none"> • the appearance of the canister, particularly at welds and in crevice locations, documented with images and video that will allow comparison in subsequent examinations

Table 6-2 Example aging management program for Localized Corrosion And Stress Corrosion Cracking Of Welded Stainless Steel Dry Storage Canisters

Element	Description
	<ul style="list-style-type: none"> changes to the size and number of any rust-colored stains as a result of iron contamination of the surface in subsequent inspections
<p>6. Acceptance Criteria</p>	<p>No indications of localized corrosion pits, etching, crevice corrosion, SCC, red-orange-colored corrosion products emanating from crevice locations, or red-orange-colored corrosion products in the vicinity of canister fabrication welds, closure welds, and welds associated with temporary attachments during canister fabrication.</p> <p>Flaws identified must be assessed in accordance with the acceptance standards identified in ASME B&PV Code Section XI, IWB-3514.</p> <p><u>Indications Requiring Additional Evaluation</u></p> <p>Confirmed or suspected areas of crevice corrosion, pitting corrosion, and SCC must be assessed in accordance with the acceptance criteria identified in ASME B&PV Code Section XI, IWB-3640.</p> <p>Although shop and handling procedures include controls to prevent iron contamination of the stainless steel surfaces, contamination does occur and is usually identified by rust-colored surface deposits. Iron contamination can exacerbate CISC in stainless steels. In accessible locations, removal of the deposits and rust stains that reveal undamaged welds (i.e., absence of pits, crack, localized attack, or etching) and the original machining/grinding marks on the stainless steel base metal, including weld heat-affected zones, may be used to confirm that localized corrosion or SCC has not been initiated.</p> <p>Indications of interest that are subject to additional examination and disposition include:</p> <ul style="list-style-type: none"> localized corrosion pits, crevice corrosion, SCC, and etching (note that these indications may be covered by obstructions (i.e., crevices)); deposits; or corrosion products discrete red-orange-colored corrosion products that are 1 mm [0.039 in] in diameter or larger, in SCC susceptible locations on the canister surface that includes areas adjacent to fabrication welds, closure welds, locations where temporary attachments may have been welded to and subsequently removed from the stainless steel dry storage canister, and the weld heat-affected zones linear appearance of any color of corrosion products of any size parallel to or traversing fabrication welds, closure welds, locations where temporary attachments may have been welded to and subsequently removed from the stainless steel dry

Table 6-2 Example aging management program for Localized Corrosion And Stress Corrosion Cracking Of Welded Stainless Steel Dry Storage Canisters

Element	Description
	<p>storage canister, and the weld heat-affected zones of these areas</p> <ul style="list-style-type: none"> • red-orange-colored corrosion products greater than 1 mm [0.039 in] in diameter or red-orange-colored corrosion tubercles of any size combined with deposit accumulations in SCC susceptible locations on the stainless steel canister • red-orange-corrosion products present at the mouth of a crevice that includes a portion of the canister surface <p>Alternative acceptance criteria may be provided. For such cases, the acceptance criteria should:</p> <ul style="list-style-type: none"> • include a quantitative basis (justifiable by operating experience, engineering analysis, consensus codes/standards) • avoid the use of nonquantifiable phrases (e.g., significant, moderate, minor, little, slight, few) • be achievable and clearly actionable
<p>7. Corrective Actions</p>	<p>Results that do not meet the acceptance criteria are addressed as conditions adverse to quality or significant conditions adverse to quality under those specific portions of the specific- or general-licensee quality assurance (QA) program approved under 10 CFR Part 72, Subpart G, or 10 CFR Part 50, Appendix B, respectively. The QA program ensures that corrective actions are completed within the specific- or general-licensee’s Corrective Action Program (CAP), and include provisions to</p> <ul style="list-style-type: none"> • perform functionality assessments • perform apparent cause evaluations and root cause evaluations • address the extent of condition • determine actions to prevent recurrence for significant conditions adverse to quality; ensure justifications for nonrepairs • trend conditions • identify operating experience actions, including modification to the existing AMP (e.g., increased frequency) • determine if the condition is reportable to the NRC per 10 CFR 72.75 <p><u>Extent of Condition</u></p> <p>Confirmation of localized corrosion or SCC may warrant inspection of additional canisters at the same ISFSI location to determine the extent of condition. Priority for additional inspections should be to canisters with similar time in service and initial loading. Canisters with confirmed localized corrosion or SCC must be evaluated for continued service. Canisters with localized corrosion or SCC that do</p>

Table 6-2 Example aging management program for Localized Corrosion And Stress Corrosion Cracking Of Welded Stainless Steel Dry Storage Canisters	
Element	Description
	not meet the prescribed evaluation criteria are not permitted to remain in service without an engineering analysis or mitigation actions.
8. Confirmation Process	<p>The confirmation process will be commensurate with the specific or general licensee QA program approved under 10 CFR Part 72, Subpart G, or 10 CFR Part 50, Appendix B. The QA program ensures that the confirmation process includes provisions to preclude repetition of significant conditions adverse to quality.</p> <p>The confirmation process describes or references procedures to:</p> <ul style="list-style-type: none"> • determine followup actions to verify effective implementation of corrective actions • monitor for adverse trends due to recurring or repetitive findings or observations
9. Administrative Controls	<p>The administrative controls are in accordance with the specific- or general- licensee QA program approved under 10 CFR Part 72, Subpart G, or 10 CFR Part 50, Appendix B, respectively. The QA program ensures that administrative controls include provisions that define:</p> <ul style="list-style-type: none"> • instrument calibration and maintenance • inspector requirements • record retention requirements • document control <p>The administrative controls describe or reference:</p> <ul style="list-style-type: none"> • methods for reporting results to NRC per 10 CFR 72.75 • frequency for updating an AMP based on site-specific, design-specific, and industrywide operational experience
10. Operating Experience	<p>The AMP references and evaluates applicable operating experience, before renewal, and will continue to do so as new operating experience is developed and made available after renewal, including:</p> <ul style="list-style-type: none"> • internal and industrywide condition reports • internal and industrywide corrective action reports • vendor-issued safety bulletins • NRC generic communications • applicable U.S. Department of Energy (DOE) or industry initiatives (e.g., EPRI- or DOE-sponsored inspections) <p>The AMP clearly identifies any degradation in the referenced operating experience as either age related or event driven, with</p>

Table 6-2 Example aging management program for Localized Corrosion And Stress Corrosion Cracking Of Welded Stainless Steel Dry Storage Canisters

Element	Description
	<p>proper justification for that assessment. Past operating experience supports the adequacy of the proposed AMP, including the method/technique, acceptance criteria, and frequency of inspection.</p> <p>The AMP references the methods for capturing operating experience from other ISFSIs with similar in-scope SSCs.</p> <p>CISCC of austenitic stainless steels is a known degradation mechanism for aqueous environments; however, operational experience in aqueous environments is not directly applicable in assessing the potential for atmospheric CISCC for austenitic stainless steel dry storage canisters. Atmospheric CISCC of austenitic stainless steels has been reported in a range of industries, including welded stainless steel components and piping in operating nuclear power plants.</p> <p><u>Spent Fuel Storage</u></p> <p>Inspections of dry storage canisters after 20 years in service have been conducted at a few ISFSI sites. Details of the inspection conducted at nuclear power plant ISFSIs are documented in EPRI and Sandia National Laboratories reports (Waldrop et al., 2016; 2014; Bryan and Enos, 2014). No evidence of localized corrosion was identified but some amount of chloride-containing salts were determined to be present and corrosion products believed to be related to iron contamination were identified at the Calvert Cliffs ISFSI.</p> <p><u>Operating Power Reactors</u></p> <p>NRC Information Notice 2012-20 (NRC, 2012) documents previous cases of atmospheric CISCC of welded stainless steel piping systems and tanks at operating reactor locations. Atmospheric CISCC growth rates determined from operational experience at both domestic and foreign nuclear power plants, including events at San Onofre, Turkey Point, St. Lucie, and Koeberg (South Africa), range from 3.6×10^{-12} m/sec to 2.9×10^{-11} m/sec for components at ambient temperatures.</p> <p><u>Relevant Literature and Testing</u></p> <p>EPRI has recently conducted a literature review of CISCC that summarizes the results of many previous laboratory investigations (Gorman et al., 2014).</p> <p>The NRC has recently published the results of a completed investigation of CISCC testing of type 304, 304L, and 316L stainless</p>

Table 6-2 Example aging management program for Localized Corrosion And Stress Corrosion Cracking Of Welded Stainless Steel Dry Storage Canisters

Element	Description
	<p>steel and welds (He et al., 2014). This study indicates that SCC was initiated at stresses just above the yield strength in tests conducted using 304 stainless steel C-ring specimens. Testing with U-bend specimens showed that CISCC was observed with the lowest simulated sea salt concentrations tested (100 mg salt/m² or ~55 mg chloride/m²) at temperatures of 52 degrees C [125.6 degrees F] using a maximum absolute humidity of 30 g/m³, which is generally accepted as being near the maximum absolute humidity in a natural environment.</p> <p>Both laboratory and field investigations have been conducted by CRIEPI and TEPCO. This includes the early work by Tokiwai et al. (1985), who reported the critical surface chloride concentrations of 8 mg/m² for CISCC on sensitized 304 stainless steel. Kosaki (2008) reported crack growth rates of 9.6×10^{-12} m/sec obtained in natural exposure tests on Miyakojima Island with type 304 base metals and welds, type 304L welds, and type 316LN welds. Hayashibara et al. (2008) reported activation energy for crack growth in type 304 stainless steel of 5.6 to 9.4 kcal/mol [23 to 39 kJ/mol], based on testing conducted at temperatures of 50 to 80 degrees C [122 to 176 degrees F].</p>
References	<p>ASME. "Boiler and Pressure Vessel Code Section XI—Rules for Inservice Inspection of Nuclear Power Plant Components." New York, New York: American Society of Mechanical Engineers. 2007.</p> <p>Bryan, C.R. and D.G. Enos. SAND2014-16383, "Analysis of Dust Samples Collected From Spent Nuclear Fuel Interim Storage Containers at Hope Creek, Delaware, and Diablo Canyon, California." Albuquerque, New Mexico: Sandia National Laboratories. July 2014.</p> <p>Cumblidge, S.E., M.T. Anderson, and S.R. Doctor. NUREG/CR-6860, "An Assessment of Visual Testing." ADAMS Accession No. ML043630040. Richland, Washington. Pacific Northwest National Laboratory. 2004.</p> <p>Cumblidge, S.E., M.T. Anderson, S.R. Doctor, F.A. Simonen, and A.J. Elliot. NUREG/CR-6943, "A Study of Remote Visual Methods to Detect Cracking in Reactor Components." ADAMS Accession No. ML073110060. Richland, Washington. Pacific Northwest National Laboratory. 2007.</p> <p>Fuhr, K., J. Broussard, and G. White. "Susceptibility Assessment Criteria for Chloride-Induced Stress Corrosion Cracking (CISCC) of Welded Stainless Steel Canisters for Dry Cask Storage Systems."</p>

Table 6-2 Example aging management program for Localized Corrosion And Stress Corrosion Cracking Of Welded Stainless Steel Dry Storage Canisters

Element	Description
	<p>EPRI-3002005371. Palo Alto, California: Electric Power Research Institute. 2015.</p> <p>Fuhr, K., J. Broussard, and G. White. "Aging Management Guidance to Address Potential Chloride-Induced Stress Corrosion Cracking of Welded Stainless Steel Canisters," EPRI-3002008193. Palo Alto, California: Electric Power Research Institute. 2017.</p> <p>Gorman, J., K. Fuhr, and J. Broussard. "Literature Review of Environmental Conditions and Chloride-Induced Degradation Relevant to Stainless Steel Canisters in Dry Cask Storage Systems." EPRI-3002002528. Palo Alto, California: Electric Power Research Institute. 2014.</p> <p>Hayashibara, H., M. Mayuzumi, Y. Mizutani, and J. Tani. "Effect of Temperature and Humidity on Atmospheric Stress Corrosion Cracking of Stainless Steel." <i>Corrosion 2008</i>. Paper 08492, Houston, Texas: NACE International. 2008.</p> <p>He, X., T.S. Mintz, R. Pabalan, L. Miller, and G. Oberson. "Assessment of Stress Corrosion Cracking Susceptibility for Austenitic Stainless Steels Exposed to Atmospheric Chloride and Non-Chloride Salts." NUREG/CR-7170. ADAMS Accession No. ML14051A417. Washington, DC. U.S. Nuclear Regulatory Commission, February 2014,</p> <p>Kosaki, A. "Evaluation Method of Corrosion Lifetime of Conventional Stainless Steel Canister Under Oceanic Air Environment." <i>Nuclear Engineering and Design</i>. Vol. 238. pp.1,233–1,240. 2008.</p> <p>NRC. "Information Notice 2012-20: "Potential Chloride-Induced Stress Corrosion Cracking of Austenitic Stainless Steel and Maintenance of Dry Cask Storage System Canisters." ADAMS Accession No. ML12319A440. Washington, DC: U.S. Nuclear Regulatory Commission. 2012.</p> <p>Selby, G. "BWR Vessel and Internals Project, Reactor Pressure Vessel and Internals Examination Guidelines." EPRI 1011689, TR-105696-R8 (BWRVIP-03) Rev. 8. Palo Alto, California: Electric Power Research Institute. 2005.</p> <p>Tokiwai, M., H. Kimura, and H. Kusanagi. "The Amount of Chlorine Contamination for Prevention of Stress Corrosion Cracking in Sensitized Type 304 Stainless Steel." <i>Corrosion Science</i>. Vol. 25, Issue 8–9. pp. 837–844. 1985.</p>

Table 6-2 Example aging management program for Localized Corrosion And Stress Corrosion Cracking Of Welded Stainless Steel Dry Storage Canisters	
Element	Description
	<p>Waldrop, K., C. Bryan, D. Enos, "Diablo Canyon Stainless Steel Dry Storage Canister Inspection," EPRI-3002002822, Palo Alto, CA: EPRI, 2016.</p> <p>Waldrop, K., W. Bracey, K. Morris, C. Bryan, and D. Enos. "Calvert Cliffs Stainless Steel Dry Storage Canister Inspection." EPRI-1025209. Palo Alto, California: Electric Power Research Institute. 2014.</p>

1

1 **6.6 Reinforced Concrete Structures**

2 An example AMP for reinforced concrete structures is provided below. The AMP consists of
3 condition monitoring, performance monitoring, and mitigation and prevention activities. The
4 program includes periodic visual inspections by personnel qualified to monitor reinforced
5 concrete for applicable aging effects, such as those described in the American Concrete
6 Institute (ACI) guides ACI 349.3R-02, ACI 201.1R-08, and American National Standards
7 Institute/American Society of Civil Engineers guidelines (ANSI/ASCE) 11-99. Identified aging
8 effects are evaluated against acceptance criteria derived from the design bases or industry
9 guides and standards, including ACI 349, ACI 318, ACI 349.3R-02 and ASME Code Section XI,
10 Subsection IWL.

11 The program also includes periodic sampling and testing of groundwater and the need to
12 assess the impact of any changes in its chemistry on below-grade concrete structures.
13 Additional activities include radiation surveys to ensure the shielding functions of the concrete
14 structure are maintained and daily inspections to ensure the air convection vents are not
15 blocked (per the requirements of the approved design bases). The program also includes
16 provisions where modifications may be appropriate.

17

Table 6-3 Example aging management program for Reinforced Concrete Structures

Element	Description
1. Scope of Program	<p>The scope of the program includes the following aging management activities:</p> <ol style="list-style-type: none"> 1. visual inspection of above-grade (readily accessible, normally inaccessible) and below-grade (underground) concrete areas (see Element 4 for sample size and justification of areas to be inspected) 2. groundwater chemistry monitoring program to identify conditions conducive to the following below-grade (underground) aging mechanisms: <ul style="list-style-type: none"> • corrosion of embedded steel • chemical attack (chloride- and sulfate-induced degradation) 3. radiation surveys¹ to: <ul style="list-style-type: none"> • ensure compliance with 10 CFR 72.104 (i.e., dose equivalent requirements beyond the controlled area during normal and off-normal conditions of storage) • monitor performance of the concrete as a neutron/gamma shield at near-system locations as an indicator of concrete degradation <p>The program provides means to adequately identify the following aging effects, as described in ACI 349.3R-02 (ACI, 2010) and SEI/ASCE 11-99 (SEI/ASCE, 2000):</p> <ul style="list-style-type: none"> • cracking or loss of material (spalling, scaling) due to Freeze and thaw degradation • cracking, loss of material (spalling, scaling), loss of strength and reduction of concrete pH (corrosion resistance of steel reinforcement) due to aggressive chemical attack • cracking and loss of strength due to reaction with aggregates • cracking, loss of material, and loss of strength due to corrosion of embedded steel • increase in porosity/permeability, loss of strength, and reduction in concrete pH due to leaching of calcium hydroxide • cracking due to differential settlement • loss of material (spalling, scaling) due to salt scaling • loss of material (spalling, scaling), loss of strength, increased porosity and permeability, and reduction in concrete pH

¹The NRC reviewer should consider the design features of the DSS when determining if radiation surveys can be excluded from the scope of this AMP on a case-by-case basis. For example, internal surfaces of a concrete overpack may be permanently blocked by a steel liner, which may prevent assessing the condition of those concrete surfaces by remote visual inspection. The NRC reviewer should evaluate any engineering justification and/or operating experience to determine if visual inspections of readily accessible and normally inaccessible (i.e., not permanently blocked) surfaces can adequately characterize the condition of the structure and provide reasonable assurance that the intended functions are maintained during the period of extended operation.

Table 6-3 Example aging management program for Reinforced Concrete Structures

Element	Description
	<p>(corrosion resistance of steel reinforcement) due to microbiological degradation</p> <p>Additional site-specific AMPs may be required for the following scenarios:</p> <ul style="list-style-type: none"> • A dewatering system is used to prevent long-term settlement. • The design bases include embedded aluminum subcomponents without a protective insulating coating. • Protective coatings are relied upon to manage the effects of aging for a subcomponent.
<p>2. Preventive Actions</p>	<p>Preventive actions include continuance of inspections to ensure that air inlet/outlet vents are not blocked and/or temperature monitoring, if applicable, to ensure design temperature limits are not exceeded (see Section 6.8, AMP on Ventilation Systems). These inspections would be part of the approved design bases and be continued for the sample size and inspection frequency identified in the respective technical specification (TS).</p> <p>Additional preventive actions are not required for structures designed and fabricated in accordance with ACI 318 (ACI, 2011) or ACI 349 (ACI, 2007a), as specified in the design bases. Otherwise, a site-specific AMP may be required.</p>
<p>3. Parameters Monitored or Inspected</p>	<p>For visual inspections, the parameters monitored or inspected quantify the following aging effects:</p> <ul style="list-style-type: none"> • cracking • loss of material (spalling, scaling) • loss of bond • increased porosity/permeability <p>AMP procedures reference the following parameters for characterizing the above aging effects, as appropriate, per the acceptance criteria:²</p> <ul style="list-style-type: none"> • affected surface area • geometry/depth of defect • cracking, crazing, delaminations, drummy areas • curling, settlements or deflections • honeycombing, bug holes • popouts and voids • exposure of embedded steel • staining/ evidence of corrosion

²The terminology is consistent with ACI standard CT-13 (ACI, 2013b).

Table 6-3 Example aging management program for Reinforced Concrete Structures

Element	Description
	<ul style="list-style-type: none"> • dusting, efflorescence of any color <p>The parameters evaluated consider any surface geometries that may support water ponding and potentially increase the rate of degradation.</p> <p>For the groundwater chemistry program, the parameters monitored or inspected include:</p> <ul style="list-style-type: none"> • water pH • concentration of chlorides and sulfates in the water <p>For radiation surveys, the parameters monitored or inspected include gamma dose rate and neutron fluence rate.</p>
<p>4. Detection of Aging Effects</p>	<p><u>Method or technique</u></p> <p>Visual inspections of readily accessible areas are performed with feeler gauges, crack comparators, or other suitable visual quantification methods per the acceptance criteria in ACI 349.3R-02 (ACI, 2010).</p> <p>Visual inspections of normally inaccessible areas are performed using a remote inspection system that has been qualified for the specific DSS and site-specific characteristics. Procedures for remote visual inspections should be demonstrated to ensure the acceptance criteria in ACI 349.3R-02 (ACI, 2010) are achievable; procedure attributes should include, for example, equipment resolution and lighting requirements and should reference applicable standards when possible.</p> <p>Groundwater chemistry is characterized using a chemical analysis method with a valid measurement range and adequate resolution and sensitivity. Procedures for groundwater chemistry analyses should be demonstrated to ensure the acceptance criteria in ASME Code Section XI, Subsection IWL, are achievable</p> <p>Radiation surveys are performed using calibrated neutron and gamma detectors with valid energy ranges, per the acceptance criteria (see Element 6).</p> <p>Procedure attributes for all inspection and monitoring activities within the scope of this program should be commensurate with 10 CFR 72.164 and 10 CFR Part 50, Appendix B, as appropriate.</p>

Table 6-3 Example aging management program for Reinforced Concrete Structures

Element	Description
	<p><u>Frequency of Inspection</u></p> <p>The schedule for visual inspections is commensurate with ACI 349.3R-02 (ACI, 2010). Alternative inspection frequencies must be adequately justified by a valid technical basis (engineering justification, operational experience data).</p> <p>Inspections of above-grade (both readily accessible and normally inaccessible) areas are conducted at least once every 5 years. The inspections of below-grade (underground) areas are opportunistic; inspections are performed when excavations occur for any reason.</p> <p>The frequency for monitoring groundwater chemistry is justified (e.g., quarterly, semiannually), per an adequate technical basis (site-specific operating experience, engineering justification).</p> <p>The frequency for radiation surveys is justified (e.g., quarterly), per an adequate technical basis (engineering justification, operating experience).</p> <p><u>Sample size</u></p> <p>Visual inspections cover 100 percent of readily accessible surfaces (or a justified coverage) of all concrete structures within the scope of renewal (e.g., all normally accessible exterior surfaces of all loaded overpacks), and 100 percent of normally inaccessible surfaces (or a justified coverage) for a justified subset of the reinforced concrete structures within the scope of renewal (e.g., interior surfaces of two overpacks, including the overpack earliest loaded and the overpack loaded with the highest heat-load canister). The extent of inspection coverage should be specified and demonstrated to sufficiently characterize the condition of the structure.</p> <p>For the groundwater chemistry program and radiation surveys, the sample size identifies and justifies specific locations where inspection or monitoring will be conducted to sufficiently characterize the condition of the structure (e.g., periodic dose rate measurements will be performed at the same locations specified in the TS for dose rate measurements at loading).</p> <p><u>Data collection:</u></p> <p>Data collection for visual inspections is commensurate with consensus standards and guides (see ACI 224.1R (ACI, 2007b) for quantitative analysis (crack width, extent), ACI 562, (ACI, 2013a), ACI 364.1R (ACI, 2007c)).</p>

Table 6-3 Example aging management program for Reinforced Concrete Structures

Element	Description
	<p>Data from all inspection and monitoring activities, including evidence of degradation and its extent and location, shall be documented on a checklist or inspection form. The results of the inspection shall be documented, including descriptions of observed aging effects and supporting sketches, photographs, or video.</p> <p>Corrective actions from AMP activities shall also be documented. An adequate clearinghouse is used for documenting inspection and monitoring operating experience.</p> <p><u>Timing</u></p> <p>Initial inspections and monitoring activities are completed before entering the period of extended operation; the activities may be part of a preapplication inspection or a general-licensee baseline inspection (see NUREG–1927, Rev. 1 (NRC, 2016)).</p>
<p>5. Monitoring and Trending</p>	<p>Monitoring and trending methods are commensurate with consensus defect evaluation guides and standards (see ACI 201.1R (ACI, 2008a), ACI 207.3R (ACI, 2008b), ACI 364.1R (ACI, 2007c), ACI 562 (ACI, 2013a), or ACI 224.1R (ACI, 2007b) for crack evaluation).</p> <p>Inspection and monitoring results are compared to those obtained during previous inspections, so that the progression of degradation can be evaluated and predicted.</p> <p>Monitoring and trending methods reference plans and procedures used to:</p> <ul style="list-style-type: none"> • establish a baseline before or at the beginning of the period of extended operation • track trending of parameters or effects not corrected in a previous inspection, for example <ul style="list-style-type: none"> — crack growth/extent — pore/void density and affected areas — dose rates
<p>6. Acceptance Criteria</p>	<p>The acceptance criteria for visual inspections are commensurate with the 3-tier quantitative criteria in ACI 349.3R-02:</p> <ul style="list-style-type: none"> • Tier 1: acceptance without further evaluation • Tier 2: acceptance after review • Tier 3: acceptance requiring further evaluation <p>All conditions not meeting the Tier 2 acceptance criteria are evaluated in the Corrective Action Program (CAP) to reasonably</p>

Table 6-3 Example aging management program for Reinforced Concrete Structures

Element	Description
	<p>ensure that the intended functions of the structure will be adequately maintained until a followup inspection, at a minimum.</p> <p>The acceptance criteria for the groundwater chemistry program are commensurate with ASME Code Section XI, Subsection IWL, which states that an aggressive below-grade environment is defined as pH < 5.5, chlorides > 500 ppm, or sulfates > 1500 ppm.</p> <p>The acceptance criteria for radiation surveys are justified and sufficient to ensure compliance with 10 CFR 72.104 and identify dose rates that statistically exceed calculated or expected dose rates at predetermined measurement locations. The adequacy of the acceptance criteria considers measured dose rates versus calculated or expected dose rates for a DSS, given the DSS contents and accounting for the decay of the source term since the DSS loading. Measurement locations should be consistent with those specified in the license or Certificate of Compliance (CoC) conditions or TS (if any) and locations where dose rates were calculated in the final safety analysis report (FSAR) and likely measured at the time of loading.</p> <p>Alternative acceptance criteria should be reviewed on a case-by-case basis. For such cases, the acceptance criteria shall:</p> <ul style="list-style-type: none"> • include a quantitative basis (justifiable by operating experience, engineering analysis, consensus codes and standards) • avoid use of nonquantifiable phrases (e.g., significant, moderate, minor, little, slight, few) • be achievable and clearly actionable
7. Corrective Actions	<p>Results that do not meet the acceptance criteria are addressed as conditions adverse to quality or significant conditions adverse to quality under those specific portions of the specific- or general-licensee QA program approved under 10 CFR Part 72, Subpart G, or 10 CFR Part 50, Appendix B, respectively. The QA program ensures that corrective actions are completed within the specific- or general-licensee's Corrective Action Program (CAP), and include provisions to</p> <ul style="list-style-type: none"> • perform functionality assessments • perform apparent cause evaluations, and root cause evaluations • address the extent of condition • determine actions to prevent recurrence for significant conditions adverse to quality; ensure justifications for nonrepairs

Table 6-3 Example aging management program for Reinforced Concrete Structures

Element	Description
	<ul style="list-style-type: none"> • trend conditions • identify operating experience actions, including modifications to the existing AMP (e.g., increased frequency) • determine if the condition is reportable to the NRC per 10 CFR 72.75 <p>Corrective actions shall be consistent with applicable consensus rehabilitation guides or standards, unless an engineering justification is provided (e.g., for cracking: ACI 224.1R, ACI 562, ACI 364.1R, and ACI RAP Bulletins; for spalling/scaling: ACI 562, ACI 364.1R, ACI 506R, and ACI RAP Bulletins).</p>
<p>8. Confirmation Process</p>	<p>The confirmation process is commensurate with the specific- or general-licensee QA program approved under 10 CFR Part 72, Subpart G, or 10 CFR Part 50, Appendix B, respectively. The QA program ensures that the confirmation process includes provisions to preclude repetition of significant conditions adverse to quality.</p> <p>The confirmation process describes or references procedures to:</p> <ul style="list-style-type: none"> • determine followup actions to verify effective implementation of corrective actions • monitor for adverse trends due to recurring or repetitive findings or observations.
<p>9. Administrative Controls</p>	<p>The administrative controls are in accordance with the specific- or general-licensee QA program approved under 10 CFR Part 72, Subpart G, or 10 CFR Part 50, Appendix B, respectively. The QA program ensures that the administrative controls include provisions that define:</p> <ul style="list-style-type: none"> • instrument calibration and maintenance • inspector requirements (commensurate with ACI 349.3R-02) • record retention requirements • document control <p>The administrative controls describe or reference:</p> <ul style="list-style-type: none"> • methods for reporting results to the NRC per 10 CFR 72.75 • frequency for updating the AMP based on industrywide operational experience
<p>10. Operating Experience</p>	<p>Structures monitoring programs using the acceptance criteria in ACI 349.3R-02 (ACI, 2010) have proven effective for aging management of concrete structures in nuclear power plants during their period of extended operation (NRC, 2010b). NUREG-1522 documents the results of a survey sponsored in 1992 by the Office of Nuclear Reactor Regulation to obtain information on the types of distress in the concrete and steel structures and components, the</p>

Table 6-3 Example aging management program for Reinforced Concrete Structures

Element	Description
	<p>type of repairs performed, and the durability of the repairs. Licensees who responded to the survey reported cracking, scaling, and leaching of concrete structures. The degradation was attributed to drying shrinkage, Freeze and thaw, and abrasion. The NUREG also describes the results of NRC staff inspections at six plants. The staff observed concrete degradation, corrosion of component support members and anchor bolts, cracks and other deterioration of masonry walls, and groundwater leakage and seepage into underground structures. The observed and reported degradations were more severe at coastal plants than those observed in inland plants, as a result of brackish and sea water. Previous reactor license renewal applicants reported similar degradation and corrective actions taken through their structures monitoring program.</p> <p>NRC Information Notice 2011-20 (NRC, 2011) documents the occurrence of alkali-silica reaction (ASR)-induced concrete degradation of a seismic Category 1 below-grade structure at the Seabrook Station power plant. The concrete used in the structure passed all industry standard ASR screening tests (ASTM, 2007, 2012) at the time of construction; however, ASR-induced degradation was identified in August 2010. The licensee completed a prompt operability determination that concluded margins to the design limits remained such that the structural integrity of the building continued to be demonstrated.</p> <p>NRC Information Notice 2013-07 documents the occurrence of Freeze and thaw cracking near the anchor blockout holes on the roof of horizontal storage modules (HSMs) at an ISFSI in Idaho. The cracking led to water migration into the concrete, resulting in efflorescence of calcium carbonate deposits. The degradation of the roofslabs was not related to age-related degradation but to a design feature leading to water accumulation. More extensive visual inspections of the HSMs also revealed map cracking on the vertical wall surfaces, random and radial cracking at the door edges in base units, and spalling at the bottom edge of shield walls. The licensee conducted nondestructive and destructive examination, which revealed adequate concrete quality and compressive strength.</p> <p>Additional visual inspections of concrete structures in DSSs have been conducted at the Calvert Cliffs ISFSI (Gellrich, 2012) and the Palisades ISFSI. Remote visual inspections of two HSMs at the Calvert Cliffs ISFSI revealed efflorescence of the concrete and the formation of calcium carbonate stalactites in the 2-inch gap between the heat shield and the concrete ceiling. These stalactites were attributed to water ingress through the outlet vent stack. A condition report was issued that did not identify an operability issue.</p>

Table 6-3 Example aging management program for Reinforced Concrete Structures

Element	Description
	<p>Inspections of the exterior surfaces of a ventilated concrete cask (VCC) and the concrete support pad at the Palisades ISFSI revealed bugholes exceeding preestablished acceptance criteria and requiring grout repair, and a void at the interface between the VCC bottom plate and the vertical VCC concrete wall. No conditions were identified to compromise the intended functions of the VCC.</p> <p>Walkdowns and visual inspections of readily accesible surfaces of concrete overpacks and HSMs are generally conducted during the initial storage period, although the acceptance criteria may vary from those in ACI 349.3R.02 (ACI, 2010). The NRC reviewer should evaluate relevant inspection results included in the renewal application, based on design and environmental similarities, and evaluate if activities in this generic AMP should be augmented as a result of those inspections.</p>
References	<p>ACI. ACI 506R-05, "Guide to Shortcrete." American Concrete Institute. 2005.</p> <p>_____. ACI 349-06, "Code Requirements for Nuclear Safety-Related Concrete Structures." American Concrete Institute. 2007a.</p> <p>_____. ACI 224.1R-07, "Causes, Evaluation, and Repair of Cracks in Concrete Structures." American Concrete Institute. 2007b.</p> <p>_____. ACI 364.1R-07, "Guide for Evaluation of Concrete Structures before Rehabilitation." American Concrete Institute. 2007c.</p> <p>_____. ACI 201.1R-08, "Guide for Conducting a Visual Inspection of Concrete in Service." American Concrete Institute. 2008a.</p> <p>_____. ACI 207.3R-94, "Practices for Evaluation of Concrete in Existing Massive Structures for Service Conditions." American Concrete Institute. 2008b.</p> <p>_____. ACI 349.3R-02, "Evaluation of Existing Nuclear Safety-Related Concrete Structures." American Concrete Institute. 2010.</p> <p>_____. ACI 318-11, "Building Code Requirements for Structural Concrete." American Concrete Institute. 2011.</p> <p>_____. ACI 562-13, "Code Requirements for Evaluation, Repair, and Rehabilitation of Concrete Buildings." American Concrete Institute. 2013a.</p> <p>_____. ACI CT-13, "ACI Concrete Terminology." American Concrete Institute. 2013b.</p>

Table 6-3 Example aging management program for Reinforced Concrete Structures

Element	Description
	<p>ASME Boiler and Pressure Vessel Code, Section XI, Subsection IWL (2013), "Requirements for Class CC Concrete Components of Light-Water-Cooled Plants"</p> <p>ASTM International. ASTM C289, "Standard Test Method for Potential Alkali-Silica Reactivity of Aggregates (Chemical Method)." West Conshohocken, Pennsylvania: American Society for Testing and Materials. 2007.</p> <p>_____. ASTM C295, "Standard Guide for Petrographic Examination of Aggregates for Concrete." West Conshohocken, Pennsylvania: American Society for Testing and Materials. 2012.</p> <p>Gellrich, G. "Calvert Cliffs Nuclear Power Plant." Letter to U.S. Nuclear Regulatory Commission, Response to Request for Supplemental Information. RE: Calvert Cliffs Independent Spent Fuel Storage Installation License Renewal Application (TAC No. L24475). ADAMS Accession No. ML12212A216. 2012.</p> <p>NRC. "Standard Review Plan for Spent Fuel Dry Storage Facilities." NUREG-1567, Rev. 0. Washington, DC. ADAMS Accession No. ML003686776. 2000.</p> <p>_____. "Standard Review Plan for Spent Fuel Dry Storage Systems at a General License Facility." NUREG-1536, Rev. 1. ADAMS Accession No. ML091060180. Washington, DC. 2010a.</p> <p>_____. NUREG-1801, "Generic Aging Lessons Learned (GALL) Report." Rev. 2, Washington DC. ADAMS Accession No. ML103490041. 2010b.</p> <p>_____. "Information Notice 2011-20, Concrete Degradation by Alkali-Silica Reaction." Washington, DC: U.S. Nuclear Regulatory Commission. ADAMS Accession No. ML112241029. 2011.</p> <p>_____. NUREG-1927, "Standard Review Plan for Renewal of Specific Licenses and Certificates of Compliance for Dry Storage of Spent Nuclear Fuel." Revision 1. Washington, DC: U.S. Nuclear Regulatory Commission. ADAMS Accession No. ML16179A148. 2016.</p> <p>SEI/ASCE 11-99, "Guideline for Structural Condition Assessment of Existing Buildings." 2000.</p>

1 **6.7 External Surfaces Monitoring Of Metallic Components**

2 An example AMP for external surfaces monitoring of metallic components is provided below.

3 The AMP manages all metallic surfaces that are directly exposed to outdoor air, concrete, or are
4 sheltered within DSS overpacks, except for stainless steel storage canisters and transfer casks,
5 which are addressed by other AMPs. The AMP is a condition monitoring program that consists
6 of periodic visual inspections to monitor for corrosion, wear, cracking, and loss of preload
7 (bolting).

8

Table 6-4 Example aging management program for External Surfaces Monitoring Of Metallic Components

Element	Description
1. Scope of Program	<p>This program manages the effects of aging for the external surfaces of steel and stainless steel components that are directly exposed to outdoor air or are sheltered within DSS overpacks (e.g., NUHOMS HSM, HI-STORM). The scope of the program includes metallic overpack exterior surfaces, dry storage canister support structures, access doors, vents, heat shields, embedments and anchorages, bolting, and other components important to safety.</p> <p>The scope of this program does not include stainless steel dry storage canisters housed within overpacks, transfer casks, or the top closure (confinement) boundary of bolted casks. The Localized Corrosion and Stress corrosion Cracking of Welded Stainless Steel Dry Storage Canisters AMP manages the effects of aging for stainless steel canisters. The Transfer Casks AMP manages the effects of aging of all transfer cask components. The Bolted Cask Seal Leakage Monitoring AMP manages the effects of aging on the integrity of the top confinement boundary of bolted spent fuel storage casks.</p> <p>Periodic visual inspections monitor for general and localized corrosion, wear, cracking, and loss of preload (bolting).</p>
2. Preventive Actions	<p>This program is a condition monitoring program to detect evidence of degradation. It does not provide guidance for the prevention of aging.</p>
3. Parameters Monitored/ Inspected	<p>This program monitors the condition of external metallic surfaces to identify general corrosion, localized corrosion, wear, and loss of preload of bolted connections. Localized corrosion of stainless steels may be a precursor to SCC.</p> <p>Parameters monitored or inspected for external metallic surfaces include:</p> <ul style="list-style-type: none"> • visual evidence of discontinuities, imperfections, and rust staining indicative of corrosion, SCC, and wear • visual evidence of loose or missing bolts, physical displacement, and other conditions indicative of loss of preload • visual evidence of coating degradation (e.g., blisters, cracking, flaking, delamination) indicative of corrosion of the base metal
4. Detection of Aging Effects	<p><u>Readily Accessible Surfaces</u></p> <p>Visual inspections are performed in accordance with ASME Code Section XI, Article IWA-2213, for VT-3 examinations. The inspections cover 100 percent of normally accessible surfaces, including the external surfaces of metallic overpacks, bolting, lightning protection system components, access doors, vents, and other metallic components.</p>

Table 6-4 Example aging management program for External Surfaces Monitoring Of Metallic Components

Element	Description
	<p><u>Normally Inaccessible Surfaces</u></p> <p>Opportunistic visual inspections are performed with remote inspection techniques on metallic surfaces within overpacks that are accessed during inspections of dry storage canisters, including heat shields, canister support structures, and other metallic components.</p> <p>The condition of metallic surfaces in contact with concrete (i.e., overpack/cask bottoms) are assessed with visual inspections on a justified frequency.</p> <p>Procedures for visual inspections should be demonstrated; procedure attributes should include, for example, equipment resolution and lighting requirements and should reference applicable standards (e.g., ASTM Code Section XI, Article IWA-2200, for VT-3 examinations). The extent of inspection coverage should be specified and demonstrated to sufficiently characterize the condition of the metallic components.</p> <p><u>Sample Size</u></p> <p>The readily accessible exterior metallic surfaces of all casks and overpacks are inspected. The inspections of normally inaccessible surfaces within overpacks is opportunistic; inspections are performed whenever the overpacks are accessed for dry storage canister inspections. Overpack and cask bottoms are inspected on a justified sample size.</p> <p><u>Frequency</u></p> <p>Inspections of readily accessible surfaces are conducted at least once every 5 years. Normally inaccessible surfaces within overpacks are inspected when those surfaces are accessed during remote inspections of dry storage canisters. Overpack and cask bottoms are inspected on a justified frequency.</p> <p><u>Data Collection</u></p> <p>Data from the examination, including evidence of degradation and its extent and location, shall be documented on a checklist or inspection form. The results of the inspection shall be documented, including descriptions of observed aging effects and supporting sketches, photographs, or video. Corrective actions resulting from each AMP inspection shall also be documented.</p> <p><u>Timing</u></p> <p>Initial inspections are completed before entering the period of extended operation.</p>

Table 6-4 Example aging management program for External Surfaces Monitoring Of Metallic Components

Element	Description
5. Monitoring and Trending	<p>Inspection results are compared to those obtained during previous inspections, so that the progression of degradation can be evaluated and predicted.</p> <p>Monitoring and trending methods reference plans and procedures used to:</p> <ul style="list-style-type: none"> • establish a baseline before or at the beginning of the period of extended operation • track trending of parameters or effects not corrected following a previous inspection, including <ul style="list-style-type: none"> — locations and size of any areas of corrosion, wear, or cracking — disposition of components with identified aging effects and the results of supplemental inspections
6. Acceptance Criteria	<p>The acceptance criteria for the visual inspections are:</p> <ul style="list-style-type: none"> • no detectable loss of material from the base metal, including uniform wall thinning, localized corrosion pits, and crevice corrosion • no red-orange-colored corrosion products on the base metal, coatings, or concrete • no coating defects (e.g., blisters, cracking, flaking, delamination) • no indications of loose bolts or hardware, displaced parts <p>If evidence of corrosion, wear, or coating degradation is identified, then the severity of the degradation must be determined using approved site-specific procedures. These may include additional visual, surface or volumetric nondestructive examination (NDE) methods to determine the loss of material and, for welded stainless steels, the presence of cracking.</p> <p>Alternative acceptance criteria are developed from system-specific design standards, industry codes or standards, or engineering evaluation. Where possible, acceptance criteria are quantitative (e.g., minimum wall thickness). Where qualitative acceptance criteria are used, the criteria are sufficiently clear to reasonably ensure that a singular decision is derived based on the observed condition, avoiding the use of ambiguous phrases (e.g., significant, moderate).</p> <p>EPRI technical reports, Technical Report (TR)-1007933, "Aging Assessment Field Guide" (EPRI, 2003), and TR-1009743, "Aging Identification and Assessment Checklist: Mechanical Components" (EPRI, 2004), provide general guidance for the evaluation of materials and the development of criteria for their acceptance when performing visual inspections.</p>

Table 6-4 Example aging management program for External Surfaces Monitoring Of Metallic Components

Element	Description
7. Corrective Actions	<p>Results that do not meet the acceptance criteria are addressed as conditions adverse to quality or significant conditions adverse to quality under those specific portions of the specific- or general- licensee QA program approved under 10 CFR Part 72, Subpart G, or 10 CFR Part 50, Appendix B, respectively. The QA program ensures that corrective actions are completed within the specific- or general- licensee’s Corrective Action Program (CAP), and include provisions to</p> <ul style="list-style-type: none"> • perform functionality assessments • perform apparent cause evaluations and root cause evaluations • address the extent of condition • determine actions to prevent recurrence for significant conditions adverse to quality; ensure justifications for nonrepairs • trend conditions • identify operating experience actions, including modification to the existing AMP (e.g., increased frequency) • determine if the condition is reportable to the NRC per 10 CFR 72.75
8. Confirmation Process	<p>The confirmation process is commensurate with the specific- or general- licensee QA program approved under 10 CFR Part 72, Subpart G, or 10 CFR Part 50, Appendix B, respectively. The QA program ensures that the confirmation process includes provisions to preclude repetition of significant conditions adverse to quality.</p> <p>The confirmation process describes or references procedures to:</p> <ul style="list-style-type: none"> • determine followup actions to verify effective implementation of corrective actions • monitor for adverse trends due to recurring or repetitive findings or observations.
9. Administrative Controls	<p>The administrative controls are addressed through those portions of the specific- or general- licensee QA program that are used to meet 10 CFR Part 72, Subpart G, or 10 CFR Part 50, Appendix B.</p>
10. Operating Experience	<p>External surface inspections through system inspections and walkdowns in support of the Maintenance Rule (10 CFR Part 50.65) have proven effective in maintaining the material condition of nuclear power plant systems.</p> <p>NRC Information Notice 2012-20 (NRC, 2012) documents cases of atmospheric CISCC of welded stainless steel piping systems and tanks at operating reactor locations. Atmospheric CISCC growth rates determined from operational experience at both domestic and foreign nuclear power plants, include events at San Onofre, Turkey Point,</p>

Table 6-4 Example aging management program for External Surfaces Monitoring Of Metallic Components	
Element	Description
	St. Lucie, and Koeberg (South Africa), range from 3.6×10^{-12} to 2.9×10^{-11} m/sec for components at ambient temperatures.
References	<p>EPRI. EPRI Technical Report 1007933, "Aging Assessment Field Guide." Palo Alto, California: Electric Power Research Institute. December 2003.</p> <p>_____. EPRI Technical Report 1009743, "Aging Identification and Assessment Checklist–Mechanical Components." Palo Alto, California: Electric Power Research Institute. August 27, 2004.</p> <p>NRC. NRC Information Notice 2012-20, "Potential Chloride-Induced Stress Corrosion Cracking of Austenitic Stainless Steel and Maintenance of Dry Cask Storage System Containers." Washington, DC: U.S. Nuclear Regulatory Commission. November 14, 2012.</p>

1

2

1 **6.8 Ventilation Systems**

2 An example AMP for ventilation systems is provided below. The AMP manages all inlet and
3 outlet vents and conduits providing convective cooling in DSSs. This is a condition monitoring
4 program that performs periodic visual inspection of vents as defined in the approved design
5 bases, with additional focused inspections to address normally unobservable vent areas, as well
6 as evidence of degradation that could result in obstructions. Temperature monitoring may be
7 used in lieu of the periodic visual surveillances to verify cooling performance.

8

Table 6-5 Example aging management program for Ventilation Systems	
Element	Description
1. Scope of Program	<p>This program manages potential loss of cooling capabilities due to blockage of the ventilation system (air inlet/outlets, convection conduits) in DSSs. Surveillance/monitoring and focused inspections of the ventilation system (i) ensure that blockage does not result in design temperature limits being exceeded and (ii) prevent unanticipated adverse degradation of components of the DSS (e.g., high-temperature dehydration of the concrete¹, hydride reorientation due to fuel cladding temperatures exceeding design-bases limits²).</p> <p><u>Visual Surveillances of Inlet and Outlet Vents</u></p> <p>The scope of the program includes continuance of surveillances (periodic walkdowns) of air inlet/outlet vents, as defined in the approved design bases (FSAR, license/CoC TS). The program provides for additional focused inspections if (i) the normally unobservable vent area exceeds the allowable blockage, and (ii) there is evidence of degradation of other components (e.g., loss of coatings, spalling or leaching of the concrete overpack) that could result in obstructions.³</p> <p><u>Temperature Monitoring</u></p> <p>The scope of the program includes temperature monitoring of DSS components in lieu of visual surveillances, as specified in the approved design bases (FSAR, license/CoC TS). Continuance of temperature monitoring provides a means to detect anomalous temperature changes in the DSS. The program further provides for focused visual inspections of the ventilation system (inlet/outlet vents, conduits) in the event that anomalous temperature changes are measured. Focused visual inspections allow for detection of degradation of other components that could result in obstructions (e.g., loss of coatings, inner spalling or leaching of the concrete overpack).</p> <p>The scope of the program does not include inspection and/or maintenance activities for aging of bird screens used to prevent vent blockage (see the External Surfaces Monitoring of Metallic Components AMP).</p>

¹See NUREG-1536/NUREG-1567 (NRC, 2010, 2002) for design criteria on maximum concrete temperatures.

²See ISG-11, Revision 3 (NRC, 2003), for cladding considerations for the transportation and storage of spent fuel.

³The approved design bases have adequately addressed the occurrence of extreme natural phenomena, such as heavy snowstorm or flooding. The QA program ensures that corrective actions are completed within the specific- or general-licensee's CAP in the event of extreme natural phenomena.

Table 6-5 Example aging management program for Ventilation Systems

Element	Description
2. Preventive Actions	<p>This program is a condition monitoring program to detect obstruction or blockages of the ventilation system that could result in design-bases temperature limits being exceeded. It does not provide guidance for the prevention of aging of components.</p>
3. Parameters Monitored/ Inspected	<p><u>Visual Surveillances of Inlet and Outlet Vents</u></p> <p>Parameters monitored or inspected include blockage or obstruction in the air inlet and outlet vents.</p> <p><u>Temperature Monitoring</u></p> <p>Parameters monitored or inspected include temperature measurements of the DSS, which could be based on (i) direct measurements of the overpack temperatures, (ii) direct measurement of the canister temperatures, (iii) a comparison of the inlet and outlet temperature difference to predicted temperature differences for each individual overpack, or (iv) other means that would identify and allow for the correction of off-normal thermal conditions that could lead to exceeding design-bases temperature limits for the concrete and/or fuel cladding.</p> <p><u>Focused Inspections</u></p> <p>Parameters monitored or inspected include (i) blockage or obstructions of the air inlets/outlets and (ii) degradation of other components (e.g., loss of coatings, inner spalling or leaching of the concrete overpack) that could result in obstructions of inaccessible convective conduits.</p>
4. Detection of Aging Effects	<p><u>Method/Technique</u></p> <p>Visual surveillances of the air inlet/outlet vents are performed during periodic walkdowns, without the need of remote equipment. Surveilling personnel should have an unobstructed view of vent areas that allows confirmation that the maximum allowable blockage is not exceeded (up to the boundary of the vent bird screen, at a minimum). The maximum allowable blockage is defined in the approved design bases (FSAR, license/CoC TS).</p> <p>Temperature monitoring is performed with qualified and calibrated measurement devices or sensors that are maintained in accordance with the site QA program.</p> <p>Focused inspections are performed with remote inspection techniques. Procedures for remote visual inspections should be demonstrated; procedure attributes should include, for example, equipment resolution and lighting requirements, in consideration of the ventilation system design.</p>

Table 6-5 Example aging management program for Ventilation Systems

Element	Description
	<p><u>Frequency of Inspection/Monitoring</u></p> <p>Visual surveillances and temperature monitoring are conducted at a frequency consistent with the approved design bases (i.e., as defined in the FSAR, or the relevant license/CoC TS). Generally, visual surveillances are conducted daily (not exceeding a 48-hour interval) and temperature monitoring is performed continuously.</p> <p>The frequency of focused inspections for vent areas should be justified based on the design (percentage of normally unobservable vent area relative to allowable blockage) and operable degradation modes of the storage system components that could lead to blockage. The frequency of focused inspections provides reasonable assurance that blockages in the normally inaccessible convective conduits will be identified before a loss of function by considering conduit-free volume relative to postulated obstructions (e.g., upon consideration of potential coating loss or concrete spalling relative to conduit-free volume). Previous operating experience may be used to justify the use of opportunistic inspections. When continuous temperature monitoring is used to verify ventilation performance, focused inspections are performed whenever anomalous temperatures are measured.</p> <p><u>Sample Size</u></p> <p>Visual surveillances include all directly observable areas of the inlet and outlet vents. Visual surveillances are performed on all loaded systems, or as justified by the approved design bases (i.e., as defined in the FSAR, or the relevant license/CoC TS).</p> <p>Temperature monitoring is performed in all loaded systems, or as justified by the approved design bases (i.e., as defined in the FSAR, or the relevant license/CoC TS).</p> <p>For focused inspections, the extent of inspection coverage should be specified and demonstrated to sufficiently characterize the condition of the ventilation system. Focused inspections include all normally unobservable vent areas exceeding the allowable blockage. The extent of inaccessible conduit inspection is justified based on the ventilation system design (conduit-free volume, accessibility) and consideration of operable degradation modes of the storage system materials. The use of continuous temperature monitoring may be used in lieu of focused inspections if anomalous temperatures are not measured.</p> <p><u>Data Collection</u></p> <p>Data collection should be consistent with site procedures in compliance with the specific- or general-licensee's QA program.</p>

Table 6-5 Example aging management program for Ventilation Systems

Element	Description
	<p><u>Timing</u></p> <p>A baseline focused inspection is conducted on a sample DSS upon entering the period of extended operation to identify any operable degradation modes that may result in an obstruction or blockage difficult to observe during a visual surveillance. The baseline-focused inspection includes 100 percent of the vents and inaccessible convective conduits of the sample DSS, or a justified volume based on design considerations (e.g., accessibility, dose rates). A baseline-focused inspection on a sample system is not necessary if temperature monitoring is used in lieu of visual surveillances.</p>
<p>5. Monitoring and Trending</p>	<p>Results from visual surveillances and temperature monitoring are trended to identify conditions (materials/environmental) leading to obstructions or blockages.</p> <p>Results from focused inspections are compared with prior inspections to monitor and trend operable degradation modes of the storage system materials that have resulted in partial blockage.</p>
<p>6. Acceptance Criteria</p>	<p>The acceptance criteria are defined to ensure that the need for corrective actions will be identified before (i) blockage results in design temperature limits being exceeded and (ii) unanticipated adverse degradation of components of the DSS results in a loss of intended function. Where possible, acceptance criteria are quantitative (e.g., 50-percent areal blockage or a specific allowed temperature range). Where qualitative acceptance criteria are used, the criteria are sufficiently clear to reasonably ensure that a singular decision is derived based on the observed condition, avoiding the use of ambiguous phrases (e.g., significant, moderate).</p> <p>The acceptance criteria for visual surveillances and focused inspections are justified based on the ventilation system design, thermal performance criteria, and consideration of operable degradation modes of the storage system materials. The acceptance criteria may be further justified by parallel maintenance activities under a separate AMP.</p> <p>The acceptance criteria for temperature monitoring are justified and conservative to the short-term temperature limits for a blocked vent condition, as defined in the approved design bases (i.e., as defined in the FSAR, or the relevant license/CoC TS).</p>
<p>7. Corrective Actions</p>	<p>Results that do not meet the acceptance criteria are addressed as conditions adverse to quality or significant conditions adverse to quality under those specific portions of the specific- or general-licensee QA program approved under 10 CFR Part 72, Subpart G, or</p>

Table 6-5 Example aging management program for Ventilation Systems

Element	Description
	<p>10 CFR Part 50, Appendix B, respectively. The QA program ensures that corrective actions are completed within the specific- or general-licensee's Corrective Action Program (CAP), and include provisions to:</p> <ul style="list-style-type: none"> • perform functionality assessments • perform apparent cause evaluations and root cause evaluations • address the extent of condition • determine actions to prevent recurrence for significant conditions adverse to quality; ensure justifications for nonrepairs • trend conditions • identify operating experience actions, including modification to the existing AMP (e.g., increased frequency) • determine if the condition is reportable to the NRC per 10 CFR 72.75
<p>8. Confirmation Process</p>	<p>The confirmation process is commensurate with the specific- or general-licensee QA program approved under 10 CFR Part 72, Subpart G, or 10 CFR Part 50, Appendix B, respectively. The QA program ensures that the confirmation process includes provisions to preclude repetition of significant conditions adverse to quality.</p> <p>The confirmation process describes or references procedures to:</p> <ul style="list-style-type: none"> • determine followup actions to verify effective implementation of corrective actions • monitor for adverse trends due to recurring or repetitive findings or observations.
<p>9. Administrative Controls</p>	<p>The administrative controls are addressed through those portions of the specific- or general-licensee QA program that are used to meet 10 CFR Part 72, Subpart G, or 10 CFR Part 50, Appendix B.</p> <p>To ensure the temperature monitoring devices will remain accurate during the period of extended operation, the electronic circuitry associated with the temperature monitoring devices should be periodically calibrated in accordance with the licensee's QA requirement in 10 CFR 72.164 and specific-license requirement in 10 CFR 72.44(c)(3)(ii). In addition, the calibration data are periodically evaluated to identify anomalous trends that could indicate degraded instrumentation or degradation in the ventilation system. All external components in the temperature measurement devices should be periodically inspected and calibrated to ensure that no degradation due to corrosion, wear, or cracking has occurred.</p>

Table 6-5 Example aging management program for Ventilation Systems

Element	Description
<p>10. Operating Experience</p>	<p>Visual surveillance of the exterior of the air inlets and outlets, inspections for ventilation blockage and temperature monitoring have been in effect at ISFSIs and have been proven effective in maintaining the convective cooling capabilities of DSSs during the initial license or certification period.</p> <p>Degradation of inner overpack materials has been observed in the field. NRC Information Notice 2013-07 (NRC, 2013) documents experience at the Three Mile Island, Unit 2, ISFSI, where water entered anchor bolt blockout holes on the roof of HSM concrete overpacks. Subsequent freeze and thaw cycles resulted in crack formation, crack growth, and efflorescence of the concrete. Inspections of two HSMs at the Calvert Cliffs ISFSI have also shown efflorescence of the concrete and the formation of calcium carbonate stalactites in the 2-inch gap between the heat shield and the concrete ceiling. These stalactites were observed near the outlet vent stack. A condition report was issued that did not identify an operability issue (CENG, 2012).</p> <p>Partial blockage of air inlet duct screens from snowfall has been identified. Decay heat from the spent fuel and/or stored heat in the overpack material (e.g., concrete) quickly melts any partial snow buildups after the snowfall has ceased. The existing activities (surveillance, monitoring, inspection) have proved adequate for natural phenomena during the initial license or certification period.</p>
<p>References</p>	<p>CENG. Letter to NRC, “Calvert Cliffs Nuclear Power Plant Independent Spent Fuel Storage Installation Material License No. SNM-2505, Docket No. 72-8, Response to Request for Supplemental Information, RE: Calvert Cliffs.” Independent Spent Fuel Storage Installation License Renewal Application, Calvert Cliffs Nuclear Power Plant, LLC. ADAMS Accession No. ML12212A216. July 27, 2012.</p> <p>NRC. NUREG–1567, “Standard Review Plan for Spent Fuel Dry Storage Facilities.” Washington DC: U.S. Nuclear Regulatory Commission. March 2000.</p> <p>_____. Interim Staff Guidance (ISG)–11, “Cladding Considerations for the Transportation and Storage of Spent Fuel.” Rev. 3. Washington DC: U.S. Nuclear Regulatory Commission. November 2003.</p> <p>_____. NUREG–1536, “Standard Review Plan for Spent Fuel Dry Storage Systems at a General License Facility.” Washington DC: U.S. Nuclear Regulatory Commission. July 2010.</p> <p>_____. Information Notice 2013-11, “Premature Degradation of Spent Fuel Storage Cask Structures and Components from</p>

Table 6-5 Example aging management program for Ventilation Systems	
Element	Description
	Environmental Moisture.” Washington, DC: U.S. Nuclear Regulatory Commission. April 16, 2013.

1

1 **6.9 Bolted Cask Seal Leakage Monitoring**

2 This AMP manages the aging of all bolted casks that employ leakage monitoring to verify the
3 integrity of the top confinement boundary. The program relies on existing pressure monitoring
4 systems to assess the integrity of cask closure seals. The program also performs periodic
5 visual inspections of normally inaccessible components under the cask protective cover to
6 monitor for corrosion, coating degradation, loose bolts, and evidence of water intrusion.

7

Table 6-6 Example aging management program for Bolted Cask Seal Leakage Monitoring

Element	Description
<p>1. Scope of Program</p>	<p>This program is used to manage the aging effects on the integrity of the top confinement boundary of bolted spent fuel storage casks to ensure that timely and appropriate corrective actions can be taken to maintain the safe storage conditions of the casks. The aging effects include loss of material as a result of corrosion of the sealing surfaces, O-rings, and bolts; loss of strength due to thermal aging and change in dimension due to creep of the metallic O-rings that results in loss of sealing forces; and loss of preload of the closure bolts.</p> <p>The specific components and systems that are typically managed by this program include the shield lid, primary lid, closure lid, protective covers, O-ring assemblies, and associated bolts and welds. The types of bolted cask designs covered by the program include TN-24, -32, -40, and -68; NAC-S/T (I26), -C28 S/T, -I28 S/T, and -STC, CASTOR V/21 and X/33; and Westinghouse MC-10 bolted casks.</p> <p>The program relies on continuous pressure-leakage monitoring to verify the integrity of the confinement boundary. In addition, the program relies on periodic visual inspections for evidence of aging that may affect the intended function of the identified SSCs and subcomponents.</p>
<p>2. Preventive Actions</p>	<p>Preventive actions include compliance with the NRC’s ISG on the materials selection for fabrication, design, and testing of casks, as described in NRC ISG-5, “Confinement Evaluation” (NRC, 1999); ISG-15, “Materials Evaluation” (NRC, 2001); and ISG-25, “Pressure and Helium Leakage Testing of the Confinement Boundary of Spent Fuel Dry Storage Systems” (NRC, 2010).</p>
<p>3. Parameters Monitored/ Inspected</p>	<p>The program relies on existing pressure-monitoring systems to assess the integrity of the cask closure seals. To verify the integrity of the seal assemblies in the bolted casks, these systems continuously monitor pressure:</p> <ul style="list-style-type: none"> • between the metallic seal assemblies in the TN-24, -32, -40, and -68; NAC-S/T (I26), -C28 S/T, -I28 S/T, and -STC, CASTOR V/21 and X/33 casks, and • inside the cask cavity in the MC-10 casks. <p>Parameters monitored/inspected for closure seal components include:</p> <ul style="list-style-type: none"> • visual evidence of loss of material from general, localized, and galvanic corrosion • visual evidence of coating degradation that could indicate corrosion of the base metal

Table 6-6 Example aging management program for Bolted Cask Seal Leakage Monitoring

Element	Description
	<ul style="list-style-type: none"> • visual evidence of clearances and physical displacements in bolted joints indicative of loss of preload or failed or missing components • visual evidence of water intrusion under the protective cover
<p>4. Detection of Aging Effects</p>	<p>Aging effects may be revealed by:</p> <ul style="list-style-type: none"> • overpressure and pressure loss (leakage) • water intrusion under protective covers • physical displacement, surface discontinuities, and imperfections indicative of loss of preload and corrosion. <p><u>Method or Technique</u></p> <p>The program credits the pressure-monitoring system, which continuously monitors the pressure between the seal assemblies in the TN-24, -32, -40, and -68; NAC-S/T (I26), -C28 S/T, -I28 S/T, and -STC, CASTOR V/21 and X/33 metal casks and inside the cask cavity of the MC-10 casks. Continuous monitoring with a pressure alarm provides a means for early detection of aging effects on the seal assemblies.</p> <p>Direct or remote VT-3 visual examination, as described in ASME Code Section XI, Article IWA-2213 (ASME, 2007), shall be performed and evaluated by personnel qualified in accordance with the requirements of IWE-2330.</p> <p><u>Frequency</u></p> <p>Pressure-monitoring systems provide continuous monitoring of the bolted cask seal integrity. Checks of system operation shall be conducted, in accordance with the established requirements for these systems. Inspection and calibration of the components of the overpressure leakage-monitoring systems shall be performed in accordance with manufacturer specifications. Opportunistic inspections of the overpressure leakage monitoring systems shall be conducted when the protective cover plate is removed for other inspection or maintenance actions.</p> <p>Visual VT-3 inspection of the normally inaccessible top sealing components in the confinement boundary, after removing the protective cover, shall be conducted on a justified frequency. This includes the condition of externally accessible surfaces of the bolts, protective covers, and protective coatings. In addition, opportunistic inspections of the top confinement boundary subcomponents shall be conducted when the protective cover is removed for other inspection or maintenance actions.</p>

Table 6-6 Example aging management program for Bolted Cask Seal Leakage Monitoring

Element	Description
	<p><u>Sample Size</u></p> <ul style="list-style-type: none"> • pressure-monitoring system: all casks • visual inspection of normally inaccessible surfaces: as justified <p><u>Data Collection</u></p> <p>Data from the examination, including the condition of the coating, locations and areas of coating degradation, and corrosion of any exposed steel surfaces shall be collected and documented on a checklist or visual inspection form. The results of the inspection shall be documented and include descriptions of observed aging effects and accompanied with sketches and/or photographs. Video coverage may also be used to document the inspection. Corrective actions resulting from each AMP inspection shall also be documented.</p> <p><u>Timing of Inspections</u></p> <p>Initial visual inspection of normally inaccessible surfaces and subcomponents shall be completed before entering the period of extended operation. Licensees may credit inspections conducted within the 5 years before the period of extended operation.</p>
<p>5. Monitoring and Trending</p>	<p>The pressure-monitoring data are trended to provide early detection of aging effects that result in leakage and to indicate when corrective action needs to be taken to maintain safe storage conditions.</p> <p>The results of visual inspections are documented, including evidence of corrosion of subcomponents, failure of protective coatings, and physical displacement of subcomponents of the cask-sealing system. Locations of all areas of degradation are documented to allow a direct comparison to prior inspection results. The inspection results will be documented and trended to identify aging-related degradation, the need for supplemental inspections, mitigation actions, and repair or replacement of subcomponents affected by aging.</p> <p>Corrective actions will be recorded and trended to evaluate the effectiveness of the actions taken.</p>
<p>6. Acceptance Criteria</p>	<p>Pressure readings should be within the range stated by the Certificate of Compliance (CoC) holder's, general licensee's, or site-specific licensee's TS. Casks with pressure-monitoring systems in the alarmed condition do not meet the acceptance criteria. The CoC holder's, general licensee's, or site-specific licensee's TS contain pressure-monitoring alarm response procedures that include criteria and specifications for corrective actions and response.</p>

Table 6-6 Example aging management program for Bolted Cask Seal Leakage Monitoring

Element	Description
	<p>For the cask-sealing subcomponents, the acceptance criteria for visual inspections are the absence of:</p> <ul style="list-style-type: none"> • coating degradation, including blistering, peeling or flaking • visual indication of corrosion on steel surfaces normally protected by a coating • loose or missing hardware • displaced subcomponents or parts <p>If coating degradation and corrosion are identified, then the severity of corrosion must be determined using approved site-specific or general licensee procedures. These may include additional visual, surface, or volumetric NDE methods to determine the loss of material. Corrosion that results in a loss of material that does not meet the design specifications is not acceptable for continued service and must be repaired or replaced.</p>
<p>7. Corrective Actions</p>	<p>Results that do not meet the acceptance criteria are addressed as conditions adverse to quality or significant conditions adverse to quality under those specific portions of the specific- or general- licensee QA program approved under 10 CFR Part 72, Subpart G, or 10 CFR Part 50, Appendix B, respectively. The QA program ensures that corrective actions are completed within the specific- or general- licensee’s Corrective Action Program (CAP), and include provisions to:</p> <ul style="list-style-type: none"> • perform functionality assessments • perform apparent cause evaluations and root cause evaluations • address the extent of condition • determine actions to prevent recurrence for significant conditions adverse to quality; ensure justifications for nonrepairs • trend conditions • identify operating experience actions, including modification to the existing AMP (e.g., increased frequency) • determine if the condition is reportable to the NRC per 10 CFR 72.75 <p>Once the low-pressure alarm is triggered, troubleshooting of the pressure leakage should be performed and, if necessary, an engineering evaluation conducted to determine whether the degradation of the seal assemblies requires immediate correction.</p>
<p>8. Confirmation Process</p>	<p>The confirmation process is commensurate with the specific- or general- licensee QA program approved under 10 CFR Part 72,</p>

Table 6-6 Example aging management program for Bolted Cask Seal Leakage Monitoring

Element	Description
	<p>Subpart G, or 10 CFR Part 50, Appendix B, respectively. The QA program ensures that the confirmation process includes provisions to preclude repetition of significant conditions adverse to quality.</p> <p>The confirmation process describes or references procedures to:</p> <ul style="list-style-type: none"> • determine followup actions to verify effective implementation of corrective actions • monitor for adverse trends due to recurring or repetitive findings or observations.
<p>9. Administrative Controls</p>	<p>The pressure-leakage monitoring system is periodically checked to ensure the system is functioning properly. Maintenance, calibration, and replacement of pressure transducers are performed in accordance with manufacturer requirements.</p> <p>The administrative controls will be commensurate with the specific or general licensee QA program and consistent with 10 CFR Part 72, Subpart G, or 10 CFR Part 50, Appendix B. The QA program ensures that inspections, evaluations, and corrective actions are completed in accordance with the specific or general licensee’s CAP. The requirements of 10 CFR Part 72, Appendix G, and 10 CFR Part 50, Appendix B, are acceptable to address the corrective actions, confirmation process, and administrative controls.</p>
<p>10. Operating Experience</p>	<p><u>Existing Operational Experience</u></p> <p>Helium leakage in two of the TN-68 bolted casks at Peach Bottom was detected in October 2010 (NRC, 2013). The root cause analyses indicated that the leakage in one cask was caused by a material defect in the weld plug that provides sealing of the drilled inter-seal passageway associated with the drain port penetration of the cask lid. The defective welds were repaired in accordance with the ASME Code and cask design requirements. In the other cask, leakage existed in the cask main lid outer closure seal. The seal leakage was caused by galvanic corrosion at the interface between the aluminum-clad cask lid seal and the stainless steel clad cask body sealing surface of the outer portion of the cask lid seal. The corrosion resulted from water infiltration through the access plate in the protective cover. The primary corrective actions involved improving the access plate design and developing a method for verifying protective cover seal integrity. Additional corrective actions included a change to the torquing process for the lid bolts and ensuring that the access plate gaskets and O-rings were inspected at installation. Corrosion of the TN-32 lid bolts and outer metallic lid seals has been observed in the Surry ISFSI owing to external water</p>

Table 6-6 Example aging management program for Bolted Cask Seal Leakage Monitoring

Element	Description
	<p>intrusion near the lid bolts and outer metallic seals, resulting in five seal replacements. One seal on a CASTOR X/33 cask has also been replaced at Surry (Virginia Electric and Power Company, 2002).</p> <p>An inspection was carried out in 2011 on the lead cask TN-40 01 at Prairie Island in conjunction with the license renewal application for the ISFSI (Schimmel, 2012). The components inspected included the carbon steel cask bottom and underlying concrete pad; the cask shell, lid, lid bolts, and trunnions; and the top neutron shield enclosure and shield bolts. In addition, the cask protective cover was removed to permit visual inspection of the protective cover, bolts, and seal; the access cover and bolts; and the overpressure tank, isolation valve and tubing, port cover, and port cover bolts. The only significant degradation observed was disbondment of approximately 25 percent of the protective coating on the bottom of the cask, minor uniform general corrosion at the upper trunnions, and a very minor rust coating on the stainless steel portions of the containment flange. In addition, the protective cover was found to have thin uniform corrosion on the flange sealing surface on the outer side of the O-ring and minor corrosion at the cover bolt holes, and the cask access cover had minor rust spots on the outside at the bolt holes. The protective cover Viton O-ring was in good condition and was not replaced, and the access cover gasket was also in good condition but was replaced. The protective cover on TN-40 cask number 13 was also removed to permit a visual inspection. Here, all components were found to be in good condition, and the only degradation noted was minor rust stains on the protective coating directly below the access cover from corrosion products dripping off the access cover.</p> <p>An inspection of an MC-10 cask was performed after about 20 years in service at Surry (Virginia Electric and Power Company, 2006). Twelve knurled nuts, which fasten the closure cover to the cask, were removed for inspection. While there was some oxidation of the outer O-ring edge, the O-ring seal surface and the areas underneath the closure cover had no cracks or indications of degradation.</p> <p>Stress relaxation and leakage tests on Helicoflex metallic seals, which are used in the CASTOR and TN cask designs, have been conducted in Germany at temperatures from room temperature to 150 degrees C [302 degrees F]. These tests found that the pressure force on the seal and its elastic recovery (or usable resilience) decrease approximately linearly when plotted against the logarithm of time, but usable lives beyond 40 years with acceptable leak rates are extrapolated. Corrosion tests were also initiated on this same seal design in 2001 with borated (2,400 ppm) water or a NaCl solution (10^{-3} mol) between the inner and outer jackets of the seal, and no</p>

Table 6-6 Example aging management program for Bolted Cask Seal Leakage Monitoring

Element	Description
	<p>increase in leakage rate has been detected to date (Völzke et al., 2012; Völzke et al., 2013). In addition, the behavior of elastomer seals at low temperature (below room temperature) has been studied to determine the minimum temperature at which these materials can function in DSS applications (Wolff et al., 2013).</p>
References	<p>ASME. “Boiler and Pressure Vessel Code Section XI—Rules for Inservice Inspection of Nuclear Power Plant Components.” New York, New York: American Society of Mechanical Engineers. 2007.</p> <p>Code of Federal Regulations. Title 10, Energy, Part 72, “Licensing Requirements for the Independent Storage of Spent Nuclear Fuel, High-Level Radioactive Waste, and Reactor-Related Greater than Class C Waste.” Washington, DC: Office of the Federal Register. 2015a</p> <p>Code of Federal Regulations. Title 10, Energy, Part 50, “Domestic Licensing of Production and Utilization Facilities.” Washington, DC: Office of the Federal Register. 2015b.</p> <p>NRC. “Confinement Evaluation.” Interim Staff Guidance-5. Rev. 1. Washington, DC: U.S. Nuclear Regulatory Commission. 1999.</p> <p>_____. “Materials Evaluation.” Interim Staff Guidance-15. Rev. 1. Washington, DC: U.S. Nuclear Regulatory Commission. 2001.</p> <p>_____. “Pressure and Helium Leakage Testing of the Confinement Boundary of Spent Fuel Dry Storage Systems.” Interim Staff Guidance-25. Washington, DC: U.S. Nuclear Regulatory Commission. 2010.</p> <p>_____. “Premature Degradation of Spent Fuel Storage Cask Structures and Components from Environmental Moisture.” Information Notice 2013-07. Washington, DC: U.S. Nuclear Regulatory Commission. 2013.</p> <p>Schimmel, M. “Prairie Island Independent Spent Fuel Storage Installation, Attachment 1 to Letter to U.S. Nuclear Regulatory Commission, Responses to Requests for Supplemental Information, RE: Prairie Island Independent Spent Fuel Storage Installation License Renewal Application.” (TAC No. L24592). ADAMS Accession No. ML12065A073. 2012.</p> <p>Virginia Electric and Power Company. “Surry Independent Spent Fuel Storage Installation License Renewal Application.” Docket No. 72–2. Richmond, Virginia: Virginia Electric and Power Company. April 29, 2002.</p>

Table 6-6 Example aging management program for Bolted Cask Seal Leakage Monitoring

Element	Description
	<p>_____. "Surry Independent Spent Fuel Storage Installation Completion of License Renewal Inspection Requirement." Docket No. 72-2, License Number SNM-2501. Richmond, Virginia: Virginia Electric and Power Company. August 22, 2006.</p> <p>Völzke, H. and D. Wolff. "Spent Fuel Storage in Dual Purpose Casks Beyond the Original Design Basis." Proceedings of the International High-Level Radioactive Waste Management Conference (IHLRWMC) April 28-May 2, 2013. La Grange Park, IL: American Nuclear Society. 2013.</p> <p>Völzke, H., U. Probst, D. Wolff, S. Nagelschmidt, and S. Schultz. "Seal and Closure Performance in Long Term Storage." Proceedings of the PSAM11 & ESREL 2012 Conference, Helsinki, Finland. 2012.</p> <p>Wolff, D., M. Jaunich, and W. Stark. "Investigating the Performance of Rubber Seals at Low Temperatures." Proceedings of the International High-Level Radioactive Waste Management Conference (IHLRWMC) April 28-May 2, 2013. La Grange Park, IL: American Nuclear Society. 2013.</p>

1

2

1 **6.10 Transfer Casks**

2 An example AMP for transfer casks is provided below. The AMP manages all transfer cask
3 subcomponents. This is a condition monitoring program that performs periodic visual
4 inspections of accessible cask internal and external surfaces to monitor for corrosion, wear, and
5 loss of preload (bolting). Steel neutron shield water jackets are monitored for wall thickness or
6 inspected for through-wall leakage.

Table 6-7 Example aging management program for Transfer Casks

Element	Description
<p>1. Scope of Program</p>	<p>This program manages loss of material due to corrosion and wear to ensure that this aging effect does not challenge the capability of the transfer casks to fulfill structural support, radiation shielding, and heat transfer functions.</p> <p>Visual inspections are performed on the accessible internal and external surfaces of steel transfer cask subcomponents that are exposed to indoor and outdoor air environments. Inaccessible steel surfaces in contact with water neutron shielding are evaluated with volumetric wall thickness measurements or inspections for through-wall leakage.</p> <p>If not addressed with a fatigue analysis, this AMP also includes inspections of trunnions for cracking.</p> <p>An additional site-specific AMP may be required to manage protective coatings that are credited in the design basis for preventing corrosion of the base metal.</p>
<p>2. Preventive Actions</p>	<p>This program is a condition-monitoring program to detect evidence of degradation. It does not provide guidance for prevention of aging.</p>
<p>3. Parameters Monitored/ Inspected</p>	<p>This program monitors the condition of internal and external steel surfaces to identify general, pitting, crevice, and galvanic corrosion, and wear. The condition of inaccessible steel internal surfaces that are continuously or intermittently exposed to a liquid neutron shield are monitored from the external side of the shield shell.</p> <p>Parameters monitored or inspected for accessible surfaces include:</p> <ul style="list-style-type: none"> • visual evidence of surface discontinuities and imperfections indicative of corrosion • visual evidence of coating degradation (e.g., blisters, cracking, flaking, delamination) indicative of corrosion of the base metal <p>Parameters monitored or inspected to evaluate inaccessible steel surfaces exposed to a liquid neutron shield include either:</p> <ul style="list-style-type: none"> • wall thickness • visual evidence of leakage on external surfaces <p>If trunnions are not addressed with a fatigue analysis, trunnion surfaces are monitored for the presence of cracks.</p>
<p>4. Detection of Aging Effects</p>	<p><u>Normally Accessible Surfaces</u></p> <p>Visual inspections are performed in accordance with the ASME Code Section XI, Article IWA-2213, for VT-3 examinations. The inspections cover 100 percent of the normally accessible steel cask</p>

Table 6-7 Example aging management program for Transfer Casks

Element	Description
	<p>surfaces, including the cask exterior, cask interior cavity, lid surfaces, and the cask bottom (during lifting or down ending).</p> <p><u>Normally Inaccessible Internal Surfaces (liquid neutron shield)</u></p> <p>Wall thicknesses of steel liquid neutron shield subcomponents are measured with ultrasonic thickness techniques. Alternatively, the condition of internal surfaces of the neutron shield shell is monitored by inspections for leakage when the shield is filled with water, following ASME Code Section XI, Article IWA-2212, VT-2 (visual) inspection requirements.</p> <p><u>Trunnions</u></p> <p>If the fatigue of trunnions is not addressed with an analysis, surface or volumetric inspection techniques are performed on 100 percent of trunnion surfaces to identify the presence of fatigue cracks.</p> <p><u>Sample Size</u></p> <p>All transfer casks are inspected.</p> <p><u>Frequency</u></p> <p>Inspections are conducted at least once every 5 years. If a transfer cask is used less frequently than once every 5 years, inspections are conducted before its use in each loading campaign.</p> <p><u>Data Collection</u></p> <p>Data from the examination, including evidence of degradation and its extent and location, shall be documented on a checklist or inspection form. The results of the inspection shall be documented, including descriptions of observed aging effects and supporting sketches, photographs, or video. Corrective actions resulting from each AMP inspection shall also be documented.</p> <p><u>Timing</u></p> <p>Initial inspections are completed before the use of the transfer casks in the first loading campaign in the period of extended operation.</p>
<p>5. Monitoring and Trending</p>	<p>Inspection results are compared to those obtained during previous inspections, so that the progression of degradation can be evaluated and predicted.</p> <p>Monitoring and trending methods reference plans/procedures used to:</p> <ul style="list-style-type: none"> • establish a baseline before or at the beginning of the period of extended operation • track trending of parameters or effects not corrected following a previous inspection

Table 6-7 Example aging management program for Transfer Casks

Element	Description
	<ul style="list-style-type: none"> — the locations, size, and depth of any areas of corrosion — the disposition of components with identified aging effects and the results of supplemental inspections
<p>6. Acceptance Criteria</p>	<p>For accessible surfaces, including trunnions, acceptance criteria are:</p> <ul style="list-style-type: none"> • no detectable loss of material from the base metal, including uniform wall thinning, localized corrosion pits, crevice corrosion, and wear scratches/gouges • no red-orange-colored corrosion products on the base metal or coatings • no coating defects (e.g., blisters, cracking, flaking, delamination) <p>For inaccessible internal surfaces, the acceptance criteria are no evidence of leakage of the water neutron shield or loss of wall thickness beyond a predetermined limit established by system-specific design standards or industry codes and standards.</p> <p>If evidence of corrosion, wear, or coating degradation are identified, then the severity of the degradation of the base metal must be determined using approved site-specific procedures. These may include additional visual, surface, or volumetric NDE methods to determine the loss of material.</p> <p>Alternative acceptance criteria are developed from system-specific design standards, industry codes or standards, or engineering evaluation. Where possible, acceptance criteria are quantitative (e.g., minimum wall thickness). Where qualitative acceptance criteria are used, the criteria are sufficiently clear to reasonably ensure that a singular decision is derived based on the observed condition, avoiding the use of ambiguous phrases (e.g., significant, moderate).</p> <p>EPRI Technical Reports, TR-1007933, "Aging Assessment Field Guide" (EPRI, 2003), and TR-1009743, "Aging Identification and Assessment Checklist: Mechanical Components" (EPRI, 2004), provide general guidance for the evaluation of materials and the development of criteria for their acceptance when performing visual inspections.</p>
<p>7. Corrective Actions</p>	<p>Results that do not meet the acceptance criteria are addressed as conditions adverse to quality or significant conditions adverse to quality under those specific portions of the specific- or general- licensee QA program approved under 10 CFR Part 72, Subpart G, or 10 CFR Part 50, Appendix B, respectively. The QA program ensures that corrective actions are completed within the specific- or general- licensee's CAP, and include provisions to:</p>

Table 6-7 Example aging management program for Transfer Casks

Element	Description
	<ul style="list-style-type: none"> • perform functionality assessments • perform apparent cause evaluations and root cause evaluations • address the extent of condition • determine actions to prevent recurrence for significant conditions adverse to quality; ensure justifications for nonrepairs • trend conditions • identify operating experience actions, including modification to the existing AMP (e.g., increased frequency) • determine if the condition is reportable to the NRC per 10 CFR 72.75
8. Confirmation Process	<p>The confirmation process is commensurate with the specific- or general-licensee QA program approved under 10 CFR Part 72, Subpart G, or 10 CFR Part 50, Appendix B, respectively. The QA program ensures that the confirmation process includes provisions to preclude repetition of significant conditions adverse to quality.</p> <p>The confirmation process describes or references procedures to:</p> <ul style="list-style-type: none"> • determine followup actions to verify effective implementation of corrective actions • monitor for adverse trends due to recurring or repetitive findings or observations.
9. Administrative Controls	<p>The administrative controls are addressed through those portions of the specific or general licensee's QA program that are used to meet 10 CFR Part 72, Subpart G, or 10 CFR Part 50, Appendix B.</p>
10. Operating Experience	<p>External surface inspections through system inspections and walkdowns in support of the Maintenance Rule (10 CFR Part 50.65) have proven effective in maintaining the material condition of nuclear power plant systems.</p>
References	<p>EPRI. "Aging Assessment Field Guide." Technical Report 1007933. Palo Alto, California: Electric Power Research Institute. December 2003.</p> <p>_____. "Aging Identification and Assessment Checklist–Mechanical Components." Technical Report 1009743. Palo Alto, California: Electric Power Research Institute. August 27, 2004.</p>

1

2

1 **6.11 High-Burnup Fuel Monitoring And Assessment**

2 An example of a High Burnup (HBU) Fuel¹ Monitoring and Assessment Program is provided
3 below. This is a licensee program that monitors and assesses data and other information
4 regarding HBU fuel performance, to confirm that the design-bases HBU fuel configuration is
5 maintained during the period of extended operation. This example HBU Fuel Monitoring and
6 Assessment Program relies on a surrogate demonstration program to provide data on HBU fuel
7 performance. Guidance for determining if a surrogate demonstration program can provide the
8 data to support a licensee's HBU Fuel Monitoring and Assessment Program is given in
9 Appendix D of NUREG-1927, Revision 1 (NRC, 2016). Although this example focuses on the
10 use of a surrogate demonstration program, a licensee may use alternative approaches that are
11 appropriately justified, including the use of test or research results and safety analyses for the
12 fuel, to demonstrate that the DSS's intended functions continue to be met during the period of
13 extended operation.

14 The aging management review is not expected to identify any aging effects that could lead to
15 fuel reconfiguration, as long as the HBU fuel is stored in a dry inert environment, temperature
16 limits are maintained, and thermal cycling is limited. Short-term testing (i.e., laboratory scale
17 testing up to a few months) and scientific analyses examining the performance of HBU fuel have
18 provided a foundation for the technical basis that storage of HBU fuel in the period of extended
19 operation may be performed safely and in compliance with regulations. However, there has
20 been relatively little operating experience, to date, with dry storage of HBU fuel.

21 Therefore, the purpose of a HBU Fuel Monitoring and Assessment Program is to monitor and
22 assess data and other information regarding HBU fuel performance to confirm there is no
23 degradation of HBU fuel that would result in an unanalyzed configuration during the period of
24 extended operation. The following description of an example HBU Fuel Monitoring and
25 Assessment Program presents the applicable information in a format using each element of an
26 effective AMP, to provide a framework for such a monitoring and assessment program.

27

¹These are fuel assemblies with discharge burnup greater than 45 gigawatt-days per metric ton of uranium (GWd/MTU).

Table 6-8 Example aging management program for High-Burnup Fuel Monitoring and Assessment

Element	Description
<p>1. Scope of the Program</p>	<p>The scope of the program provides a description of (i) the design bases characteristics of the HBU fuel, (ii) the surrogate demonstration program that will be used to provide data on the applicable design-bases HBU fuel performance, and (iii) how the parameters of the surrogate demonstration program are applicable to the design-bases HBU fuel.</p> <p>Aging effects will be determined for material/environment combinations per an alternative surrogate demonstration program meeting the guidance in Appendix D of NUREG–1927, Revision 1 (NRC, 2016).</p> <p>Example language to address this “scope of the program” element follows: Fuel stored in a [define cask/canister model] is limited to an assembly average burnup of [define design-bases limit] GWd/MTU. The cladding materials for the HBU fuel are [define types of cladding], and the fuel is stored in a dry helium environment. HBU fuel was first placed into dry storage in a [define cask/canister model] on [start date of storage term of first storage of HBU fuel].</p> <p>The program relies on the joint EPRI and DOE HBU Dry Storage Cask Research and Development Project (HDRP) (EPRI, 2014), conducted in accordance with the guidance in Appendix D of NUREG–1927, Revision 1, as a surrogate demonstration program that monitors the performance of HBU fuel in dry storage.</p> <p>The HDRP is a program designed to collect data from an SNF storage system containing HBU fuel in a dry helium environment. The program entails loading and storing an AREVA TN-32 bolted lid cask (the “Research Project Cask”) at Dominion Virginia Power’s North Anna Power Station with intact HBU fuel (of nominal burnups ranging between 53 GWd/MTU and 58 GWd/MTU). The fuel to be used in the program includes four kinds of cladding (Zircaloy-4, low-tin Zircaloy-4, ZIRLO™, and M5™). The Research Project Cask is to be licensed to the temperature limits contained in ISG-11, Rev. 3 (NRC, 2003), and loaded such that the fuel cladding temperature is as close to the limit as practicable. [If an alternative surrogate demonstration program is used, provide a description of the program.]</p> <p>The parameters of the surrogate demonstration program are applicable to the design-bases HBU fuel, as the (i) maximum burnup of the design-bases HBU fuel [define value] is less than the burnup of the fuel in the surrogate demonstration program [define value], (ii) the cladding type of the design-bases HBU fuel [define type] is the same as the surrogate demonstration program [define type], and (iii) the temperatures in the surrogate demonstration program [define</p>

Table 6-8 Example aging management program for High-Burnup Fuel Monitoring and Assessment	
Element	Description
	values] bound the design bases temperature/heat load of the loaded systems [define values].
2. Preventive Actions	<p>There are no specific preventive actions associated with this HBU Fuel Monitoring and Assessment Program. However, the applicant should discuss the design-bases characteristics of the licensed/certified DSS, in terms of initial cask loading operations, to show the HBU fuel is stored in a dry inert environment.</p> <p>Example language follows:</p> <p>During the initial loading operations of the cask/canister, the design and ISFSI TS require that the fuel be stored in a dry inert environment. TS [name and number] demonstrates that the cask/canister cavity is dry by maintaining a cavity absolute pressure less than or equal to [value] for a [time period] with the cask/canister isolated from the vacuum pump. TS [name and number], requires that the cask/canister then be backfilled with helium. These two TS requirements ensure that the HBU fuel is stored in an inert environment, thus preventing cladding degradation due to oxidation mechanisms. TS [name and number] also requires that the helium environment be established within [time] hours of commencing cask/canister draining. The cask/canister is loaded in accordance with the criteria of ISG-11, Revision 3 (NRC, 2003).</p>
3. Parameters Monitored or Inspected	The applicant identifies the parameters monitored and inspected in a surrogate demonstration program that are applicable to its particular design-bases HBU fuel and describes how this meets the guidance of Appendix D of NUREG-1927, Revision 1.
4. Detection of Aging Effects	The applicant identifies the detection of aging effects in a surrogate demonstration program that are applicable to its particular design-bases HBU fuel and describes how this meets the guidance of Appendix D of NUREG-1927, Revision 1.
5. Monitoring and Trending	<p>As information/data from a surrogate demonstration program or from other sources (such as testing or research results and scientific analyses) become available, the licensee will monitor, evaluate, and trend the information via its operating experience program and/or the CAP to determine what actions should be taken.</p> <p>The licensee will evaluate the information/data from a surrogate demonstration program or from other sources to determine whether the acceptance criteria in Element 6 are met.</p> <ul style="list-style-type: none"> • If all of the acceptance criteria are met, no further assessment is needed. • If any of the acceptance criteria are not met, the licensee must conduct additional assessments and implement appropriate corrective actions (see Element 7).

Table 6-8 Example aging management program for High-Burnup Fuel Monitoring and Assessment	
Element	Description
	Formal evaluations of the aggregate information from a surrogate demonstration program and other available domestic or international operating experience (including data from monitoring and inspection programs, NRC-generated communications, and other information) will be performed at specific points in time during the period of extended operation, as delineated in Table B-4 of NUREG-1927, Revision 1.
6. Acceptance Criteria	<p>The HBU Fuel Monitoring and Assessment Program acceptance criteria are:</p> <ul style="list-style-type: none"> • hydrogen content—Maximum hydrogen content of the cover gas over the approved storage period should be extrapolated from the gas measurements to be less than the design-bases limit for hydrogen content. • moisture content—The moisture content in the cask/canister, accounting for measurement uncertainty, should be less than the expected upper-bound moisture content per the design-bases drying process¹. • fuel condition/performance²—nondestructive examination (e.g., fission gas analysis) and destructive examination (e.g., to obtain data on creep, fission gas release, hydride reorientation, cladding oxidation, and cladding mechanical properties) should confirm the design-bases fuel condition (i.e., no changes to the analyzed fuel configuration considered in the safety analyses of the approved design bases). <p>The applicant should provide information on the design-bases characteristics of the DSS, with regard to these criteria. The applicant should reference the source of specific values, or explain any assumptions made, for defining design-bases characteristics of the fuel condition/performance.</p>
7. Corrective Actions	<p>The corrective actions are in accordance with the specific or general licensee QA program and consistent with 10 CFR Part 72, Subpart G, or 10 CFR Part 50, Appendix B, respectively.</p> <p>Corrective actions should be implemented if data from a surrogate demonstration program or other sources of information indicate that</p>

¹The applicant will need to provide the expected upper-bound moisture content based on its design-bases drying process. If the design-bases drying process involves a vacuum drying method of evacuating a cask/canister to less than or equal to 3 torr and maintaining a constant pressure for 30 minutes after the cask/canister is isolated from the vacuum pump, the expected water content is about 0.43 gram-mole. (See NRC, 2010.)

²While it is not a fuel performance criterion, the spatial distribution and time history of the temperature must be known to evaluate the relationship between the performance of the rods in a surrogate demonstration program and the HBU fuel rod behavior expected in the cask.

Table 6-8 Example aging management program for High-Burnup Fuel Monitoring and Assessment

Element	Description
	<p>any of the HBU Fuel Monitoring and Assessment Program acceptance criteria (in Element 6) are not met.</p> <p>If any of the acceptance criteria are not met, the licensee will:</p> <ul style="list-style-type: none"> (i) assess fuel performance (impacts on fuel and changes to fuel configuration) (ii) assess the design-bases safety analyses, considering degraded fuel performance (and any changes to fuel configuration), to determine the ability of the DSS to continue to perform its intended functions under normal, off-normal, and accident conditions. <p>The licensee will determine what corrective actions should be taken to:</p> <ul style="list-style-type: none"> (i) manage fuel performance, if any (ii) manage impacts related to degraded fuel performance to ensure that all intended functions for the DSS are met. <p>In addition, the licensee will obtain the necessary NRC approval in the appropriate licensing/certification process for modification of the design bases to address any conditions outside of the approved design bases.</p>
8. Confirmation Process	<p>The confirmation process is commensurate with the specific- or general-licensee QA program approved under 10 CFR Part 72, Subpart G, or 10 CFR Part 50, Appendix B, respectively. The QA program ensures that the confirmation process includes provisions to preclude repetition of significant conditions adverse to quality.</p> <p>The confirmation process describes or references procedures to:</p> <ul style="list-style-type: none"> • determine followup actions to verify effective implementation of corrective actions • monitor for adverse trends due to recurring or repetitive findings or observations.
9. Administrative Controls	<p>The administrative controls are in accordance with the specific or general licensee QA program and consistent with 10 CFR Part 72, Subpart G, or 10 CFR Part 50, Appendix B, respectively. The QA program ensures that the administrative controls include provisions that define:</p> <ul style="list-style-type: none"> • formal review and approval processes • record retention requirements • document control

Table 6-8 Example aging management program for High-Burnup Fuel Monitoring and Assessment

Element	Description
10. Operating Experience	<p>The program references and evaluates applicable operating experience, including:</p> <ul style="list-style-type: none"> • internal and industrywide condition reports • internal and industrywide corrective action reports • vendor-issued safety bulletins • NRC Information Notices • applicable DOE or industry initiatives (e.g., HDRP) • applicable research (e.g., Oak Ridge National Laboratory studies on bending responses of the fuel, Argonne National Laboratory and Central Research Institute of Electric Power Industry studies on hydride reorientation effects) <p>The review of operating experience clearly identifies any HBU fuel degradation as either age related or event driven, with proper justification for that assessment. Past operating experience supports the adequacy of the HBU Fuel Monitoring and Assessment Program.</p> <p>Surrogate demonstration programs with storage conditions and fuel types similar to those in the licensed/certified DSS that meet the guidance in Appendix D of NUREG–1927, Revision 1, are a viable method to obtain operating experience.</p> <p>New data/research on fuel performance from both domestic and international sources that are relevant to the licensed/certified HBU fuel in the DSS should be evaluated on a periodic basis.</p>
References	<p>EPRI. “HBU Dry Storage Cask Research and Development Project Final Test Plan.” DOE Contract No.: DE-NE-0000593. Palo Alto, California: Electric Power Research Institute. 2014.</p> <p>NRC. “NRC Interim Staff Guidance 11, “Cladding Considerations for the Transportation and Storage of Spent Fuel.” Rev. 3. ADAMS Accession No. ML033230335. Washington, DC: U.S. Nuclear Regulatory Commission. November 17, 2003.</p> <p>_____. NUREG–1536, “Standard Review Plan for Spent Fuel Dry Storage Systems at a General License Facility.” Rev. 1. ADAMS Accession No. ML101040620. Washington, DC. U.S. Nuclear Regulatory Commission. 2010.</p> <p>_____. NUREG–1927, “Standard Review Plan for Renewal of Specific Licenses and Certificates of Compliance for Dry Storage of Spent Nuclear Fuel.” Revision 1. ADAMS Accession No. ML16179A148. Washington, DC: U.S. Nuclear Regulatory Commission. 2016. .</p>

NRC FORM 335 (12-2010) NRCMD 3.7 <p style="text-align: center;">BIBLIOGRAPHIC DATA SHEET <i>(See instructions on the reverse)</i></p>	<p style="text-align: center;">U.S. NUCLEAR REGULATORY COMMISSION</p> 1. REPORT NUMBER (Assigned by NRC, Add Vol., Supp., Rev., and Addendum Numbers, if any.) <p style="text-align: center;">NUREG-2214</p>				
2. TITLE AND SUBTITLE Managing Aging Processes in Storage (MAPS) Report Draft Report for Comment	3. DATE REPORT PUBLISHED <table border="1" style="width: 100%;"> <tr> <td style="text-align: center;">MONTH</td> <td style="text-align: center;">YEAR</td> </tr> <tr> <td style="text-align: center;">October</td> <td style="text-align: center;">2017</td> </tr> </table> 4. FIN OR GRANT NUMBER 	MONTH	YEAR	October	2017
MONTH	YEAR				
October	2017				
5. AUTHOR(S) 	6. TYPE OF REPORT <p style="text-align: center;">Technical</p> 7. PERIOD COVERED (Inclusive Dates) 				
8. PERFORMING ORGANIZATION - NAME AND ADDRESS (If NRC, provide Division, Office or Region, U. S. Nuclear Regulatory Commission, and mailing address; if contractor, provide name and mailing address.) Division of Spent Fuel Management Office of Nuclear Material Safety and Safeguards U.S. Nuclear Regulatory Commission Washington, DC 20555-0001					
9. SPONSORING ORGANIZATION - NAME AND ADDRESS (If NRC, type "Same as above", if contractor, provide NRC Division, Office or Region, U. S. Nuclear Regulatory Commission, and mailing address.) Same as above					
10. SUPPLEMENTARY NOTES 					
11. ABSTRACT (200 words or less) This Managing Aging Processes in Storage (MAPS) Report provides guidance for the U.S. Nuclear Regulatory Commission (NRC) technical reviewer. It establishes a technical basis for the safety review of renewal applications for specific licenses of independent spent fuel storage installations and Certificates of Compliance for dry storage systems, as codified in Title 10 of the Code of Federal Regulations Part 72. "Licensing Requirements for the Independent Storage of Spent Nuclear Fuel, High Level Radioactive Waste, and Reactor Related Greater Than Class C Waste." The MAPS Report evaluates known aging degradation mechanisms to determine if they could affect the ability of dry storage system components to fulfill their safety functions in the 20- to 60 year period of extended operation. The guidance also provides examples of aging management programs that are considered generically acceptable to address the credible aging mechanisms to ensure that the design bases of dry storage systems will be maintained. An applicant for a renewed license or Certificate of Compliance may reference the information in the MAPS Report to support its aging management review and proposed aging management programs.					
12. KEY WORDS/DESCRIPTORS (List words or phrases that will assist researchers in locating the report.) Dry Storage System Aging Management Spent Fuel Aging Degradation Mechanisms	13. AVAILABILITY STATEMENT <p style="text-align: center;">unlimited</p> 14. SECURITY CLASSIFICATION <i>(This Page)</i> <p style="text-align: center;">unclassified</p> <i>(This Report)</i> <p style="text-align: center;">unclassified</p> 15. NUMBER OF PAGES 16. PRICE 				



Federal Recycling Program



**UNITED STATES
NUCLEAR REGULATORY COMMISSION**
WASHINGTON, DC 20555-0001

OFFICIAL BUSINESS



**NUREG-2214
Draft**

Managing Aging Processes In Storage (MAPS) Report

October 2017